

KITAGAWA GAS DETECTOR TUBE SYSTEM HANDBOOK

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1. KITAGAWA GAS DETECTOR TUBE SYSTEM

1.1 HISTORY OF KITAGAWA GAS DETECTOR TUBES

The history of Kitagawa gas detector tubes goes back to 1946, when a study was undertaken to determine the efficiencies of desulfurisers in the production of ammonium sulphate, then used as a fertilizer. While ammonia, a raw material of ammonium sulphate, was produced by synthesizing hydrogen and nitrogen with the aid of a catalyst under high pressure, the catalyst was deteriorated by hydrogen sulphide mixed in the hydrogen as an impurity.

It was determined that a new method of gas detection was needed for a process control environment. This new method would need to meet the following requirements:

- (1) Hydrogen sulphide can be measured with high accuracy.
- (2) The detector must be small and light enough to carry.
- (3) The detector must be easy to operate, requiring no special skills.
- (4) Measurement results should be obtained fast, with no calculation required.
- (5) No liquid reagents were to be used.
- (6) The detector could not use electrical power to prevent possible gas/vapour explosion.

In solving these strict requirements, the basic concept of the Kitagawa gas detector tube, using a dry chemical reaction, was born. The detector tube made use of the discolouration reaction of a detecting reagent (lead acetate) adsorbed on the surface of solid granules (such as silica gel). Gas measurement was made by the colour change of the detecting reagent. As the hydrogen sulphide was passed through the thin glass detector tube filled with the white detecting reagent, the tube turned black, and the degree of the discolouration increased in proportion to the hydrogen sulphide concentration. This new method was quickly adopted by a number of ammonia synthesis plants, where its performance was confirmed with as accuracy of plus or minus 5 per cent.

The highly successful development of the hydrogen sulphide detector tube suggested the possibility of detecting other gases. There quickly followed the development of a phosphide detector tube for quality control of acetylene production, and an ammonia detector tube for quality control of ammonium sulphide production. In this way the Kitagawa gas detector tube was established as a new gas detection method enabling anyone to quickly, and with a high degree of accuracy detect and measure concentrations of gases in different process environments.

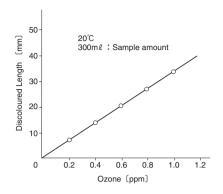
Kitagawa detector tubes, first used for process and quality control, quickly penetrated into the field of industrial hygiene in the 1950's when detector tubes were developed for the measurement of chlorine leakage in electrolytic soda plants and for carbon monoxide build-up in coal mines.

Today, the Kitagawa Toxic Gas Detector System represents well over 200 different kinds of detector tubes, as part of a complete sampling and analysis system to detect and measure toxic gases and materials. Each production lot of tubes undergoes strict quality control and testing to ensure their performance.

1.2 GAS DETECTOR TUBE METHOD

1.2.1 PRINCIPLE

Kitagawa gas detector tubes are made from high-quality borosilicate glass tubes with a uniform inside diameter. Inside, each tube is packed with a gas detecting reagent. When both ends of a detector tube are broken, inserted into the Model AP-20 Kitagawa Gas Aspirating Pump, and an air sample is pulled through the tube by means of pulling back on the pump handle, the detecting reagent changes colour. The length of the discoloured layer is proportional to the concentration of the gas or vapour in the sample because of the fixed volume of sample, which is always 100 ml. The relationship between the length of the discoloured layer and the concentration of gas is shown in FIG. 1.1. Graduations printed on the tube showing the gas concentration make it fast and easy to take a reading with Kitagawa detector tubes.



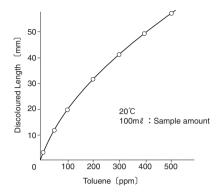


FIG. 1.1 DISCOLOURED LENGTH VS. GAS

1.2.2 DETECTING REAGENT

The detecting reagent is a reactive chemical adsorbed by granules (the carrier) of a highly purified silica gel, activated alumina, silica sand or silica glass. This packing changes colour by the chemical reaction with the target gas. Because of this, it is important to select the reactive reagent that meets the following criteria:

- (1) The colour-developing reagent must be stable when absorbed on the carrier and have a long shelf life.
- (2) The reagent must show a sharp colour change reaction with the target gas and be acceptable in selectivity
- (3) The detecting reagent must give a strong colour contrast both before and after the colour change reaction.
- (4) The detecting reagent must react with the target gas in the sample quickly and develop a discoloured length with a clear end point.

1.3 ADVANTAGES OF THE GAS DETECTOR TUBE METHOD

The Kitagawa gas detector tube method has several advantages over other methods of gas detection:

- (1) As the gas detector tube method requires no chemical analyzers, reagents, etc., advance preparations are not needed; detector tubes are always ready for use
- (2) The gas detector tube method is well-suited for use at the work site because it is small, light weight, and needs only a small sample volume to determine the concentration of gas or vapour in a sample.
- (3) The operating procedures are simple, allowing anyone to make measurements without the need for individual sampling technique.
- (4) The results of measurements are available in just minutes, so fast action can be taken when needed.
- (5) Since no electrical power source is required, detector tubes can be used even when flammable gases are present.
- (6) Different types of detector tubes are available for different gases and measuring ranges, from 0.1 ppm to more than 10 %, making the system flexible tor different sampling situations.

1.4 CONDITIONS THAT AFFECT GAS DETECTOR TUBE RESULTS

1.4.1 SAMPLE VOLUME

Graduations on the detector tube showing the concentration of gas are generally calibrated to $100 \, \text{m}\,\ell\ell\,(100\,\text{cc})$, the quantity of gas passed through the detector tube in one pump stroke. Some detector tube graduations are calibrated to $200\,\text{m}\,\ell$ or $300\,\text{m}\,\ell$ (two or three pump strokes) . The term pump "stroke" simply means the number of times the sample pump is operated (pump handle pulled back all the way and allowed to automatically lock) .

The discoloured layer generally lengthens in proportion to the quantity of gas passed through the detector tube. When a gas or vapor is present in concentrations lower than the lowest graduation on the detector tube, the approximate concentration value can be found by increasing the sample volume to several hundred millilitres (several pump strokes) and dividing the concentration value read on the detector tube by the number of pump strokes. It should be noted, however, that the quantity of gas or vapour detected may not be exactly proportional to the length of the discoloured layer obtained due to the effects of moisture (RH) and a depleted oxidizer.

1.4.2 TEMPERATURE

The temperature of the gas detector tube (normally the same temperature in the sampling environment) can also affect the measured concentration of gas or vapour for the following reasons:

- (1) An increase or decrease in the sample volume caused by a temperature change
- (2) A change in the quantity of gas adsorbed by the detecting reagent
- (3) A change in the rate in which the detecting reagent reacts with the gas or vapour

These effects can overlap one another, causing the length of the stain in the detector tube to increase or decrease. Or they can offset one another, having no effect on the measured value. Those Kitagawa gas detector tubes that can be affected by temperature changes are provided with a temperature correction table as part of the instruction sheet packed with each box of detector tubes

1.4.3 HUMIDITY

While nearly all Kitagawa gas detector tubes are not affected by unsaturated water vapour, some detector tubes are formulated with (a) both a dehumidifying reagent and detection reagent together in tube, or (b) a separate dehumidifying tube that is connected to the detector tube with a connector supplied in each box of tubes. However, water droplets can enter a detector tube; water can even condense inside the detector tube due to temperature changes. Both these situations can have a serious effect on both the gas or vapour readings obtained and the actual colour change of the detector tube. For these reasons, taking readings with Kitagawa Water Vapour tubes prior to gas or vapour sampling is recommended.

1.4.4 COEXISTING GASES OR VAPOURS

While the detecting reagents in Kitagawa gas detector tubes are formulated to react uniquely with the gas or vapour to be measured, it can also show a similar reaction (colour change) with another gas or gases having similar chemical properties. It is necessary to carefully consider this when taking gas or vapour readings with any gas detector tube.

A coexisting gas or vapour can have the following effects on gas detector tubes:

- (1) It may not discolour the detecting reagent, but can give a higher or lower reading.
- (2) It can discolour the detecting reagent in a similar way, giving a higher reading than the actual concentration.
- (3) It can give an entirely different colour change in the tube.
- (4) It can give the discoloured length with an indistinct end point.

1.4.5 SHELF LIFE

Gas detector tubes use chemical reagents that inevitably deteriorate over time. Because of this, if detector tubes are used after the stated shelf life expiration date (stamped on each box of tubes),

they may not show an accurate gas or vapour concentration. While the highest purity reagents are used in formulating Kitagawa detector tubes to ensure the longest shelf life possible, complex chemistries used to detect some gases and vapours make it important to use tubes within the stamped expiration date.

1.4.6 DETECTOR TUBE STORAGE

All Kitagawa detector tubes should be stored in a cool, dark place to ensure adequate shelf life. Store tubes at a temperature of 25 $^{\circ}\mathrm{C}$ (77 $^{\circ}\mathrm{F})$ or less out of direct sunshine in case a specific temperature is not shown on a box. Some detector tubes are required to store in a refrigerated place at a temperature between 0 $^{\circ}\mathrm{C}$ and 10 $^{\circ}\mathrm{C}$ (32-50 $^{\circ}\mathrm{F})$ as stated on the detector tube box. Detector tubes stored at high temperatures or in direct sunlight may show a performance deterioration even if they are within the stated shelf life.

2. COMPONENTS AND OPERATION

2.1 KITAGAWA GAS DETECTOR TUBES

2.1.1 TYPES OF GAS DETECTOR TUBES

Kitagawa precision gas detector tubes vary in type according to the gas or vapour to be detected and the required detection range. They are used for measuring gases in different concentration ranges. As gas or vapour concentrations can vary from high to low, Kitagawa gas detector tubes vary in detectable concentration from high to low.

Detector tube Models vary from "H", "A", "B", "C", "D", and "U" according to concentration ranges detectable, with "H" tubes detecting the highest ranges and the "U" tubes detecting ultralow ranges of gases and vapours. Some detector tubes employ a multi-tube arrangement, two tubes connected in tandem (a pretreat or "scrubbing" tube. and a detector tube) in order to detect the gas or vapour of interest.

All Kitagawa gas detector tubes are of three basic types:

- (1) Direct Reading
- (2) Concentration Chart
- (3) Colour Intensity

2.1.2 LENGTH OF STAIN DETECTOR TUBES

(1) DIRECT READING TUBES

Kitagawa direct reading detector tubes have gas concentration scales printed directly on the tubes themselves. This makes it convenient for taking gas or vapour measurements on-site because the tube is read just like a thermometer.



FIG. 2.1 DIRECT READING TUBE

(2) CONCENTRATION CHART TUBES

As the concentration chart shown in FIG. 2.2 can be drawn with calibration lines at close intervals, the resolution is improved over printed scale tubes. The chart, through the use of its slanted top and bottom lines, corrects for any difference in packing length.

2.1.3 COLOUR INTENSITY TUBES

When a gas or vapour with little reactivity or a low reaction rate is detected, tube discolouration is nearly uniform throughout the tube's length, with no clear end-stain border line. As this tube colour change shows a colour intensity proportional to concentration of the gas, that concentration can be measured by comparing the tube's colour stain to a colour standard chart. (see FIG. 2. 3). This is the basis of the Colour intensity method for detector tubes. The Length of Stain detector tube method exceeds the Colour Intensity method in measurement accuracy. However, trace gas concentrations below the minimum detectable concentration of Length of Stain detector tubes can be measured by Colour Intensity tubes by passing a large quantity of sample through the detector tube.

2.1.4 DETECTOR TUBES AND PRETREAT TUBES

Two types of pretreat tubes are provided for sampling convenience:

(1) pretreat tubes filled with an eliminating reagent or dehumidifying reagent for eliminating interfering gases or moisture in the sample, and (2) pretreat tubes filled with an oxidizing reagent for breaking down the sample gas before it enters the detector tube. For example: when a trichloroethylene detector tube is used, trichloroethylene is broken down in the pretreat tube to generate chlorine gas; the chlorine gas is passed through and detected by the chlorine-detecting reagent in the detector tube. The tube's scale graduations are calibrated by trichloroethylene in parts per million (ppm). All pretreat tubes are connected to their detector tubes as shown in FIG 2.4

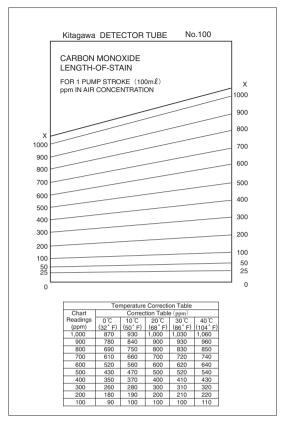


FIG.2.2 CONCENTRATION CHART

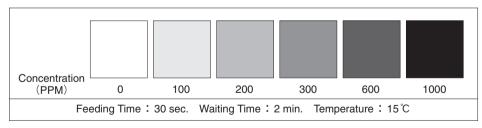


FIG.2.3 COLOUR STANDARD CHART

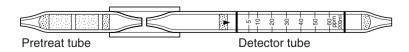


FIG.2.4 CONNECTION WITH PRETREAT TUBE

2.2 MODEL AP-20 SAMPLING PUMP KIT

The Kitagawa Model AP-20 consists of an aspirating pump, accessories, and a compact, lightweight carrying case (FIG.2.5, Tubes are optional.).



FIG.2.5 MODEL AP-20

The Model AP-20 is provided with a sample flow indicator, detector tube tip cutter and storage, and a two-stage locking mechanism on the pump handle that allows sampling of either 50 m ℓ or 100 m ℓ sample volumes. The carrying case can be used as a portable type.

2.2.1 MODEL AP-20 ASPIRATING PUMP CONSTRUCTION

(1) SAMPLE FLOW INDICATOR

The Model AP-20 pump includes an exclusive sample flow finish indicator as an aid in sampling. Before sampling, the red flow indicator is sticking out from the indicator base. As a sample is drawn through a detector tube, the red flow indicator will be sticking out from the indicator base. As sampling progresses, this red flow indicator sticks out fully. Sampling cycle is completed when the red flow indicator sticks out fully from the indicator base. With this handy sampling aid, it is not necessary to always time each sample, allowing multiple tasks to be performed. (See FIG. 2.7)

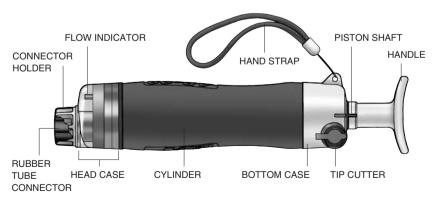


FIG. 2.6 CONSTRUCTION AND PART NAMES

(2) DETECTOR TUBE TIP STORAGE

Glass ends of detector tubes broken off in the tip cutter collect in the tip storage compartment. This reduces the possibility of glass cuts scatter. (See FIG. 2.8)

(3) TWO-STAGE PUMP MECHANISM

Model AP-20 Aspirating Pump is provided with convenient grooves on the pump shaft at the 50 ml and 100 ml mark. When the red line on the pump shaft and bottom case are lined up and the handle is pulled straight out, the pump will automatically lock at either the 50 ml or 100 ml mark. To unlock the pump after sampling, simply turn the handle 90 degrees.

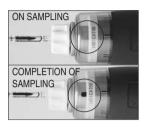






FIG.2.8 TIP STORAGE

2.2.2 CHECKING PRIOR TO USE (Leakage Test)

If there is leakage in the aspirating pump, (a) a lower reading can be obtained, or (b) the detecting reagent in the tube may not change colour. The Kitagawa Aspirating Pump should always be checked for sample leakage into the pump ("inboard" leakage) prior to use. If there is leakage into the pump, the full 100 m ℓ sample volume required will not pass through the detector tube during sampling. This could result in possible low readings on the tube which could lead to injury by understating a hazardous condition. The procedure is as follows:

- ① Insert a sealed, unbroken detector tube into the rubber tube connector. Align the red line on the bottom case and that of shaft, and pull the handle to full stroke locked position.
- Wait one minute. Unlock the handle by turning it 1/4 turn (90 degrees), and check to see if the handle returns to the initial position. When the lock is released under full vacuum, the handle tends to snap back quickly. To prevent possible damage to the locking device allow the handle to return slowly by holding the cylinder and handle securely. If the handle returns completely to the original position, the performance is satisfactory.
- 3) If the pump fails the leak test, the following are possible causes.
 - A) A loose inlet connector holder.
 - B) Cracks and deterioration of the rubber tube connector.
 - C) Deterioration of the grease.

To correct a leaking pump, consult the "MAINTENANCE" section.

2.2.3 MAINTENANCE

(1) RUBBER TUBE CONNECTOR

When the rubber tube connector is sufficiently worn to make tube insertion difficult, or does not hold detector tubes properly, replace the rubber tube connector.

(2) APPLICATION OF NEW VACUUM GREASE

- 1) Pull the handle part way and turn the bottom case counterclockwise to remove it.
- 2 Pull the piston out from the cylinder.
- ③ Wipe off the old grease and dirt from the piston and inside the cylinder using a clean paper towel. Apply a thin coat of vacuum grease to the rubber gasket of the piston. When wiping off the old grease, be careful not to scratch the inside walls of the cylinder. Reassemble pump.

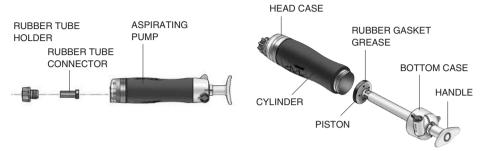


FIG. 2.10 RUBBER TUBE CONNECTOR

FIG. 2.11 TO REMOVE THE PISTON

2.3 GAS CONCENTRATION MEASUREMENT

2.3.1 KITAGAWA DETECTOR TUBE BOXES

Always confirm before sampling that the gas detector tube chosen is suitable for the type of sampling and environment. Be sure to read the necessary items for measurement printed on each tube box. An instruction sheet is enclosed in each tube box. Read the instruction sheet before sampling, as new improvement in chemical reactions and sampling methods will always be described there.

(1) On the tube box (See FIG. 2.12)

- 1 Expiration date of shelf life
- 2 Tube number (catalogue number)
- 3 Gas or vapour to be detected
- 4 Measuring range
- ⑤ Quantity of tubes in the box
- **(6)** Storage instructions (Refer to 1.4.6)
- (7) Manufacturing lot number

(2) In the instruction sheet (See FIG. 2.13)

- (1) Performance of the tube
- 2 Sampling and measurement procedures
- ③ Correction for ambient conditions
- 4 Interferences by single existence and coexistence
- (5) Chemical reaction in the detector tube
- 6 Disposal of used tubes
- 7 Properties of the measuring gas
- Inspection of the aspirating pump
- ① Users responsibility



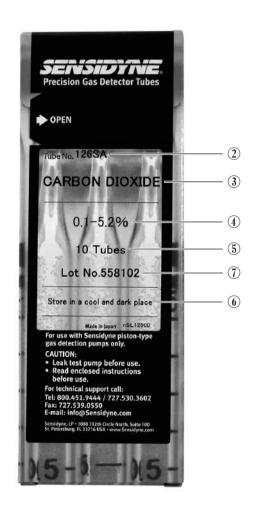


FIG. 2.12 DETECTOR TUBE BOX

INSTRUCTION MANUAL No.126SA CARBON DIOXIDE DETECTOR TUBE

- ★READ CAREFULLY THIS INSTRUCTION MANUAL AND THE INSTRUCTIONS OF THE ASPIRATING PUMP PRIOR TO USING THIS PRODUCT.
- ★DO NOT DISCARD THIS INSTRUCTION MANUAL UNTIL ALL THE TUBES IN THIS BOX ARE USED UP

1. PERFORMANCE:

(1)

Measuring Range: 0.2 - 5.2% 0.1 - 2.6% (*) and Pump Stroke: 1/2 pump stroke 1 pump stroke (*) Graduations on the detector tube are based on 1 pump stroke.

Sampling Time:	2.5 minutes	5 minutes
Colour Change:	Purple blue → Pale	pink
Detectable Limit:	0.01% (100 ppm)	(1 pump stroke)
Operating Temperature:	0 - 40°C (32-104°F	(No temperature correction is necessary.)
Aspirating Pump:	Model AP-20S, AP	P-1S, 400B OR 400A

A CAUTION

- 1. THE DETECTOR TUBE CONTAINS CHEMICAL REAGENTS.
- 2. DO NOT TOUCH THESE REAGENTS DIRECTLY ONCE TUBES WERE BROKEN.
- 3. KEEP THE TUBES OUT OF THE REACH OF CHILDREN.

NOTICE

- 1. USE ONLY WITH PUMP MODELS AP-20S, AP-1S, 400B OR 400A. OTHERWISE, CONSIDERABLE ERROR IN INDICATION MAY OCCUR.
- 2. BEFORE TESTING, CHECK THE ASPIRATING PUMP FOR LEAKS. (REFER TO ITEM 8. INSPECTION OF ASPIRATING PUMP.) ANY PUMPS SHOWING SIGNS OF LEAKAGE SHOULD BE CORRECTED BEFORE USE.
- 3. DO NOT USE THIS TUBE OUTSIDE THE STATED OPERATING TEMPERATURE RANGE.
- 4. STORE TUBES IN A COOL AND DARK PLACE (0-25 °C/32-77°F), AND USE BEFORE EXPIRATION DATE PRINTED ON THE TOP OF THE BOX.
- 5. PRIOR TO USE, READ CAREFULLY ITEM 9. USER RESPONSIBILITY.
- 6. READ THE CONCENTRATION IMMEDIATELY AFTER MEASUREMENT.

2. SAMPLING AND MEASUREMENT:

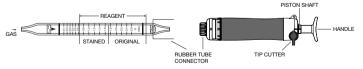


Fig. 1

(2)

1 Break both ends of detector tube.

▲ CAUTION SAFETY GLASSES AND GLOVES SHOULD BE WORN TO PREVENT INJURY FROM SPLINTERING GLASS.

- ② Insert the detector tube into the aspirating pump securely as shown in Fig.1. (Arrow mark shall point to the pump.)
- 3 Align the guide marks on the shaft and stopper of the aspirating pump.
- ④ Pull the pump handle at a full stroke until it locks and wait for 5 minutes or until the completion of sampling is confirmed with the flow indicator of the pump. (See descriptions about the flow indicator in the instruction manual of the pump.)
- ⑤ On completion of sampling, read the scale at the maximum point of the stained layer.
- **(6)** When the concentration is over the scale range, a 1/2 pump stroke can be used to determine concentrations of 0.2 to 5.2 ppm.
 - 1) Remove the detector tube from the pump.
 - 2) Turn the pump handle right or left by $1/4(90^{\circ})$, push back it toward to the pump.
 - 3) Insert the new tube into the aspirating pump.
 - 4) Pull the pump handle at a 1/2 stroke until it locks and wait for 2.5 minutes or until the completion of sampling is confirmed with the flow indicator of the pump.
 - 5) On completion of sampling, read the scale at the maximum point of the stained layer.
 - 6) Then multiply the reading value by 2.

SPECIAL NOTE:

(3)

- I. The scale is calibrated at 20℃ (68°F), 50 %R.H. and 1013hPa. Readings obtained in other circumstances should be corrected.(REFER TO ITEM 3. CORRECTION FOR AMBIENT CONDITIONS)
- II. When the maximum point of the stained layer is unclear or oblique, read the scale at the centre between the longest and shortest points.

3. CORRECTION FOR AMBIENT CONDITIONS:

① Temperature; Correct the tube reading by following temperature correction table.

	Temperature Correction Table								
Tube	Corrected Concentration (%)								
Readings (%)	0°C (32°F)	10 ℃ (50 °F)	20°C (68° F)	30°C (86°F)	40 ℃ (104 °F)				
2.6	2.85	2.75	2.6	2.5	2.4				
2.4	2.6	2.5	2.4	2.3	2.2				
2.2	2.4	2.3	2.2	2.1	2.0				
2.0	2.2	2.1	2.0	1.9	1.8				
1.8	2.0	1.9	1.8	1.7	1.6				
1.6	1.75	1.7	1.6	1.5	1.4				
1.4	1.55	1.5	1.4	1.35	1.3				
1.2	1.3	1.25	1.2	1.15	1.1				
1.0	1.1	1.05	1.0	0.95	0.9				
0.8	0.9	0.85	0.8	0.75	0.7				
0.6	0.65	0.6	0.6	0.6	0.55				
0.5	0.55	0.5	0.5	0.5	0.45				
0.4	0.45	0.4	0.4	0.4	0.35				

- 2 Humidity; No corrections is necessary.
- 3 Atmospheric Pressure;

True concentration = $\frac{\text{Temperature corrected}}{\text{concentration}} \times \frac{1013}{\text{Atmospheric pressure (in hPa)}}$

4. INTERFERENCES:

Coexistence of Nitrogen, Oxygen, Carbon Monoxide, and/or Ethylene does not affect the accuracy of readings.

Coexistence of less than 500ppm of Sulphur Dioxide,100ppm of Hydrogen Sulphide, 200ppm of Hydrogen Cyanide,100ppm of Chlorine or 500ppm of Ammonia does not affect the accuracy of readings. Coexistence of more than 10ppm of Nitrogen Dioxide will give lower readings.

(5) 5. CHEMICAL REACTION IN THE DETECTOR TUBE:

By reacting with alkali, pH indicator is discolouerd.

CO2+ 2NaOH → Na2CO3+H2O

6 6. DISPOSAL OF TUBE :

USED TUBES SHOULD BE DISPOSED CAREFULLY ACCORDING TO RELEVANT REGULATIONS, IF ANY.

⑦ 7. HAZARDOUS AND DANGEROUS PROPERTIES OF CARBON DIOXIDE :

T.L.V.◆: 5000 ppm

Explosive range in air: -

◆Threshold Limit Value established by the American Conference of Governmental Industrial Hygienists, 2010.

(8) 8. INSPECTION OF ASPIRATING PUMP:

Checking for leaks:

- ① Insert a sealed, unbroken detector tube into the pump.
- ② Align the guide marks on the shaft and stopper of the pump.
- 3 Pull the handle to full stroke (100ml) and wait for 1 minute as it is.
- ① Unlock the handle and allow it to return slowly into the pump by holding the cylinder and handle securely.

▲ CAUTION! HANDLE WILL TEND TO SNAP BACK INTO THE PUMP QUICKLY.

(3) If the handle returns completely to the original position, the performance is satisfactory. Otherwise, refer to maintenance procedures in the instruction manual of the pump to correct the leakage.

9. USER RESPONSIBILITY:

It is the sole responsibility of the user of this equipment to ensure that the equipment is operated, maintained, and repaired in strict accordance with these instructions and the instructions provided with each Model AP-20, AP-20S, 400B, AP-1, AP-1S or 400A aspirating pump, and that detector tubes are not used beyond their expiration date or have a colour change different to that stated in the Performance specifications.

The Manufacturer and Manufacturer's Distributors shall not be otherwise liable for any incorrect measurement or any damages, whether damages result from negligence or otherwise.

* Product specifications are subject to change without any prior notice.

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2.3.2 OPERATION FOR MEASUREMENT

As the method of operating the Kitagawa gas detector tube varies somewhat according to the kind of tubes to be used, read the instruction sheet for the gas detector tube carefully and follow the instructions.

(1) Prepare Aspirating pump.

Check the pump for leaks in accordance with "CHECKING PRIOR TO USE". (Refer to page 8)

(2) Cut both ends of the gas detector tube.

Insert the tip of the gas detector tube into the tip cutter and scratch the tip of tube by rotating it for one revolution, then pull it toward you. (The glass tip can be thrown away by removing the tip cutter cap.)





(3) Connect the gas detector tube to the aspirating pump.

The sample gas must be drawn through the gas detector tube in the correct direction. Insert the gas detector tube into the rubber tube connector with the tube's directional arrow pointing toward the pump.



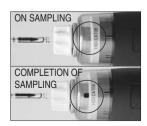
(4) Pull the handle.

Align the red line on the bottom case and that on the shaft and pull the pump handle to its full 100cc locking position. If the sample calls for a half stroke, pull out the pump handle until 50cc line appears, and shaft will be locked at 50cc.



(5) Draw the sample gas.

Draw the sample gas for the specified time at the desired sampling point, and confirm with the flow indicator that the sample is completed. The sample time required for each detector tube is stated clearly in the instruction sheet.



(6) Return the handle.

When the sample is completed, turn the handle 1/4 turn (90 degrees) clockwise or counterclockwise to unlock the handle. Confirm that the handle remains extended. (If the handle returns part way, the sample is incomplete, and this will cause a low reading). Some detector tubes require extra pump strokes (i.e., more than 100cc of air). In this case, push back the handle and repeat the operation.



(7) Read the concentration.

Remove the gas detector tube from the aspirating pump after a prescribed quantity of gas is aspirated and read the concentration of gas at the maximum end of the discolored layer against graduations on the direct reading detector tube or by using the concentration chart for the concentration chart method tube.

Read the concentration chart in the following way. Bring the end of the detecting reagent on the gas inlet side in line with the 0-0 line on the concentration chart and the other end in line with the X-X line respectively to read out gas concentration at the end of the discoloured layer. When the end is oblique, read at the middle point of the oblique discoloured layer.



FIG. 2.19 READOUT OF DIRECT READING TUBE

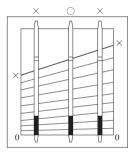


FIG. 2.20 READOUT OF CONCENTRATION CHART TUBE

2.3.3 CORRECTION

(1) Temperature correction

As graduations showing the concentration of gas on the gas detector tube are generally calibrated at 20° C, the reading value may have to be corrected according to the kind of the gas detector tube and the measuring range. (Refer to examples shown in FIG. 2.21 and 2.22.)

The temperature at which the gas detector tube is used generally ranges from 0 to 40 $^{\circ}$ C and it is impossible to obtain an accurate measured value at temperatures outside of the range. The temperature means that of the gas detector tube (usually the temperature of the sampled gas.) , and even if the temperature of sample gas varies somewhat, it will not affect the measured value because of a small thermal capacity of gas. Temperature correction need not be applied to the gas detector tube to which the temperature correction chart is not attached.

Example 1: When the reading value is 500 ppm at 10° C, the true concentration is 470 ppm.

Scale	True Concentration (ppm)					
Readings (ppm)	0 ℃ (32 °F)	10°C (50° F)	20˚C (68˚ F)	30 ℃ (86°F)	40 ℃ (104 °F)	
1,000	870	930	1,000	1,030	1,060	
900	780	840	900	930	960	
700	610	660	700	720	740	
600	520	560	600	620	640	
(500)—	430	► (470)	500	520	540	
400	350	370	400	410	430	
300	260	280	300	310	320	
200	180	190	200	210	220	
100	90	100	100	100	110	

FIG. 2.21 TEMPERATURE CORRECTION

Example 2 : When the reading value is 550 ppm at 25 $^{\circ}$ C, the true concentration is 560 ppm which is found by proportional allotment of each concentration and temperature as shown in FIG. 2.22.

Scale	True Concentration (ppm)						
Readings (ppm)	0 ℃ (32 °F)	10℃ (50°F)	20℃ (68°F)	30℃ (86°F)	40 ℃ (104°F)		
1,000	870	930	1,000	1,030	1,060		
900	780	840	900	930	960		
700	610	660	700	720	740		
600	520	560	600	620	640		
500	430	470	500	520	540		
400	350	370	400	410	430		
300	260	280	300 /	310	320		
200	180	190	200	210	220		
100	90	100	100	100	110		

()			
Scale Readings	20℃	(25°C)	30℃
600	600	(610)	620
(550)	(550)	(560)	(570)
500	500	(510)	520
(450)	(450)	(457.5)	(465)
400	400	(405)	410
- 10			

Numerals in parentheses are determined by proportional allotment.

FIG. 2.22 CORRECTION UNDER EXPANSION

(2) Pressure correction

When gas is measured in an atmosphere under pressure or reduced pressure, a true concentration can be obtained by following equation.

True concentration = Reading value X 101.3/Atmospheric pressure (in kpa)

True concentration = Reading value X 760 / Atmospheric pressure (in mmHg)

P (mmHg): Atmospheric pressure in the place where gas is measured.

2.3.4 SPECIAL PRECAUTIONS

- (1) When the concentration chart tube is used, be sure to use the concentration chart found in the box.
- (2) It is desirable to read the concentration immediately after measurement because the discoloured layer gets longer gradually or fades after measurement in some gas detector tubes. It is recommended to mark the end of the discoloured layer by a soft pen for keeping a record of the concentration.
- (3) If the end of the discoloured layer is slanted, read the numerical value at the middle of the oblique layer.

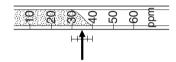


FIG. 2.23 OBLIQUE LAYER (Channeling)

2.4 SPECIAL DETECTOR TUBES FOR IONS AND TWA 2.4.1 ION DETECTOR TUBES

- (1) The ion detector tubes, used for measuring the concentration of ions in the water, include tubes for sulphide, chloride, iron, cyanide, residual chlorine and salinity detector tubes. The concentration of an ion is measured by three different method as follows: (1) Injection method (see FIG. 2.24): Sample solution taken by a syringe is injected into the detector tube.
- (2) Immersion method (see FIG.2.25): One end of the detector tube is immersed in the sample solution to draw it up by capillarity action.
- (3) Direct sampling method (see FIG. 2.26): Sample solution is drawn into the tube end by a rubber bulb (an extra option). The concentration of the ion is read by the direct reading method or the concentration chart method as in case of the gas detector tubes.

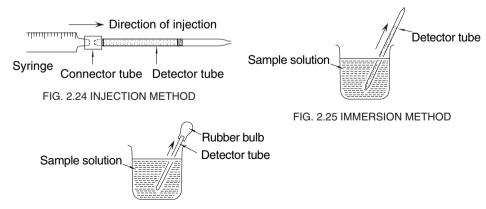


FIG. 2.26 DIRECT SAMPLING METHOD

2.4.2 TWA DETECTOR TUBES

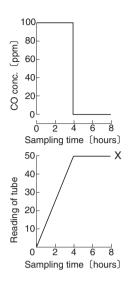
The TWA detector tube is designed to produce a stain corresponding to a time-concentration relationship. For instance, when 200 ppm of gas is passed through the detector tube for one hour and when 100 ppm of gas is passed through the detector tube at the same flow rate for two hours the two conditions will produce equal stain lengths. Therefore, the length of the discoloured layer is given in terms of integrated value of gas measured, making it possible to derive the TWA concentration for the time during which gas is passed through the detector tube from the length of the discoloured layer. This detector tube is most suited for measuring the concentration of gas to which a person is exposed.

All the examples in which gas is passed through the detector tube as shown on the upper tier in FIG. 2.27-2.30 give a TWA concentration of 50 ppm after the lapse of 8 hours. When gas is passed through the TWA detector tube in these examples, the same reading value (X) is given by the detector tube after 8 hours in all the cases, though the reading value varies with the lapse of time as shown on lower tier.

(1) How to use

- ① Connect the detector tube to the personal sampler or an equivalent, sampler and set the flow using a soap film flow meter to the specified flow rate (6 to 10 m \ell /min).
- ② Place the detector tube and sampler in the sample area and draw the sample gas for the required time.
- (3) Since graduations on the detector tube are calibrated in the value obtained for eight hours (480 minutes), convert the reading value obtained for a shorter time by the following equation.

TWA value = Reading value on detector tube
$$\times \frac{480}{\text{Sampling time (minutes)}}$$



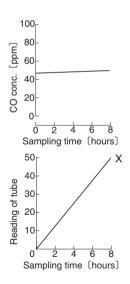
100 80 CO conc. [ppm] 60 40 20 0 4 6 Sampling time (hours) 50₁ 40 Reading of tube 30 20 10 Sampling time (hours)

FIG. 2.27 SAMPLE DELIVERY MODE A

FIG. 2.28 SAMPLE DELIVERY MODE B

TWA value =
$$\frac{100 \times 4 + 0 \times 4}{8}$$
 = 50 ppm

TWA value =
$$\frac{100 \times 2 + 0 \times 4 + 100 \times 2}{8} = 50 \text{ ppm}$$



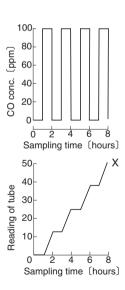


FIG. 2.29 SAMPLE DELIVERY MODE C

FIG. 2.30 SAMPLE DELIVERY MODE D

TWA value =
$$\frac{50 \times 8}{8}$$
 = 50 ppm TWA value = $\frac{0 \times 1 + 100 \times 1 + 0 \times 1 + 100 \times 1 + 0 \times 1 + 100 \times 1 + 0 \times 1 + 100 \times 1}{8}$ = 50 ppm

3. DATA OF KITAGAWA GAS DETECTOR TUBES

3.1 EXPLANATION OF THE TUBE TABLE

3.1.1 PERFORMANCE

An external view with printed graduations and specifications are described.

3.1.2 RELATIVE STANDARD DEVIATION

Errors which affect the measuring value are composed of a systematic error and an incidental error. The incidental error, however, originates in quality control throughout gas detector tube manufacturing process and it cannot be decreased by any caution in operation and handling by the user.

The incidental error of the Kitagawa gas detector tube is controlled as the relative standard deviation, shown in every tube table. The relative standard deviation is defined as a percentage in which a standard deviation is divided by a mean value of readings. Further, the Kitagawa gas detector tube is being controlled by dividing the full measuring range into three ranges, low, middle and high ranges, and giving quality control for the every range. (Some kinds of tubes are controlled with four divided ranges.) These controlled values are set forth as "RSD-low" for the low measuring range, "RSD-mid." for the middle range and "RSD-high" for the high range.

3.1.3 CHEMICAL REACTION

Chemical reaction for the measuring gas and reagents are shown.

3.1.4 CALIBRATION OF THE TUBE

The analytical method of gas, which is used for calibration and final accuracy testing of the tube, is set forth.

3.1.5 INTERFERENCE & CROSS SENSITIVITY

Coexistent gases, which give interferences to the measurement, are described. On the other hand, some of listed gases herein may be possible to detect in quantitative or qualitative analysis.

© MARK

In cases where gas concentration is read using a conversion chart as shown in the tube instruction sheet, a mark $\mathbb C$ is affixed after the tube number in this brochure. For example: $190U \mathbb C$. However, this mark $\mathbb C$ is shown only in the brochure and does not appear on the printed tube box or in the instrustion sheet. When ordering such tubes, it is unnecessary to include the $\mathbb C$ mark on your purchase orders.

3.2 KITAGAWA SHORT-TERM DETECTOR TUBES

ACETALDEHYDE



1. PERFORMANCE

1) Measuring range 0.004-1.0%Number of pump stroke $1(100\text{m}\ell)$

2) Sampling time : 1 minute/1 pump stroke 3) Detectable limit : 0.0005 % (5ppm)

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Concentration chart method

8) Colour change : Yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

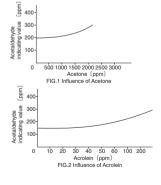
By reacting with Hydroxylamine hydrochloride, Hydrogen chloride is liberated and PH indicator discoloured. CH₃CHO + NH₂OH • HCI → HCI + CH₃CH = NOH + H₂O

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substa	nce	Interference	ppm	Coexistence
Acetone	FIG.1	Similar stain is produced	1400	Higher readings are given.
Acrolein	FIG.2	"	35	"
Methyl ethyl ke	tone	"	900	"
Methyl isobutyl	ketone	"	2900	"



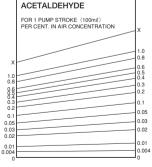
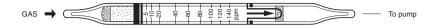


Chart	Corrected Concentration (%)						
Readings (%)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)		
1.0		1.6	1.0	0.50	0.30		
0.8	1.9	1.3	0.8	0.40	0.20		
0.6	1.6	1.05	0.6	0.30	0.16		
0.5	1.45	0.9	0.5	0.25	0.14		
0.4	1.3	0.8	0.4	0.20	0.13		
0.3	1.2	0.65	0.3	0.15	0.11		
0.2	0.95	0.45	0.2	0.10	0.08		
0.1	0.6	0.2	0.1	0.07	0.05		
0.05	0.25	0.09	0.05	0.04	0.03		
0.03	0.08	0.05	0.03	0.025	0.01		
0.02	0.03	0.025	0.02	0.015	0.007		
0.01	0.02	0.015	0.01	0.007	0.004		

ACETALDEHYDE



1. PERFORMANCE

1) Measuring range : 5-140 ppm Number of pump strokes $1(100m\ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 2 ppm

4) Shelf life 2 years (Necessary to store in refrigerated conditions ; $0 \sim 10 \, \text{°C}$)

5) Operating temperature : 10 ~ 40 °C

0~80% R.H. (See "HUMIDITY CORRECTION TABLE") 6) Operating humidity 7) Reading Direct reading from the scale calibrated by 1 pump stroke : Yellow → Pink

8) Colour change

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

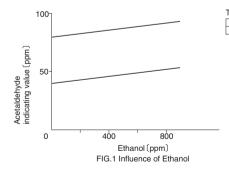
By reacting with Hydroxylamine hydrochloride, Hydrogen chloride is liberated and PH indicator discoloured. $CH_3CHO + (NH_2OH)_3 \cdot H_3PO_4 \rightarrow H_3PO_4 + CH_3CH = NOH + H_2O$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

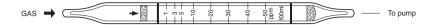
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Other Aldehydes	Similar atain is produced.	Higher readings are given.
Ketones	"	"
Ethanol FIG. 1	The accuracy of readings is not affected.	"



FABLE OF COEFFICIENT FOR HUMIDITY CORRECTION(50 % R.H.)									
Humidity(%R.H.)									
Coefficient	1.70	1.40	1.20	1.10	1.05	1.00	0.97	0.95	0.93

ACETIC ACID



1. PERFORMANCE

1) Measuring range 1-50 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 0.2 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

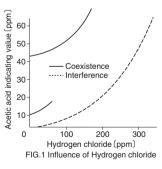
By reacting with alkali, PH indicator is discoloured. CH₃COOH + Na₂SiOnH₂O → CH₃CO₂Na + H₂O

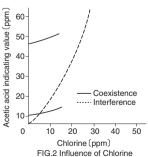
4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance pp		ppm	Interference	ppm	Coexistence
Sulphur dioxide			Similar stain is produced.	1/20 × Acetic acid	Higher readings are given.
Nitrogen dioxide		300	"	10	Unclear stain is produced.
Hydrogen chloride FIG.1			Pink stain is produced.	2 × Acetic acid	Higher readings are given.
Chlorine	FIG.2		Yellow stain is produced.	5	"





Tube	Corrected Concentration (ppm)						
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)		
50	63	57	50	45	40		
40	50	45	40	36	31		
30	37	33	30	27	23		
20	24	22	20	18	16		
10	12	11	10	9	8		
5	6	6	5	5	4		
3	4	4	3	3	2		
1	1	1	1	1	1		

ACETIC ANHYDRIDE



1. PERFORMANCE

7) Colour change

1) Measuring range 1-15 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Acetic anhydride concentration is determined by using a conversion chart

at 1 pump stroke.

∴ Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured.

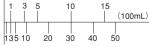
4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide		Similar stain is produced. Higher readings are given.	
Nitrogen dioxide	300	"	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	//
Acetic acid		Similar stain is produced.	"





No.216S tube reading (ppm)

ACETONE



1. PERFORMANCE

: 1.0-5.0 % 0.1-2.0% 1) Measuring range Number of pump strokes $1/2(50 \text{m} \ell)$ 1(100ml) 2) Sampling time 1.5 minute/1 pump stroke 3) Detectable limit $0.02\%(200 \text{ ppm})(100\text{m}\ell)$

4) Shelf life 3 years 5) Operating temperature 0~40℃

6) Temperature compensation Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange → brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3COCH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

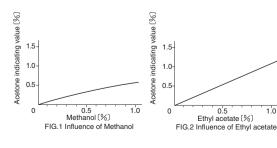
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence		
Alcohols	FIG.1	Similar stain is produced.		Higher readings are given.	
Other ketones		"		"	
Aromatic hydrocarbons		"		"	
Esters	FIG.2	"		"	
Halogenated hydrocarbons	S	Whole reagent is slightly discoloured to pale brown.	0.5%	"	

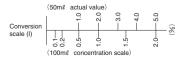
^{*}Methanol is indicated with half the sensitivity and Ethyl acetate has the same sensitivity with Acetone. (NOTE)

- 1) If the discolouration is over the scale, replace the tube with new one and pull the handle at half stroke (to 50mℓ line). And read a figure from the scale on the tube.
- 2) Correct the reading value with the TEMPERATURE CORRECTION TABLE first, and convert the value into an actual concentration by using the conversion scale shown in the instruction sheet.



TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (%)				
Readings (%)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)
2.0	_	2.38	2.00	1.78	1.60
1.5	2.20	1.76	1.50	1.30	1.16
1.0	1.44	1.18	1.00	0.86	0.76
0.5	0.72	0.60	0.50	0.42	0.36
0.2	0.30	0.25	0.20	0.16	0.14
0.1	0.16	0.12	0.10	0.08	0.08



1.0

ACETONE



1. PERFORMANCE

1) Measuring range 0.01-4.0%Number of pump strokes $1(100m\ell)$

2) Sampling time : 1 minute/1 pump stroke 3) Detectable limit : 0.001 % (10 ppm)

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10 \,^{\circ}\text{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Hydrogen chloride of Hydroxylamine hydrochloride is liberated and acidfied, and PH indicator is discoloured. CH3COCH3 + NH2OH ⋅ HCI→HCI + (CH3) 2C = NOH + H2O

4. CALIBRATION OF THE TUBE

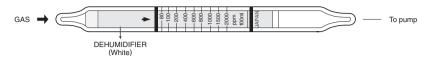
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acrolein	Similar stain is produced.	20	Higher readings are given.
Acetaldehyde	"	30	"
Methyl ethyl ketone	"	150	"
Methyl isobutyl ketone	"	400	"

Tube		Corrected Concentration (%)					
Readings	0℃	10 °C	20℃	30℃	40°C		
(%)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)		
4.0		_	4.0	3.2	2.7		
3.5		_	3.5	2.8	2.3		
3.0	_	4.1	3.0	2.4	2.0		
2.5	4.8	3.7	2.5	2.0	1.7		
2.0	3.9	2.8	2.0	1.6	1.4		
1.5	2.9	2.0	1.5	1.2	1.0		
1.0	1.8	1.3	1.0	0.8	0.7		
0.7	1.1	0.9	0.7	0.6	0.5		
0.5	0.8	0.7	0.5	0.4	0.3		
0.3	0.5	0.4	0.3	0.3	0.2		
0.1	0.16	0.12	0.1	0.08	0.05		

ACETONE



1. PERFORMANCE

1) Measuring range : 125-5,000 ppm 50-2,000 ppm 20-800 ppm Number of pump strokes $1/2(50m\ell)$ $1(100m\ell)$ $2(200m\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit $10 \text{ ppm} (200 \text{m} \ell)$ 4) Shelf life 2 years5) Operating temperature $0 \sim 40 \text{ C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3COCH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Brown stain is produced.	Higher readings are given.
Other Ketones	"	"
Esters	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons (more than C ₃)	Whole layer is discoloured to Brown.	If the top of discoloured layer is possible to read, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	"

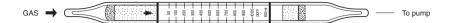
(NOTE)

In case of 1/2 or 2 pump strokes, following formula is available for the actual concentration.

1/2 pump strokes: Actual concentration = $2.5 \times$ Temperature corrected value 2 pump strokes: Actual concentration = $0.4 \times$ Temperature corrected value

Tube	C	Corrected Concentration (ppm)					
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)		
2,000	3,500	2,800	2,000	1,600	1,300		
1,500	2,570	2,000	1,500	1,250	1,000		
1,000	1,670	1,250	1,000	850	700		
800	1,300	1,000	800	700	600		
600	1,000	800	600	550	450		
400	650	500	400	350	300		
200	300	200	200	200	200		
100	200	100	100	100	100		
50	100	50	50	50	50		

ACETYLENE



1. PERFORMANCE

1) Measuring range : 50-1000 ppmNumber of pump strokes $: 1(100\text{m}\ell)$

2) Sampling time : 3 minutes/1 pump stroke

3) Detectable limit : 10 ppm 4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→ Brownish blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Molybadate is reduced and molybdeum blue is produced. $HC \equiv CH + PdSO_4 + (NH_4) 2MoO_4 \rightarrow Mo_3O_8$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Carbon monoxide	Whole layer is discoloured to Green or Blue.	50	Whole layer is discoloured to Green or Blue and higher readings are given.
Hydrogen (over 40 ℃)	"	10%	Whole layer is discoloured to Green and higher readings are given.
Unsaturated hydrocarbons such as Ethylene, Propylene, Butylene, etc.	Similar stain is produced.	10	Higher readings are given.
Saturated hydrocarbons such as Propane, Butane, etc. (more than C ₃)	"	10	"
Butadiene	Original colour is faded to white.	25	Original colour is faded to White and lower readings are given.
Hydrogen sulphide	Whole layer is discoloured to Black.	10	Black stain is produced.
Ammonia	Original colour is faded to White.		Original colour is faded to White and lower readings are given.
Hydrogen cyanide	"		Blue stain is produced and higher reading are given.
Chlorine	Yellowish orange or Yellowish brown stain is produced.		Higher readings are given.
Nitrogen dioxide	"		"
Carbon disulphide	"		"

ACETYLENE · ETHYLENE

1. PERFORMANCE

1) Measuring range:

 $\begin{array}{lll} \mbox{Acetylene} & 20\mbox{-}300 \mbox{ ppm} \\ \mbox{Ethylene} & 200\mbox{-}2000 \mbox{ ppm} \\ \mbox{Number of pump strokes} & 1 (100\mbox{m}\ell) \end{array}$

2) Sampling time : 3 minutes/1 pump stroke with orifice

3) Detectable limit:

Acetylene 0.1 ppm Ethylene 1 ppm 4) Shelf life 1 year 5) Operating temperature $10 \sim 40^{\circ}$ C

6) Temperature compensation:

Acetylene No temperature correction is necessary.

Ethylene Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change:

Acetylene Yellow→Dark brown Ethylene Pale yellow→Blue

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 10 %

3. CHEMICAL REACTION

Acetylene detector tube : $HC \equiv CH + K_2Pd(SO_3)_2 \rightarrow Pd$

Ethylene detector tube : $H_2C = CH_2 + PdSO_4 + (NH_4)_2MoO_4 \rightarrow Mo_3O_8$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene detector tube			
Carbon monoxide		10	Higher readings are given.
Hydrogen		5000	"
Ethylene		2000	"
Ethylene detector tube			
Carbon monoxide		1350	Higher readings are given.
Acetylene		370	"
Propylene	Blue stain is produced.		"

Tube	Corrected Concentration (%)					
Readings (ppm)	10 °C (50 °F)	20 °C (68 °F)	30 °C 40 °C (86 °F) (104 °F)			
2000	1550	2000				
1800	1400	1800	2050			
1600	1300	1600	1900			
1400	1150	1400	1600			
1200	1000	1200	1400			
1000	900	1000	1200			
800	750	800	950			
600	600	600	700			

ACROLEIN



1. PERFORMANCE

1) Measuring range 0.005-1.8 % Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke 3) Detectable limit : 0.0005 % (5 ppm)

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Concentration chart method

8) Colour change : Yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3.CHEMICAL REACTION

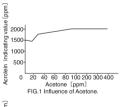
By reacting with Hydroxylamine hydrochloride, Hydrogen chloride is liberated and PH indicator is discoloured. CH₂ = CHCHO + NH₂OH ⋅ HCI → HCI + CH₂ = CHCH = NOH + H₂O

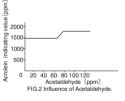
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Acetone	FIG.1	Similar stain is produced.	20	Higher readings are given.
Acetaldehyde	FIG.2	"	70	"
Methyl ethyl ketone	FIG.3	"	60	"
Methyl isobutyl ketone)	"	500	"





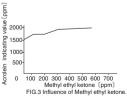
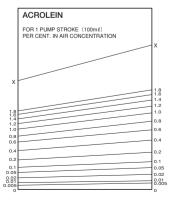


Chart		Corrected Concentration (%)				
Readings (%)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30°C (86°F)	40 °C (104 °F)	
1.8	_	2.3	1.8	1.3	1.0	
1.6	_	2.1	1.6	1.1	0.9	
1.4	_	1.85	1.4	1.0	0.75	
1.2	2.2	1.6	1.2	0.85	0.65	
1.0	1.9	1.35	1.0	0.7	0.5	
0.8	1.55	1.1	0.8	0.55	0.4	
0.6	1.2	0.85	0.6	0.4	0.3	
0.4	0.85	0.6	0.4	0.2	0.15	
0.2	0.5	0.3	0.2	0.12	0.1	
0.1	0.2	0.15	0.1	0.06	0.04	
0.05	0.1	0.08	0.05	0.03	0.015	
0.02	0.07	0.05	0.02	0.02	0.01	
0.01	0.03	0.02	0.01	0.01	0.005	



ACRYLIC ACID



1. PERFORMANCE

1) Measuring range 1-50 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Acrylic acid concentration is determined by using a conversion chart at

1 pump stroke.

7) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

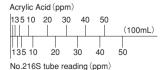
By reacting with alkali, PH indicator is discoloured. CH₂ = CHCHO + NaOH → C₂H₃COONa + H₂O

4. CALIBRATION OF THE TUBE

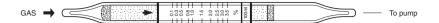
VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence	
Sulphur dioxide		Similar stain is produced.	Higher readings are given.	
Nitrogen dioxide	300	//	The top of discoloured layer becomes unclear.	
Hydrogen chloride		Pink stain is produced.	Higher readings are given.	
Chlorine		Blueish yellow stain is produced.	"	
Acetic acid		Similar stain is produced.	"	



ACRYLONITRILE



1. PERFORMANCE

1) Measuring range 0.1-3.5%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 0.001%(10 ppm)4) Shelf life 3 years5) Operating temperature $0 \sim 40\%$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange→Dark green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = CHCN + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene	Whole layer is discoloured to Brown.	3%	Whole layer is discoloured to Brown and higher readings are given.
Propane	"	0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	Higher readings are given.

Tube	Corrected Concentration (%)								
Readings (%)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 ℃ (86 °F)	40 ℃ (104 °F)				
3.5	5.5	4.6	3.5	3.2	2.8				
3.0	4.7	3.9	3.0	2.7	2.4				
2.5	3.9	3.2	2.5	2.2	2.0				
2.0	3.1	2.6	2.0	1.8	1.6				
1.5	2.4	1.9	1.5	1.4	1.2				
1.0	1.6	1.3	1.0	0.9	0.8				
0.8	1.3	1.0	0.8	0.7	0.7				
0.5	0.8	0.7	0.5	0.5	0.4				
0.3	0.5	0.4	0.3	0.3	0.3				
0.1	0.2	0.1	0.1	0.1	0.1				

ACRYLONITRILE



1. PERFORMANCE

1) Measuring range 10-500 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION COEFFICIENT TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = CHCN + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

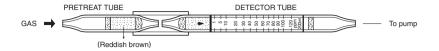
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Alcohols	Similar stain is produced.		Higher readings are given.
Esters	"		"
Ketones	"		"
Aromatic hydrocarbons	"		"
Halogenated hydrocarbons			If the maximum end point of Pale blue stain is discernable, the accuracy of the readings is not affected.

TEMPERATURE CORRECTION COEFFICIENT TABLE

Tube		Correction Coefficient				
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)	
500	_	-	1.00	0.90	0.78	
400	_	1.25	1.00	0.90	0.78	
300	1.60	1.25	1.00	0.90	0.78	
200	1.60	1.25	1.00	0.90	0.78	
100	1.55	1.25	1.00	0.90	0.80	
50	1.34	1.20	1.00	0.94	0.88	
30	1.30	1.17	1.00	0.97	0.90	
10	1.30	1.10	1.00	1.00	0.90	

ACRYLONITRILE



1. PERFORMANCE

1) Measuring range : 1-120 ppm Number of pump strokes : $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit : 0.5 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen cyanide is produced. This Hydrogen cyanide reacts with Mercuric chloride, then Hydrogen chloride is liberated and PH indicator is discoloured.

 $CH_2 = CHCN + CrO_3 + H_2SO_4 \rightarrow HCN$

 $HCN + HgCI_2 {\rightarrow} HCI$

4. CALIBRATION OF THE TUBE

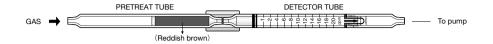
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen cyanide	Similar stain is produced.	2	Higher readings are given.
Methyl ethyl ketone		600	Lower readings are given.
Styrene		less than 350	The accuracy of readings is not affected.
Butadiene		200	Lower readings are given.

Scale		True Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40°C (104°F)	
120	165	140	120	104	90	
100	142	117	100	87	77	
90	127	105	90	79	70	
80	112	93	80	70	62	
70	98	81	70	62	55	
60	84	70	60	53	48	
50	69	58	50	45	41	
40	55	46	40	37	34	
30	41	34	30	28	27	
20	26	22	20	20	20	
10	12	10	10	10	10	

ACRYLONITRILE



1. PERFORMANCE

1) Measuring range : 1-20 ppm 0.5-10 ppm 0.25-5 ppm 0.2-4 ppm Number of pump strokes 1(100ml) $2(200 \text{m} \ell)$ 4 (400m l) 5 (500ml)

2) Sampling time : 1.5 minutes/1 pump stroke 3) Detectable limit : 0.05ppm (500m $\ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions : $0 \sim 10^{\circ}$ C)

5) Operating temperature : 0 ~ 40 ℃

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation: Direct reading from the scale calibrated by 1 pump stroke 7) Reading : Yellow → Red

8) Colour change

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen cyanide is produced, Hydrogen cyanide reacts with Mercuric chloride, then Hydrogen chloride is liberated and PH indicator is discoloured.

 $CH_2 = CHCN + CrO_3 + H_2SO_4 \rightarrow HCN$ HCN + HgCI₂→HCI

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Butadiene	Less than 350 ppm		The accuracy of readings is not affected.
Toluene	Less than 600 ppm		"
Hexane	Less than 800 ppm		"
Styrene	Less than 720 ppm		"
Hydrogen cyanide		Similar stain is produced.	Higher readings are given.

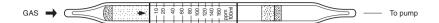
(NOTE)

In case of 2 to 5 pump strokes, following formula is available for actual concentration.

Actual concentration = Temperature corrected value $\times \frac{1}{\text{Number of pump strokes}}$

Tube	Corrected Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 ℃ (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
20	26	22.5	20	18	17
18	23	20	18	16.5	15.5
16	20.5	18	16	15	14
14	18	15.5	14	13	12
12	15.5	13.5	12	- 11	10.5
10	13	11	10	9.5	9
8	10.5	9	8	7.5	7
6	8	7	6	6	5.5
4	5.5	4.5	4	4	4
2	3	2	2	2	2
1	1.5	1	1	1	1

ALLYL ALCOHOL



1. PERFORMANCE

1) Measuring range 20-500 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (10 \sim 40 $^{\circ}$ C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading Graduations printed on the tube are calibrated by Methyl methacrylate at 1 pump stroke and Allvl alcohol concentration is determined by using a converstion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

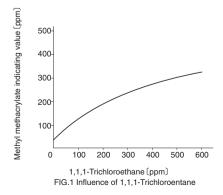
 $CH_2 = CHCH_2OH + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

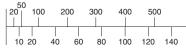
Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Pale brown.	"
Aromatic hydrocarbons	"	"
Esters	"	"
Ketones	"	"
Halogenated hydrocarbons FIG.1	"	"



TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)				
Readings (ppm)	0°C (32°F)	10 ℃ (50 °F)	20°C (68°F)	30 °C (86°F)	40 °C (104 °F)
500	_	600	500	430	380
400	_	480	400	350	300
300	480	360	300	260	230
200	320	240	200	170	150
100	240	120	100	90	80
50	80	60	50	43	38
20	32	24	20	17	15

Concentration of Allyl alcohol (ppm)



No.184S tube reading (ppm)



1. PERFORMANCE

1) Measuring range 0.5-10%Number of pump strokes $1(100m\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit ∴ 0.04 % 4) Shelf life ∴ 3 years 5) Operating temperature ∴ 0 ~ 40 °C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pink→Gray · Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $NH_3 + H_3PO_4 \rightarrow (NH_4)_2HPO_4$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



1. PERFORMANCE

1) Measuring range 50-900 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

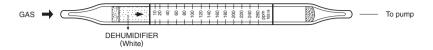
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $NH_3 + H_3PO_4 \rightarrow (NH_4)_2HPO_4$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substance	Interference	ppm	Coexistence
Amines	Similar stain is produced.		Higher readings are given.
Chlorine	The accuracy of readings is not affected.	2	Lower readings are given.
Sulphur dioxide	"	NH₃conc.×1/4	"



1. PERFORMANCE

1) Measuring range Number of pump strokes 10-260 ppm 5-130 ppm 1 $(100m\ell)$ 2 $(200m\ell)$ 2) Sampling time 1 minute/1 pump stroke 10.5 ppm $(200m\ell)$

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $NH_3 + H_3PO_4 \rightarrow (NH_4)_2HPO_4$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

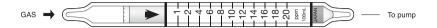
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Amines	Similar stain is produced.		Higher readings are given.
Sulphur dioxide	The accuracy of readings is not affected.	NH₃ conc. × 1/5	Lower readings are given.
Chlorine	"	2	"

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $1/2 \times$ Reading value



1. PERFORMANCE

1) Measuring range : 1-20 ppm 0.5-10 ppm 0.2-4 ppm Number of pump strokes $1(100 \text{m} \ell)$ $2(200 \text{m} \ell)$ $5(500 \text{m} \ell)$

2) Sampling time ∴ 1 minute/1 pump stroke 3) Detectable limit ∴ 0.1 ppm (100m ℓ)

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. NH₃ + H₃PO₄→ (NH₄)₂HPO₄

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.

(NOTE)

When the concentration is below 1 ppm, 2 to 5 pump strokes can be used to determine the lower concentration. Following formula is available for actual concentration.

 $Actual\ concentration = Reading\ value \times \frac{1}{Number\ of\ pump\ strokes}$



1. PERFORMANCE

1) Measuring range : 10-200 ppm 5-100 ppm 1-20 ppm Number of pump strokes $1/2 \, (50 \text{m} \, \ell)$ $1 \, (100 \text{m} \, \ell)$ $5 \, (500 \text{m} \, \ell)$

2) Sampling time ∴ 1 minute/1 pump stroke 3) Detectable limit ∴ 0.2 ppm (500mℓ)

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. NH₃ + H₃PO₄ → (NH₄)₂HPO₄

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Amines	Similar stain is produced.	Higher readings are given.		
Chlorine	The accuracy of readings is not affected.	Lower readings are given.		
Sulphur dioxide	"	"		

(NOTE)

When the concentration is below 5 ppm, 5 pump strokes can be used to determine the lower concentration and following formula is available to obtain the actual concentration.

Actual concentration = Temperature corrected concentration \times 1/5

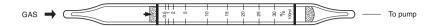
When the concentration is over 100 ppm, 1/2 pump strokes can be used to determine the higher concentration and following formula is available to obtain the actual concentration.

Actual concentration = Temperature corrected concentration \times 2

COEFFICINET TABLE FOR TEMRERATURE CORRECTION (AT 20 °C)

Temperature (°C)	0	1	2	3	4	5	6	7	8	9	10~40
Coefficient	0.90	0.91	0.92	0.93	0.94	0.95	0.96	0.97	0.98	0.99	1.00

Actual concentration = Reading value X Coefficient for temperature correction



1. PERFORMANCE

1) Measuring range 0.5-30%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 0.01%4) Shelf life 3 years 5) Operating temperature $0 \sim 40\%$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pink→Blue · Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Ammonia reacts with Cobalt chloride (II) and complex salt is produced.

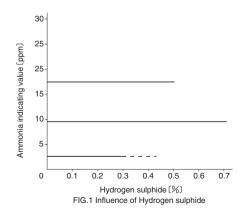
 $NH_3 + \alpha CoCI_2 + \beta H_2O \rightarrow CoCI_2 \cdot \alpha NH_3 \cdot \beta H_2O$

4. CALIBRATION OF THE TUBE

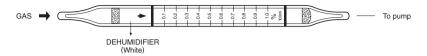
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Hydrogen sulphide FIG.1	0.01%	Whole layer is discoloured to Black.	0.3%	If the maximum end point of the stain is discernable, the accuracy of readings is not affected.
Amines				The accuracy of readings is not affected.



Tube	Corrected Concentration (%)					
Readings (%)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40°C (104°F)	
30	-	35.0	30.0	27.0	25.0	
25	35.0	30.0	25.0	23.0	21.0	
20	28.0	24.0	20.0	18.0	17.0	
15	22.0	18.0	15.0	13.0	12.0	
10	15.0	12.0	10.0	9.0	8.0	
5	8.0	6.0	5.0	4.5	4.0	
2	3.0	2.5	2.0	1.9	1.8	
1	1.2	1.2	1.0	1.0	0.9	
0.5	0.5	0.5	0.5	0.5	0.5	



1. PERFORMANCE

1) Measuring range 0.1-1.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 0.5 minutes/1 pump stroke

3) Detectable limit 0.01%4) Shelf life 2 years 5) Operating temperature $0 \sim 40\%$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid: 5% RSD-high: 5%

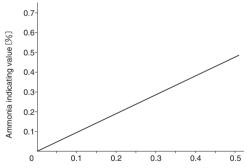
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $NH_3 + H_3PO_4 \rightarrow (NH_4)_2HPO_4$

4. CALIBRATION OF THE TUBE

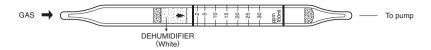
STANDARD GAS CYLINDER METHOD

Substance		Interference	ppm	Coexistence
Amines FIG.1		Similar stain is produced.		Higher readings are given.
Sulphur dioxide		The accuracy of readings is not affected.	200	Lower readings are given.



Monoethyl amine / Monomethyl amine [%]
FIG.1 Influence of Monoethyl amine / Monomethyl amine

ANILINE



1. PERFORMANCE

1) Measuring range : 2-30 ppm 1-15 ppm Number of pump strokes 1 (100ml) $2(200 \text{m} \ell)$ 2) Sampling time : 0.5 minutes/1 pump stroke 3) Detectable limit $0.05 \text{ ppm} (200 \text{m} \ell)$

4) Shelf life 3 years 5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with p-Dimethylamino-benzaldehyde, Azomethine is produced.

$$NH_2 + CHO - N(CH_3)_2 \longrightarrow N = CH - N(CH_3)_2 + H_2O$$

4. CALIBRATION OF THE TUBE

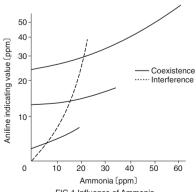
DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

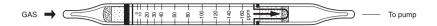
Substance	Interference	ppm	Coexistence	
Toluidine	Similar stain is produced.	Aniline conc.X1/3	Higher readings are given.	
Ammonia FIG.	The accuracy of readings is not affected.	The same conc. of Aniline	"	
Paraffin amines	"		"	
Aromatic amines	"		"	

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration. Actual concentration = $1/2 \times$ Reading value.



ARSINE



1. PERFORMANCE

1) Measuring range 5-160 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

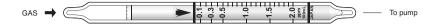
Mercury chloride (II) is reduced and Mercury is liberated. $AsH_3 + HgCI_2 + H_2O {\rightarrow} Hg + H_3AsO_3 + HCI$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substance	Interference	ppm	Coexistence
Hydrogen selenide	Similar stain is produced.	5	Higher readings are given.
Hydrogen sulphide	"	5	"
Phosphine	"	5	"

ARSINE



1. PERFORMANCE

1) Measuring range Number of pump strokes 2) Sampling time 3) Detectable limit $\begin{array}{c} 0.1-2.0 \text{ ppm} & 0.05-0.5 \text{ ppm} \\ 1 \left(100\text{m}\,\ell\right) & 2\left(200\text{m}\,\ell\right) \\ \vdots & 1 \text{ minute/1 pump stroke} \\ \vdots & 0.02 \text{ ppm} \left(200\text{m}\,\ell\right) \end{array}$

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : The tube scale is calibrated based on Phosphine at 1 pump stroke and the tube

has the same sensitivity for Arsine.

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercury chloride (II), Hydrogen chloride is produced and PH indicator is discoloured. $AsH_3 + HgCI_2 \rightarrow As(HgCI)_3 + HCI$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

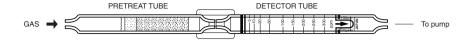
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Hydrogen selenide	Similar stain is produced.	Higher readings are given.
Mercaptans	"	"
Hydrogen sulphide	"	"
Hydrogen cyanide	Whole reagent is changed to Red.	"
Sulphur dioxide	"	Whole reagent is changed to Pale red, but Purplish red stain indicates Arsine concentration.

(NOTE)

When the concentration is below 0.5 ppm, 2 pump strokes can be used to determine the lower concentration with the following formula;

Actual concentration = $1/2 \times$ Reading value



1. PERFORMANCE

1) Measuring range 5-300 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 3 minutes/1 pump stroke

3) Detectable limit $3 \text{ ppm} (100 \text{m} \ell)$ 4) Shelf life 2 years5) Operating temperature $10 \sim 40 \, ^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Greenish brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. $C_6H_6 + I_2O_5 + H_2SO_4 \cdot nSO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Toluene	20	Yellowish brown stain is	150	Higher readings are given.
Xylene	60	"	300	"
Hexane	200	Whole reagent is discoloured to Dark grey.	800	Whole reagent is discoloured to Dark grey and higher readings are given.

(NOTE)

Aromatic hydrocarbons except Benzene is eliminated.

Tube	Corrected Concentration (ppm)					
Readings (ppm)	10 °C (32 °F)	15 ℃ (50 °F)	20 ℃ (68 °F)	30 °C (80 °F)	40 °C (104 °F)	
300	_	_	300	270	240	
250	_	390	250	225	200	
200	_	290	200	180	160	
150	300	225	150	135	120	
100	225	163	100	90	80	
50	135	93	50	45	40	
20	70	45	20	18	16	
10	45	28	10	9	8	
5	30	18	5	4.5	4	



1. PERFORMANCE

2) Sampling time : 4 minutes/2 pump strokes 3) Detectable limit : 0.2 ppm (400mℓ)

4) Shelf life 2 years5) Operating temperature 0.2 ppm 0.2 ppm 0.2 ppm

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : White→Greenish brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. $C_6H_6 + I_2O_5 + H_2SO_4 \cdot nSO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

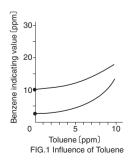
5. INTERFERENCE AND CROSS SENSITIVITY

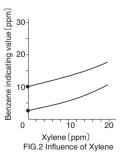
Substanc	е	Interference	ppm	Coexistence
Toluene	FIG.1	Yellowish brown stain is produced.		Higher readings are given.
Xylene	FIG.2	"		"
Carbon monoxide			50	Whole reagent is discoloured to Pale brown, the top of discoloured layer becomes unclear and higher readings are given.
Hexane			100	"

(NOTE)

In case of 1 or 4 pump strokes, following formula is available for the actual concentration.

 $Actual\ concentration = Temprature\ corrected\ value \times \frac{\textit{\angle}}{\text{Number of pump strokes}}$





Tube	Corrected Concentration (ppm)						
Readings (ppm)	0 °C (32 °F)	10 ℃ (50 °F)	20 ℃ (68 °F)	30 °C (86 °F)	40°C (104°F)		
50	38	44	50	_	- 1		
40	30	35	40	45	50		
30	23	26	30	34	38		
20	15	18	20	23	25		
10	8	9	10	11	13		
5	4	5	5	6	6		



1. PERFORMANCE

1) Measuring range : 1-75 ppm 0.2-15 ppm 0.1-7.5 ppm Number of pump strokes $1\,(100\mathrm{m}\,\ell)$ $5\,(500\mathrm{m}\,\ell)$ $10\,(1000\mathrm{m}\,\ell)$

2) Sampling time $\hbox{$:$ 10 minutes/5 pump strokes}$ 3) Detectable limit $\hbox{$:$ 0.02 ppm (1000m\ell)$ }$

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 5 pump strokes

8) Colour change : White→Greenish brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. $C_6H_6 + I_2O_5 + H_2SO_4 \cdot nSO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Toluene			Higher readings are given.
Xylene			"
Carbon monoxide		2.0	Whole reagent is changed to Pale brown, unclear stain is produced and higher readings are given.
Hexane		2.0	"

(NOTE)

In case of 1 or 10 pump strokes, following formula is available for the actual concentration.

Actual concentration = Tmperature corrected value $\times \frac{5}{\text{Number of pump strokes}}$

CORRECTION FOR AMBIENT CONDITIONS:

Temperature; In order to correct for temperature, multiply the tube reading with the following factors.

Temperature (℃)	0	1	2	3	4	5	6	7	8	9
Correction Factor	0.69	0.71	0.72	0.73	0.75	0.76	0.78	0.79	0.81	0.82
Temperature (°C)	10	11	12	13	14	15	16	17	18	19
Correction Factor	0.84	0.85	0.87	0.88	0.90	0.92	0.93	0.95	0.97	0.98
Temperature (°C)	20	21	22	23	24	25	26	27	28	29
Correction Factor	1.00	1.02	1.04	1.05	1.07	1.09	1.11	1.13	1.14	1.16
Temperature (°C)	30	31	32	33	34	35	36	37	38	39
Correction Factor	1.18	1.20	1.22	1.24	1.26	1.28	1.30	1.32	1.34	1.36
Temperature (°C)	40			•		•			•	
Correction Factor	1.39	1								



1. PERFORMANCE

1) Measuring range 1-80 ppm 0.2-1 ppm $1(100\text{m}\ell)$ 5 $(500\text{m}\ell)$ 2) Sampling time 2 minutes/1 pump stroke 3) Detectable limit 2 minutes/1 pm 5.00 pm 2 minutes/1 pm 6.00 pm 2 minutes/1 pm/1 pm/1

4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 ~ 40 °C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. $C_6H_6 + I_2O_5 + H_2SO_4 \cdot nSO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Toluene	Similar stain is produced.	1000	Higher readings are given.
Xylene	"	11	"
Ethyl benzene	"	"	"
Carbon monoxide		2	Whole reagent is changed to Pale brown, indiscernible stain is produced and higher readings are given.
Hexane		"	Very light brown is produced.

TEMPERATURE CORRECTION TABLE

Temperature; To correct for temperature, multiply the tube reading by the following factors.

Temperature (°C)	0	1	2	3	4	5	6	7	8	9
Correction Factor	0.66	0.66	0.66	0.66	0.66	0.66	0.66	0.66	0.66	0.66
Temperature (°C)	10	11	12	13	14	15	16	17	18	19
Correction Factor	0.66	0.70	0.73	0.76	0.79	0.83	0.86	0.89	0.93	0.96
Temperature (°C)	20	21	22	23	24	25	26	27	28	29
Correction Factor	1.00	1.04	1.08	1.11	1.15	1.19	1.23	1.27	1.31	1.35
Temperature (°C)	30	31	32	33	34	35	36	37	38	39
Correction Factor	1.40	1.44	1.48	1.53	1.57	1.62	1.66	1.71	1.71	1.71
Temperature (°C)	40									
Correction Factor	1.71									

BROMINE



1. PERFORMANCE

1) Measuring range : 1-20 ppm Number of pump strokes 1(100ml)

: 2 minutes/1 pump stroke 2) Sampling time

3) Detectable limit 0.1 ppm 4) Shelf life 2 years 0 ~ 40 °C 5) Operating temperature

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation

7) Reading Concentration chart method

8) Colour change : White→Orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

o-Toluidine is oxidized and Orthoguinone is produced.

$$Br_2 + H_2N - \bigcirc \longrightarrow NH_2 \longrightarrow BrNH_2 \longrightarrow NH_2Br$$

$$CH_3 \qquad CH_3 \qquad CH_3 \qquad CH_3$$

Coexistence

10

Chlorine (ppm)

FIG.2 Influence of Chlorine

····· Interference

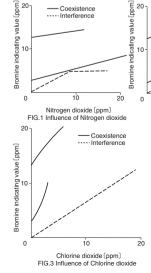
4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

10

Substance		Interference	ppm	Coexistence
Nitrogen dioxide	FIG.1	Similar stain is produced.		Higher readings are given.
Chlorine	FIG.2	"	1	"
Chlorine dioxide	FIG.3	"	5	"



Coexistence

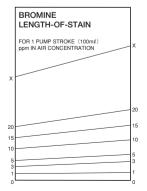
····· Interference

20

TEMPERATURE CORRECTION TABLE Chart Corrected Concentration (ppm) Reading (ppm) 20°C (68°F) 30 °C (86 °F) 10 °C (50 °F) 40°C (104°F) 20 18 15 10

20







1. PERFORMANCE

1) Measuring range 0.03-2.6%Number of pump strokes $1(100\text{m}\,\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 0.001%(10 ppm)4) Shelf life 3 years5) Operating temperature $0 \sim 40\%$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change

☐ Brownish orange → Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = CHCH = CH_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

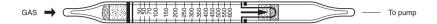
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene	Whole regent is discoloured to Brown.	3%	Higher readings are given.
Propane	"	0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	"

Tube	Corrcted Concentration (%)				
Readings (%)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
2.6	3.3	3.0	2.6	2.5	2.4
2.4	3.0	2.7	2.4	2.3	2.2
2.2	2.7	2.5	2.2	2.1	2.0
2.0	2.4	2.2	2.0	1.9	1.8
1.8	2.1	2.0	1.8	1.8	1.7
1.6	1.9	1.8	1.6	1.6	1.5
1.4	1.6	1.5	1.4	1.4	1.3
1.2	1.4	1.3	1.2	1.2	1.1
1.0	1.1	1.1	1.0	1.0	0.9
0.8	0.9	0.8	0.8	0.8	0.7
0.6	0.7	0.6	0.6	0.6	0.6
0.4	0.4	0.4	0.4	0.4	0.4



1. PERFORMANCE

1) Measuring range 30-600 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale yellow→ White

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Molybdate is reduced and molybdeum blue is produced. $CH_2 = CHCH = CH_2 + PdSO_4 + (NH_4)_2MoO_4 \rightarrow Mo_3O_8$

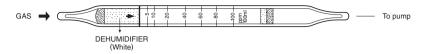
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Carbon monoxide	Whole layer is discoloured to Green or Blue.	
Hydrogen sulphide	Black stain is produced.	
Saturated Hydrocarbons (more than C ₃)	Whole layer is discoloured to Blue.	
Ethylene	Blue stain is produced.	
Propylene	"	
Butylene	"	
Benzene	Yellowish orange stain is produced.	
Ammonia	Similar stain is produced.	Higher readings are given.
Hydrogen cyanide	"	"

Tube		Corrected Concentration (ppm)			
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
600	910	740	600	550	510
550	820	670	550	510	470
500	740	600	500	460	430
450	650	530	450	420	390
400	570	470	400	370	350
350	490	400	350	330	310
300	410	340	300	280	270
250	330	280	250	240	220
200	250	220	200	190	180
150	180	160	150	140	140
100	110	110	100	100	90



1. PERFORMANCE

3) Detectable limit $\begin{array}{c} \text{3.5 ppm} (200\text{m}\,\ell) \\ \text{4) Shelf life} \\ \text{5) Operating temperature} \\ \end{array}$ $\begin{array}{c} \text{1.9 ppm} (200\text{m}\,\ell) \\ \text{1.1 pear} \\ \text{2.0 } \text$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_2 = CHCH = CH_2 + Cr^6 + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

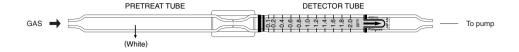
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interfe	erence	Coexistence
Esters	Similar stain is produc	ed.	Higher readings are given.
Alcohols	,	<i>Y</i>	"
Ketones	,	,	"
Aromatic hydrocarbons (more than C ₃)	Whole reagent is changed to Pale brown.		"

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $1/2 \times$ Reading value



1. PERFORMANCE

1) Measuring range Number of pump strokes 2) Sampling time 1.0.1-2.0 ppm $4(400\text{m}\ell)$ $1(100\text{m}\ell)$ $1(100\text{m}\ell)$ 12 minutes/4 pump strokes

3) Detectable limit $0.02 \text{ ppm } (400 \text{m} \ell)$ 4) Shelf life 3 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 4 pump strokes

7) Colour change : Pink→White

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Potassium permanganate is reduced.

 $CH_2 = CHCH = CH_2 + KMnO_4 \rightarrow White reaction products$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	Coexistence
Hydrogen sulphide	Similar stain is produced.	Higher readings are given.
Isobutylene	"	"
Ammonia	The accuracy of readings is not affected.	Higher readings with unclear maximum point of stained layer will be given.

n-BUTANE



1. PERFORMANCE

1) Measuring range 0.05-0.6 % Number of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 4 minutes/1 pump stroke 3) Detectable limit : 0.002 % (20 ppm)

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Orange→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 15 % RSD-mid.: 15 % RSD-high: 15 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

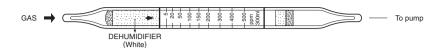
CH₃ $(CH_2)_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	Coexistence
Toluene	Similar stain is produced.	Higher readings are given.
Hexane	"	"
Trichloroethylene	"	"
Ethyl alcohol	The accuracy of readings is not affected.	

1-BUTANOL



1. PERFORMANCE

1) Measuring range : 5-100 ppm Number of pump strokes 3(300ml)

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit 2 ppm 4) Shelf life 2 years 5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

Necessary (See"TEMPERATURE CORRECTION TABLE") 6) Temperature compensation:

7) Reading Graduations printed on the tube are calibrated by Ethyl cellosolve at 3 pump strokes

and 1-Butanol concentration is determined by using a conversion chart.

8) Colour change : Yellow → Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

CH₃CH₂CH₂CH₂OH + Cr^{6+} + $H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Alcohols FIG.		Similar stain is produced.	Higher readings are given.
Toluene	200	Whole reagent is changed to Pale blue.	"
Hexane	1,000	The accuracy of readings is not affected.	The accuracy of readings is not affected.
Trichloroethylene	1,000	"	"
Ethyl acetate FIG.:	1,000	"	"

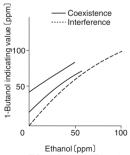
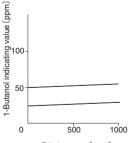
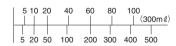


FIG.1 Influence of Ethanol



Ethyl acetate (ppm) FIG.2 Influence of Ethyl acetate

1-Butanol (ppm)



No.190U Tube reading (ppm)

Conversion	С	Corrected Concentration (ppm)				
Value (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)	
100	_	_	100	85	77	
80	_	_	80	70	63	
60	_	80	60	53	50	
40	75	50	40	35	33	
20	30	23	20	18	16	
10	13	- 11	10	9	8	
5	5	5	5	5	5	

2-BUTANOL



1. PERFORMANCE

1) Measuring range Number of pump strokes 2) Sampling time 10-300 ppm 4-120 ppm $2(200m\ell)$ $4(400m\ell)$ 3 minutes/2 pump strokes

3) Detectable limit $\begin{array}{c} : \ 3 \ ppm (200m \, \ell) \\ 4) \ Shelf \ life \\ 5) \ Operating \ temperature \\ \end{array} \begin{array}{c} : \ 2 \ years \\ : \ 10 \, {\sim} \, 40 \, {\,}^{\circ}\!{\rm C} \\ \end{array}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3CH_2CH(OH)CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

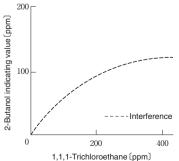
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence
Alcohols		Brown stain is produced.	Higher readings are given.
Ethers		"	"
Aliphatic hydrocarbons (more than C ₃)		Whole reagent is change to Pale brown.	"
Aromatic hydrocarbons		"	"
Esters		"	"
Ketones		"	"
Halogenated hydrocarbons	FIG.1	"	"

(NOTE)

In case of 4 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2/5 \times$ Temperature corrected value



1,1,1-Trichloroethane (ppm)
FIG.2 Influence of 1,1,1-Trichloroethane

TEMPERATURE CORRECTION TABLE

Tube		Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30°C (86°F)	35 ~ 40 °C (95 ~ 104 °F)	
300	480	360	300	270	250	240	
250	390	290	250	230	220	210	
200	300	230	200	190	190	180	
150	200	160	150	150	140	140	
100	110	100	100	100	100	100	
50	50	50	50	50	50	50	

tert-BUTANOL



1. PERFORMANCE

7) Colour change

1) Measuring range 20-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

tert-Butanol concentration is determined by using a conversion chart

at 1 pump stroke : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

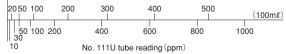
 $(CH_3)_3COH + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

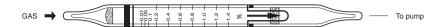
GAS CHROMATOGRAPHY

Substance	Interference	Coexistence	
Alcohols	Similar or brown stain is produced.	Higher readings are given.	
Esters	"	"	
Ketones	"	"	
Aromatic hydrocarbons	"	"	
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.	
Halogenated hydrocarbons	"	"	





BUTYL ACETATE



1. PERFORMANCE

1) Measuring range 0.01-1.0%Number of pump strokes $2(200m\ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit : 10 ppm 4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \, ^{\circ}$ C

6) Reading Graduations printed on the tube are calibrated by Methyl ethyl ketone at 2 pump strokes and Butyl acetate concentration is determined by using a conversion chart.

7) Colour change : Orange→Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Dichromate is reduced.

 $CH_{3}CO_{2}\ (CH_{2})_{3}CH_{3}+Cr^{6\,+}+H_{2}SO_{4} {\longrightarrow} Cr^{3\,+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

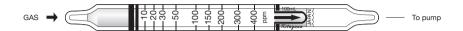
Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is clanged to Brown.
Propane		0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons		50	Higher readings are given.

Butyl acetate (%)



No.139SB Tube reading (%)

BUTYL ACETATE



1. PERFORMANCE

1) Measuring range : 10-400 ppm Number of pump strokes : $1(100m\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 10 ppm 4) Shelf life : 1 year 5) Operating temperature : $10 \sim 40 \, \text{°C}$

6) Temperature compensation
7) Reading
Colour change
Correction TABLE
Direct reading from the scale calibrated by 1 pump stroke
Pale yellow—Pale blue (The top of discoloured layer is to Brown.)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

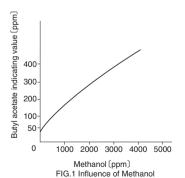
3. CHEMICAL REACTION

Dichromate is reduced. $CH_3CO_2C_4H_9 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	Coexistence
Esters	Similar stain is produced.	Higher readings are given.
Alcohols FIG.1	"	"
Ketones	"	"
Aromatic hydrocarbons	Whole reagent is changed to Pale brown.	"



TEMPERATURE CORRECTION TABLE

Tube		Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (80 °F)	35 ℃ (95 °F)	40 °C (95 °F)
400	-	-	400	235	180	145	120
300			300	185	145	120	100
200		345	200	140	110	95	80
150		255	150	115	95	80	70
100		145	100	80	70	60	50
50	90	65	50	40	35	30	30
30	45	35	30	25	25	20	20
20	25	25	20	20	15	15	15
10	15	15	10	10	5	5	5

BUTYL ACRYLATE



1. PERFORMANCE

1) Measuring range 2-60 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit : 0.5 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = CHCO_2 (CH_2)_3 CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

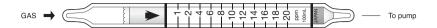
Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Dark brown.	"
Halogenated hydrocarbons	"	"
Esters	"	"
Aromatic hydrocarbons	"	"

(NOTE)

Butyl acrylate has the same sensitivity at 20 °C to Methyl acrylate.

Tube	С	Corrected Concentration (ppm)					
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 ℃ (86 °F)	40 °C (104 °F)		
60	105	80	60	45	32		
40	68	54	40	30	22		
20	32	26	20	15	11		
10	15	13	10	8	5		
5	8	7	5	4	3		

105SDC Di-n-BUTYL AMINE



1. PERFORMANCE

1) Measuring range : 2-20 ppm Number of pump strokes 1 (100m l)

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 4) Shelf life 3 years 5) Operating temperature : 15 ~ 25 °C

6) Reading Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke

and Di-n-Butyl amine concentration is determined by using a conversion

chart at 1 pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

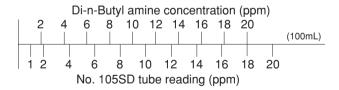
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $(C_4H_9)_2NH + H_3PO_4 \rightarrow (R_2NH_2)_3PO_4$

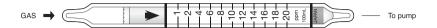
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



n-BUTYL AMINE



1. PERFORMANCE

1) Measuring range 1-20 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke and n-Butyl amine concentration is determined by using a conversion chart at

1 pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

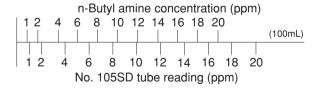
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. C₄H₉NH₂ + H₃PO₄ → (R₂NH₃)₃PO₄

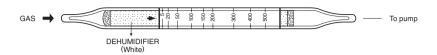
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



BUTYL CELLOSOLVE



1. PERFORMANCE

1) Measuring range : 10-1,000 ppm Number of pump strokes : $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 35 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading The tube is calibrated based on Ethyl cellosove. Butyl cellosolve concentration is determined by two times of the reading value after temperature correction.

8 Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $HOCH_2CH_2OC_4H_9 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

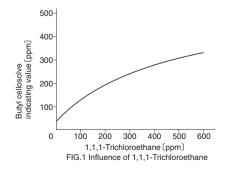
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Esters	"	"
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Brown.	"
Aromatic hydrocarbons	"	"
Ketones	"	"
Halogenated hydrocarbons FIG.1	"	"

(NOTE)

Multiply the corrected value with Ethyl cellosolve temperature correction table by 2.



Tube	Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	25℃ (77°F)	30 °C (80 °F)	35 ℃ (95 °F)
500	800	620	500	410	335	265
400	620	490	400	325	260	200
300	450	365	300	245	195	145
200	290	245	200	160	125	95
150	215	185	150	120	90	70
100	145	125	100	80	60	45
50	75	65	50	40	30	25
20	30	25	20	15	12	10
5	10	7	5	4	3	2

BUTYL ETHER



1. PERFORMANCE

1) Measuring range : 10-1,200 ppm Number of pump strokes : $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

Butyl ether concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $(CH_3CH_2CH_2CH_2)_2O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

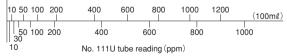
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

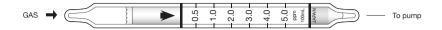
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar or brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	"

Butyl ether concentration (ppm)



TERT-BUTYL MERCAPTAN



1. PERFORMANCE

1) Measuring range Number of pump strokes Sampling time Speech Selection 1.10 ppm 0.5-5 ppm $1/2(50m\ell)$ $1(100m\ell)$ Substituting 1.10 ppm $1/2(50m\ell)$ $1(100m\ell)$ Stroke Speech Selection 1.10 ppm $1/2(50m\ell)$ $1/2(50m\ell)$ Speech Selection 1.10 ppm $1/2(50m\ell)$ $1/2(50m\ell)$

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : The tube scale is calibrated based on Methyl mercaptan at 1 pump stroke and

the tube has the same sensitivity for tert-Butyl mercaptan

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3.CHEMICAL REACTION

By reacting with Mercuric chloride, Hydrogen chloride is produced and PH indicator is discoloured. (CH₃)₃CSH + HgCl₂→(CH₃)₃CS (HgCl) + HCl

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

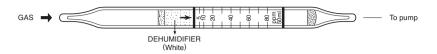
Substance	Interference	Coexistence
Arsine	Similar stain is produced.	Higher readings are given.
Hydrogen selenide	"	"
Phosphine	"	"
Hydrogen sulphide	"	"
Hydrogen cyanide	Whole reagent is changed to Red.	"
Sulphur dioxide		Whole reagent is changed to Pale red, but Pink stain indicates Mercaptans conc.

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times$ Reading value

tert-BUTYL MERCAPTAN



1. PERFORMANCE

1) Measuring range Solution Strokes Solution Strokes Solution Sol

3) Detectable limit $1 \text{ ppm} (100\text{m} \ell)$ 4) Shelf life 2 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : The tube scale is calibrated based on Ethyl mercaptan at 1/2 pump strokes

and the tube has the same sensitivity for tert-Butyl mercaptan

7) Colour change : Yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride, Hydrogen chloride is produced and PH indicator is discoloured. (CH₃)₃CSH + HgCl₂→(CH₃)₃S(HgCl)+ HCl

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Hydrogen sulphide	Similar stain is produced.	Higher readings are given.
Phosphine	"	"
Other mercaptans	"	"
Arsine	"	"
Hydrogen selenide	"	"
Hydrogen cyanide	"	"
Nitrogen dioxide	The accuracy of readings is not affected.	Lower readings are given.
Ammonia	"	"
Sulphur dioxide	Whole layer is discoloured to Pale red.	The accuracy of reading is not affected if the maximum end point of stained layer is discernable.

(NOTE)

In case of a 1 pump stroke, following formula is available for the actual concentration.

Actual concentration = $0.5 \times$ Reading value

BUTYL METHACRYLATE



1. PERFORMANCE

7) Colour change

1) Measuring range 20-1,000 ppmNumber of pump strokes $1(100\text{m}\,\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

Butyl methacrylate concentration is determined by using a conversion chart

at 1 pump stroke : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = C(CH_3)CO_2C_4H_9 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

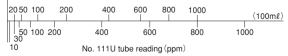
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar or brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	"

Butyl methacrylate concentration (ppm)



tert-BUTYL METHYL ETHER



1. PERFORMANCE

1) Measuring range 25-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

tert-Butyl methyl ether concentration is determined by using a conversion chart at 1 pump stroke

7) Colour change : Yellow → Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3OC(CH_3)_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

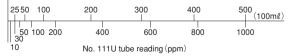
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar or brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	"

tert-Butyl methyl ether concentration (ppm)



BUTYRIC ACID



1. PERFORMANCE

1) Measuring range 3-60 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Butyric acid concentration is determined by using a conversion chart at

1 pump stroke.

7) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

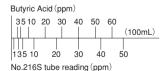
By reacting with alkali, PH indicator is discoloured. CH₃CH₂CH₂COOH + NaOH → C₃H₇COONa + H₂O

4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm Interference		Coexistence
Sulphur dioxide		Similar stain is produced.	Higher readings are given.
Nitrogen dioxide	300	"	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	"
Acetic acid		Similar stain is produced.	"





1. PERFORMANCE

1) Measuring range $\begin{array}{ccc} \text{Number of pump strokes} & \vdots & 0.03\text{-}0.7\,\% & 100\text{-}1\text{,}500 \text{ ppm} \\ & 1\,(100\text{m}\,\ell) & 3\,(300\text{m}\,\ell) \\ \text{2) Sampling time} & \vdots & 5 \text{ minutes/1 pump stroke} \\ \end{array}$

3) Detectable limit $\therefore 20 \text{ ppm} (300 \text{m} \ell)$ 4) Shelf life $\therefore 2 \text{ years}$ 5) Operating temperature $\therefore 0 \sim 40 \text{ °C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Concentration chart method 8) Colour change : Purplish blue → Pale pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid: 5% RSD-high: 5%

3. CHEMICAL REACTION

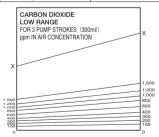
By reacting with alkali, PH indicator is discoloured. CO₂ + 2NaOH → Na₂CO₃ + H₂O

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

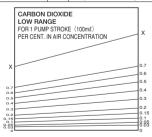
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference		Coexistence
Chlorine	The accuracy of readings not affected.		The accuracy of readings is not affected.
Hydrogen sulphide	"		"
Nitrogen dioxide	"		"
Sulphur dioxide	"		"
Hydrogen cyanide	Similar stain is produced.	120	Higher readings are given.



TEMPERATURE CORRECTION TABLE (3 pump strokes)

Chart		Corrected Concentration (ppm)					
Readings (ppm)	0°C (32°F)	10°C (50°F)	20°C (68°F)	30°C (86°F)	40°C (104°F)		
1,500	1,800	1,650	1,500	1,400	1,350		
1,400	1,700	1,550	1,400	1,300	1,250		
1,300	1,550	1,400	1,300	1,250	1,150		
1,200	1,450	1,300	1,200	1,150	1,100		
1,100	1,300	1,200	1,100	1,050	1,000		
1,000	1,200	1,100	1,000	950	900		
900	1,100	1,000	900	850	800		
800	950	900	800	750	700		
700	850	750	700	650	600		
600	700	650	600	550	500		
500	600	550	500	500	450		
400	500	450	400	400	350		
300	350	300	300	300	250		
200	250	200	200	200	200		



TEMPERATURE CORRECTION TABLE (1 pump stroke)

				,	
Chart		Correcte	d Concenti	ration (%)	
Readings	0,0	10 °C	20°C	30°C	40°C
(%)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)
0.7	0.8	0.75	0.7	0.65	0.6
0.6	0.7	0.65	0.6	0.55	0.5
0.5	0.6	0.55	0.5	0.45	0.4
0.4	0.45	0.4	0.4	0.4	0.35
0.3	0.35	0.3	0.3	0.3	0.25
0.2	0.25	0.2	0.2	0.2	0.2



1. PERFORMANCE

1) Measuring range Number of pump strokes $1/2(50m\ell) = 1(100m\ell)$ 2) Sampling time $1/2(50m\ell) = 1(100m\ell)$ 3) Detectable limit $1/2(50m\ell) = 1(100m\ell)$

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale blue → Pale pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. CO₂ + 2NaOH → Na₂CO₃ + H₂O

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

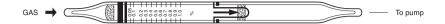
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference		Coexistence
Chlorine	The accuracy of readings not affected.		The accuracy of readings is not affected.
Hydrogen sulphide	"		"
Nitrogen dioxide	"		"
Sulphur dioxide	"		"
Hydrogen cyanide	Similar stain is produced.	120	Higher readings are given.

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times$ Reading value



1. PERFORMANCE

1) Measuring range 0.05-1.0%Number of pump strokes $1(100\text{m}\,\ell)$

2) Sampling time : 5 minutes/1 pump stroke 3) Detectable limit : 0.005 % (50 ppm)

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Purple blue → Pale pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. CO₂ + 2NaOH → Na₂CO₃ + H₂O

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference		Coexistence
Chlorine	The accuracy of readings not affected.		The accuracy of readings is not affected.
Hydrogen sulphide	"		"
Nitrogen dioxide	"		"
Sulphur dioxide	"		"
Hydrogen cyanide	Similar stain is produced.	120	Higher readings are given.

Tube		Corrected Concentration (%)					
Readings (%)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)		
1.0	1.16	1.09	1.00	0.94	0.89		
0.9	1.05	0.98	0.90	0.85	0.80		
0.8	0.93	0.87	0.80	0.75	0.70		
0.7	0.81	0.76	0.70	0.66	0.62		
0.6	0.70	0.65	0.60	0.57	0.53		
0.5	0.58	0.54	0.50	0.47	0.44		
0.4	0.46	0.43	0.40	0.38	0.35		
0.3	0.36	0.33	0.30	0.28	0.26		
0.2	0.24	0.22	0.20	0.19	0.17		
0.1	0.12	0.11	0.10	0.09	0.08		



1. PERFORMANCE

1) Measuring range 100-2000 ppm 200-4000 ppm Number of pump strokes $1(100\text{m}\ell)$ $1/2(50\text{m}\ell)$ 2) Sampling time 200-4000 ppm $1/2(50\text{m}\ell)$ $1/2(50\text{m}\ell)$

3) Detectable limit $: 5 \text{ ppm} (100 \text{m} \ell)$ 4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \text{ }^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump strokes

7) Colour change : Pink→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $CO_2 + 2KOH \rightarrow K_2CO_3 + H_2O$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen cyanide	Similar stain is produced.	1000	Higher readings are given.
Hydrogen chloride	"		The accuracy of readings is not affected.
Hydrogen sulphide	"	10	"
Nitrogen dioxide	"	5	"
Sulphur dioxide	"	100	"
Chlorine	Original colour is faded to White.	15	"
Ammonia	The accuracy of readings is not affected.		"

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times \text{Reading value}$



1. PERFORMANCE

1) Measuring range 0.02-0.7% 0.04-1.4%Number of pump strokes $1(100\text{m}\ell) 1/2(50\text{m}\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke3) Detectable limit $0.001\%(10 \text{ ppm})(100\text{m}\ell)$

4) Shelf life : 1 year 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pink→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured.

 $CO_2 + 2NaOH \rightarrow Na_2CO_3 + H_2O$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen cyanide	Similar stain is produced.	1000	Higher readings are given.
Hydrogen chloride	"	30	The accuracy of readings is not affected.
Hydrogen sulphide	"	10	"
Nitrogen dioxide	"	5	"
Sulphur dioxide	"	100	"
Chlorine	Original colour is faded to White.	15	"
Ammonia	The accuracy of readings is not affected.		"

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times \text{Reading value}$



1. PERFORMANCE

1) Measuring range 1-20%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 0.2%4) Shelf life 2 years 5) Operating temperature $0 \sim 40\%$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pink→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

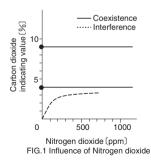
By reacting with alkali, PH indicator is discoloured. $CO_2 + 2KOH \rightarrow K_2CO_3 + H_2O$

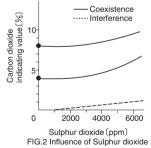
4. CALIBRATION OF THE TUBE

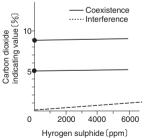
STANDARD GAS CYLINDER METHOD

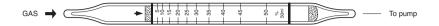
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		ppm	Interference	ppm Coexistence	
Nitrogen dioxide	FIG.1	50	White stain is produced.	50	Similar stain is produced, but if there is more than 3% of CO ₂ , the accuracy of readings is not affected.
Sulphur dioxide	FIG.2	3,000	Similar stain is produced.	3,000	Higher readings are given.
Hydrogen sulphide	FIG.3	4,000	"	3,000	"









1. PERFORMANCE

1) Measuring range : 5-50%Number of pump strokes $1/2(50m\ell)$

2) Sampling time : 1.5 minutes/1/2 pump stroke

3) Detectable limit $\therefore 1\%$ 4) Shelf life $\therefore 2$ years $\therefore 0 \sim 40\%$

6) Reading : Direct reading from the scale calibrated by 1/2 pump stroke

7) Colour change : White→Purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. CO₂ + N₂NNH₂→H₂NNHCOOH

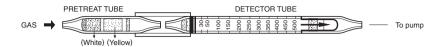
4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Sulphur dioxide	The accuracy of readings is not affected.	The accuracy of readings is not affected.
Ammonia	"	"
Hydrogen chloride	"	"
Chlorine	"	"
Hydrogen cyanide	"	"
Nitrogen dioxide	"	"
Hydrogen sulphide	"	"

CARBON DISULPHIDE



1. PERFORMANCE

1) Measuring range : 30-500 ppm Number of pump strokes $1(100m\ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit 10 ppm

4) Shelf life 2 years (Necessary to store in refrigerated conditions; $0 \sim 10 \,^{\circ}\text{C}$)

5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation Direct reading from the scale calibrated by 1 pump stroke 7) Reading

8) Colour change : Pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Sulphur dioxide is produced by an Oxidizer. By reacting between this Sulphur dioxide and alkali, PH indicator is discoloured.

 $CS_2 + CrO_3 + H_2SO_4 \rightarrow SO_2$ $SO_2 + 2NaOH \rightarrow Na_2SO_3 + H_2O$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Interference ppm	
Sulphur dioxide FIG.1	50	Similar stain is produced.		Higher readings are given.
Hydrogen sulphide FIG.2	400	"	400	"
Chlorine		White stain is produced.		"

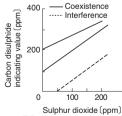
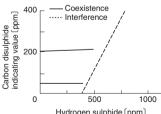


FIG.1 Influence of Sulphur dioxide



Hydrogen sulphide (ppm) FIG.2 Influence of Hydrogen sulphide

rube	Corrected Concentration (ppm)							
Readings (ppm)	0 °C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)			
500	600	550	500	490	470			
450	540	500	450	440	430			
400	480	440	400	390	370			
350	420 380 350		350	340	330			
300	360	330	300	290	280			
250	300	270	250	240	230			
200	240	220 200		190	180			
150	180	170	150	140	130			
100	120	110	100	95	90			
50	60	55	50	50	50			
30	35	30	30	30	30			

CARBON DISULPHIDE



1. PERFORMANCE

1) Measuring range 2-50 ppm 0.8-20 ppm Number of pump strokes $2(200\text{m}\ell)$ $4(400\text{m}\ell)$ 2) Sampling time 4 minutes/2 pump strokes

3) Detectable limit : $0.3 \text{ ppm} (400 \text{m} \ell)$

4) Shelf life : 3 years (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes.

8) Colour change : Pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Sulphur dioxide is produced by an Oxidizer. By reacting between this Sulphur dioxide and alkali, PH indicator is discoloured.

 $\begin{aligned} &CS_2 + CrO_3 + H_2SO_4 {\rightarrow} SO_2 \\ &SO_2 + 2NaOH {\rightarrow} Na_2SO_3 + H_2O \end{aligned}$

4. CALIBRATION OF THE TUBE

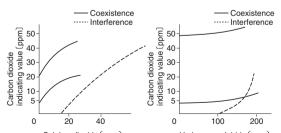
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference p		Coexistence
Sulphur dioxide FIG.1	15	Similar stain is produced.		Higher readings are given.
Hydrogen sulphide FIG.2	100	"	120	"
Chlorine		Pale pink stain is produced.		"

(NOTE)

In case of 4 pump strokes, following formula is available for the actual concentration. Actual concentration $= 2/5 \times$ Temperature corrected value



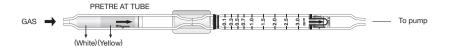
Sulphur dioxide (ppm) FIG.1 Influence of Sulphur dioxide FIG.2 I

Hydrogen sulphide (ppm) FIG.2 Influence of Hydrogen sulphide

TEMPERATURE CORRECTION TABLE

Tube		Corrected Concentration (ppm)							
Readings (ppm)	0 °C (32 °F)	10 ℃ (50 °F)	20 ℃ (68 °F)	30 °C (86 °F)	40 °C (104 °F)				
50	72	59	50	43	38				
45	65	54	45	45 39					
40	58	48	40	34	30				
35	51	42	35	35 30					
30	44	36	30	26	22				
25	37	30	25	25 21					
20	31	25	20	17	14				
15	24	19	15	13	10				
10	16	12	10	8	7				
5	9	7	5	4	3				
2	3	3	2	2	1				

CARBON DISULPHIDE



1. PERFORMANCE

1) Measuring range 0.1-3.0 ppm Number of pump strokes 0.2-6.4 ppm 0.1-3.0 ppm 0.1-

3) Detectable limit : $0.02 \text{ ppm} (400 \text{m} \ell)$

4) Shelf life : 1 year 5) Operating temperature : $0 \sim 40 \, ^{\circ}\mathrm{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 4 pump strokes

8) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Sulphur dioxide is produced by an Oxidizer. Sulphur dioxide reacts with alkari and PH indicator is discoloured.

 $CS_2 + CrO_3 + H_2SO_4 \rightarrow SO_2$ $SO_2 + 2NaOH \rightarrow Na_2SO_3 + H_2O$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Sulphur dioxide		2	Higher readings are given.
Hydrogen sulphide		40	"
Chlorine		0.2	White stain is produced from the inlet side of the tube, and higher readings are given.
Nitrogen dioxide	The accuracy of readings is not affected.		
Carbon tetrachloride	"		

TEMPERATURE CORRECTION TABLE (2 pump strokes)

Tube	Corrected Concentration (ppm)							
Readings (ppm)	0°C 10°C 20°C (32°F) (50°F) (68°F)		30 ℃ (86 °F)	40 °C (104 °F)				
3.0	_	_	6.40	8.75	5.20			
2.5	_	6.75	5.40	4.80	4.35			
2.0	6.50	5.35	4.30	3.80	3.40			
1.5	4.90	3.95	3.20	2.80	2.50			
1.0	3.20	2.50	2.05	1.80	1.60			
0.7	2.20	1.60	1.35	1.20	1.05			
0.5	1.50	1.10	0.92	0.80	0.70			
0.3	0.85	0.62	0.53	0.44	0.37			
0.1	0.28	0.24	0.20	0.16	0.13			

TEMPERATURE CORRECTION TABLE (4 pump strokes)

	Tube	(Corrected	Concentra	ation (ppm)
2.5 3.40 2.90 2.50 2.30 2.05 2.0 2.70 2.30 2.00 1.80 1.65 1.5 2.00 1.70 1.50 1.35 1.20 1.0 1.30 1.15 1.00 0.90 0.80 0.7 0.90 0.80 0.70 0.62 0.55 0.5 0.64 0.57 0.50 0.44 0.40 0.3 0.38 0.34 0.30 0.26 0.24 0.26 0.24			10 °C (50 °F)	20 °C (68 °F)		40 °C (104 °F)
2.0 2.70 2.30 2.00 1.80 1.65 1.5 2.00 1.70 1.50 1.35 1.20 1.0 1.30 1.15 1.00 0.90 0.80 0.70 0.90 0.80 0.70 0.62 0.55 0.5 0.64 0.57 0.50 0.44 0.40 0.3 0.38 0.34 0.30 0.26 0.24	3.0	_	3.45	3.00	2.75	2.45
1.5 2.00 1.70 1.50 1.35 1.20 1.0 1.30 1.15 1.00 0.90 0.80 0.7 0.90 0.80 0.70 0.62 0.55 0.5 0.64 0.57 0.50 0.44 0.40 0.3 0.38 0.34 0.30 0.26 0.24	2.5	3.40	2.90	2.50	2.30	2.05
1.0 1.30 1.15 1.00 0.90 0.80 0.7 0.90 0.80 0.70 0.62 0.55 0.5 0.64 0.57 0.50 0.44 0.40 0.3 0.38 0.34 0.30 0.26 0.24	2.0	2.70	2.30	2.00	1.80	1.65
0.7 0.90 0.80 0.70 0.62 0.55 0.5 0.64 0.57 0.50 0.44 0.40 0.3 0.38 0.34 0.30 0.26 0.24	1.5	2.00	1.70	1.50	1.35	1.20
0.5 0.64 0.57 0.50 0.44 0.40 0.3 0.38 0.34 0.30 0.26 0.24	1.0	1.30	1.15	1.00	0.90	0.80
0.3 0.38 0.34 0.30 0.26 0.24	0.7	0.90	0.80	0.70	0.62	0.55
	0.5	0.64	0.57	0.50	0.44	0.40
0.1 0.12 0.11 0.10 0.09 0.08	0.3	0.38	0.34	0.30	0.26	0.24
	0.1	0.12	0.11	0.10	0.09	0.08



1. PERFORMANCE

1) Measuring range $\begin{array}{ccc} \text{1.25-1,000 ppm} & \text{5-300 ppm} \\ \text{Number of pump strokes} & \text{1.} & \text{100m} \ell \text{)} & \text{3.} & \text{300m} \ell \text{)} \\ \text{2) Sampling time} & \text{3.3 minutes/1 pump stroke} \\ \end{array}$

3) Detectable limit $\begin{array}{c} : 1 \text{ ppm} (300 \text{m} \ell) \\ \text{4) Shelf life} \\ \text{5) Operating temperature} \end{array} \begin{array}{c} : 1 \text{ ppm} (300 \text{m} \ell) \\ : 3 \text{ years} \\ : 0 \sim 40 \, \text{°C} \\ \end{array}$

6) Reading : Concentration chart method 7) Colour change : Yellow→ Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Pottasium disulphate palladate (II) is reduced, and Palladium is liberated. $CO + K_2Pd(SO_3)_2 \rightarrow Pd$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Ethylene	5,000	Pale grey stain is produced.	5,000	The top of discoloured layer becomes unclear and higher readings are given.
Hydrogen	5,000	Greyish yellow stain is produced.	5,000	Whole layer is discoloured to Greyish yellow and the top of discoloured layer becomes unclear.
Acetylene	1.5	Dark green stain isproduced.	CO conc. × 1/5	Higher readings are given.
Sulphur dioxide	100	Original colour is faded.	"	"
Nitrogen dioxide	dioxide The accuracy of readings is not affected.		"	"

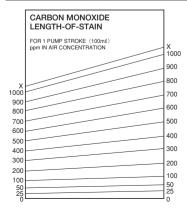
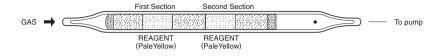


Chart		Corrected Concentration (ppm)						
Readings	0°C	10 °C	20 °C	30 °C	40 °C			
(ppm)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)			
1,000	870	930	1,000	1,030	1,060			
900	780	840	900	930	960			
800	690	750	800	830	850			
700	610	660	700	720	740			
600	520	560	600	620	640			
500	430	470	500	520	540			
400	350	370	400	410	430			
300	260	280	300	310	320			
200	180	190	200	210	220			
100	90	100	100	100	110			



1. PERFORMANCE

1) Measuring range : 10-1,000 ppm

This tube is calibrated based on the sampling time not related with number

of pump strokes.

2) Sampling time : 0.5-5.0 minutes 3) Detectable limit : 10 ppm (5.0 minutes)

4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 °C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE)

7) Reading : Colour intensity method using colour standard chart shown in below

8) Colour change : Pale yellow → Green to Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: RSD-mid.: RSD-high:

3. CHEMICAL REACTION

Molybdate is reduced and molybdeum blue is produced.

 $CO + PdSO_4 + (NH_4)_2MoO_4 \rightarrow Mo_3O_8$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence	
Nitrogen dioxide		1	Lower readings are given.	
Hydrogen sulphide	Blue or Black stain is produced.	1,000 Black stain is produced and hig readings are given.		
Ethylene	"	5	"	
Hydrogen (over 40 ℃)		10%	Whole reagent is discoloured to Blue and higher readings are given.	

6. NOTE

- In case of determining CO concentration, it is necessary to take a waiting time related to the sampling time and ambient temperature. After the waiting time, compare the second section of reagent with the colour standard chart.
- 2) In case of spontaneous combustion of coal mines, Ethylene and CO are produced. The first section is partially stained to deep blue from the top of the section according to Ethylene concentration. Whole areas of first and second sections are equally stained to Greenish blue according to CO concentration. The first section is available to detect less than 5 ppm of Ethylene.

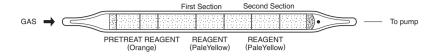
TEMPERATURE CORRECTION TABLE

Temperature	0 °C (32 °F)	5°C (41°F)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (86 °F)	35 °C (96 °F)	40 °C (104 °F)
Waiting time (minute) Sampling time up to 1 minutes Sampling time of 2 minutes	5 7	5 7	3 4	2	2	1 2	1	1	1
Chart Readings (ppm) 100 200 300 600 1.000	400 800 1,200 2,300 3,900	250 500 800 1,500 2,500	150 300 450 900 1.500	100 200 300 600 1.000	140 200 400	50 100 150 300 500	40 70 100 200 300	30 50 80 150 250	20 40 60 120 200

Color Standards for CARBON MONOXIDE



Feeding Time : 30 sec.
Waiting Time : 2 min.
Temperature : 15 ℃



1. PERFORMANCE

1) Measuring range : 10-1,000 ppm

This tube is calibrated based on the sampling time not related with number of

pump strokes.

2) Sampling time : 0.5-5 minutes
3) Detectable limit : 10 ppm (5.0 minutes)

4) Shelf life 2 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

 $\textbf{6) Temperature compensation} \hspace*{0.2cm} \vdots \hspace*{0.2cm} \textbf{Necessary} (\textbf{See "TEMPERATURE CORRECTION TABLE"}) \\$

7) Reading : Colour intensity method using colour standard chart shown in below

8) Colour change : Pale yellow→Green to Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: RSD-mid.: RSD-high:

3. CHEMICAL REACTION

Molybdate is reduced and molybdeum blue is produced.

 $CO + PdSO_4 + (NH_4)_2MoO_4 \rightarrow Mo_3O_8$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen sulphide	Black stain is produced.	1,000	Black stain is produced and higher readings are given.
Hydrogen (over 40 °C)	Similar stain is produced.	10%	Whole reagent is discoloured to Blue and higher readings are given.

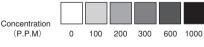
6. NOTE

- No.106C is specially designed to eliminate an influence of Nitrogen dioxide from readings, which is produced after blasting, by providing a pretreat reagent. The pretreat reagent is available to eliminate Nitrogen dioxide up to 300 ppm.
- 2) The function of first and second section of detecting reagents is the same with the No.106B tube.

TEMPERATURE CORRECTION TABLE

Temperature	0 °C	5℃	10 °C	15 °C	20 °C	25℃	30 °C	35℃	40 °C
	(32 °F)	(41°F)	(50 °F)	(59 °F)	(68 °F)	(77°F)	(86 °F)	(96°F)	(104 °F)
Waiting time (minute) Sampling time up to 1 minute Sampling time of 2 minute	5 7	5 7	3 4	2	2	1 2	1	1 1	1 1
Chart Readings (ppm)									
100	400	250	150	100	70	50	40	30	20
200	800	500	300	200	140	100	70	50	40
300	1,200	800	450	300	200	150	100	80	60
600	2,300	1,500	900	600	400	300	200	150	120
1,000	3,900	2,500	1,500	1,000	700	500	300	250	200

Color Standards for CARBON MONOXIDE



Feeding Time : 30 sec. Waiting Time : 2 min. Temperature : 15 ℃



1. PERFORMANCE

1) Measuring range 10-250 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 9 minutes/3 pump strokes

3) Detectable limit $1 \text{ ppm}(300\text{m}\ell)$ 4) Shelf life 2 years5) Operating temperature $0 \sim 40 \text{ C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Potassium disulphate palladate (II) is reduced, and Palladium is liberated. $CO + K_2Pd(SO_3)_2 \rightarrow Pd$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	%	Coexistence
Hydrogen sulphide	5	Dark red stain is produced.	CO conc. ×1/5	Higher readings are given.
Sulphur dioxide	120	Pale yellow stain is produced.	CO conc. ×1/5	"
Nitrogen dioxide		The accuracy of readings is not affected.	CO conc. ×1/5	"
Ethylene	0.5			"
Alcohols	1			"
Ketones	1			"
Olefinic hydrocarbons	1			"
Aromatic hydrocarbons	1			"
Halogenated hydrocarbons	1			"
Ammonia	1			"
Aliphatic hydrocarbons	1			Lower readings are given.

Tube	Corrected Concentration (ppm)								
Readings (ppm)	m) (32°F) (50°F)		20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)				
250	210	230	250	275	300				
200	170	185	200	220	240				
150	130	140	150	165	180				
100	80	90	100	110	120				
75	65	70	75	80	85				
50	40	45	50	55	60				
25	20	25	25	25	30				



1. PERFORMANCE

1) Measuring range : 20-1,000 ppm 5-50 ppm 40-2,000 ppm Number of pump strokes $1 (100 \text{m} \ell)$ $4 (400 \text{m} \ell)$ $1/2 (50 \text{m} \ell)$

2) Sampling time : 3 minutes/1 pump stroke

3) Detectable limit $2 \text{ ppm}(400\text{m}\ell)$ 4) Shelf life 3 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Potassium disulphate palladate (II) is reduced, and Palladium is liberated. $CO + K_2Pd(SO_3)_2 \rightarrow Pd$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Ethylene	5,000	Pale grey stain is produced.	5,000	The top of discoloured layer becomes unclear and higher readings are given.
Hydrogen	5,000	Greyish yellow stain is produced.	5,000	Whole layer is discoloured to Greyish yellow and the top of discoloured layer becomes unclear.
Acetylene	1.5	Dark green stain is produced.	CO conc. × 1/5	Higher readings are given.
Sulphur dioxide	100	Original colour is faded.	"	"
Nitrogen dioxide		The accuracy of readings is not affected.	"	"

(NOTE)

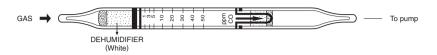
When the concentration is below 50 ppm, 4 pump strokes can be used to determine the lower concentration with the following formula;

Actual concentration = $1/4 \times$ Temperature corrected value

When the concentration is over 1,000 ppm, 1/2 pump strokes can be used to determine higher concentration with the following formula;

Actual concentration = $2 \times$ Temperature corrected value

Tube	Corrected Concentration (ppm)								
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)				
1,000	870	930	1,000	1,030	1,060				
900	780	840	900	930	960				
800	690	750	800	830	850				
700	610	680	700	720	720				
600	520	560	600	620	640				
500	430	470	500	520	540				
400	350	370	400	410	430				
300	260	280	300	310	320				
200	180	190	200	210	220				
100	00	100	100	100	110				



1. PERFORMANCE

1) Measuring range Number of pump strokes 1 (100m ℓ)
2) Sampling time 4 minutes 2) Detectable limit 0.5 ppm
4) Shelf life 1 year 5) Operating temperature 5 \sim 40 °C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange → Reddish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $CO + K_2Pd(SO_3)_2 \rightarrow Pd$

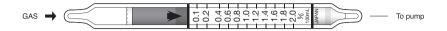
4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Sulphur dioxide	Similar stain is produced.	Higher readings are given.
Formic acid	"	"
Acetylene	Whole reagent is change to Pale red.	The maximum end point of the stain is indiscernible and higher readings are given.
Hydrogen sulphide	Dark reddish purple and red stain are produced.	Higher readings are given.
Hydrogen		"

Tube	Corrected Concentration (ppm)											
Readings (ppm)	5 °C (41 °F)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (86 °F)	35 ℃ (95 °F)	40 °C (104 °F)				
50	28	40	45	50	55	60	65	70				
40	23	32	36	40	44	48	52	56				
30	17	24	27	30	33	36	39	42				
20	12	16	18	20	22	24	26	28				
10	7	9	10	10	11	12	13	13				
5	4	5	5	5	5	6	6	6				



1. PERFORMANCE

1) Measuring range 0.1-2.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 3 minutes/1 pump stroke

3) Detectable limit 0.02%4) Shelf life 1 year5) Operating temperature $0 \sim 40\%$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

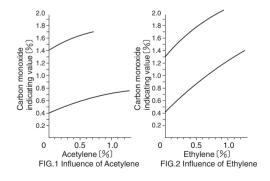
lodine pentoxide is reduced. $CO + I_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	%	Coexistence		
Acetylene	FIG.1	Similar stain is produced.	0.3	Higher reading are given.		
Ethylene	FIG.2	"	0.15	"		
Hexane		Speckled stain is produced.	0.1	The maximum point of the stained layer		
Isobutane		"	0.2	becomes unclear and higher readings are		
Propane		"	15	given.		



Tube		Corrected	Concent		
Readings	0°C	10 °C	20°C	30°C	40°C
(%)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)
2.0	1.24	1.52	2.00	2.98	_
1.8	1.14	1.38	1.80	2.64	_
1.6	1.04	1.25	1.60	2.28	2.86
1.4	0.93	1.11	1.40	1.90	2.50
1.2	0.82	0.97	1.20	1.62	2.12
1.0	0.71	0.82	1.00	1.33	1.75
0.8	0.60	0.68	0.80	1.02	1.38
0.6	0.47	0.53	0.60	0.73	0.98
0.4	0.33	0.37	0.40	0.48	0.58
0.2	0.17	0.18	0.20	0.22	0.25
0.1	0.08	0.09	0.10	0.11	0.12



1. PERFORMANCE

1) Measuring range Number of pump strokes 2) Sampling time 3) Detectable limit 4) Shelf life 5) Operating temperature : 30-500 ppm 1 $(100\text{m}\ell)$ 2 1 minute : 10 ppm 4) Shelf life : 1.5 years 5) Operating temperature $: 0 \sim 60 \text{ }^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Potassium disulphate palladate (II) is reduced, and Palladium is liberated. $CO + K_2Pd(SO_3)_2 \rightarrow Pd$

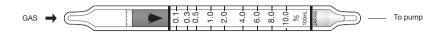
4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Sulphur dioxide	The accuracy of readings is not affected.	CO conc.X 1/2	Higher reading are given.
Ammonia	White stain is produced.	CO conc. X 100	"
Acetylene	Similar stain is produced.	CO conc. X 1/20	"
Hydrogen sulphide	Brown stain is produced.	CO conc. X 1/2	"

Tube	Corrected Concentration (ppm)								
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)	50 °C (122 °F)	60°C (140°F)		
500	400	450	500	550	600	650	700		
400	320	360	400	440	480	520	560		
300	240	270	300	330	360	380	400		
200	160	180	200	220	240	260	280		
100	80	90	100	110	120	130	140		
50	40	50	50	50	60	65	70		
30	30	30	30	30	30	35	40		



1. PERFORMANCE

1) Measuring range 0.2-20% 0.1-10%Number of pump strokes $1/2(50m\ell)$ $1(100m\ell)$ 2) Sampling time 0.2m 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10% 0.1-10%

3) Detectable limit 0.01% (14) Shelf life 3 years $0 \sim 40\%$

6) Temperature compensation : Necessary (0 \sim 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Dark brown

2. RELATIVE STANDARD DEVIATON

RSD-low: 15 % RSD-mid.: 15 % RSD-high: 10 %

3. CHEMICAL REACTION

lodine pentoxide is reduced CO + I₂O₅ + H₂SO₄→I₂

4. CALIBRATION OF THE TUBE

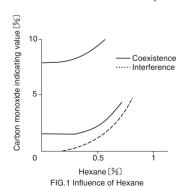
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence
Acetylene	Similar stain is produced.	2	Higher reading are given.
Ethylene	"	2	"
Isobutane	Speckled stain is produced.	0.5	"
Propane	"		The accuracy of reading is not affected.
Hexane FIG.1	Similar stain is produced.	0.4	The top of discoloured layer becomes unclear and higher readings are given.

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration. Actual concentration $= 2 \times$ Temperature corrected value



TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (%)					
Readings (%)	0°C (32°F)	5 ℃ (50 °F)	10 °C (68 °F)	20-40 °C (104 °F)		
	(32 F)	(50 F)	(68 F)	(104 F)		
10.0	4.0	5.6	8.0	10.0		
8.0	3.0	4.5	6.1	8.0		
6.0	2.3	3.2	4.3	6.0		
4.0	1.6	2.0	2.6	4.0		
2.0	1.0	1.2	1.5	2.0		
1.0	0.6	0.7	0.8	1.0		
0.5	0.4	0.5	0.5	0.5		

CARBON TETRACHLORIDE

1. PERFORMANCE

1) Measuring range 5-60 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : $0.5 \text{ ppm} (100 \text{m} \ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10 \,^{\circ}\text{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Carbon tetrachloride is decomposed and Phosgene is produced. By reacting with this Phosgene,

4-(p-Nitrobenzyl)-phridine and Benzylaniline, dyestuff is produced.

: White→Red

$$CCI_4 + I_2O_5 + F \cdot H_2SO_4 \longrightarrow COCI_2$$

$$\begin{array}{c} CH_2 \longrightarrow CH_2 \longrightarrow NH \\ NO_2 & N \\ \vdots \\ H \end{array} + \begin{array}{c} CH_2 \longrightarrow NH \\ \\ COCI_2 \longrightarrow DYESTUFF \end{array}$$

4. CALIBRATION OF THE TUBE

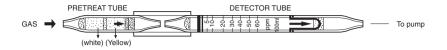
DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Phosgene	Similar stain is produced.	2	Higher reading are given.
Chlorine		12	"
Trichloroethylene		2	"

Tube	Corrected Concentration (ppm)				
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 ℃ (86*F)	40 ℃ (104 °F)
60	108	75	60	50	44
50	90	62.5	50	41	36.5
40	72	50	40	33	29
30	54	37.5	30	25	22
20	36	25	20	16.5	14.5
10	18	12.5	10	8	7
5	9	6	5	4	3.5

CARBONYL SULPHIDE



1. PERFORMANCE

1) Measuring range 5-60 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit ∴ 2 ppm 4) Shelf life ∴ 3 years 5) Operating temperature ∴ 0 ~ 40 °C

6) Temperature compensation : Necessary ($0 \sim 20^{\circ}$ C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By discomposing with an Oxidizer, Sulphur dioxide is produced. Sulphur dioxide reacts with Alkali and PH indicator is discoloured.

 $\begin{aligned} &COS + CrO_3 + H_2SO_4 \rightarrow SO_2 \\ &SO_2 + 2NaOH \rightarrow Na_2SO_3 + H_2O \end{aligned}$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence
Sulphur dioxide			Higher readings are given.
Hydrogen sulphide			"
Carbon disulphide			"
n-Butane		0.1	Lower readings are given.

Tube	Correct	Corrected Concentration (ppm)			
Readings (ppm)	0 °C (32 °F)	10 ℃ (50 °F)	20-40 °C (68-104 °F)		
60	85	70	60		
50	75	60	50		
40	60	45	40		
30	45	35	30		
20	30	25	20		
10	15	12	10		
- 5	7	6	5		

CHLORINE



1. PERFORMANCE

1) Measuring range : 1-40 ppm Number of pump strokes : $1(100 \text{m}\ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 0.3 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Yellowish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

0-Toluidine is oxidized and Orthoquinone is produced. $CI_2 + H_2N - \bigcirc - \bigcirc - NH_2 \rightarrow CINH_2 = \bigcirc = \bigcirc = NH_2CI$ $CH_3 \qquad CH_3 \qquad CH_3 \qquad CH_3$

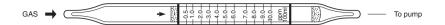
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm Interference		ppm	Coexistence
Bromine	0.1	Similar stain is produced.	1	Higher readings are given.
Chlorine dioxide	0.3	"	1	"
Nitrogen dioxide		Pale yellow stain is produced.	Chlorine conc. X 1/2	The maximum end point of the stain is indiscernible and higher readings are given.

CHLORINE



1. PERFORMANCE

2) Sampling time ∴ 1 minute/1 pump stroke 3) Detectable limit ∴ 0.06 ppm (500mℓ)

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Pale orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

0-Toluidine is oxidized and Orthoquinone is produced.

$$H_2N - \bigcirc - \bigcirc - NH_2 + CI_2 \longrightarrow CINH_2 = \bigcirc = \bigcirc = NH_2CI_3$$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Bromine	Pale yellow stain is produced.	1	Higher readings are given.
Chlorine dioxide	"	1	"
Nitrogen dioxide	"	Chlorine conc. X 1/5	"
Nitrogen trichloride	"	5	"

(NOTE)

When the concentration is below 2 ppm, 4 or 5 pump strokes can be used to determine the lower concentration with the following formula;

Actual concentration = Reading value $\times \frac{1}{\text{Number of pump strokes}}$

CHLORINE



1. PERFORMANCE

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct readingfrom the scale calibrated by 1 pump stroke

7) Colour change : White→Pale purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with 3, 3'-Dimethylnaphthidine, Nitroso-compound is produced. CI₂ + 3, 3'-Dimethylnaphthidine → Nitroso-compound

4. CALIBRATION OF THE TUBE

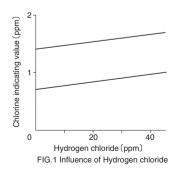
PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Hydrogen chloride	FIG.1	The accuracy of readings is not affected.	Chlorine conc. X 20	Higher readings are given.
Nitrogen dioxide	FIG.2	Similar stain is produced.		"

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration. Actual concentration = $1/2 \times \text{Reading value}$



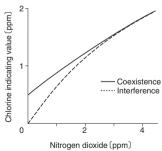


FIG.2 Influence of Nitrogen dioxide

CHLORINE DIOXIDE



1. PERFORMANCE

1) Measuring range : 1-20 ppm Number of pump strokes $1(100m\ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 0.3 ppm 4) Shelf life 2 years 5) Operating temperature : 0 ~ 40 ℃

6) Temperature compensation: Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading Concentration chart method 8) Colour change : White→Reddish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

0-Toluidine is oxidized and Orthoguinone is produced.

4. CALIBRATION OF THE TUBE

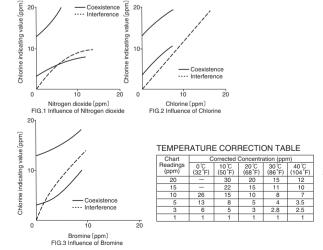
Coexistence

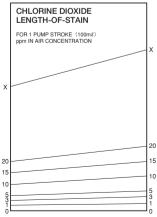
ABSORPTIOMETRIC METHOD

20

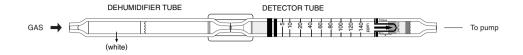
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Nitrogen dioxide	FIG.1	Similar stain is produced.	1	Higher readings are given.
Chlorine	FIG.2	"	1	"
Bromine	FIG.3	"	1	"





CHLOROBENZENE



1. PERFORMANCE

1) Measuring range Number of pump strokes Sampling time Specification 1.5 - 140 ppm Specification $1.00 \text{m} \ell$ Specification $1.00 \text{m$

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Pale brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 20 % RSD-mid.: 10 % RSD-high: 10 %

3. CHEMICAL REACTION

lodine pentoxide is reduced. $C_6H_5CI + I_2O_5 + H_2SO_4 \cdot nSO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Ethyl benzene	Similar stain is produced.		Higher readings are given.
Toluene	"		"
Xylene	"		"
Benzene	"		"
Carbon monoxide	Whole reagent is changed to the similar stain's colour.	50	"
Hexane	"	100	"

(NOTE)

When the concentration is below 5 ppm, 5 pump strokes can be used to determine the lower concentration with the following formula;

Actual concentration = $1/5 \times$ Reading value

CHLOROFORM



1. PERFORMANCE

1) Measuring range : 70-500 ppm 35-250 ppm 23-167 ppm Number of pump strokes $2(200 \text{m}\ell)$ 3(300m ℓ) 4(400m ℓ)

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit : 20 ppm $(400 \text{m} \ell)$

4) Shelf life : 2 years (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $10 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : White→Yellowish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 20% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

Chlorine is produced by an Oxidizer. By reacting between this Chlorine and o-Toluidine, yellow holoquinone is produced.

 $CHCl_3 + I_2O_5 + CrO_3 + H_2SO_4 \cdot nSO_3 \rightarrow Cl_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Halogens		Similar stain is produced.		Higher readings are given.
Halogenated hydrocarbons		"		"
n-Hexane	FIG.2	The accuracy of readings is not affected.	200	Lower readings are given.

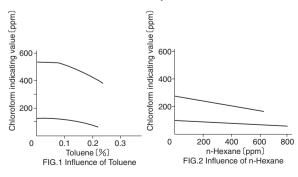
(NOTE)

In case of 3 pump strokes, following formula is available for the actual concentration.

Actual concentration = $1/2 \times$ Temperature corrected value

In case of 4 pump strokes, following formula is available for the actual concentration.

Actual concentration = $1/3 \times$ Temperature corrected value



Tube	Corrected Concentration (ppm)				
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	30 ℃ (95 °F)	40 °C (95 °F)
500	_		500	335	260
400	_	520	400	280	215
300	590	390	300	215	165
200	380	260	200	145	115
100	180	130	100	80	65
70	120	85	70	60	50

CHLOROPICRIN



1. PERFORMANCE

3) Detectable limit : $0.02 \text{ ppm}(200\text{m}\ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10 \, ^{\circ}\mathrm{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $CI_3CNO_2 + CrO_5 + H_2SO_4 \rightarrow COCI_2$

COCI₂ + Nitro-benzyl pyridine → DYESTUFF

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

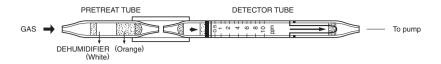
Substance	Interference	Coexistence		
Carbon tetrachloride	Similar stain is produced.	Higher readings are given.		
Phosgene	"	"		
Aliphatic hydrocarbons	The accuracy of readings is not affected.	The accuracy of readings is not affected.		

(NOTE)

When the concentration is below the scale ranges, 2 pump strokes can be used to determine lower concentration with the following formula;

Actual concentration = $1/2 \times$ Reading value

CHLOROPRENE



1. PERFORMANCE

1) Measuring range Number of pump strokes 1-20 ppm 0.5-10 ppm $1 (100m\ell) 2 (200m\ell)$ 2) Sampling time 3) Detectable limit : $0.01 \text{ ppm} (200m\ell)$

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (0-20 °C) (See"TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Greenish yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured.

 $CH_2 = CCICH = CH_2 + CrO_3 + H_2SO_4 \cdot nSO_3 \rightarrow HCI$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference		Coexistence	
Acetylene The accuracy of readings is not affected.		4%	Lower readings are given.	
Ethylene	"	200	"	
Vinyl chloride Similar stain is produced.			Higher readings are given.	

(NOTE)

In case of 1 pump stroke, following formula is available for the actual concentration.

Actual concentration = $2 \times$ Temperature corrected value

Tube	Corrected Concentration (ppm)							
Readings (ppm)	0 °C (32 °F)	5℃ (41°F)	10 °C (50 °F)	15 °C (59 °F)	20 °C ~ 40 °C (68 °F ~ 104 °F)			
10.0	80.0	23.0	16.5	12.5	10.0			
8.0	64.0	18.5	13.0	10.5	8.0			
6.0	48.0	14.0	10.0	7.5	6.0			
4.0	15.0	8.8	6.5	5.0	4.0			
2.0	8.0	4.0	3.0	2.5	2.0			
1.0	3.0	1.7	1.6	1.5	1.0			
0.5	0.5	0.5	0.5	0.5	0.5			

CRESOL



1. PERFORMANCE

1) Measuring range 0.5-25.0 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit 0.3 ppm4) Shelf life 0.3 ppm5) Operating temperature 0.3 ppm0.3 ppm

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : The tube scale is calibrated based on Phenol at 2 pump strokes and the tube has

the same sensitivity for Cresol.
8) Colour change : Pale yellow→Pale brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

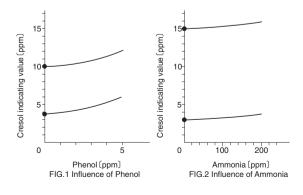
Cresol is oxidized and the polymer is produced. $CH_3C_6H_4OH + Ce^{4} \rightarrow CH_3C_6H_4O \rightarrow (CH_3C_6H_4O)_n$

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

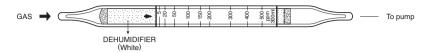
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference pp		Coexistence	
Phenols FIG.1 Ammonia FIG.2		Similar stain is produced.	2.5	Higher readings are given.	
		White stain is produced.	ed. 200 Discolouration of gas inlet faded and higher reading are		
Aliphatic amines Aromatic amines		"	50	"	
		Blue stain is produced.	50	Two layers discolouration of Pale brown and blue are produced and higher reading are given.	



Tube		Corrected Concentration (ppm)				
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 ℃ (68 °F)	30 °C (95 °F)	40 ℃ (104 °F)	
25.0	31.2	27.8	25.0	21.8	18.8	
20.0	24.5	22.3	20.0	17.5	15.0	
15.0	18.4	16.7	15.0	13.1	11.3	
10.0	12.3	11.1	10.0	8.8	7.5	
5.0	6.1	5.6	5.0	4.4	3.8	
3.0	3.7	3.3	3.0	2.6	2.3	
1.0	1.2	1.1	1.0	0.9	0.8	
0.5	0.5	0.5	0.5	0.5	0.5	

CROTONALDEHYDE



1. PERFORMANCE

1) Measuring range 2-40 ppmNumber of pump strokes $3(300 \text{m}\ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl cellosolve at 3 pump strokes and

Crotonaldehyde concentration is determined by using a conversion chart

at 3 pump strokes

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3CH = CHCHO + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

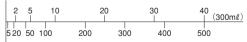
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Brown.	"
Aromatic hydrocarbons	"	"
Esters	"	"
Ketones	"	"
Halogenated hydrocarbons	"	"

Crotonaldehyde concentration (ppm)



No. 190U tube reading (ppm)

CUMENE



1. PERFORMANCE

1) Measuring range 20-140 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

Cumene concentration is determined by using a conversion chart

at 1 pump stroke
7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $C_6H_5CH(CH_3)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

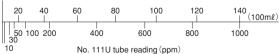
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Alcohols	Similar or brown stain is produced.	Higher readings are given.		
Esters	"	"		
Ketones	"	"		
Aromatic hydrocarbons	"	"		
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.		
Halogenated hydrocarbons	"	"		





CYCLOHEXANE



1. PERFORMANCE

1) Measuring range 0.01-0.6 % Number of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke
3) Detectable limit : 0.005 % (50 ppm)

4) Shelf life : 3 years5) Operating temperature $: 10 \sim 40 \, \text{°C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange→Dark green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $C_6H_{12} + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

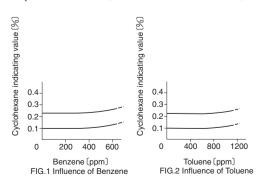
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

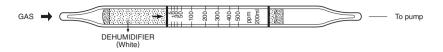
Substance		Interference	ppm	Coexistence
Benzene FIG.1		Black stain is produced.	400	Higher readings are given.
Toluene FIG.2		// 800		"
Xylene		"	2,000	"
Acetylene		llene Whole layer is discoloured to Dark green.		"
Ethylene		Similar stain is produced.		"
Aliphatic hydrocarbons (more than C ₃)		"		"

^{*} In presence of Alcohols, Ketones or Esters less than 6 %, the accuracy of readings is not affected.



Tube	Corrected Concentration (%)					
Readings	10 °C	15 °C	20°C	30 °C	40°C	
(%)	(50°F)	(59°F)	(68°F)	(86°F)	(104°F)	
0.6	0.73	0.67	0.60	0.53	0.49	
0.5	0.62	0.54	0.50	0.45	0.41	
0.4	0.48	0.43	0.40	0.36	0.34	
0.3	0.34	0.32	0.30	0.28	0.26	
0.2	0.22	0.21	0.20	0.19	0.18	
0.1	0.10	0.10	0.10	0.10	0.10	

CYCLOHEXANOL



1. PERFORMANCE

1) Measuring range 5-500 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

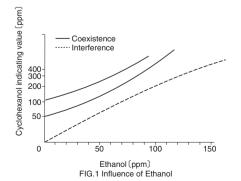
Chromium oxide is reduced. $C_6H_{11}OH + C_7e^6 + H_2SO_4 \rightarrow C_7e^3 + C_7$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

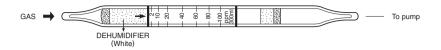
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Aliphatic hydrocarbons	Whole reagent is changed to Pale	The accuracy of reading is not	
(more than C ₃)	brown.	affected.	
Aromatic hydrocarbons	"	"	
Halogenated hydrocarbons	"	"	
Alcohols FIG.1	Similar stain is produced.	Higher readings are given.	
Esters	Pale ringed stain is produced near the bottom of the reagent.	The accuracy of reading is not affected.	



Lube	Corrected Concentration (ppm)							
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (80 °F)	35 ℃ (95 °F)	40 °C (95 °F)	
500			500	380	280	210	160	
400		520	400	300	230	170	130	
300		400	300	220	170	130	100	
200		270	200	140	110	90	70	
100	230	140	100	90	70	50	40	
50	90	70	50	40	35	20	15	
30	50	40	30	25	18	14	10	
10	18	13	10	8	6	4	2	
5	9	7	5	4	3	2	1	

CYCLOHEXANONE



1. PERFORMANCE

1) Measuring range 2-100 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

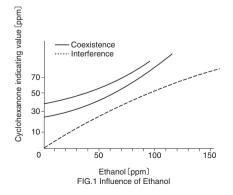
Chromium oxide is reduced. $C_6H_{10}O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to pale brown.	The accuracy of readings is not affected if the maximum end poir of the pale blue stain is discernable	
Aromatic hydrocarbons	"	"	
Halogenated hydrocarbons	"	"	
Alcohols FIG.1	Similar stain is produced.	Higher readings are given.	
Esters	Pale brown stain is produced from the zero end of the detecting reagent (inlet side of the tube).	The accuracy of readings is not affected.	



Tube	Corrected Concentration (ppm)					
Readings (ppm)	0°C (32°F)	10 ℃ (50 °F)	20℃ (68°F)	30 °C (86 °F)	40 ℃ (104 °F)	
100	-	-	100	75	60	
80	-	120	80	62	50	
60	110	84	60	46	37	
40	70	52	40	30	25	
20	30	26	20	16	13	
10	18	14	10	8	7	
2	4	3	2	2	1	

CYCLOHEXENE



1. PERFORMANCE

1) Measuring range 20-300 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

Cyclohexene concentration is determined by using a conversion chart

at 1 pump stroke
7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

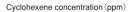
3. CHEMICAL REACTION

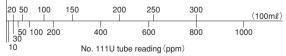
Chromium oxide is reduced. $C_6H_{10} + C_7r^{6+} + H_2SO_4 \rightarrow C_7r^{3+}$

4. CALIBRATION OF THE TUBE

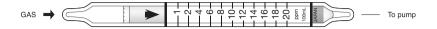
GAS CHROMATOGRAPHY

Substance	Interference	Coexistence		
Alcohols	Similar or brown stain is produced.	Higher readings are given.		
Esters	"	"		
Ketones	"	"		
Aromatic hydrocarbons	"	"		
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.		
Halogenated hydrocarbons	"	"		





CYCLOHEXYL AMINE



1. PERFORMANCE

1) Measuring range 1-20 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $C_6H_{11}NH_2+H_3PO_4\rightarrow (R_2NH_3)_3PO_4$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.

DECAHYDRONAPHTHALENE



1. PERFORMANCE

1) Measuring range 20-200 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

Decahydronaphthalene concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $C_{10}H_{18} + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

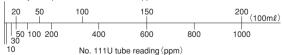
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar or brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	//

Decahydronaphthalene concentration (ppm)



n-DECANE



1. PERFORMANCE

1) Measuring range : 5-90 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke

and n-Decane concentration is determined by using a conversion chart

at 1 pump stroke
7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

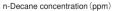
Chromium oxide is reduced.

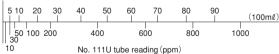
CH₃ (CH₂)₈CH₃ + Cr^{6+} + $H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

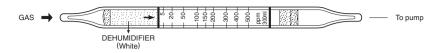
GAS CHROMATOGRAPHY

Substance	Interference	Coexistence
Alcohols	Similar or brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	"





DIACETONE ALCOHOL



1. PERFORMANCE

1) Measuring range 10-250 ppmNumber of pump strokes $3(300\text{m}\ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 40 °C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading Graduations printed on the tube are calibrated by Ethyl cellosolve at 3 pump strokes and Diacetone alcohol concentration is determined by using a conversion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

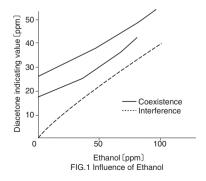
 $(CH_3)_2C(OH)CH_2COCH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Alcohols FIG.1	Similar stain is produced.	Higher readings are given.	
Halogenated hydrocarbons	"		
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	It the top of Pale blue stain is clear,the accuracy of readings is not affected.	
Aromatic hydrocarbons	"	"	
Esters	"	"	



TEMPERATURE CORRECTION TABLE

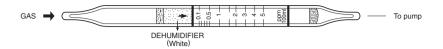
Conver.		Corrected Concentration (ppm)					
value (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (86 °F)	35 ℃ (95 °F)	40 °C (104 °F)
250	_	380	250	170	130	90	70
200	440	300	200	140	100	80	60
150	330	210	150	110	80	60	50
100	200	130	100	80	60	40	30
50	80	60	50	40	30	20	16
30	50	40	30	23	18	12	8
10	16	14	10	8	6	4	3

Diacetone alcohol (ppm)



No.190U Tube reading (ppm)

DIBORANE



1. PERFORMANCE

1) Measuring range : 0.1-5 ppm 0.05-2.5 ppm 0.02-1.0 ppm Number of pump strokes $1 \, (100 \text{m} \, \ell)$ $2 \, (200 \text{m} \, \ell)$ $5 \, (500 \text{m} \, \ell)$

2) Sampling time ∴ 1 minute/1 pump stroke 3) Detectable limit ∴ 0.01 ppm (500m ℓ)

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale yellow→Reddish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Mercuric cholride, Hydrogen chloride is liberated and PH indicator is discoloured. B2H6 + HgCl2→HCI

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

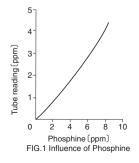
Substance		Interference	Coexistence
Arsine	FIG.2		Higher readings are given.
Phosphine	FIG.1		"
Hydrogen selenide			"
Monosilane			(Boundery is unclear.)
Disilane			" (")
Monogermane			Not affected.

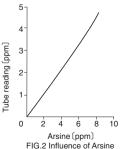
(NOTE)

In case of 2 and 5 pump strokes, the following formula is available for the actual concentration.

2 pump strokes : Actual concentration = Temperature corrected value \div 2

5 pump strokes: Actual concentration = Temperature corrected value ÷ 5





TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)					
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)	
5	_	9	5	3.5	2.5	
4	_	7	4	3	2	
3	_	5	3	2.5	1.5	
2	8	3	2	1.5	1.3	
1	1.5	1	1	1	0.8	
0.5	0.5	0.5	0.5	0.5	0.5	
0.1	0.1	0.1	0.1	0.1	0.1	

0-DICHLOROBENZENE



1. PERFORMANCE

1) Measuring range 5-100 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→ Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

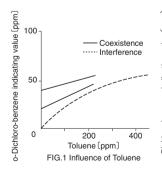
3. CHEMICAL REACTION

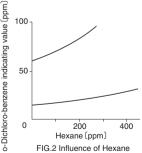
Iodine pentoxide is reduced. $C_6H_4CI_2 + I_2O_5 + H_2SO_4 \cdot nSO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

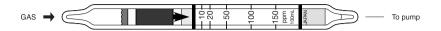
GAS CHROMATOGRAPHY

Substance		Interference	Coexistence	
Alcohols		Similar stains are produced.	Higher readings are given.	
Aliphatic hydrocarbons (more than C ₃) FIG.3		Whole reagent is changed to Dark brown.	"	
Halogenated hydracarbon	ns	"	"	
Ethyl alcohol		"	"	
Aromatic hydrocarbons	FIG.1,2	"	"	





P-DICHLOROBENZENE



1. PERFORMANCE

1) Measuring range 10-150 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 6 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Blueish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 15 % RSD-mid.: 10 % RSD-high: 10 %

3. CHEMICAL REACTION

By reacting with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. C₆H₄CI₂ + PbO₂ + H₂SO₄ → HCI

4. CALIBRATION OF THE TUBE

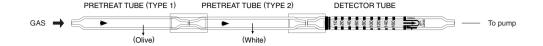
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Benzene	The accuracy of reading is not affected.	The accuracy of reading is not affected.
Toluene	"	"
Xylene	"	
Trichloroethylene	Similar stain is produced.	Higher readings are given.
Tetrachloroethylene	"	"
Chlorobenzene	"	"
o-Dichlorobenzene	"	"

Tube		Corrected Concentration (ppm)							
Readings (ppm)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	25℃ (77°F)	30 °C (86 °F)	35 ℃ (95 °F)	40 °C (104 °F)		
150		_	150	108	66	48	30		
100		_	100	72	46	37	26		
50	110	80	50	28	26	21	16		
20	50	27	20	14	10	8	6		
10	22	15	10	8	6	5	4		

1,1-DICHLOROETHANE



1. PERFORMANCE

1) Measuring range 10-160 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 3 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature $0 \sim 40 \,^{\circ}$ C $20-80 \,^{\circ}$ R.H.

7) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 8) Reading : Direct reading from the scale calibrated by 1 pump stroke

9) Colour change : White→Purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 10%

3. CHEMICAL REACTION

Chlorine is produced by decomposing with an Oxidizer.

By reacting between this Chlorine and 3, 3'-Dimethylnaphthidine, Nitroso-compound is produced.

 $CH_3CHCI_2 + CrO_3 + H_2SO_4 \cdot nSO_3 \rightarrow CI_2$

CI₂ + 3, 3'-Dimethylnaphthidine → Nitroso-compound

4. CALIBRATION OF THE TUBE

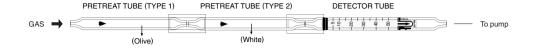
DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Nitrogen oxides	Similar stain is produced.		Higher readings are given.
Halogens	"		"
Halogenated hydrocarbons	"		"
Alcohols	The accuracy of readings is not affected.	400	Lower readings are given.
Hexane	"	20	"
Toluene	"	20	"

Tube	ation (ppr	n)			
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
160	230	195	160	118	75
140	200	170	140	105	70
120	170	145	120	90	60
100	140	120	100	75	50
80	110	95	80	60	40
60	80	70	60	48	35
40	50	45	40	33	25
20	25	23	20	18	15
10	10	10	10	10	10

1,2-DICHLOROETHANE



1. PERFORMANCE

1) Measuring range 5-50 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 2 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10 \, ^{\circ}\mathrm{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 15 % RSD-mid.: 10 % RSD-high: 10 %

3. CHEMICAL REACTION

Chlorine is produced by decomposing with an Oxidizer.

By reacting between this Chlorine and 3, 3'-Dimethylnaphthidine, Nitroso-compound is produced.

 $CICH_2CH_2CI + CrO_3 + H_2SO_4 \cdot nSO_3 \rightarrow CI_2$

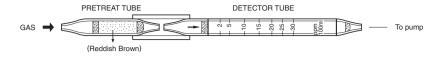
 $CI_2 + 3$, 3'-Dimethylnaphthidine \rightarrow Nitroso-compound

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

Substance	Interference	ppm	Coexistence
Nitrogen oxides	Similar stain is produced.		Higher readings are given.
Halogens	"		"
Halogenated hydrocarbons	"		"
Hexane	The accuracy of readings is not affected.	100	Lower readings are given.

2,2'-D,ICHLOROETHYL ETHER



1. PERFORMANCE

1) Measuring range 2-30 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 40 seconds/1 pump stroke

3) Detectable limit : 0.5 ppm4) Shelf life : 1 year5) Operating temperature $: 0 \sim 40 \text{ °C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Greenish yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. (CICH₂CH₂)₂O + CrO₃ + H₂SO₄→ 2HCI

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Halogenated hydrocarbons	Similar stain is produced.		Higher readings are given.	
Aliphatic hydrocarbons (more than C ₃)	The accuracy of readings is not affected.	less than 500	The accuracy of readings is not affected.	
Aromatic hydrocarbons	"	less than 300	"	
Alcohols	"	less than 100	"	
Esters	"	less than 300	"	

Tube	Corrected Concentration (ppm)						1)	
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (86 °F)	35 °C (95 °F)	40°C (104°F)
30	_	49	37	30	24	19	14	9
25	45	35	30	25	20	16	12	8
20	32	27	23	20	17	13	10	6
15	21	19	17	15	13	10	7	5
10	14	13	12	10	9	7	5	4
5	7	6	6	5	4	3	2	1
2	3	3	2	2	2	1	1	1

1,2-DICHLOROETHYLENE



1. PERFORMANCE

: 4.2-84 ppm 9.2-184 ppm 20-400 ppm 42-840 ppm 1) Measuring range Number of pump strokes 4 (400ml) $2(200m\ell)$ $1(100m\ell)$ $1/2(50 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke 3) Detectable limit : 0.5 ppm (4 pump strokes)

4) Shelf life 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : 5 ~ 40 °C

6) Temperature compensation: Necessary (See "TEMPERATURE CORRECTION TABLE") Direct reading from the scale calibrated by 1 pump stroke 7) Reading : Yellow → Red

8) Colour change

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Hydrogen chloride is produced by an Oxidizer and PH indicator is discoloured.

 $CICH = CHCI + PbO_2 + H_2SO_4 \rightarrow HC1$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Trichloroethylene	Similar stain is produced.	3	Higher readings are given.
Vinyl chloride	"	300	"
Hydrogen chloride	"	10	"
Chlorine	Pale red stain is produced.	15	"

(NOTE)

This tube is calibrated based on Cis-1,2-Dichloroethylene.

In case of trans-1,2-Dichloroethylene measurement, correct the reading value by the temperature correction table if necessary, then multiply the value by 1.1.

Tube		Concentration (ppm)
Readings (ppm)	5℃ (41*F)	10 °C ~ 40 °C (50 °F ~ 104 °F)
400	475	400
350	415	350
300	355	300
250	295	250
200	235	200
150	175	150
100	115	100
50	55	50
20	20	20

DICHLOROMETHANE



1. PERFORMANCE

1) Measuring range 30-1,000 ppm 10-200 ppm Number of pump strokes $2(200\text{m}\ell)$ $4(400\text{m}\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit : 5 ppm $(400 \text{m} \ell)$

4) Shelf life : 2 years (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $5 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : White→Reddish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

Chlorine is produced by an Oxidizer. By reacting between this Chlorine and o-Toluidine, Orthoquinone is produced. CH2CI2 + I2O5 + CrO3 + H2SO4 ⋅ nSO3 → CI2

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

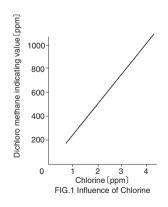
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Halogens or Halogenated hydrocarbons FIG.1	Similar stain is produced.	Higher readings are given.

(NOTE)

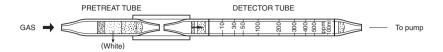
When the concentration is below 200 ppm, 4 pump strokes can be used to determine the lower concentration Following formula is available for the actual concentration.

Actual concentration = $1/3 \times$ Temperature corrected value.



Tube		Corrected Concentration (ppm)						
Readings (ppm)	5℃ (41°F)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (86 °F)	35 ℃ (95 °F)	40 ℃ (104 °F)
1000	_	_	1230	1000	820	670	550	450
800	_	1190	990	800	660	530	440	360
600	1120	900	740	600	500	400	330	270
400	720	600	500	400	330	260	220	180
200	360	300	250	200	165	135	110	90
100	170	145	120	100	80	65	50	45
30	50	45	35	30	25	20	15	10

1,3-DICHLOROPROPANE



1. PERFORMANCE

1) Measuring range 10-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 1 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10 \, ^{\circ}\text{C}$)

5) Operating temperature : $0 \sim 40 \, \text{°C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, chlorine is produced. Yellowish Orthoquinone is produced by reacting between this Chlorine and o-Toulidine.

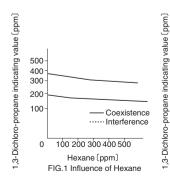
$$\begin{array}{c} CH_2CICH_2CH_2CI + I_2O_5 + C_7O_3 + H_2SO_4 \cdot nSO_3 \rightarrow CI_2 \\ H_2N - \bigcirc - \bigcirc - NH_2 + CI_2 \rightarrow CINH_2 = \bigcirc = \bigcirc = NH_2C \\ CH_3 \qquad CH_3 \qquad CH_3 \qquad CH_3 \qquad CH_3 \end{array}$$

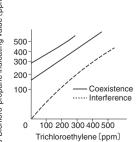
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

HITEHI EHENGE AND GROOG GENOTIVITI						
Substance	Interference	Coexistence				
Alcohols	The accuracy of readings is not affected.	The accuracy of readings is not affected.				
Esters	"	"				
Aromatic hydrocarbons	"	"				
Aliphatic hydrocarbons FIG.1	"	"				
Halogenated hydrocarbons FIG.2	Similar stain is produced.	Higher readings are given.				





TEMPERATURE CORRECTION TABLE Tube | Corrected Concentration (ppm) | Readings | 0 °C | 10 °C | 20 °C | 30 °C | 40

Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 ℃ (68 °F)	30 ℃ (86 °F)	40 °C (104 °F)
500		690	500	420	350
400	_	550	400	330	280
300	_	400	300	240	200
200	380	270	200	170	130
100	180	130	100	80	60
50	90	70	50	40	30
30	60	40	30	20	15
10	20	14	10	7	5

1,3-DICHLOROPROPENE



1. PERFORMANCE

1) Measuring range 0.5-10 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 0.1 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \text{ }^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellowish green → Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. C₃H₄CI₂ + C₇O₃ + H₂SO₄ → HCI

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

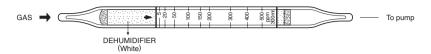
Substance	Interference	ppm	Coexistence
Chloropicrin	The accuracy of readings is not affected.	1800	The accuracy of readings is not affected.
Methyl isothiocyanate	"	600	"

TEMPERATURE CORRECTION TABLE

Temperature; To correct for temperature, multiply the tube reading by the following factors.

Temperature (°C)	0	1	2	3	4	5	6	7	8	9
Correction Factor	2.00	1.91	1.83	1.75	1.66	1.57	1.49	1.42	1.36	1.30
Temperature (°C)	10	11	12	13	14	15	16	17	18	19
Correction Factor	1.25	1.21	1.18	1.15	1.12	1.10	1.08	1.06	1.04	1.02
Temperature (°C)	20	21	22	23	24	25	26	27	28	29
Correction Factor	1.00	0.98	0.96	0.95	0.94	0.93	0.92	0.91	0.90	0.89
Temperature (°C)	30	31	32	33	34	35	36	37	38	39
Correction Factor	0.88	0.88	0.88	0.88	0.88	0.88	0.88	0.88	0.88	0.88
Temperature (°C)	40									
Correction Factor	0.88									

DICYCLOPENTADIENE



1. PERFORMANCE

1) Measuring range 2-60 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl cellosolve at 3 pump strokes and

Dicyclopentadiene concentration is determined by using a conversion chart at 3 pump strokes

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced. $C_{10}H_{12} + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

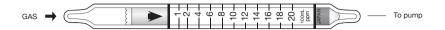
Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Aliphatic hydrocarbons (more than C ₃)	"	"
Ketones	"	"
Aromatic hydrocarbons	Whole reagent is changed to Brown.	"
Esters	"	"
Halogenated hydrocarbons	"	"

Dicyclopentadiene concentration (ppm)



No. 190U tube reading (ppm)

DIETHYL AMINE



1. PERFORMANCE

1) Measuring range 1-20 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 0.3 ppm4) Shelf life 3 years5) Operating temperature $0 \sim 40 \text{ °C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. (CH₃CH₂)NH + H₃PO₄→ [(C₂H₅)₂NH]₂HPO₄

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

Substance	Interference	Coexistence
Ammonia	Similar stain is produced.	Higher readings are given.
Other Amines	"	"

DIETHYL BENZENE



1. PERFORMANCE

1) Measuring range 10-180 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

Diethyl benzene concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $C_6H_4 (C_2H_5)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

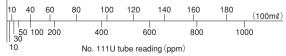
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

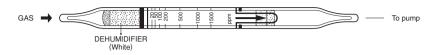
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar or brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	"

Diethyl benzene concentration (ppm)



DIISOBUTYL KETONE



1. PERFORMANCE

1) Measuring range \therefore 20-1,000 ppm Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : Graduations printed on the tube are calibrated by Metyl ethyl ketone at 1 pump

stroke and Diisobutyl ketone concentration is determined by using a conversion

chart at 1 pump stroke.

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

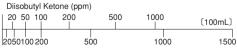
 $((CH_3)_2CHCH_2)_2CO + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

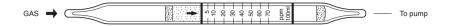
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohol	Similar or Brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Halogenated hydrocarbons		Whole reagent is changed to Brown, but if the maximum end point of the Pale blue stain is discernable, the accuracy of reading is not affected.
Aliphatic hydrocarbons		"



No. 139U tube reading (ppm)

N, N-DIMETHYL ACETAMIDE



1. PERFORMANCE

1) Measuring range 5-70 ppmNumber of pump strokes $2(200\text{m}\ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 1 ppm 4) Shelf life : 1 year 5) Operating temperature : $10 \sim 40 \, ^{\circ}$ C

6) Temperature compensation : Necessary under 20 °C (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

By reacting with Alkali, Dimethyl amine is produced. Dimethyl amine reacts on Phosphoric acid and PH indicator is discoloured.

 $CH_3CON(CH_3)_2 \rightarrow HN(CH_3)_2$

 $2HN(CH_3)_2 + H_3PO_4 \rightarrow ((CH_3)_2NH_2)_2HPO_4$

4. CALIBRATION OF THE TUBE

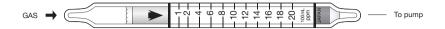
DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Carbon dioxide	The accuracy of readings is not affected.	0.17%	Lower readings are given.
Sulphur dioxide	"		
Ammonia	Similar stain is produced.		Higher readings are given.
Amines	"		"
Hydrazine	"		"

Tube	Correct	ed Concentratio	
Readings (ppm)	10 °C (50 °F)	15 °C (59 °F)	20-40 °C (68-104 °F)
70	120	84	70
60	102	70	60
50	85	60	50
40	67	47	40
30	50	36	30
20	35	23	20
10	17	12	10
- 5	10	6	5

DIMETHYL AMINE



1. PERFORMANCE

1) Measuring range 1-20 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 0.2 ppm4) Shelf life 3 years5) Operating temperature $0 \sim 40 \text{ °C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

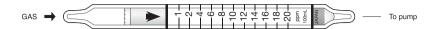
By reacting with Phosphoric acid, PH indicator is discoloured. (CH₃)₂NH + H₃PO₄→(CH₃NH₃)₂HPO₄

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Ammonia	Similar stain is produced.	Higher readings are given.
Other Amines	"	"

N, N-DIMETHYL ANILINE



1. PERFORMANCE

1) Measuring range 0.5-9 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke

and N, N-Dimethyl aniline concentration is determined by using a conversion

chart at 1 pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

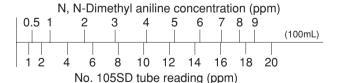
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. C_6H_5N (CH₃) $_2+H_3PO_4 \rightarrow (R_3NH)_3PO_4$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



DIMETHYL ETHER



1. PERFORMANCE

1) Measuring range 0.01-1.2%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke 3) Detectable limit : 0.001 % (10 ppm)

4) Shelf life \vdots 3 years 5) Operating temperature \vdots 0 \sim 40 $^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Orange → Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10 % RSD-mid.: 10 % RSD-high: 10 %

3. CHEMICAL REACTION

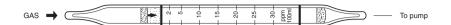
Chromium oxide is reduced. $CH_3OCH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is changed to Brown.
Propane		0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	Higher readings are given.

N, N-DIMETHYL FORMAMIDE



1. PERFORMANCE

: 2-30 ppm 1-5 ppm 1) Measuring range Number of pump strokes 1 (100ml) $2(200m\ell)$ 2) Sampling time 1 minute/1 pump stroke 3) Detectable limit $0.2 \text{ ppm} (200 \text{m} \ell)$

4) Shelf life 2 years 5) Operating temperature 10 ~ 40 ℃

Necessary $(0 \sim 20 \, ^{\circ}\text{C})$ (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation:

Direct reading from the scale calibrated by 1 pump stroke 7) Reading

8) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with alkali, Amine is produced. Further, PH indicator is discoloured by reacting together with phosphoric acid.

 $HCON(CH_3)_2 \rightarrow HN(CH_3)_2$ $2HN(CH_3)_2 + H_3PO_4 \rightarrow ((CH_3)_2NH_2)_2HPO_4$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Sulphur dioxide	FIG.1	The accuracy of readings is not affected.	200	Lower readings are given.
Carbon dioxide	FIG.2	"	0.1%	"
Chlorine		"		
Ammonia		Similar stain is produced.		Higher readings are given.
Amines		"	"	"
Hydrazine		"	"	"

(NOTE)

When the concectration is below 5 ppm, 2 pump strokes can be used to determine the lower concentration. Following formula is available for the actual concentration.

Actual concentration = $1/2 \times$ Temperature corrected value

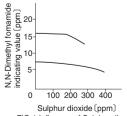
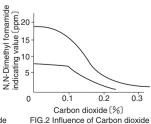


FIG.1 Influence of Sulphur dioxide



TEMPERATURE CORRECTION TABLE

Tube	Correct	on (ppm)	
Readings (ppm)	10 ℃ (50 °F)	15 ℃ (59 °F)	20-40°C (68-104°F)
30	81	40	30
25	67	33	25
20	54	27	20
15	40	20	15
10	27	13	10
5	13	7	5
2	5	3	2

1,4-DIOXANE



1. PERFORMANCE

1) Measuring range 0.05-2.5%Number of pump strokes $2(200m\ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit \therefore 10 ppm 4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 °C

6) Reading Graduations printed on the tube are calibrated by Methyl ethyl ketone at 2 pump strokes and 1,4-Dioxane concentration is determined by using a conversion chart.

7) Colour change : Orange → Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Dichromate is reduced. $O \stackrel{\text{CH}_2\text{CH}_2}{\sim} O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

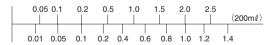
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is discoloured to Brown.
Propane		0.2%	"
Other organic or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	Higher reading are given.

Dioxane (%)



No.139SB Tube reading (%)

1,4-DIOXANE



1. PERFORMANCE

1) Measuring range 20-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Graduations printed on the tube are calibrated by Methyl alcohol at 1 pump stroke

and 1,4-Dioxane concentration is determined by using a conversion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15 % RSD-mid.: 10 % RSD-high: 10 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

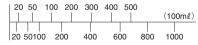
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence
Alcohols	FIG.1	Similar stain is produced.	Higher readings are given.
Esters	FIG.2		"
Aliphatic hydrocarbons (more than C ₃)		Whole reagent is changed to Pale brown.	"
Aromatic hydrocarbons		"	"
Ketones		"	"
Halogenated hydrocarbor	าร	"	"

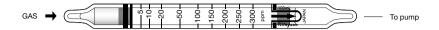
1,4-Dioxane concentration (ppm)



No.119U Tube reading (ppm)

Tube	Corrected Concentration (%)				
Readings (%)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
(/0/	(32 F)	(30 F)	(00 17)	(00 17)	
500	700	600	500	470	440
400	480	440	400	370	350
300	370	330	300	280	260
200	250	220	200	190	170
100	130	110	100	90	85
50	65	60	50	45	40
20	25	23	20	18	15

DIVINYL BENZENE



1. PERFORMANCE

1) Measuring range 5-50 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : The tube scale is calibrated based on Styrene at 1 pump stroke and

Divinyl benzene concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 10%

3. CHEMICAL REACTION

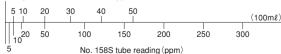
A polymer of Divinyl benzene is produced by Sulphuric acid.

4. CALIBRATION OF THE TUBE

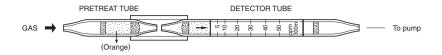
GAS CHROMATOGRAPHY

Substance	Interference	ppm	Coexistence
Acrylonitrile	The accuracy of readings is not affected.	400	Lower readings are given.
Butadiene	Similar stain is produced and higher readings are given.	5	Uneven discolouration is produced and higher readings are given.
Formaldehyde	"	15	Yellowish orange stain is produced and higher readings are given.
Acetaldehyde	"	350	Similar stain is produced and higher readings are given.
Methyl alcohol	The accuracy of readings is not affected.	0.35%	Pale discolouration is produced and higher readings are given.
Ethyl alcohol	"	0.18%	"
Ethyl acetate	"	700	"
Butyl acetate	"	700	"





EPICHLOROHYDRIN



1. PERFORMANCE

1) Measuring range $3(300 \text{m} \ell)$ 3.

2) Sampling time : 2 minutes/3 pump strokes

3) Detectable limit : 1 ppm4) Shelf life : 1 year5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Greenish yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured.

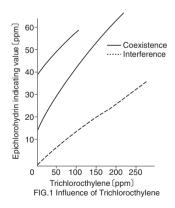
$$CH2CHCH2Cl + CrO3 + H2SO4 \rightarrow HCl$$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Halogenated hydrocarbons FIG.1	Similar stain is produced.	Higher readings are given.



Tube	Corrected Concentration (ppm)						
Readings (ppm)	0°C (32°F)	5°C (41°F)	10 ℃ (50 °F)	15 °C (59 °F)	20 ℃ (68 °F)	30 °C (86°F)	40 °C (104 °F)
50	_	-	70	60	50	42	37
40	_	-	57	47	40	34	30
30	_	61	42	35	30	25	23
20	80	39	29	24	20	17	15
10	33	19	15	12	10	9	8
5	15	10	8	6	5	5	1

ETHYL ACETATE



1. PERFORMANCE

1) Measuring range 0.1-5.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 0.001%(10 ppm)4) Shelf life 3 years5) Operating temperature $0 \sim 40\%$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange→Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3CO_2C_2H_5 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is discoloured to Brown.
Propane		0.2%	"
Other organic or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	Higher reading are given.

Tube	Corrected Concentration (%)				
Readings	0°C	10 °C	20 °C	30 °C	40 °C
(%)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)
5.0	_	_	5.0	4.0	3.0
4.5	_	_	4.5	3.6	2.7
4.0	-	5.8	4.0	3.2	2.4
3.5	-	5.0	3.5	2.8	2.1
3.0	-	4.2	3.0	2.3	1.7
2.5	-	3.5	2.5	1.9	1.4
2.0	-	2.6	2.0	1.5	1.1
1.5	2.6	1.9	1.5	1.1	0.7
1.0	1.5	1.2	1.0	0.7	0.4
0.5	0.6	0.6	0.5	0.3	0.2
0.3	0.4	0.3	0.3	0.2	0.1
0.1	0.1	0.1	0.1	0.1	0.1

ETHYL ACETATE



1. PERFORMANCE

1) Measuring range 10-1,000 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 2 years5) Operating temperature $: 10 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION COEFFICIENT TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3CO_2C_2H_5 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher reading are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons (more than C ₃)	"	"
Halogenated hydrocarbons FIG.1	Whole reagent is discoloured to Pale brown.	If the maximum end point of the brown stain is discernable, the accuracy of readings is not affected.

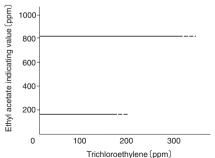
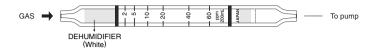


FIG.1 Influence of Trichloroethylene

TEMPERATURE CORRECTION COEFFICIENT TABLE

Tube	Correction Coefficient (at 20 °C)						
Readings (ppm)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	25℃ (77°F)	30 °C (86 °F)	35 °C (95 °F)	40 °C (104 °F)
1,000	1.33	1.17	1.00	0.87	0.74	0.64	0.53
800	1.38	1.19	1.00	0.86	0.73	0.63	0.53
600	1.40	1.20	1.00	0.86	0.72	0.63	0.53
400	1.40	1.20	1.00	0.85	0.70	0.58	0.46
200	1.40	1.20	1.00	0.84	0.68	0.55	0.42
100	1.50	1.25	1.00	0.81	0.62	0.48	0.33
50	1.50	1.25	1.00	0.77	0.54	0.43	0.32
30	1.50	1.25	1.00	0.77	0.53	0.42	0.30
10	1.50	1.25	1.00	0.75	0.50	0.40	0.30

ETHYL ACRYLATE



1. PERFORMANCE

1) Measuring range \cdot 5-60 ppm Number of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit 0.5 ppm4) Shelf life: 2 years5) Operating temperature $0 \sim 40 \text{ °C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Graduations printed on the tube are calibrated by Methyl acrylate at 2 pump strokes

and Ethyl acrylate concentration is determined by using a conversion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = CHCO_2CH_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

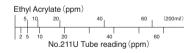
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

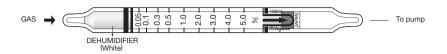
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Alcohols	Similar stain is produced.	Higher reading are given.	
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is discoloured to Dark brown.	"	
Halogenated hydrocarbons	"	"	
Esters	"	"	
Aromatic hydrocarbons	"	"	

Tube	Corrected Concentration (ppm)					
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)	
60	80	70	60	52	44	
40	52	46	40	34	28	
20	27	23	20	17	15	
10	14	12	10	9	7	
5	7	6	5	4	3	



ETHYL ALCOHOL (ETHANOL)



1. PERFORMANCE

1) Measuring range 0.05-5.0%Number of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 0.01%4) Shelf life 3 years 5) Operating temperature $0 \sim 40\%$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellowish orange → Light green (The top of discoloured layer is Brown, but read

at the top of Pale green.)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $C_2H_5OH + C_7f^6 + H_3PO_4 \rightarrow C_7f^3 + C_7f^3$

4. CALIBRATION OF THE TUBE

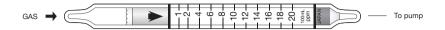
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Aliphatic hydrocarbons (over C ₃)	Similar stain is produced.	Higher reading are given.	
Alcohols	"	"	
Esters	"	"	
Ketones	"	"	
Aromatic hydrocarbons	"	"	
Halogenated hydrocarbons	Pale brown stain is produced.	If the maximum end point of the stain is discernable, the accuracy of readings is not affected.	

Tube	Corrected Concentration (%)					
Readings (%)	0°C (32°F)	10 °C (50 °F)	20°C (68°F)	30 °C (86 °F)	40 °C (104 °F)	
5.0	_	_	5.0	3.9	3.2	
4.0	_	_	4.0	3.2	2.6	
3.0	_	_	3.0	2.4	2.0	
2.0	_	_	2.0	1.6	1.3	
1.0	_	1.9	1.0	0.8	0.7	
0.5	_	0.8	0.5	0.4	0.3	
0.3	0.9	0.4	0.3	0.3	0.2	
0.1	0.1	0.1	0.1	0.1	0.1	

ETHYL AMINE



1. PERFORMANCE

1) Measuring range 1-20 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 0.2 ppm4) Shelf life 3 years5) Operating temperature $0 \sim 40 \text{ °C}$

6) Reading : The tube scale is calibrated based on Methyl amine at 1pump stroke and the

tube has the same sensitivity for Ethyl amine.

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, Ph indicator is discoloured. $CH_3CH_2NH_2 + H_3PO_4 {\longrightarrow} (CH_3NH_3) \, {}_2HPO_4$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence	
Ammonia	Similar stain is produced.	Higher reading are given.	
Other amines	"	"	

ETHYL BENZENE



1. PERFORMANCE

1) Measuring range 10-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 3 ppm4) Shelf life : 1.5 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (0 \sim 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. $C_6H_5C_2H_5 + I_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence	
Benzene	Similar stain is produced.		Higher readings are given.	
Toluene	"		"	
Xylene	"		"	
Methanol	The accuracy of readings is not affected.	1%	"	
Hexane	Pale brown stain is produced.	0.1%	The maximum end point of the stain is indiscernable and higher readings are given.	

TEINI ETIATOTIE GOTTILOTION TABL							
Tube	Corrected Concentration (ppm)						
Readings (ppm)	0°C (32°F)	5℃ (50°F)	10 °C (68 °F)	20 °C (86 °F)			
500	-	-	600	500			
400	590	520	465	400			
300	390	360	330	300			
200	240	225	215	200			
100	120	112	105	100			

ETBE (ETHYL-tert-BUTYL ETHER)



1. PERFORMANCE

1) Measuring range 1-60 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit 0.5 ppm4) Shelf life 1 year5) Operating temperature $0 \sim 40 \text{ C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Pale yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $C_2H_5OC(CH_3)_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

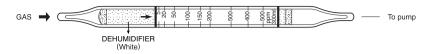
Substance	Interference	Coexistence
Ethyl alcohol	Similar stain is produced.	Higher readings are given.

TEMPERATURE CORRECTION TABLE

Temperature; To correct for temperature, multiply the tube reading by the following factors.

			_	-	-					
Temperature (℃)	0	1	2	3	4	5	6	7	8	9
Correction Factor	1.86	1.81	1.75	1.70	1.65	1.60	1.55	1.50	1.46	1.41
Temperature (℃)	10	11	12	13	14	15	16	17	18	19
Correction Factor	1.37	1.33	1.29	1.25	1.21	1.17	1.13	1.10	1.06	1.03
Temperature (℃)	20	21	22	23	24	25	26	27	28	29
Correction Factor	1.00	0.97	0.95	0.92	0.90	0.88	0.86	0.84	0.82	0.81
Temperature (℃)	30	31	32	33	34	35	36	37	38	39
Correction Factor	0.79	0.78	0.77	0.76	0.76	0.75	0.75	0.74	0.74	0.73
Temperature (℃)	40									
Correction Factor	0.73	1								

ETHYL CELLOSOLVE



1. PERFORMANCE

1) Measuring range 5-500 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 35 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

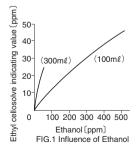
Chromium oxide is reduced. $C_2H_5OCH_2CH_2OH + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

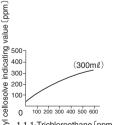
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence
Alcohols F	FIG.1	Similar stain is produced.	Higher readings are given.
Ethers		"	"
Aliphatic hydrocarbons (more than C ₃)		Whole reagent is changed to Brown.	"
Aromatic hydrocarbons		"	"
Esters		"	"
Ketones		"	"
Halogenated hydrocarbons F	FIG.2	//	"

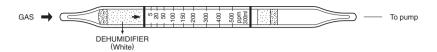




1,1,1-Trichloroethane (ppm) FIG.2 Influence of 1,1,1-Trichloroethane

						Corrected Concentration (ppin)							
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20℃ (68°F)	25 °C (77 °F)	30 °C (80 °F)	35℃ (95°F)							
500	800	620	500	410	340	270							
400	620	490	400	330	260	200							
300	450	370	300	250	200	150							
200	290	250	200	160	130	100							
150	220	190	150	120	90	70							
100	150	130	100	80	60	50							
50	80	70	50	40	30	30							
20	30	25	20	15	12	10							
5	10	7	5	4	3	2							
	500 400 300 200 150 100 50 20	500 800 400 620 300 450 200 290 150 220 100 150 50 80 20 30	500 800 620 400 620 490 300 450 370 200 290 250 150 220 190 100 150 130 50 80 70 20 30 25	500 800 620 500 400 620 490 400 400 620 490 400 300 450 370 300 200 290 250 200 150 220 190 150 150 130 100 50 80 70 50 20 30 25 20	500 800 620 500 410 400 620 490 400 330 300 450 370 300 250 200 290 250 200 180 150 220 190 150 120 100 150 130 100 80 50 80 70 50 40 20 30 25 20 15	500 800 620 500 410 340 400 620 490 400 330 260 300 450 370 300 250 200 200 290 250 200 160 130 150 220 190 150 120 90 100 150 130 100 80 60 50 80 70 50 40 30 20 30 25 20 15 12							

ETHYL CELLOSOLVE ACETATE



1. PERFORMANCE

1) Measuring range 5-150 ppmNumber of pump strokes $3(300\text{m}\ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 35 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading Graduations printed on the tube are calibrated by Ethyl cellosolve at 3 pump strokes and Ethyl cellosolve acetate concentration is determined by using a conversion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

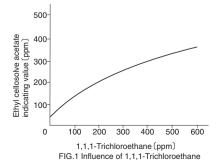
 $CH_3CO_2CH_2CH_2OC_2H_5 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

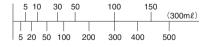
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Pale Brown.	"
Aromatic hydrocarbons	"	"
Esters	"	"
Ketones	"	"
Halogenated hydrocarbons FIG.1	"	"



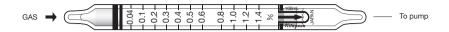
Ethyl cellosolve acetate (ppm)



No.190U Tube reading (ppm)

Conversion	Corrected Concentration (ppm)					
Vaiue (ppm)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (80 °F)	35 °C (95 °F)
150	230	190	150	120	90	75
100	160	130	100	80	70	60
50	80	60	50	40	35	30
20	25	23	20	18	16	14
10	10	10	10	10	8	7
5	5	5	5	5	5	5

ETHYL ETHER



1. PERFORMANCE

1) Measuring range 0.04-1.4 % Number of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 0.001%(10 ppm)4) Shelf life 3 years5) Operating temperature 0.001%(10 ppm)

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange→Dark green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $(C_2H_5)_2O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

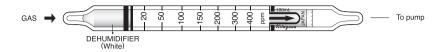
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is changed to Brown.
Propane		0.2%	"
Other organic gases and va- pours except Halogenated hydrocarbons		50	Higher readings are given.

Tube	Corrected Concentration (%)				
Readings (%)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
1.4	1.8	1.6	1.4	1.3	1.1
1.2	1.5	1.4	1.2	1.1	0.9
1.0	1.2	1.1	1.0	0.9	0.8
0.8	0.9	0.9	0.8	0.7	0.6
0.6	0.6	0.6	0.6	0.6	0.5
0.5	0.5	0.5	0.5	0.5	0.4
0.4	0.4	0.4	0.4	0.4	0.3
0.3	0.3	0.3	0.3	0.3	0.3

ETHYL ETHER



1. PERFORMANCE

1) Measuring range 20-400 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 1 ppm 4) Shelf life : 2 years 5) Operating temperature : $10 \sim 40 \, \text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid: 10% RSD-high: 5%

3. CHEMICAL REACTION

Potassium dichromate is reduced. $(C_2H_5)_2O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	Whole reagent is changed to Pale brown.	"

Tube	Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	20 ℃ (68 °F)	30 °C (86 °F)	40 °C (104 °F)		
400	640	400	290	230		
300	480	300	220	180		
200	320	200	150	120		
150	240	150	110	90		
100	160	100	80	60		
50	90	50	40	30		
20	40	20	20	10		

ETHYLENE



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \text{Number of pump strokes} \\ \text{Number of pump strokes} \\ \text{2) Sampling time} \\ \text{3) Detectable limit} \\ \end{array} \begin{array}{c} \text{: } 0.5\text{--}100 \text{ ppm} \\ 1\,(100\text{m}\,\ell) \\ \text{: } 2\,\text{minutes}/1\,\text{pump stroke} \\ \text{: } 0.01\,\text{ppm}\,(500\text{m}\,\ell) \\ \end{array}$

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Colour intensity method 7) Colour change : Pale yellow → Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: RSD-mid.: RSD-high:

3. CHEMICAL REACTION

Molybdate is reduced and molybdeum blue is produced. $H_2C = CH_2 + PdSO_4 + (NH_4) \ _2MoO_4 \rightarrow Mo_3O_8$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen (over 40 ℃)	Similar stain is produced.	10%	Whole reagent is discoloured to Blue and higher readings are given.
Saturated hydrocarbons	"		Higher readings are given.
Acetylene	Dark blue stain is produced.		"
Carbon monoxide	Green or Blue stain is produced.		Lower readings are given.
Hydrogen sulphide	Black stain is produced.	1,000	"
Hydrogen cyanide	Original colour is faded to White.		"
Benzene	Yellowish orange or Yellowish brown stain is produced.		
Carbon disulphide	"		
Chlorine	"		
Nitrogen dioxide	"	1	
Ammonia	Original colour fades to White.		Lower readings are given.

ETHYLENE



1. PERFORMANCE

1) Measuring range \therefore 20-1,200 ppm Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 3 minutes/1 pump stroke

3) Detectable limit \therefore 10 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

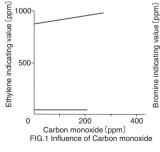
Molybdate is reduced and molybdeum blue is produced. $H_2C = CH_2 + PdSO_4 + (NH_4) 2MoO_4 \rightarrow Mo_3O_8$

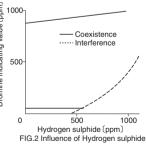
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		ppm	Interference	Coexistence
Carbon monoxide	FIG.1	10	Whole layer is discoloured to Green.	Higher readings are given.
Hydrogen sulphide	FIG.2	500	Black stain is produced.	"
Acetylene		20	Similar stain is produced.	"
Propylene			"	"





TEMPERATURE CORRECTION TABLE

Lube	Corrected Concentration (ppm)				
Readings	0℃	10 °C	20 °C	30 °C	40 °C
(ppm)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)
1,200	950	1,100	1,200	1,300	1,350
1,000	800	900	1,000	1,050	1,100
800	650	720	800	820	840
600	500	550	600	600	600
400	400	400	400	400	400
200	200	200	200	180	150
100	100	100	100	50	
50	70	60	50		
20	40	30	20	_	_

ETHYLENE

GAS → To pump

1. PERFORMANCE

1) Measuring range : 1-200 ppm Number of pump strokes $4(400 \text{m} \ell)$

2) Sampling time : 12 minutes/4 pump strokes

3) Detectable limit : 0.1 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (0 \sim 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 4 pump strokes

8) Colour change : Yellow→Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Molybdate is reduced and molybdeum blue is produced. $H_2C = CH_2 + PdSO_4 + (NH_4) 2MoO_4 \rightarrow Mo_3O_8$

4. CALIBRATION OF THE TUBE

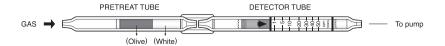
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Acetylene	Dark brown stain is produced.	Higher readings are given.
Carbon monoxide	Similar stain is produced.	"
Hydrogen sulphide	Black stain is produced.	"
Propylene	Similar stain is produced.	"

Tube	Corrected Concentration (ppm)			
Readings	.0℃	10 °C	20 ℃~40 ℃	
(ppm)	(32°F)	(50°F)	(68°F~104°F)	
200	220	210	200	
180	200	190	180	
160	180	170	160	
140	160	150	140	
120	130	140	120	
100	120	110	100	
80	105	92.5	80	
60	85	72.5	60	
40	70	65	40	
20	55	40	20	
10	40	25	10	
5	25	15	5	
3	18	12.5	3	
1	7	4	1	

ETHYLENE DIBROMIDE



1. PERFORMANCE

1) Measuring range 1-50 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 0.2 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; 0 \sim 10 $^{\circ}\mathrm{C})$

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Bromine is produced by an Oxidizer. By reacting between this Bromine and o-Toluidine, Orthoquinone is produced BrCH₂CH₂Br + I₂O₅ + CrO₃ + H₂SO₄ \rightarrow Br₂

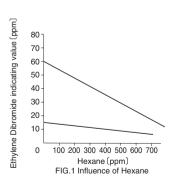
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

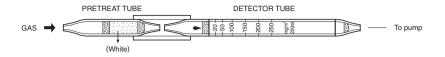
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference		Coexistence	
Halogens	Similar stain is produced.		Higher readings are given.	
Halogenated hydrocarbons	"		"	
Hexane FIG.1	The accuracy of readings is not affected.	200	Lower readings are given.	

Tube	Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (80 °F)	35℃ (95°F)
50	-	-	82	50	42	39
40	_	80	58	40	35	33
30	98	56	40	30	27	26
20	50	40	30	20	18	18
10	16	14	12	10	10	10
5	7	7	6	5	5	5
1	1	1	- 1	- 1	- 1	- 1



ETHYLENE GLYCOL



1. PERFORMANCE

1) Measuring range $20-250 \text{ mg/m}^3$ Number of pump strokes $2(200\text{m}\ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit 5 mg/m^3 4) Shelf life 1.5 years5) Operating temperature $20 \sim 40 \, \text{°C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Pink→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Formic acid is produced and PH indicator is discoloured.

HOCH₂CH₂OH + HIO₄ → 2HCHO + HIO₃ + H₂O HCHO + HIO₄ + H₂SO₄ → HCOOH + HIO₃ HCOOH + N₈OH → N₈ (HCOO) + H₂O

4. CALIBRATION OF THE TUBE

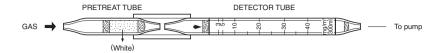
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Aldehydes	Similar stain is produced.	Higher readings are given.
Sulphur dioxide	"	"
Ethylene oxide	"	"
Hydrogen sulphide	Orange/Yellow stain is produced.	"

Tube			Concentrati	on (mg/m³)			
Readings (mg/m³)	20 °C (68 °F)	22°C (71.6°F)	25℃ (77 F)	30°C (86°F)	40 ℃ (104 F)		
250	- (001)	370	250	200	155		
200	-	260	200	165	130		
150	270	170	150	125	105		
100	120	110	100	85	75		
50	60	55	50	45	40		
20	23	20	20	18	15		

ETHYLENE GLYCOL



1. PERFORMANCE

1) Measuring range $3-40 \text{mg/m}^3$ Number of pump strokes $3(300 \text{m}\ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit 1 Img/m^3 4) Shelf life 2 years5) Operating temperature $15 \sim 40 \text{ }^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Formic acid is produced and PH indicator is discoloured.

HOCH₂CH₂OH + HIO₄ → 2HCHO + HIO₃ + H₂O HCHO + HIO₄ + H₂SO₄ → HCOOH + HIO₃ HCOOH + N₈OH → N₈(HCOO) + H₂O

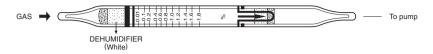
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Aldehydes	Similar stain is produced.	Higher readings are given.
Sulphur dioxide	Brownish yellow stain is produced.	"
Hydrogen sulphide	"	"

Tube	Corrected Concentration (mg/m³)						
Readings	15°C	20°C	30°C	40°C			
(mg/m³)	(59°F)	(68°F)	(86°F)	(104 F)			
40	67	40	32	25			
30	50	30	24	19			
20	34	20	16	12			
10	17	10	8	6			
5	8	5	4	3			
9		9	2.5	2			



1. PERFORMANCE

1) Measuring range 1.0-4.0% 0.01-1.8 % Number of pump strokes $1/2(50m\ell)$ 1 $(100m\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke 3) Detectable limit 1.005%(50 ppm)

3) Detectable limit 0.005% (3) Shelf life 3 years 0.005% (2) Operating temperature 0.005% (3) 0.005% (4) 0.005% (5)

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_2CH_2O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

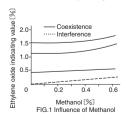
GAS CHROMATOGRAPHY

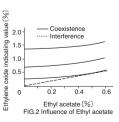
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference ppm		Coexistence	
Alcohols FIG.1		Similar stain is produced.		Higher readings are given.	
Esters	FIG.2	"		"	
Ketones		"		"	
Aromatic hydrocarbons		"		"	
Halogenated hydr	rocarbons	Whole reagent is discoloured to Pale brown.	0.5%	"	

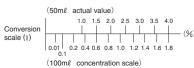
Ethanol is indicated with half the sensitivity, Ethyl acetate has the same sensitivity with Ethylene oxide. (NOTE)

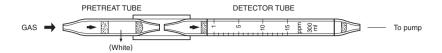
- If the discolouration exceeds the scale, replace the tube with new one and pull the handle at half stroke (to 50mℓ line). And read the concentration.
- Correct the reading value with the temperature correction table first, and convert the value into an actual concentration by using the conversion scale shown in the instruction sheet.





Tube	Corrected Concentration (%)				
Readings	0.0	10℃	20 °C	30℃	40℃
(%)	(32°F)	(50°F)	(68°F)	(86 °F)	(104°F)
1.8	-	2.00	1.80	1.70	1.64
1.6	2.06	1.78	1.60	1.50	1.46
1.4	1.84	1.58	1.40	1.35	1.28
1.2	1.60	1.34	1.20	1.14	1.09
1.0	1.31	1.12	1.00	0.93	0.88
0.8	1.08	0.91	0.80	0.73	0.68
0.6	0.86	0.70	0.60	0.54	0.49
0.4	0.63	0.49	0.40	0.34	0.30
0.2	0.36	0.26	0.20	0.16	0.13
0.1	0.20	0.14	0.10	0.08	0.06
0.01	0.05	0.02	0.01	0.01	0.01





1. PERFORMANCE

1) Measuring range 1-15 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit 0.5 ppm 4) Shelf life 0.5 2 years 5) Operating temperature 0.5 10 - 40 C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Formic acid is produced and PH indicator is discoloured.

CH2CH2O + H2SO4→HOCH2CH2OH

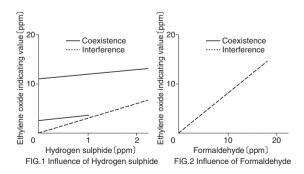
HOCH₂CH₂OH + HIO₄ → 2HCHO + HIO₃ + H₂O HCHO + HIO₄ + H₂SO₄ → HCOOH + HIO₃ HCOOH + N₈OH → N₈ (HCOO) + H₂O

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence
Aldehydes	FIG.2	Similar stain is produced.	Higher readings are given.
Sulphur dioxide		Pale yellow stain is produced.	"
Hydrogen sulphide	FIG.1	"	"



Tube	Corrected Concentration (ppm)						
Readings (ppm)	10 °C (50 °F)	15-25 ℃ (59-77 °F)	30 °C (86 °F)	40 °C (77 °F)			
15	19.0	15.0	13.0	10.0			
10	12.5	10.0	8.5	7.0			
5	6.0	5.0	4.0	3.5			
1	1.0	1.0	1.0	0.5			



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \text{130-2,600 ppm} \\ \text{Number of pump strokes} \end{array} \begin{array}{c} \text{130-2,600 ppm} \\ 1/2\left(50\text{m}\,\ell\right) \\ \text{1}\left(100\text{m}\,\ell\right) \\ \text{1.5 minutes/1 pump stroke} \end{array}$

3) Detectable limit $\begin{array}{c} : 10 \text{ ppm} (100 \text{m} \ell) \\ \text{4) Shelf life} \\ : 3 \text{ years} \\ \text{5) Operating temperature} \\ \vdots \ 10 \sim 40 \, ^{\circ}\text{C} \\ \end{array}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_2CH_2O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Alcohols	Similar stain is produced.	Higher readings are given.		
Esters	"	"		
Ethers	"	"		
Ketones	"	"		
Aliphatic hydrocarbons (more than C ₃)	"	"		
Aromatic hydrocarbons	"	"		
Halogenated hydrocarbons	"	"		

(NOTE)

In case of 1/2 pump strokes, the following formula is available for the actual concentration. Actual concentration = $2.6 \times$ Reading value

Tube		Corrected Concentration (ppm)						
Readings	10 °C	15 °C	20°C	25 °C	30 °C	35 °C	40°C	
(ppm)	(50°F)	(59°F)	(68°F)	(77°F)	(86°F)	(95°F)	(104°F)	
1000	980	990	1000	1080	1180	1350	1570	
800	780	790	800	880	970	1130	1310	
600	550	575	600	670	750	880	1050	
400	330	360	400	460	520	650	790	
200	150	175	200	240	280	330	430	
100	85	85	100	115	120	140	170	
50	45	45	50	50	55	60	70	

 $[\]ast$ In case of 1/2 pump strokes, no temperature correction is necessary at less than 20 $^{\circ}\!C$.



1. PERFORMANCE

1) Measuring range 5-100 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \vdots 3 ppm 4) Shelf life \vdots 3 years 5) Operating temperature \vdots 10 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_2CH_2O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

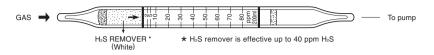
PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Esters	"	"
Ketones	"	"
Aliphatic hydrocarbons (more than C ₃)	"	"
Aromatic hydrocarbons	"	"
Halogen hydrocarbons	"	"

Tube		Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 ℃ (86 °F)	35 °C (95 °F)	40 °C (104 °F)
100	70	79	100	122	_	_	_
80	57	64	80	97	120	_	_
60	44	48	60	73	92	115	140
40	31	33	40	46	60	78	92
30	25	26	30	33	40	50	60
20	17	18	20	22	26	28	30
10	9	10	10	10	10	11	11
5	5	5	5	5	5	5	5

ETHYL MERCAPTAN



1. PERFORMANCE

1) Measuring range : 4-160 ppm 2-80 ppm 1-40 ppm Number of pump strokes : $1(100m\ell)$ 2 $(200m\ell)$ 4 $(400m\ell)$

2) Sampling time : 3 minutes/2 pump strokes 3) Detectable limit : 0.2 ppm (200mℓ)

4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 °C

6) Temperature compensation : Necessary (0 \sim 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : White → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Ethyl mercaptan reacts with Palladium sulphate. $C_2H_5SH + PdSO_4 \rightarrow (C_2H_5S)_2Pd + H_2SO_4$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Carbon monoxide	150	Dark grey stain is produced.		
Ethylene	200	"		
Hydrogen sulphide	40	Dark brown stain is produced.		
Acetylene	20	Pale brown stain is produced.		
Methyl mercaptan	1	Reddish yellow stain is produced.		
Methyl sulphide			1	Lower readings are given.
Nitrogen dioxide			1 //	
Chlorine			0.2	"

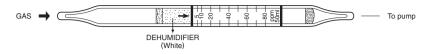
(NOTE)

- 1) Max. 40 ppm of Hydrogen sulphide is eliminated by pretreat reagent.
- 2) In case of 1 or 4 pump strokes, following formula is available for the actual concentration.

 $Actual concentration = Temperature corrected value \times \frac{2}{\text{Number of pump strokes}}$

Tube	Corrected Concentration (ppm)					
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20-40 °C (68-104 °F)			
80	91	85	80			
70	80	75	70			
60	69	64	60			
50	57	53	50			
40	45	42	40			
30	33	31	30			
20	21	20	20			

ETHYL MERCAPTAN



1. PERFORMANCE

1) Measuring range Solution Strokes Solution Strokes Solution Sol

3) Detectable limit $1 \text{ ppm} (100\text{m}\ell)$ 4) Shelf life 2 years5) Operating temperature $0 \sim 40 \text{ }^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1/2 pump strokes

7) Colour change : Yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride, Hydrogen chloride is produced and PH indicator is discoloured. $C_2H_5SH + H_9Cl_2 \rightarrow C_2H_5S\left(H_9Cl\right) + HCl$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Hydrogen sulphide	Similar stain is produced.	Higher readings are given.		
Phosphine	"	"		
Other mercaptans	"	"		
Arsine	"	"		
Hydrogen selenide	"	"		
Hydrogen cyanide	"	"		
Nitrogen dioxide	The accuracy of readings is not affected.	Lower readings are given.		
Ammonia	"	"		
Sulphur dioxide	Whole layer is discoloured to Pale red.	The accuracy of readings is not affected if the maximum end point of the pink stain is discernable.		

(NOTE)

In case of a 1 pump stroke, following formula is available for the actual concentration.

Acutual concentration = $0.5 \times$ Reading value

ETHYL MERCAPTAN



1. PERFORMANCE

4) Shelf life \therefore 2 years \therefore 0 \sim 40 $^{\circ}$ C

6) Reading : The tube scale is calibrated based on Methyl mercaptan at 1 pump stroke and

the tube has the same sensitivity for Ethyl mercaptan.

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric cholride, Hydrogen chloride is produced and PH indicator is discoloured. C2HsSH + HgCI2→RS (HgCI) + HCI

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Arsine	Similar stain is produced.	Higher readings are given.
Hydrogen selenide	"	"
Phosphine	"	"
Hydrogen sulphide	"	"
Hydrogen cyanide	Whole reagent is discoloured to Red.	"
Sulphur dioxide		Whole reagent is changed to Pale red, but Pink stain indicates Mercaptans conc.

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times \text{Reading value}$

ETHYL METHACRYLATE



1. PERFORMANCE

1) Measuring range 20-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

Ethyl methacrylate concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 5 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = C(CH_3)CO_2C_2H_5 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

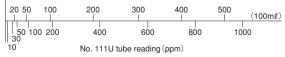
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

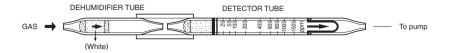
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Alcohols	Similar or brown stain is produced.	Higher readings are given.		
Esters	"	"		
Ketones	"	"		
Aromatic hydrocarbons	"	"		
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.		
Halogenated hydrocarbons	"	"		

Ethyl methacrylate concentration (ppm)



FORMALDEHYDE



1. PERFORMANCE

1) Measuring range 20-1,500 ppm Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minute s/1 pump stroke

3) Detectable limit : 5 ppm

4) Shelf life : 2 years (Necessary to store in refrigerated conditions ; $0 \sim 10 \, ^{\circ}\text{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

: Yellow → Pink

8) Colour change

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Hydroxylamine phosphate, Phosphoric acid is liberated and PH indicator is discoloured. $HCHO + (NH_2OH)_3 \cdot H_3PO_4 \rightarrow H_3PO_4 + HCH = NOH + H_2O$

4. CALIBRATION OF THE TUBE

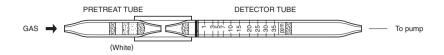
ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference pp		Coexistence
Other Aldehydes	Similar stain is produced.		Higher readings are given.
Ketones	"		"
Methanol	The accuracy of readings is not affected.	Formal- dehyde conc. ×1/2	Lower readings are given.

Tube	Corrected Concentration (ppm)								
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20°C (68°F)	25 °C (77 °F)	30 °C (86 °F)	35 °C (95 °F)	40 °C (104 °F)		
1500	-	_	1500	900	690	530	410		
1000	-	-	1000	700	550	430	340		
800	_	1300	800	600	490	390	300		
600	1200	900	600	490	410	330	250		
400	600	520	400	350	290	230	180		
200	240	230	200	180	150	130	100		
130	100	100	100	90	80	70	60		
50	50	50	50	45	40	35	30		
20	20	20	20	20	20	20	20		

FORMALDEHYDE



1. PERFORMANCE

1) Measuring range : 1-35 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 3 minutes/3 pump strokes

3) Detectable limit : 0.5 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 3 pump strokes

7) Colour change : White→Brownish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

By reacting with Aromatic compounds, a polymer is produced.

$$HCHO + \bigcirc + H_2SO_4 \longrightarrow \left[\bigcirc \bigcap_{H}^{C} \bigcap_{n} \right]_n + H_2O$$

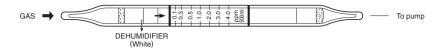
4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Styrene	Similar stain is produced.		Higher readings are given.
Trichloroethylene		500	"
Ethyl acetate		1,000	"
Ether		1,000	"
Acetaldehyde	Similar stain is produced.	1	"

FORMALDEHYDE



1. PERFORMANCE

1) Measuring range 0.05-2.0 ppm 0.05-2.0 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; 0 \sim 10 $^{\circ}\mathrm{C})$

5) Operating temperature : $10 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 5 pump strokes

8) Colour change : Yellowish orange → Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Hydroxylamine phosphate, Phosphoric acid is liberated and PH indicator is discoloured. HCHO + (NH₂OH)₃ ⋅ H₃PO₄→ H₃PO₄ + HCH = NOH + H₂O

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Ammonia		The accuracy of readings is not affected.	10	Lower readings are given. Inlet side is faded the discoloured layer.
Nitrogen dioxide	3	Similar stain is produced.	3	Higher readings are given. The top of discoloured layer becomes unclear.
Acetaldehyde		"		Higher readings are given.
Toluene		The accuracy of readings is not affected.		The accuracy of readings is not affected.
Methanol		"		

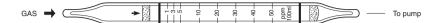
TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)					
Readings (ppm)	10 ℃ (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)		
4.0	6.4	4.0	2.4	1.6		
3.5	5.6	3.5	2.1	1.4		
3.0	4.8	3.0	1.8	1.2		
2.5	4.0	2.5	1.5	1.0		
2.0	3.2	2.0	1.2	0.8		
1.5	2.4	1.5	0.9	0.6		
1.0	1.6	1.0	0.6	0.4		
0.5	0.8	0.5	0.3	0.2		
0.3	0.5	0.3	0.18	0.12		
0.1	0.16	0.1	0.06	0.04		

(NOTE)

In case of 10 pump strokes, following formula is available for the actual concentration. Actual concentration = $1/2 \times$ Temperature corrected value

FORMIC ACID



1. PERFORMANCE

1) Measuring range : 1-50 ppm Number of pump strokes $1(100m\ell)$

2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit $0.1 \text{ ppm} (100 \text{m} \ell)$ 4) Shelf life 3 years 5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

Necessary $(0 \sim 20 \, ^{\circ}\mathbb{C})$ (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation

7) Reading Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. HCOOH + NaOH → HCOONa + H2O

4. CALIBRATION OF THE TUBE

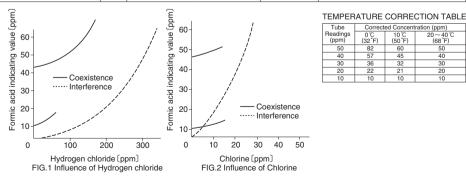
VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance ppm		Interference	ppm	Coexistence
Sulphur dioxide		Similar stain is produced.	HCO₂H conc. × 1/20	Higher readings are given.
Nitrogen dioxide		"	10	The top of discoloured layer becomes unclear.
Hydrogen chloride FIG.1		Pink stain is produced.	HCO₂Hconc. ×2	Higher readings are given.
Chlorine FIG.2		Yellow stain is produced.	5	"
Acetic acid		Similar stain is produced.		"

40

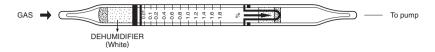
30



(NOTE)

This tube scale is calibrated based on Acetic acid and the same scale is available for Formic acid.

FURAN



1. PERFORMANCE

1) Measuring range Number of pump strokes 1/2 ($50m\ell$) 1 ($100m\ell$) 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit : 10 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Ethylene oxide at 1 pump stroke

and Furan concentration is determined by using a conversion chart.

7) Colour change : Orange→Black

2. RELATIVE STANDARD DEVIATION

RSD-low: 15 % RSD-mid.: 10 % RSD-high: 5 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

$$\boxed{ \bigcirc } + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

FIG.1 Influence of Toluene

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Aromatic hydrocarbons	FIG.1	Similar stain is produced.		Higher readings are given.
Esters	FIG.2	"		"
Ketones		"		
Alcohols	FIG.3	"		"
Halogenated hydrocarbo	ons	Whole reagent is changed to Pale brown.	0.5%	"

(NOTE)

In case of 1/2 pump strokes, following conversion scale is available for the actual concentration.

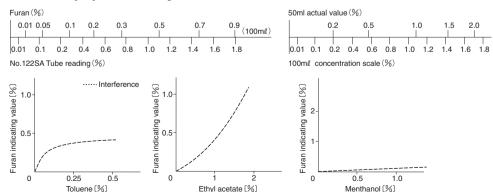
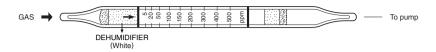


FIG.3 Influence of Menthanol

FIG.2 Influence of Ethyl acetate

FURFURAL



1. PERFORMANCE

1) Measuring range 2-60 ppmNumber of pump strokes $3(300 \text{ m}\ell)$

2) Sampling time : 4.5 minutes / 3 pump strokes

3) Detectable limit 0.5 ppm4) Shelf life 3 years5) Operating temperature $0 \sim 40 \text{ °C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Graduations printed on the tube are calibrated by Ethyl cellosolve at 3 pump strokes and Furfural concentration is determined by using a conversion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

$$CH = CH$$

$$\downarrow \qquad \qquad + H_2SO_4 + Cr^{6+} \rightarrow Cr^{3+}$$

$$CH = C - CHO$$

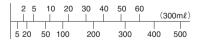
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Alcohols		Similar stains are produced and higher readings are given.	
Toluene	200	Whole reagent is changed to Pale orange.	Unclear stains are produced.

Furfural (ppm)



No.190U Tube reading (ppm)

Conver.	Corrected Concentration (ppm)				
value (ppm)	0 °C (32 °F)	10 ℃ (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
60			60	50	45
50		71	50	45	40
40		53	40	35	35
30	63	35	30	30	30
20	25	22	20	20	20
15	17	15	15	15	15
10	10	10	10	10	10
5	5	5	5	5	5
2	2	2	2	2	2

FURFURYL ALCOHOL



1. PERFORMANCE

1) Measuring range 5-25 ppmNumber of pump strokes $5(500\text{m}\ell)$

2) Sampling time : 5 minutes/5 pump strokes

3) Detectable limit : 3 ppm 4) Shelf life : 1 year 5) Operating temperature : $10 \sim 40 \, \text{°C}$

6) Operating Humidity : Not affected at $20 \sim 90 \% RH$

7) Reading : Direct reading from the scale calibrated by 5 pump strokes

8) Colour change : White→Black

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Chromic acid, Polymer is produced.

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Ethyl acetate		The accuracy of readings is not affected.
Hexane		"
Methanol		"
Toluene		"

(NOTE)

An original colour of white in reagent sometimes could be pale orange,

but it does not affect the tube readings.

GASOLINE



1. PERFORMANCE

1) Measuring range 0.05-0.6 % Number of pump strokes $1(100 \text{m} \ell)$

(This detector tube is calibrated by Hexane.)

2) Sampling time 1 minute/1 pump stroke3) Detectable limit 0.005%(50 ppm)4) Shelf life 3 years5) Operating temperature $10 \sim 40\%$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange→Dark green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. Gasoline $+ Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

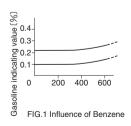
4. CALIBRATION OF THE TUBE

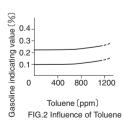
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Benzene	FIG.1	Black stain is produced.	400	Higher readings are given.
Toluene	FIG.2	"	800	"
Xylene		"	2,000	"
Acetylene		Similar stain is produced.		"
Ethylene		"		"
Cyclohexane		"		"
Aliphatic hydrocarbons (more t	than C ₃)	"		"

In presence of Alcohols, Ketones or Esters less than 6 %, the accuracy of readings is not affected.





Tube	Corrected Concentration (%)				
Readings (%)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	30 ℃ (86 °F)	40 °C (104 °F)
0.6	0.73	0.65	0.60	0.52	0.48
0.5	0.60	0.55	0.50	0.45	0.42
0.4	0.48	0.44	0.40	0.37	0.35
0.3	0.35	0.32	0.30	0.28	0.27
0.2	0.22	0.21	0.20	0.19	0.18
0.1	0.10	0.10	0.10	0.10	0.10

GENERAL HYDROCARBONS



1. PERFORMANCE

1) Measuring range : 50-1,400 ppm (as n-Hexane)

Number of pump strokes $1(100\text{m}\ell)$

2 pump strokes (200mℓ) are required for Kerosine and Mineral tupentine

determination.

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (refer to "Table 2. Temperature CorrectionTable")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5% (Controlled on n-Hexane)

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3 (CH_2)_4 CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence	
Aromatic hydrocarbons			The bottom of the discoloured layer is changed to Black and higher readings are given.	
Alcohols		6	Higher readings are given.	
Esters		6	"	
Ketones		6	"	

It has no influence on readings even if Alcohols, Esters or Ketones each co-exists up to 6 % .

(NOTE)

1) Determine the concentration of objective gas by multiplication with the figure shown in Table 1 after temperature correction.

Table.1 Coefficient Chart

Name of Gas	Figure	Name of Gas	Figure
Isobutane	0.8	Heptane	1.5
Pentane	0.8	Octane	2.0
n-Hexane	1.0	Cyclohexane	1.0

Table.2
TEMPERATURE CORRECTION TABLE (20 °C standard)

Tube	Corrected Concentration (ppm)							
Readings	0℃	10 °C	20°C	30 °C	40 °C			
1400	1630	1530	1400	1270	1180			
1200	1400	1320	1200	1090	1010			
1000	1170	1100	1000	910	840			
800	930	870	800	720	670			
600	700	660	600	550	500			
400	460	430	400	360	330			
200	220	210	200	180	170			
100	100	100	100	100	100			

Unit: ppm

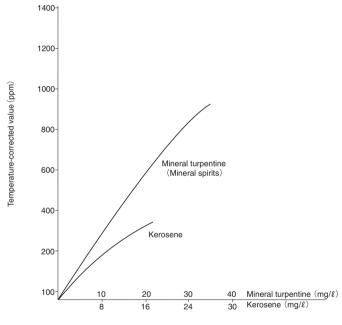
Example) For measuring Heptane at 10 °C of temperature

Reading concentration : 600 ppm
Concentration on temperature correction : 660 ppm
Concentration of Heptane : 990 ppm

- 2) Measurement of mixed solvents:
- (1) Take 2 pump strokes and use the following conversion graph to measure Kerosene or Mineral turpentine (Mineral spirits).
- (2) After temperature correction for the reading of the gas detector tube with the Table 2, determine the concentration from FIG.1 conversion graph.

Instance)

For measuring mineral turpentine at 40 °C
Reading Concentration 600 ppm
Concentration on temperature correction 500 ppm
Concentration of mineral tupentine 16mg /ℓ



HEPTANE



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \text{100-2,000 ppm} \\ \text{Number of pump strokes} \end{array}$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 15 ppm 4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : The tube scale is calibrated based on n-Hexane at 1 pump stroke and

Heptane concentration is determined by using a conversion chart at 1 pump stroke

8) Colour change : Orange → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

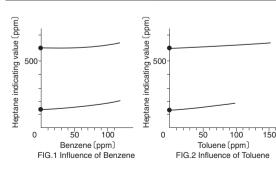
CH₃ (CH₂)₅CH₃ + Cr^{6+} + $H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

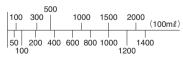
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Alcohols	Similar stain is produced.	6%	
Ketones	"	"	"
Esters	"	"	"
Aromatic hydrocarbons FIG.1	,2		The bottom of the discoloured layer is stained to Black and higher readings are given.
Aliphatic hydrocarbons (more than C ₃)	"		Higher readings are given.



Heptane concentration (ppm)



No.113SB Tube reading (ppm)

Tube	Corrected Concentration (ppm)					
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)	
2000	_		2000	1680	1460	
1600	_	2000	1600	1380	1200	
1200	2000	1480	1200	1040	920	
800	1200	940	800	720	660	
400	520	460	400	360	320	
200	260	230	200	180	160	
100	100	100	100	100	100	

n-HEXANE



1. PERFORMANCE

1) Measuring range 0.01-0.32% 0.05-0.6% Number of pump strokes 0.01-0.32% 1/2 0.05-0.6%

4) Shelf life : 3 years 5) Operating temperature : $10 \sim 40 \, ^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange→Dark green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3 (CH_2)_4 CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

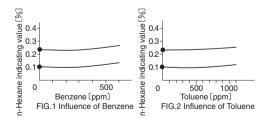
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Aliphatic hydrocarbons (more than C ₃)	Similar stain is produced.		Higher readings are given.
Acetylene	"		"
Ethylene	"		"
Cyclohexane	"		"
Benzene FIG.1	"	400	"
Toluene FIG.2	"	800	"
Xylene	"	2,000	"

In presence of Alcohols, Ketones or Esters less than 6 %, the accuracy of readings is not affected.



TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (%)				
Readings	10 °C	15 °C	20 °C	30°C	.40 °C
(%)	(50°F)	(59°F)	(68°F)	(86°F)	(104°F)
0.6	0.73	0.65	0.60	0.52	0.48
0.5	0.60	0.55	0.50	0.45	0.42
0.4	0.48	0.44	0.40	0.37	0.35
0.3	0.35	0.32	0.30	0.28	0.27
0.2	0.22	0.21	0.20	0.19	0.18
0.1	0.10	0.10	0.10	0.10	0.10

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration. Actual concentration $= 2.2 \times \text{Tmperature corrected value}$

n-HEXANE



1. PERFORMANCE

1) Measuring range : 50-1,400 ppm Number of pump strokes 1 (100ml)

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 5 ppm 4) Shelf life 2 years 5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation 7) Reading Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

CH₃ $(CH_2)_4CH_3 + Cr^{6+} + H_2SO_4 \cdot nSO_3 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Aromatic hydrocarbons FIG.1,2	Black stain is produced.	Higher readings are given.
Aliphatic hydrocarbons (more than C ₃)	Similar stain is produced.	"

In presence of Alcohols, Ketones or Esters less than 6 %, the accuracy of readings is not affected.

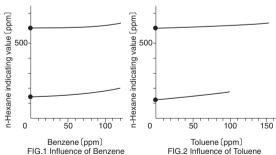
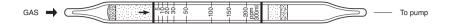


FIG.2 Influence of Toluene

Tube	Co	orrected (Concentra	tion (ppn	1)
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
1400	1630	1530	1400	1270	1180
1200	1400	1320	1200	1090	1010
1000	1170	1100	1000	910	840
800	930	870	800	720	670
600	700	660	600	550	500
400	460	430	400	360	330
200	220	210	200	180	170
100	100	100	100	100	100

n-HEXANE



1. PERFORMANCE

1) Measuring range Summer of pump strokes Sumpling time S-208 ppm S-200 ppm $1 (100 \text{m} \ell)$ $3 (300 \text{m} \ell)$ 6 minutes/3 pump strokes

3) Detectable limit $2 \text{ ppm}(300\text{m}\ell)$ 4) Shelf life 2 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (TEMPERATURE CORRECTION TABLE is provided.)
7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

CH₃ (CH₂)₄CH₃ + Cr⁶⁺ + H₂SO₄ · SO₄ \rightarrow Cr³⁺

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Methyl alcohol		500	Not affected.
Ethyl acetate		500	"
Methyl ethyl ketone		500	"
Toluene	Brown stain is produced.		Higher readings are given.

(NOTE)

In case of 1 pump stroke, following formula is available for the actual concentration.

Actual concentration = Temperature corrected value × 4

Tube		Corrected Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)	
200	250	225	200	185	170	
150	180	165	150	145	135	
100	120	110	100	95	85	
50	60	55	50	45	40	
30	35	30	30	30	25	
20	20	20	20	20	20	
10	10	10	10	10	10	
- 5	5	5	5	5	5	

HYDRAZINE



1. PERFORMANCE

2) Sampling time : 4 minutes/4 pump strokes

3) Detectable limit : 0.02 ppm (800mℓ)

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Yellow→Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $N_2H_4 + H_2SO_4 \rightarrow N_2H_4 \cdot H_2SO_4$

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.
Ammonia	"	"

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration.

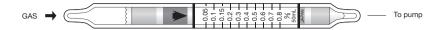
Actual concentration = Temperature corrected value \times 2

In case of 8 pump strokes, following formula is available for the actual concentration.

Actual concentration = Temperature corrected value \times 1/2

Tube		Corrected Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 ℃ (86 °F)	40 ℃ (104 °F)	
5.0	7.0	6.0	5.0	4.5	4.0	
4.0	5.6	4.7	4.0	3.6	3.2	
3.0	4.2	3.6	3.0	2.7	2.4	
2.0	2.8	2.4	2.0	1.8	1.6	
1.0	1.3	1.2	1.0	0.9	0.8	
0.5	0.5	0.5	0.5	0.45	0.4	
0.1	0.1	0.1	0.1	0.1	0.1	

HYDROGEN



1. PERFORMANCE

1) Measuring range 0.05-0.8 % Number of pump strokes $1/2(50 \text{m} \ell)$

2) Sampling time : 0.5minutes/1/2 pump stroke

3) Detectable limit 0.03%4) Shelf life 3 years 5) Operating temperature $0 \sim 40\%$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1/2 pump strokes 8) Colour change : Yellow → Blue (over 0.1 %) or Yellowish green (below 0.1 %)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Oxygen in Atomsphere, water vapour is produced. This Water vapour reacts with Magnesium perchlorate and PH indicator is discoloured.

 $H_2 + O_2 \rightarrow H_2O$

 $H_2O + Mg(CIO_4)_2 \rightarrow Mg(CIO_4)_2 \cdot H_2O$

4. CALIBRATION OF THE TUBE

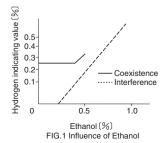
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance %		%	Interference		Coexistence
Ethanol	FIG.1	0.25	Similar stain is produced.	0.4%	Higher readings are given.
Carbon monoxide			The accuracy of readings is not affected.	500	Lower readings are given.

6. SPECIAL NOTE

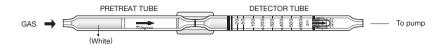
- The tube will not respond in the absence of oxygen.
- 2) When the concentration of Hydrogen is 12 to 16 %, pretreat reagent gives a heat but is not dangerous for use in hazardous area.
- 3) When the concentration of Hydrogen is over 40 %, the reading value may be indicated below 0.8 %. In this case, the bottom of the discoloured layer becomes dark purple. In order to make sure that the concentration is extremely high such as 40 %, measure the gas concentration with connecting 2 tubes. If both tubes are discoloured to green, extremely high concentration Hydrogen exists.



TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (%)									
Readings (%)	0°C (32°F)	5°C (41°F)	10 ℃ (50 °F)	15 °C (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (86 °F)	35 ℃ (95 °F)	40 °C (104 °F)	
0.8		_		- 1	0.80	0.68	0.58	0.51	0.45	
0.7		_		1.00	0.70	0.60	0.52	0.45	0.40	
0.6		_	1.00	0.80	0.60	0.52	0.44	0.39	0.35	
0.5		_	0.80	0.65	0.50	0.44	0.37	0.33	0.30	
0.4		_	0.62	0.51	0.40	0.35	0.30	0.27	0.25	
0.3		0.70	0.46	0.37	0.30	0.26	0.23	0.21	0.19	
0.2	0.65	0.47	0.30	0.25	0.20	0.18	0.16	0.14	0.13	
0.15	0.46	0.34	0.22	0.19	0.15	0.13	0.12	0.11	0.10	
0.1	0.28	0.21	0.15	0.12	0.10	0.09	0.08	0.08	0.07	
0.06	0.13	0.10	0.07	0.06	0.05	0.05	0.05	0.05	0.05	

HYDROGEN CHLORIDE



1. PERFORMANCE

1) Measuring range : 40-1,200 ppm 20-600 ppm Number of pump strokes $1/2(50 \text{m} \ell)$ 1(100ml) 2) Sampling time : 1.5 minutes/1 pump stroke

5 ppm (100mℓ) 3) Detectable limit 4) Shelf life 2 years 5) Operating temperature : 0 ~ 40 °C

6) Reading Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Purple → Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

PH indicator is discoloured by Hydrogen chloride.

4. CALIBRATION OF THE TUBE

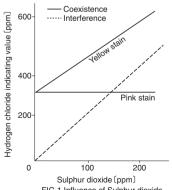
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence		
Sulphur dioxide FIG.1	Yellow stain is produced.		Yellow - Pink double - layer stains are produced and this pink stain indicates Hydrogen chloride concentration.		
Chlorine	"				
Hydrogen sulphide		500	The accuracy of readings is not affected.		

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration. Actual concentration = $2 \times$ Reading value



HYDROGEN CHLORIDE



1. PERFORMANCE

1) Measuring range : 4-40 ppm 2-20 ppm 0.4-4 ppm Number of pump strokes $1/2(50 \text{m}\ell)$ $1(100 \text{m}\ell)$ $5(500 \text{m}\ell)$

2) Sampling time : 1 minute/1 pump stroke
3) Detectable limit : 0.2 ppm (500mℓ)
4) Shelf life : 3 years
5) Operating temperature : 0 ~ 40 °C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Yellowish green→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

PH indicator is discoloured by Hydrogen chloride.

4. CALIBRATION OF THE TUBE

COLOURIMETRY METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide	200	Similar pale stain is produced but can be distinguished from clear discoloration by Hydrogen chloride.	The accuracy of readings is not affected.
Nitric acid	High conc.	"	"
Nitrogen dioxide	100	"	"
Chlorine	1	Similar stain is produced.	Higher readings are given.

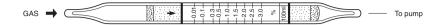
(NOTE)

In case of 1/2 and 5 pump strokes, the following formula is available for the actual concentration.

1/2 pump strokes; Actual concentration = Reading value $\times 2$

5 pump strokes; Actual concentration = Reading value $\times \frac{1}{5}$

HYDROGEN CYANIDE



1. PERFORMANCE

1) Measuring range 0.01-3.0%Number of pump strokes $1(100\text{m}\,\ell)$

2) Sampling time 2 minutes/1 pump stroke 3) Detectable limit 0.001% (10 ppm)

4) Shelf life : 3 years 5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Yellow → Brownish red

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Hydrogen cyanide reacts with Sodium picrate.

$$NaCN + O_{2}N \xrightarrow{ONa} NO_{2} \xrightarrow{O_{2}N} ONa \\ NO_{2} \xrightarrow{NO_{2}} NC \xrightarrow{ONa} NO_{2}$$

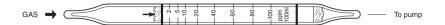
4. CALIBRATION OF THE TUBE

IODOMETRIC TITRATION METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Sulphur dioxide	Similar stain is produced.	200	Higher readings are given.
Hydrogen sulphide	"	100	"
Dicyan	"		"
Carbon disulphide	"		"
Acetone	"		"

HYDROGEN CYANIDE



1. PERFORMANCE

1) Measuring range 2-100 ppm 0.5-25 ppm Number of pump strokes $1(100\text{m}\ell)$ $4(400\text{m}\ell)$ 2) Sampling time 1 in minute/1 pump stroke 3) Detectable limit 2-100 ppm 0.5-25 ppm $1(100\text{m}\ell)$ 2-100 ppm $1(100\text{m}\ell)$ $1(1000\text{m}\ell)$ $1(100\text{m}\ell)$ $1(1000\text{m}\ell)$ $1(1000\text{m}\ell)$ $1(1000\text{m}\ell)$

4) Shelf life : 2 years (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Yellow→Red

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercury chloride, Hydrogen chloride is liberated and PH indicator is discoloured. HCN + HgCl₂→HCl

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

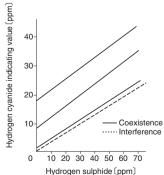
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Sulphur dioxide		Similar stain is produced.	1	Higher readings are given.
Phosphine		"	1	"
Hydrogen sulphide	FIG.1	"	3	"
Ammonia		The accuracy of readings is not affected.	5	Lower readings are given.

(NOTE)

In case of 4 pump strokes, following formula is available for the actual concentration.

Actual concentration = Reading value $\times \frac{1}{4}$



Hydrogen sulphide [ppm] FIG.1 Influence of Hydrogen sulphide

HUMIDITY CORRECTION TABLE

Tube Readings	Corrected Concentration (ppm)						
(ppm)	10%R.H.	30%R.H.	50%R.H.	70%R.H.	90%R.H.		
100	91.0	95.0	100.0	105.0	111.0		
80	73.0	76.0	80.0	84.0	88.5		
60	54.5	57.0	60.0	63.0	66.0		
40	36.0	38.0	40.0	42.0	44.5		
20	18.0	19.0	20.0	21.0	22.5		
10	8.4	9.2	10.0	10.8	11.6		
5	4.2	4.6	5.0	5.4	5.8		

HYDROGEN CYANIDE



1. PERFORMANCE

1) Measuring range : 0.3-8 ppm Number of pump strokes $3(300m\ell)$

2) Sampling time : 4.5 minute s/3 pump strokes

3) Detectable limit : 0.15 ppm (300mℓ)

4) Shelf life 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : 0 ~ 40 °C

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation: Direct reading from the scale calibrated by 3 pump strokes 7) Reading : Yellow → Red

8) Colour change

2. RELATIVE STANDARD DEVIATION RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Hydrogen cyanide reacts with Mercuric chloride and liberates hydrogen chloride, then PH indicator is discoloured. $HCN + HgCl_2 \rightarrow HCl$

4. CALIBRATION OF THE TUBE

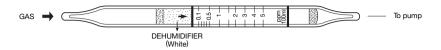
ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Sulphur dioxide	Red stain is produced.	1.0	Higher readings are given.
Hydrogen sulphide	"		"
Acrylonitrile	"		"
Phosphine	"		"
Ammonia		2.0	Lower readings are given.

Tube	corrected Concentration (ppm)						
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)		
8	10.0	8.8	8.0	7.5	7.0		
7	8.3	7.6	7.0	6.6	6.2		
6	6.9	6.4	6.0	5.6	5.4		
5	5.8	5.2	5.0	4.8	4.5		
4	4.5	4.2	4.0	3.8	3.7		
3	3.4	3.2	3.0	2.9	2.8		
2	22	2.1	2.0	1.9	1.8		
1	1.1	1.1	1.0	0.9	0.9		
0.6	0.7	0.7	0.6	0.5	0.5		
0.3	0.3	0.3	0.3	0.3	0.3		

HYDROGEN SELENIDE



1. PERFORMANCE

 $\begin{array}{cccc} \text{1) Measuring range} & \text{1} \cdot 1\text{-20 ppm} & 0.5\text{-10 ppm} \\ \text{Number of pump strokes} & 1 \left(100\text{m}\ell\right) & 2 \left(200\text{m}\ell\right) \\ \text{2) Sampling time} & \text{1} \cdot 1 \text{ minute/1 pump stroke} \\ \end{array}$

3) Detectable limit :—
4) Shelf life : 2 years
5) Operating temperature : $15 \sim 20 \, ^{\circ}$ C

6) Reading : The tube scale is calibrated based on Diborane at 1 pump stroke and Hydrogen selenide concentration is determined by using a conversion chart at 1 and 2

pump strokes

7) Colour change : Pale yellow→Reddish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Mercuric cholride, Hydrogen chloride is liberated and PH indicator is discoloured. $H_2Se + H_3CI_2 \rightarrow 2HCI$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Arsine	Similar stain is produced.	Higher readings are given.
Phosphine	"	"
Monosilane	The maximum end point of stained layer is indiscernible.	"
Disilane	"	"
Monogermane	The accuracy of readings is not affected.	The accuracy of readings is not affected.

Hydrogen selenide concentration (ppm)



No.242S Tube reading (ppm)

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration.

Actual concentration = Converted value \times 1/2

HYDROGEN FLUORIDE



1. PERFORMANCE

1) Measuring range : 0.5-30 ppm 0.25-15 ppm 0.17-2 ppm Number of pump strokes 3 $(300 \text{m}\ell)$ 6 $(600 \text{m}\ell)$ 9 $(900 \text{m}\ell)$

2) Sampling time 1 minute/1 pump stroke 3) Detectable limit 2.0.05 ppm (900mℓ) 4) Shelf life 3 years 5) Operating temperature 0 ~ 40 °C

6) Temperature and humidity

compensation : Necessary (See "TEMP. /R. H. CORRECTION COEFFICIENT TABLE")

7) Reading Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Greenish yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

PH indicator is discoloured by Hydrogen fluoride.

4. CALIBRATION OF THE TUBE

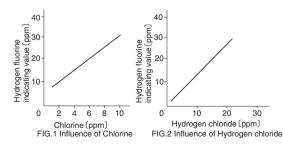
PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence
Chlorine	FIG.1	Similar stain is produced.	Higher readings are given.
Hydrogen chloride	FIG.2	"	"

(NOTE)

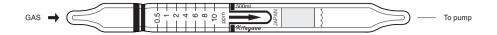
- 1) This detector tube is affected by ambient temperature and humidity, therefore, it is necessary to compensate the reading of gas detector tube with the following equation and correction coefficient table. Actual concentration = Reading Value × Correction Coefficient
- In case of 6 pump strokes, following formula is available for the actual concentration. Actual concentration = 1/2 × Temp./R.H. corrected value
- 3) In case of 9 pump strokes, following formula is available for the actual concentration. Actual concentration = 1/3 × Temp./R.H. corrected value



TEMP./R.H.CORRECTION COEFFICIENT TABLE

Relative	Temperature						
Humidity (%)	0°C (32°F)	10 °C (50 °F)	20°C (68°F)	30°C (86°F)	40°C (104°F)		
30	0.9	0.7	0.6	0.55	0.5		
40	1.6	1.0	0.8	0.65	0.6		
50	2.6	1.3	1.0	0.8	0.7		
60	_	2.2	1.5	1.1	8.0		
70	_	_	2.7	1.6	0.9		
80	_	_	_	2.9	1.1		

HYDROGEN PEROXIDE



1. PERFORMANCE

1) Measuring range 0.5-10 ppmNumber of pump strokes $5(500\text{m}\ell)$

2) Sampling time : 7.5 minutes/5 pump strokes

3) Detectable limit : 0.2 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10 \,^{\circ}\text{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "NOTE")

7) Reading : Direct reading from the scale calibrated by 5 pump strokes

8) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Titanium sulphate, yellow complex is generated. H₂O₂ + T₁ (SO₄)₂ → Yellow compound

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence		
Chlorine	The accuracy of reading is not affected.		The accuracy of reading is not affected.		
Ozone	"		"		
Nitrogen dioxide	"		"		
Acetaldehyde	"		"		
Formaldehyde	"	10	Lower readings are given.		

(NOTE)

The scale is calibrat based on the temperature of 20 $^{\circ}$ C (68 $^{\circ}$ F). Readings obtained in other temperature circumstances should be corrected with the following temperature correction coefficient table.

TEMPERATURE CORRECTION COEFFICIENT TABLE (AT 20 °C)

Temperature (°C)	0	1	2	3	4	5	6	7	8	9
Correction Factor	1.35	1.32	1.28	1.25	1.23	1.20	1.17	1.15	1.13	1.11
Temperature (°C)	10	11	12	13	14	15				
Correction Factor	1.09	1.07	1.06	1.05	1.03	1.02				

Actual concentration = Reading value \times Coefficient for temperature correction.

HYDROGEN SELENIDE



1. PERFORMANCE

1) Measuring range Sumbler of pump strokes Sampling time S-600 ppm $1 \cdot 120 \text{ ppm}$ $1 \cdot 120 \text{ ppm}$ $1 \cdot 100 \text{m} \ell)$ $5 \cdot (500 \text{m} \ell)$ $1 \cdot 1.5 \text{ minutes}/1 \text{ pump stroke}$

3) Detectable limit $\begin{array}{c} \text{3.5 ppm} \left(500\text{m}\ell\right) \\ \text{4) Shelf life} \\ \text{5) Operating temperature} \\ \end{array}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Gold chloride, Colloidal gold is liberated. $H_2Se + AuCI_3 \rightarrow Au$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence				
Acetylene	3%	Higher readings are given.	Higher readings are given.				
Carbon monoxide	0.1%	"	"				
Sulphur dioxide		Pale blue stain is produced.	"				
Hydrogen sulphide	10	Brown stain is produced.	"				
Mercury vapours		Similar stain is produced.	"				
Arsine	10	"	"				
Iron carbonyl	10	"	"				
Nickel carbonyl	10	"	"				

(NOTE)

When the concentration is below 5 ppm, 5 pump strokes can be used to determine the lower concentration with the following formula.

Actual concentration = $1/5 \times$ Reading value



1. PERFORMANCE

1) Measuring range : 0.025-0.2 gr/100 cf 0.05-0.4 gr/100 cfNumber of pump strokes : $1(100 \text{m} \ell)$ $1/2 (50 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke
3) Detectable limit : 0.005 gr/100 cf
4) Shelf life : 2 years
5) Operating temperature : 0 ~ 40 °C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reaction with Mercury chloride (II) , Hydrogen chloride is produced and PH indicator is discoloured. $H_2S + H_3Cl_2 \rightarrow H_3$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

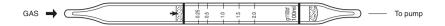
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Sulphur dioxide	Whole reagent is changed to Pale red, but Purplish red stain is indicated H ₂ S concentration.	Higher readings are given.
Hydrogen selenide	Similar stain is produced.	"
Arsine	"	"
Mercaptans	"	"
Phosphine	"	"
Hydrogen cyanide	Whole reagent is changed to red.	"

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times$ Reading value



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \hbox{: } 0.25\text{-2 gr}/100 \text{ cf} \\ \hbox{Number of pump strokes} \end{array} \begin{array}{c} \hbox{: } 0.25\text{-2 gr}/100 \text{ cf} \\ \hbox{1}/2 (50\text{m}\,\ell) \end{array}$

2) Sampling time : 1 minute/1 pump stroke
3) Detectable limit : 0.05 gr/100 cf
4) Shelf life : 3 years
5) Operating temperature : 0 ~ 40 °C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Pale brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $H_2S + Pb(CH_3COO)_2 \rightarrow PbS + 2CH_3COOH$

4. CALIBRATION OF THE TUBE

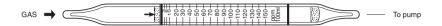
PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Sulphur dioxide	The accuracy of readings is not affected.	Higher readings are given.	
Mercaptans	"	"	
Nitrogen dioxide	"	"	

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration. Actual concentration $= 2 \times \text{Reading value}$



1. PERFORMANCE

1) Measuring range : 3-150 ppm 1-50 ppm 0.75-37.5 ppm 6-300 ppm Number of pump strokes 1(100ml) 3(300ml) 4 (400ml) $1/2(50 \text{m} \ell)$

2) Sampling time 1 minute/1 pump stroke 3) Detectable limit $0.3 \text{ ppm} (300 \text{m} \ell)$

4) Shelf life 3 years 5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading Direct reading from the scale calibrated by 1 pump stroke

7) Colour change White→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Lead acetate (II), Lead sulphide is produced. $H_2S + Pb(CH_3CO_2)_2 \rightarrow PbS + 2CH_3CO_2H$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference ppm		Coexistence	
Sulphur dioxide	FIG.1	The accuracy of readings is not affected.	12	Higher readings are given.	
Mercaptans	FIG.2	"	550	"	
Nitrogen dioxide	FIG.3	"	2	Lower readings are given.	

(NOTE)

In case of 1/2, 3 or 4 pump strokes, following formula is available for the actual concentration.

Actual concentration = Reading value $\times \frac{1}{\text{Number of pump strokes}}$

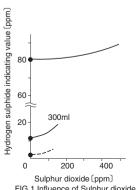


FIG.1 Influence of Sulphur dioxide

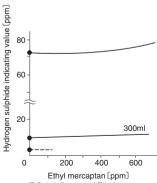
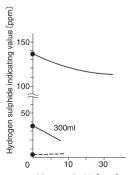


FIG.2 Influence of Ethyl mercaptan



Nitrogen dioxide (ppm) FIG.3 Influence of Nitrogen dioxide



1. PERFORMANCE

1) Measuring range 50-1,600 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit \therefore 20 ppm 4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow → Dark blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Palladium sulphate and Ammonium molybdate, Palladium sulphate is produced. H₂S + PdSO₄ + (NH₄)₂MoO₄ → PdS

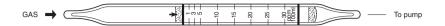
4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Carbon monoxide		10	Whole reagent is changed to Blue and higher readings are given.
Ethylene		5	Similar stains are produced and higher readings are given.
Propylene		5	"
Butylene		5	"
Acetylene		5	"
Methyl acetylene		5	"
Hydrogen cyanide			White stain is produced and the discolouration by Hydrogen sulphide is interfered.
Ammonia			"
Sulphur dioxide		less than 6%	The accuracy of readings is not affected.

Tube	Corrected Concentration (%)					
Readings	.0℃	10 °C	20 °C	30°C	40°C	
(%)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)	
1600	1400	1500	1600	1700	1700	
1400	1300	1300	1400	1500	1500	
1200	1100	1200	1200	1300	1300	
1000	900	1000	1000	1000	1100	
800	700	800	800	800	900	
600	500	600	600	600	700	
400	400	400	400	400	400	
300	300	300	300	300	300	
200	200	200	200	200	200	



1. PERFORMANCE

1) Measuring range Sumbler of pump strokes 1.2 (50 m ℓ) 1 (100 m ℓ) 2) Sampling time 1.3 Detectable limit 1.3 Detectable limit 1.3 Dependence 1.3 Depende

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \, ^{\circ}\mathrm{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Pale brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Lead acetate (II), Lead sulphide is produced. H₂S + Pb (CH₃CO₂)₂→PbS + 2CH₃CO₂H

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

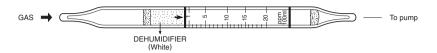
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Sulphur dioxide	FIG.1	The accuracy of readings is not affected.	10	Higher readings are given.
Mercaptans	FIG.2	"	300	"
Nitrogen dioxide	FIG.3	"	2	Lower readings are given.

(NOTE)

When the concentration is over 30ppm, 1/2 pump strokes can be used to determine higher concentration with the following formula;

Actual concentration = $2 \times$ Reading value



1. PERFORMANCE

1) Measuring range : 2-40 ppm 1-20 ppm 0.5-10 ppm Number of pump strokes 1/2(50ml) 1(100ml) $2(200m\ell)$

2) Sampling time 1 minute/1 pump stroke 3) Detectable limit $0.2 \text{ ppm} (200 \text{m} \ell)$

4) Shelf life 2 years 5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading Direct reading from the scale calibrated by 1 pump stroke

7) Colour change Yellow → Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with mercuric chloride, Hydrogen chloride is produced and PH indicator is discoloured. $H_2S + H_gCI_2 \rightarrow H_gS + 2HCI$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

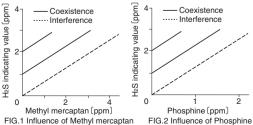
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Phosphine	Similar stain is produced.	Higher readings are given.
Mercaptans	"	"
Arsine	"	Higher readings are given.
Hydrogen selenide	"	"
Hydrogen cyanide	"	"
Nitrogen dioxide	The accuracy of readings is not affected.	Lower readings are given.
Ammonia	Pale brown stain is produced.	"
Sulphur dioxide	"	If the maximum end point of the pink stain is discernable, the accuracy of readings is not affected.

(NOTE)

In case of 1/2 and 2 pump strokes, the following equation is available for the actual concentration.

1/2 pump strokes: Actual concentration = Reading value $\times 2$ 2 pump strokes: Actual concentration = Reading value \div 2





1. PERFORMANCE

1) Measuring range : 100-2,000 ppm 50-1,000 ppm 25-500 ppm Number of pump strokes $1/2(50\text{m}\ell)$ $1(100\text{m}\ell)$ $2(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke 1 minute/1/2 pump strokes

3) Detectable limit $0.5 \text{ ppm} (100 \text{m} \ell)$ 4) Shelf life 3 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Black

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $H_2S + 2AgNO_3 \rightarrow Ag_2S$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Sulphur dioxide	The accuracy of readings is not affected.	5000	Lower readings are given.
Mercaptans	Pale yellow stain is produced.	50	The mixmum end point of the stain is indiscernible, and higher readings are given.

(NOTE)

In case of 1/2 and 2 pump strokes, following formula is available for the actual concentration.

1/2 pump strokes : Actual concentration = $2 \times$ Reading value 2 pump strokes : Actual concentration = $0.5 \times$ Reading value



1. PERFORMANCE

1) Measuring range 0.1-4.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time ∴ 1 minute/1 pump stroke 3) Detectable limit ∴ 0.005 % (50mℓ)

4) Shelf life 3 years $0 \sim 40 \, \text{C}$

6) Reading Direct reading from the scale calibrated by 1 pump stroke

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

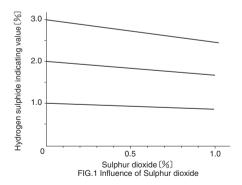
By reacting with Cupric sulphate (II), Cupric sulphide is produced. $H_2S + CuSO_4 \rightarrow CuS$

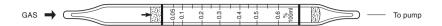
4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence
Sulphur dioxide FIG.1	The accuracy of readings is not affected.	0.5	Lower readings are given.





1. PERFORMANCE

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Cupric sulphate (II), Cupric sulphide is produced. H2S + CuSO₄ \rightarrow CuS

4. CALIBRATION OF THE TUBE

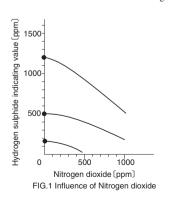
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	%	Coexistence
Sulphur dioxide	FIG.1	The accuracy of readings is not affected.	0.3	Higher readings are given.
Ammonia		Blue stain is produced.		The accuracy of readings is not affected.
Methyl mercaptan		Pale yellow stain is produced.		"

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration. Actual concentration $= 2 \times \text{Reading value}$



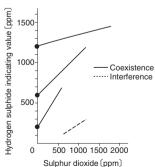


FIG.2 Influence of Sulphur dioxide



1. PERFORMANCE

1) Measuring range Number of pump strokes 2) Sampling time 1.2 ($50m\ell$) $1 (100m\ell)$ $1 (100m\ell)$ 3) Detectable limit 1 minute/1 pump stroke 1 minute/1 pump stroke $1 0.05 ppm (<math>100m\ell$)

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercury chloride (II), Hydrogen chloride is produced and PH indicator is discoloured. H2S + HgCI2→HgS + HCI

4. CALIBRATION OF THE TUBE

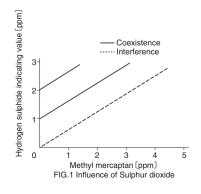
PERMEATION TUBE METHOD

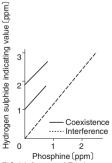
5. INTERFERENCE AND CROSS SENSITIVITY

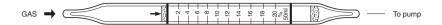
Substance	Interference	Coexistence
Sulphur dioxide FIG.1	Whole reagent is changed to Pale red, but Purplish red stain is indicated H ₂ S concentration.	
Hydrogen selenide	Similar stain is produced.	Higher readings are given.
Arsine	"	"
Mercaptans FIG.2	"	"
Phosphine	"	"
Hydrogen cyanide	Whole reagent is changed to Red.	"

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration. Actual concentration $= 2 \times \text{Reading value}$







1. PERFORMANCE

1) Measuring range \therefore 2-20 % Number of pump strokes $1/2 (50 \text{m} \ell)$

2) Sampling time : 1 minute/1/2 pump strokes

3) Detectable limit 0.05%4) Shelf life 3 years 5) Operating temperature $0 \sim 40\%$

6) Reading : Direct reading from the scale calibrated by 1/2 pump stroke

7) Colour change : Pale blue → Black

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Cupric sulphate (II), Cupric sulphide is produced. $H_2S + CuSO_4 \rightarrow CuS$

4. CALIBRATION OF THE TUBE

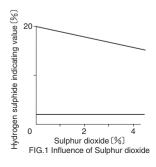
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	%	Coexistence
Sulphur dioxide FIG.1		The accuracy of readings is not affected.	2	Lower readings are given.
Carbon dioxide		"		The accuracy of readings is not affected.

(NOTE)

- 1) Where the discolouration is over the scale, non-reacted Hydrogen sulphide remains in the pump. This residual Hydrogen sulphide is not exhausted even if the pump handle is pushed back, and it still remains in the pump. When the handle is pulled next, it is exhausted toward the operator from the stopper side of the pump. Accordingly, air purging of the pump inside by pulling/pushing the handle several times should be done in fresh air where there is not Hydrogen sulphide, because Hydrogen sulphide is very toxic. During the air purging, protection apparatus such as a gas mask should be worn.
- 2) Maximum sampling volume of the gas shall be approximately $70m\ell$.





1. PERFORMANCE

1) Measuring range Summber of pump strokes Sampling time S-40 % 2.5-5 % $1/2(50m\ell)$ 1 $(100m\ell)$ 2) Sampling time 1.1 minute/1/2 pump strokes

3) Shelf life \therefore 3 years \therefore 0 \sim 40 $^{\circ}$ C

5) Reading : Direct reading from the scale calibrated by 1/2 pump stroke

6) Colour change : Pale blue → Black

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Copper sulphate (II), Cupric sulphide is produced. $H_2S + CuSO_4 \rightarrow CuS$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

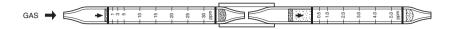
Substance	Interference	%	Coexistence
Sulphur dioxide	The accuracy of readings is not affected.	8	Lower readings are given.
Carbon dioxide	"		The accuracy of readings is not affected.

(NOTE)

In case of 1 pump stroke, following formula is available for the actual concentration.

Actual concentration = $2 \times \text{Reading value}$

HYDROGEN SULPHIDE · MERCAPTANS



1. PERFORMANCE

1) Measuring range:

Hydrogen sulphide 1-30 ppm Mercaptans 0.5-5 ppm Number of pump strokes $1(100 \text{m}\ell)$

2) Sampling time : 3 minutes/1 pump stroke

3) Detectable limit

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change:

Hydrogen sulphide White→Dark brown Mercaptans Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10 % RSD-mid.: 10 % RSD-high: 10 %

3. CHEMICAL REACTION

Hydrogen sulphide detector tube : H₂S + Pb (CH₃CO₂)₂→PbS + 2CH₃CO₂H

Mercaptans detector tube : $R \cdot SH + HgCI_2 \rightarrow RS (HgCI) + HCI$

4. CALIBRATION OF THE TUBE

Hydrogen sulphide STANDARD GAS CYLINDER METHOD Mercaptans STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
H ₂ S detector tube			
Sulphur dioxide	The accuracy of readings is not affected.	H ₂ S conc. × 1/3	The maximum end point of the stain becomes indiscernable and higher readings are given.
Nitrogen dioxide	"	H ₂ S conc. × 1/5	Lower readings are given.
R·SH detector tube			
Nitrogen dioxide	The accuracy of readings is not affected.	2	Lower readings are given.
Ammonia	"	R·SH conc. ×10	The stain from the gas inlet side is faded and lower readings are given.
Hydrogen sulphide		30	The maximum end point of the stain becomes indiscernable.

ISOBUTANE



1. PERFORMANCE

1) Measuring range 50-1,200 ppm Number of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 10 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by n-Hexane at 1 pump stroke and Isobutane concentration is determined by using a conversion chart.

7) Colour change : Orange → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

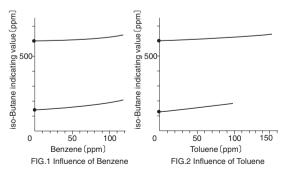
Chromium oxide is reduced $(CH_3)_3CH = Cr^{6+} \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

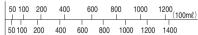
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Alcohols	Similar stain is produced.	6%	Higher readings are given.
Ketones	"	"	"
Esters	"	"	"
Aromatic hydrocarbons FIG.1.2			The top of the discoloured layer is stained to Black and higher readings are given.
Aliphatic hydrocarbons (more than C ₃)	Similar stain is produced.		Higher readings are given.



Isobutane concentration (ppm)



No.113SB Tube reading (ppm)

ISOBUTYL ACETATE



1. PERFORMANCE

1) Measuring range 0.01-1.4%Number of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit 0.001%(10 ppm)4) Shelf life 3 years5) Operating temperature $0 \sim 40\%$

6) Reading Graduations printed on the tube are calibrated by Methyl ethyl ketone at 2 pump strokes and Isobutyl acetate concentration is determined by using a conversion chart.

7) Colour change : Orange → Pale brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Dichromate is reduced

 $CH_3CO_2CH_2CH(CH_3)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

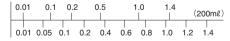
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is changed to Pale brown.
Propane		0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	Higher readings are given.

Isobutyl acetate (%)



No.139SB Tube reading (%)

ISOBUTYL ACETATE



1. PERFORMANCE

1) Measuring range 10-400 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 3 ppm 4) Shelf life : 1 year 5) Operating temperature : $10 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale yellow→Pale blue (The top of discoloured layer is Brown, but read at the

top of Pale blue.)

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 5 % RSD-high : 5 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

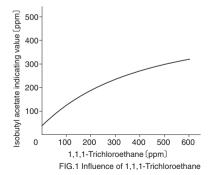
 $CH_3CO_2CH_2CH(CH_3)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence
Esters		Similar stain is produced.	Higher readings are given.
Alcohols		"	"
Ketones		"	"
Aromatic hydrocarbons	FIG.1	"	"



Tube		Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (86 °F)	35 °C (95 °F)	40 °C (104 °F)
400	-	-	400	285	240	210	190
300	-	-	300	225	190	170	160
200	-	275	200	165	140	130	120
150	_	195	150	125	110	103	95
100	_	120	100	90	80	75	70
50	70	55	50	45	40	40	35
30	40	35	30	25	20	20	20
20	25	20	20	20	15	15	15
10	15	10	10	10	5	5	5

ISOBUTYL ACRYLATE



1. PERFORMANCE

1) Measuring range 5-60 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit : 0.5 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

 Reading
 Graduations printed on the tube are calibrated by Methyl acrylate at 2 pump strokes and Isobutyl acrylate concentration is determined by using a conversion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

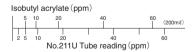
 $CH_2 = CHCO_2CH_2CH(CH_3)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

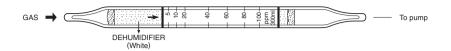
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Esters	Whole reagent is discolured to Dark brown	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons (more than C ₃)	"	"
Halogenated hydrocarbons	"	"



Conversion	С	Corrected Concentration (ppm)				
Value (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 ℃ (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)	
60	_	90	60	48	40	
40	115	57	40	32	27	
20	50	27	20	16	14	
10	20	13	10	8	6	
5	10	6	5	4	3	

ISOBUTYL ALCOHOL



1. PERFORMANCE

1) Measuring range 5-100 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

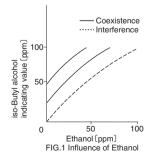
 $(CH_3)_2CHCH_2OH + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

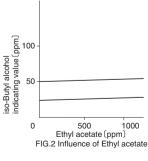
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance ppm		ppm	Interference	Coexistence
Alcohols	FIG.1		Similar stain is produced.	Higher readings are given.
Toluene		200	Whole reagent is changed to Pale blue.	"
Hexane		1,000	The accuracy of readings is not affected.	The accuracy of readings is not affected.
Trichloroethylene 1,000		1,000	"	"
Ethyl acetate	FIG.2	1,000	"	"





Tube	С	Corrected Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)	
100	_	_	100	85	80	
80	_	100	80	75	60	
60	95	70	60	53	50	
40	58	47	40	36	34	
20	27	22	20	18	17	
10	13	11	10	9	8	
5	5	5	5	5	5	

ISOBUTYLENE



1. PERFORMANCE

1) Measuring range 0.03-2.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : —
4) Shelf life : 2 years
5) Scale calibration temperature : 20 °C

6) Reading : Graduations printed on the tube are calibrated by n-Hexane at 1 pump stroke

and Isobutylene concentration is determined by using a conversion chart at

1 pump stroke.

7) Colour change : Orange → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 5 %

3. CHEMICAL REACTION

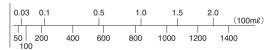
Chromium oxide is reduced

CH₃ (CH₂)₅CH₃ + Cr⁶⁺ + H₂SO₄ \rightarrow Cr³⁺

4. CALIBRATION OF THE TUBE

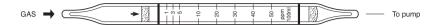
GAS CHROMATOGRAPHY

Isobutylene concentration (ppm)



No.113SB Tube reading (ppm)

ISOBUTYRIC ACID



1. PERFORMANCE

1) Measuring range 3-50 ppm Number of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Isobutyric acid concentration is determined by using a conversion chart at 1 pump stroke.

7) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

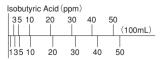
By reacting with alkali, PH indicator is discoloured. CH₃CH₂CH₂COOH + NaOH → C₃H₇COONa + H₂O

4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

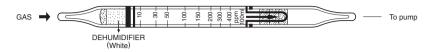
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Sulphur dioxide		Similar stain is produced.	HCO₂H conc. × 1/20	Higher readings are given.
Nitrogen dioxide	300	"	10	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	HCO₂H conc. ×2	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	5	"
Acetic acid		Similar stain is produced.		"



No.216S tube reading (ppm)

ISOPENTYL ACETATE



1. PERFORMANCE

1) Measuring range : 10-400 ppm Number of pump strokes : $1(100m\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \therefore 4 ppm 4) Shelf life \therefore 1 year 5) Operating temperature \therefore 10 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale yellow Pale blue (The top of discoloured layer is Brown, but read at the

top of Pale blue.)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Potassium dichromate is reduced

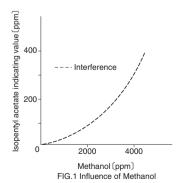
 $CH_3CO_2 (CH_2)_2CH (CH_3)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

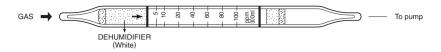
Substance		Interference	Coexistence
Esters		Similar stain is produced.	Higher readings are given.
Alcohols	FIG.1	"	"
Ketones		"	"
Aromatic hydrocarbon	S	Whole reagent is changed to Pale brown.	"



TEMPERATURE CORRECTION TABLE

Tube		Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 ℃ (68 °F)	25℃ (77°F)	30 °C (86 °F)	35 ℃ (95 °F)	40 °C (104 °F)
400			400	240	190	160	140
300		_	300	200	160	140	120
200		360	200	150	125	110	95
150	_	230	150	120	100	90	80
100	_	135	100	85	75	67	60
50	80	60	50	45	40	35	35
30	40	35	30	30	25	25	25
10	10	10	10	10	10	10	10

ISOPENTYL ALCOHOL



1. PERFORMANCE

1) Measuring range 5-100 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

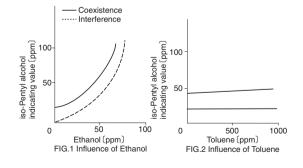
Chromium oxide is reduced. $(CH_3)_2CH(CH_2)_2OH + Cr^{6\,+} + H_2SO_4 {\longrightarrow} Cr^{3\,+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

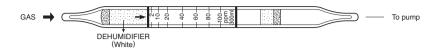
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		ppm	Interference	Coexistence
Alcohols	Alcohols FIG.1 Similar stain is produced.		Higher readings are given.	
Toluene FIG.2		200	Whole reagent is changed to Pale blue.	"
Hexane		1,000		"
Trichloroethyler	ne	11		"
Ethyl acetate		"		"



Tube	Corrected Concentration (ppm)						
Readings (ppm)	0 °C (32 °F)	10 ℃ (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)		
100			100	80	70		
80			80	65	55		
60		83	60	50	45		
40	90	50	40	33	30		
20	32	23	20	16	15		
10	14	11	10	9	8		
5	5	5	5	5	5		

ISOPHORONE



1. PERFORMANCE

7) Colour change

1) Measuring range 5-80 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : The tube scale is calibrated based on Cyclohexanone at 3 pump strokes and

Isophorone concentration is determined by using a conversion chart

at 3 pump strokes ∴ Yellow → Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $C_9H_{14}O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

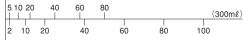
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

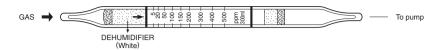
Substance	Interference	Coexistence	
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to pale brown.	The accuracy of readings is not affected if the maximum end point of the blue stain is discernable.	
Aromatic hydrocarbons	"	"	
Halogenated hydrocarbons	"	"	
Alcohols	Similar stain is produced.	Higher readings are given.	
Esters	Pale brown stain is produced from the zero end of the detecting reagent (inlet side of the tube).	The accuracy of readings is not affected.	

Isophorone concentration (ppm)



No. 197U tube reading (ppm)

ISOPRENE



1. PERFORMANCE

1) Measuring range 1-16 ppmNumber of pump strokes $3(300\text{m}\ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit : 0.5 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Graduations printed on the tube are calibrated by Ethyl cellosolve at 3 pump strokes

and Isoprene concentration is determined by using a conversion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH = C(CH_3)CH = CH_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Pale brown.	"
Aromatic hydrocarbons	"	"
Esters	"	"
Ketones	"	"
Halogenated hydrocarbons	"	"

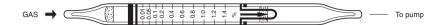
Isoprene (ppm)



No.190U Tube reading (ppm)

Conversion	Corrected Concentration (ppm)					
Value (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 ℃ (86 °F)	40 °C (104 °F)	
16	20.5	18.0	16.0	14.5	13.0	
14	18.0	15.5	14.0	12.5	11.0	
12	15.5	13.5	12.0	10.5	9.5	
10	12.5	11.0	10.0	9.0	8.0	
8	10.0	9.0	8.0	7.5	6.5	
6	7.5	6.5	6.0	6.0	5.0	
4	5.0	4.0	4.0	4.0	3.5	
2	2.0	2.0	2.0	2.0	2.0	
1	1.0	1.0	1.0	1.0	1.0	

ISOPROPYL ACETATE



1. PERFORMANCE

1) Measuring range 0.01-1.2 % Number of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit : 10 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Methyl ethyl ketone at 2 pump strokes and Isopropyl acetate concentration is determined by using a conversion chart.

7) Colour change : Orange→Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Dichromate is reduced $CH_3CO_2CH(CH_3)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is changed to brown.
Propane		0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	Higher readings are given.

Isopropyl acetate (%)



No.139SB Tube reading (%)

ISOPROPYL ACETATE



1. PERFORMANCE

1) Measuring range : 10-1,000 ppm Number of pump strokes : $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 2 years5) Operating temperature $: 10 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION COEFFICIENT TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3CO_2C_2H_5 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	It the top of brown stain is clear, the accuracy of readings is not affected.
Halogenated hydrocarbons FIG.1	"	"

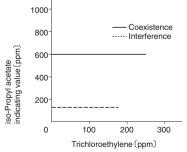


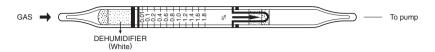
FIG.1 Influence of Trichloroethylene

TEMPERATURE CORRECTION COEFFICIENT TABLE

-tl--- 0 - -#I-I--- (-t 00 °C)

- 1	rube	Correction Coefficient (at 20 C)						
	Readings (ppm)	10 ℃ (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 ℃ (86 °F)	35℃ (95°F)	40 °C (104 °F)
Γ	1,000	1.33	1.17	1.00	0.87	0.74	0.64	0.53
Γ	800	1.38	1.19	1.00	0.86	0.73	0.63	0.53
Γ	600	1.40	1.20	1.00	0.86	0.72	0.63	0.53
Γ	400	1.40	1.20	1.00	0.85	0.70	0.58	0.46
Γ	200	1.40	1.20	1.00	0.84	0.68	0.55	0.42
Γ	100	1.50	1.25	1.00	0.81	0.62	0.48	0.33
Γ	50	1.50	1.25	1.00	0.77	0.54	0.43	0.32
Γ	30	1.50	1.25	1.00	0.77	0.53	0.42	0.30
Г	10	1.50	1.25	1.00	0.75	0.50	0.40	0.30

ISOPROPYL ALCOHOL



1. PERFORMANCE

1) Measuring range 0.05-2.5%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 100 ppm 4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \, ^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Graduations printed on the tube are calibrated by Ethylene oxide at 1 pump stroke

and Isopropyl alcohol is determined by using a conversion chart.

8) Colour change : Orange→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3CH(OH)CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

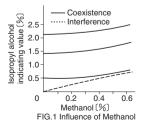
GAS CHROMATOGRAPHY

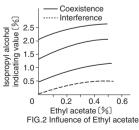
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference %		Coexistence	
Alcohols FIG.1		Similar stain is produced.		Higher readings are given.	
Esters	FIG.2	"		"	
Ketones		"		"	
Aromatic hydrocarbo	ons	"		"	
Aliphatic hydrocarbons		Whole reagent is discoloured to Pale brown.	0.5	"	

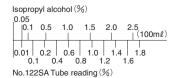
(NOTE)

Methanol and Ethyl acetate have the same sensitivity as Isopropyl alcohol. Methyl ethyl ketone has 3/4 sensitivity of Isopropyl alcohol.

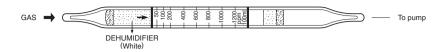




Conversion	Corrected Concentration (ppm)					
Value	0℃	0°C 10°C 20°C 30°C				
(%)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)	
2.5			2.50	2.10	2.00	
2.0		_	2.00	1.70	1.62	
1.5	_	_	1.50	1.28	1.21	
1.0	_	1.43	1.00	0.85	0.80	
0.5	1.00	0.62	0.50	0.42	0.38	
0.1	0.16	0.14	0.10	0.09	0.08	
0.05	0.09	0.06	0.05	0.04	0.03	



ISOPROPYL ALCOHOL



1. PERFORMANCE

1) Measuring range 50-1,200 ppm 20-480 ppm Number of pump strokes $1(100\text{m}\ell)$ $2(200\text{m}\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit $: 5 \text{ ppm} (200 \text{m} \ell)$ 4) Shelf life : 2 years5) Operating temperature $: 10 \sim 40 \, ^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke.

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3CH(OH)CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

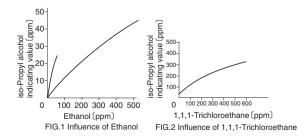
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

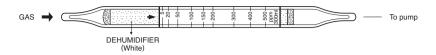
Substance	Interference	Coexistence	
Alcohols FIG	1 Similar stain is produced.	Higher readings are given.	
Ethers	"	"	
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is discoloured to Pale brown.	"	
Aromatic hydrocarbons	"	"	
Esters	"	"	
Ketones	"	"	
Halogenated hydrocarbons FIG	2	"	

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration. Actual concentration $= 2/5 \times \text{Reading value}$



ISOPROPYL CELLOSOLVE



1. PERFORMANCE

7) Colour change

1) Measuring range ∴ 5-350 ppm Number of pump strokes 3 (300m Ø)

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : The tube scale is calibrated based on Ethyl cellosolve at 3 pump strokes and

Isopropyl cellosolve concentration is determined by using a conversion chart

at 3 pump strokes ∴ Yellow → Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10 % RSD-mid.: 10 % RSD-high: 10 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $(CH_3)_2HCO(CH_2)_2COH + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Alcohols	Similar stain is produced.	Higher readings are given.		
Ethers	"	"		
Aliphatic hydrocarbons (more than C ₃)	,	"		
Ketones	"	"		
Aromatic hydrocarbons	Whole reagent is changed to Brown.	"		
Esters	"	"		
Halogenated hydrocarbons	"	"		

Isopropyl cellosolve concentration (ppm)



No. 190U tube reading (ppm)

ISOPROPYL ETHER



1. PERFORMANCE

1) Measuring range 30-800 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

 $\hbox{\it 6) Reading } \qquad \hbox{\it i. The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and} \\$

Isopropyl ether concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 5 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $[(CH_3)_2CH]_2O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

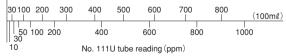
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

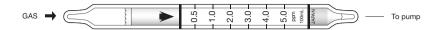
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar or brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	"

Isopropyl ether concentration (ppm)



ISOPROPYL MERCAPTAN



1. PERFORMANCE

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : The tube scale is calibrated based on Methyl mercaptan at 1 pump stroke and

the tube has the same sensitivity for Isopropyl mercaptan.

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride, Hydrogen chloride is produced and PH indicator is discoloured. RSH + HgCI₂→RS(HgCI) + CI

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

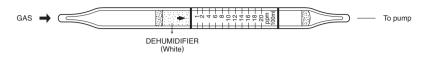
Substance	Interference	Coexistence
Arsine	Similar stain is produced.	Higher readings are given.
Hydrogen selenide	"	"
Phosphine	"	"
Hydrogen sulphide	"	"
Hydrogen cyanide	Whole reagent is discoloured to Red.	"
Sulphur dioxide		Whole reagent is changed to Pale red, but Pink stain indicates Mercaptans conc.

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times$ Reading value

ISOPROPYLAMINE



1. PERFORMANCE

1) Measuring range : 1-12 ppmNumber of pump strokes $: 1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Diethyl amine at 1 pump stroke and

Isopropylamine concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Pale purple→Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. (CH₃)₂CHNH₂ + H₃PO₄ → [(CH₃)₃NH]₂HPO₄

4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Ammonia	Similar stains are produced.	Higher readings are given.
Other amines	"	"





No. 222S tube reading (ppm)

ISOVALERIC ACID



1. PERFORMANCE

1) Measuring range 3-50 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years 5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Isovaleric acid concentration is determined by using a conversion chart

at 1 pump stroke.
7) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

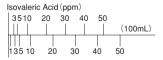
By reacting with alkali, PH indicator is discoloured. (CH₃)₂CHCH₂COOH + N_aOH → (CH₃)₂CHCH₂COON_a + H₂O

4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide		Similar stain is produced.	Higher readings are given.
Nitrogen dioxide	300	"	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	"
Acetic acid		Similar stain is produced.	"



No.216S tube reading (ppm)

MALEIC ANHYDRIDE



1. PERFORMANCE

1) Measuring range 0.2-10 ppmNumber of pump strokes 0.2-10 ppm0.2-10 ppm

2) Sampling time : 6 minutes/4 pump strokes

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : Direct reading from the scale calibrated by 4 pump strokes, then multiply the

tube reading by 0.2 (tube reading \times 0.2).

7) Colour change : Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured.

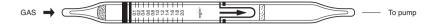
4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide		Similar stain is produced.	Higher readings are given.
Nitrogen dioxide	300	"	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	"
Acetic acid		Similar stain is produced.	"

MERCURY VAPOUR



1. PERFORMANCE

1) Measuring range $0.5-10mg/m^3$ $0.1-2.0mg/m^3$ Number of pump strokes $0.5-10mg/m^3$ $0.1-2.0mg/m^3$ $0.5-10mg/m^3$ $0.5-10mg/m^3$ 0.5-1

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 5 pump strokes

7) Colour change : Grey→Pale orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

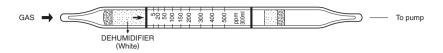
Substance	Interference	ppm	Coexistence
Chlorine	The accuracy of readings is not affected.	0.1	Lower readings are given.
Hydrogen chloride	"	0.5	"
Nitrogen dioxide	Brown stain is produced.	0.1	"
Hydrogen Sulphide	"	0.5	"

Coexistence of less than 2ppm of Chlorine does not affect the readings if Tube No.109SB Chlorine detector tube is connected as a pretreat tube.

(NOTE)

In case of 1 pump stroke, actual concentration is calculated with five times of reading value.

MESITYL OXIDE



1. PERFORMANCE

1) Measuring range 5-100 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (0 \sim 20 $^{\circ}\mathrm{C}$) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Graduations printed on the tube are calibrated by Ethyl cellosolve at 3 pump strokes

and Mesityl oxide is determined by using a conversion chart.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3COCH = C(CH_3)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

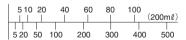
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Alcohols		Similar stain is produced.		Higher readings are given.
Toluene	200	Whole reagent is discoloured to Brown.	500	The top of discoloured layer becomes unclear.
Hexane		The accracy of reading is not affected. 1,000ppm	less than affected.	The accracy of reading is not
Ethyl Acetate		"		"
Trichloroethylene		"		"

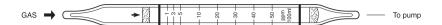
Mesityl oxide (ppm)



No.190U Tube reading (ppm)

Conversion	Corrected Concentration (ppm)					
Value (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C ~ 40 °C (68 °F) (104 °F)			
100	_	_	100			
80	_	95	80			
60	100	70	60			
40	60	45	40			
20	30	25	20			
10	15	12	10			
5	7	6	5			

METHACRYLIC ACID



1. PERFORMANCE

7) Colour change

1) Measuring range : 1-50 ppm Number of pump strokes : $1(100 \text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years 5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Methacrylic acid concentration is determined by using a conversion chart

at 1 pump stroke.

∴ Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

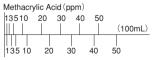
By reacting with alkali, PH indicator is discoloured. $CH_2 = C(CH_3) COOH + NaOH \rightarrow CH_2 = C(CH_3) COONa + H_2O$

4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide		Similar stain is produced.	Higher readings are given.
Nitrogen dioxide	300	//	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	"
Acetic acid		Similar stain is produced.	"



No.216S tube reading (ppm)

1-METHOXY-2-PROPANOL



1. PERFORMANCE

7) Colour change

1) Measuring range 10-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : The tube scale is calibrated based on Cyclohexanone at 3 pump strokes and

1-Methoxy-2-propanol concentration is determined by using a conversion chart

at 1 pump stroke ∴ Yellow → Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

CH₃CHOHCH₃OCH₃ + Cr^{6+} + $H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to pale brown.	The accuracy of readings is not affected if the maximum end point of the blue stain is discernable.		
Aromatic hydrocarbons	"	"		
Halogenated hydrocarbons	"	/ Higher readings are given.		
Alcohols	Similar stain is produced.			
Esters	Pale brown stain is produced from the zero end of the detecting reagent (inlet side of the tube).	The accuracy of readings is not affected.		

1-Methoxy-2-propanol concentration (ppm)

10 	50 	100	20	00	30	00	40	00	50	00 (100mℓ))
2	10	20	4	0	6	0	8	0	10	00	

No. 197U tube reading (ppm)

METHYL ACETATE



1. PERFORMANCE

1) Measuring range 0.1-3.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $0 \sim 40^{\circ}$ C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

Methyl acetate concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Orange → Dark green

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 5 %

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3CO_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

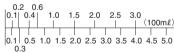
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is changed to Brown.
Propane		0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	Higher readings are given.

Methyl acetate concentration (%)



No.111SA Tube reading (%)

METHYL ACRYLATE



1. PERFORMANCE

1) Measuring range 2-60 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit 0.5 ppm4) Shelf life 2 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = CHCO_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

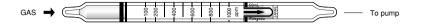
DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to dark Brown.	"
Halogenated hydrocarbons	"	"
Esters	"	"
Aromatic hydrocarbons	"	"

Tube	C	Corrected Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 ℃ (86 °F)	40 ℃ (104 °F)	
60	88	70	60	54	48	
40	54	46	40	36	32	
20	28	24	20	18	16	
10	14	12	10	9	8	
5	7	6	5	4	3	
2	2	2	2	2	2	

METHANOL IN LP GAS



1. PERFORMANCE

1) Measuring range ∴ 100-1,000 ppmv Number of pump strokes 1/2 (50mℓ)

2) Sampling time : 30 seconds/1/2 pump strokes

3) Detectable limit \therefore 20 ppmv 4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1/2 pump stroke

8) Colour change : Yellow→Blue or Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Perchlorate, PH indicator is discoloured. CH₃OH + Mg (CIO₄)₂→ Alkaline compound

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppmv	Interference		
Hydrogen sulphide	0.4	The accuracy of readings is not affected.		
Sulphur dioxide	0.5	"		
Toluene	10	"		

TEMPERATURE CORRECTION TABLE

Temperature; Multiply the reading by the correction factors to correct for temperature.

Temperature (°C)	0	1	2	3	4	5	6	7	8	9
Correction Factor	1.40	1.38	1.36	1.34	1.32	1.30	1.28	1.26	1.24	1.22
Temperature (°C)	10	11	12	13	14	15	16	17	18	19
Correction Factor	1.20	1.18	1.16	1.14	1.12	1.10	1.08	1.06	1.04	1.02
Temperature (°C)	20	21	22	23	24	25	26	27	28	29
Correction Factor	1.00	0.98	0.96	0.94	0.92	0.90	0.88	0.86	0.84	0.82
Temperature (°C)	30	31	32	33	34	35	36	37	38	39
Correction Factor	0.80	0.80	0.79	0.79	0.78	0.78	0.77	0.77	0.76	0.76
Temperature (°C)	40									
Correction Factor	0.75									

METHYL ALCOHOL (METHANOL)



1. PERFORMANCE

1) Measuring range 0.05-6.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke
3) Detectable limit : 0.01 % (100 ppm)

4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow orange → Pale green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Dichromate is reduced $CH_3OH + Cr^{6+} + H_3SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

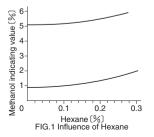
GAS CHROMATOGRAPHY

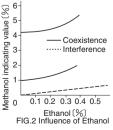
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Aliphatic hydrocarbons (more than C ₃) FIG.1	Similar stain is produced.		Higher readings are given.
Alcohols FIG.2	"		"
Esters	"	50	"
Ketones	"		"
Aromatic hydrocarbons	"		"
Halogenated hydrocarbons	Reagent is discoloured to Brown slightly.		If the top of the discolouration can be obtained, the accuracy of readings is not affected.

(NOTE)

Although the top of light green stain changes to Brown gradually, read the concentration on the top of the light green stain shortly.





TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (%)				
Readings (%)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40°C (104°F)
6.0	-	-	6.0	4.3	3.3
5.0	-	-	5.0	3.7	2.8
4.0	-	-	4.0	3.0	2.3
3.0	-	5.1	3.0	2.3	1.8
2.0	-	3.0	2.0	1.6	1.2
1.0	2.7	1.4	1.0	0.8	0.6
0.5	1.0	0.7	0.5	0.4	0.3
0.1	0.1	0.1	0.1	0.1	0.1

METHYL ALCOHOL (METHANOL)



1. PERFORMANCE

1) Measuring range \therefore 20-1,000 ppm Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (0 \sim 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

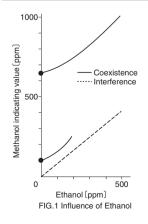
Chromium oxide is reduced $CH_3OH + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

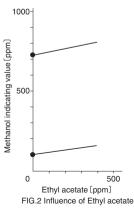
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence	
Alcohols	FIG.1	Similar stain is produced.	Higher readings are given.	
Esters	FIG.2		"	
Aliphatic hydrocarbons (more than C ₃)		Whole reagent is changed to Pale brown.	"	
Aromatic hydrocarbons		"	"	
Ketones		"	"	
Halogenated hydrocarbo	ns	"	"	

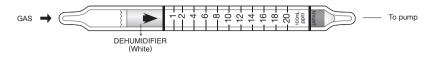




TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)				
Readings (ppm)	0 ℃ (32*F)	10 ℃ (50 °F)	20-40 °C (68-104 °F)		
1000	1200	1100	1000		
800	960	880	800		
600	720	660	600		
400	480	440	400		
200	240	220	200		
100	120	110	100		
50	60	55	50		
20	24	22	20		

METHYL AMINE



1. PERFORMANCE

1) Measuring range 1-20 ppm Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 0.2 ppm4) Shelf life 3 years5) Operating temperature $0 \sim 40 \text{ }^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. CH₃NH₂ + H₃PO₄ → (CH₃NH₃)₂HPO₄

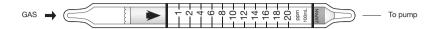
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Ammonia	Similar stain is produced.	Higher readings are given	
Other amines	"	"	

N-METHYL ANILINE



1. PERFORMANCE

1) Measuring range 0.5-6.0 ppmNumber of pump strokes 0.5-6.0 ppm0.5-6.0 ppm

2) Sampling time : 2 minutes/2 pump strokes

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : The tube scale is calibrated based on Ammonia at 1 pump stroke and

N-Methyl aniline concentration is determined by using a conversion chart

at 2 pump strokes

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. C6H5NHCH3 + H3PO4→ (R2NH2)3PO4

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Amines	Similar stain is produced.	Higher readings are given.	

N-Methyl aniline concentration (ppm)



No. 105SD tube reading (ppm)



1. PERFORMANCE

1) Measuring range $3-70 \text{ g/m}^3$ Number of pump strokes $1/2 (50 \text{m} \ell)$

2) Sampling time : 2 minutes/1/2 pump strokes

3) Detectable limit 0.6 g/m^3 4) Shelf life 2 years 5) Operating temperature $24 \sim 45 \text{ C}$

6) Reading : Direct reading from the scale calibrated by 1/2 pump strokes

7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $CH_3Br + CrO_3 + I_2O_5 + H_2SO_4 \rightarrow Br_2$ $Br_2 + Pd$ compound $\rightarrow PdBr_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Chloropicrin		100	The bottom of the stained layer becomes light brown, but it does not affect the reading.



1. PERFORMANCE

1) Measuring range : 10-500ppm Number of pump strokes : $1(100m\ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 1 ppm

4) Shelf life : 3 years (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Reddish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Bromine is produced. Bromine reacts with o-Toluidine and red Orthoquinone is produced.

 $CH_3Br + I_2O_5 + CrO_3 + H_2SO_4 \cdot nSO_3 \rightarrow Br_2$

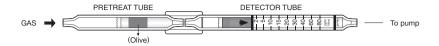
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Chlorine	1	Similar stain is produced.	1	Higher readings are given.
Bromine	1	"	1	"
Nitrogen dioxide	1	"	1	"
Dichloromethane	500	"	500	"
Chloroform	50	"	50	"
Ethylene dibromide	50	"	50	"
Trichloroethylene	50	"	20	"
Tetrachloroethylene	50	"	40	"

Tube		Corrected Concentration (ppm)							
Readings (ppm)	0°C (32°F)	5 °C (41 °F)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)		
500	_	-	_	620	500	430	390		
400	_	_	630	475	400	360	325		
300	_	675	430	345	300	280	260		
200	580	320	235	210	200	190	175		
150	260	185	160	150	150	140	130		
100	120	110	100	100	100	95	90		
50	50	50	50	50	50	50	50		
10	10	10	10	10	10	10	10		



1. PERFORMANCE

1) Measuring range : 2-80 ppm 1-25 ppm 0.4-10 ppm Number of pump strokes $1\,(100\mathrm{m}\,\ell)$ $2\,(200\mathrm{m}\,\ell)$ $4\,(400\mathrm{m}\,\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : $0.2 \text{ ppm} (400 \text{m} \ell)$

4) Shelf life : 3 years (Necessary to store in refrigerated conditions; $0 \sim 10 \, ^{\circ}\text{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Bromine is produced. Bromine reacts with *o*-Toluidine and yellow Orthoquinone is produced.

$$CH_3Br + L_2O_5 + CrO_3 + H_2SO_4 \cdot nSO_3 \rightarrow Br_2$$

$$3Br_2H_2N - \bigcirc - \bigcirc - NH_2 \rightarrow BrH_2N = \bigcirc = \bigcirc = NH_2Br$$

$$CH_3 \qquad CH_3$$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

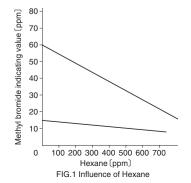
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence	
Halogen		Similar stain is produced.	Higher readings are given.		
Halogenated hydrocarbons		"		"	
Hexane F	IG.1	The accuracy of readings is not affected.	200	Lower readings are given.	

(NOTE)

2 pump strokes can be used with the following formula to measure the range of 1-25 ppm; Actual concentration = $1/2 \times$ Temperature corrected value

4 pump strokes can be used with the following formula to measure the range of 0.4-10 ppm; Actual concentration = $1/5 \times$ Temperature corrected value



Tube	С	Corrected Concentration (ppm)						
Readings (ppm)	0°C (32°F)	5℃ (41°F)	10 °C (50 °F)	20°C (68°F)	30°C (86°F)	40 °C (104 °F)		
80	_	140	98	80	75	73		
60	145	76	67	60	57	56		
40	44	43	42	40	40	40		
30	30	30	30	30	30	30		



1. PERFORMANCE

1) Measuring range : 8.8-22 ppm 0.5-10 ppm 0.1-0.5 ppm Number of pump strokes 1/2 (50mℓ) 1 (100mℓ) 3 (300mℓ)

2) Sampling time : 2.5 minutes/1 pump stroke, 1.5 minutes/ 1/2 pump strokes

3) Detectable limit : $0.03 \text{ ppm} (300 \text{m} \ell)$

4) Shelf life : 1 year 5) Operating temperature : $10 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Bromine is produced.

Bromine reacts with 3,3-Dimethylnaphthidine and Bromine compound is produced.

 $CH_3Br + CrO_3 + H_2SO_4 \cdot nSO_3 \rightarrow Br_2$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

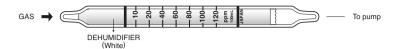
Substance	Interference		
Carbon dioxide	The accuracy of readings is not affected.		

(NOTE)

1/2 pump srtokes can be used with the following formula to measure the range of 8.8-22 ppm; Actual concentration = $2.2 \times \text{Tube}$ reading

3 pump srtokes can be used with the following formula to measure the range of 0.1-0.5 ppm; Actual concentration = $0.2 \times \text{Tube}$ reading

METHYL BUTYL KETONE



1. PERFORMANCE

1) Measuring range : 5-80 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Vinyl acetate at 1 pump stroke and

Methyl butyl ketone concentration is determined by using a conversion chart

at 2 pump strokes

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

CH₃ (CH₂)₃COCH₃ + Cr^{6+} + $H_2SO_4 \rightarrow Cr^{3+}$

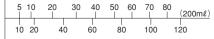
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

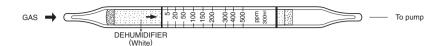
Substance	Interference	ppm	Coexistence
Acetic acid			The accuracy of readings is not affected.
Ethylene	Pale brown or pale blue stain is produced.	150	Lower readings are given.
Alcohols	Similar stain is produced.		Higher readings are given.
Ethers	"		"
Aliphatic hydrocarbons (more than C ₃)	Whole layer is discoloured to pale brown.		"
Aromatic hydrocarbons	"		"
Halogenated hydrocarbons	"		"
Esters	"		"

Methyl butyl ketone concentration (ppm)



No. 237S tube reading (ppm)

METHYL CELLOSOLVE



1. PERFORMANCE

1) Measuring range 5-500 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : The tube scale is calibrated based on Ethyl cellosolve at 3 pump strokes and the

tube has the same sensitivity for Methyl cellosolve.

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

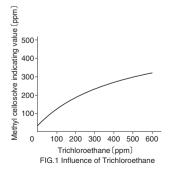
 $HOCH_2CH_2OCH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

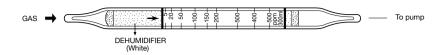
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Alcohols	Similar stain is produced.	Higher readings are given.	
Ethers	"	"	
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Brown.	"	
Aromatic hydrocarbons	"	"	
Esters	"	"	
Ketones	"	"	
Halogenated hydrocarbons FIG.1	"	"	



Tube	Corrected Concentration (ppm)					
Readings (ppm)	10 °C (50 °F)	15 °C (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 °C (86 °F)	35℃ (95°F)
500	800	620	500	410	335	265
400	620	490	400	330	260	200
300	450	370	300	250	200	150
200	290	250	200	160	130	100
150	220	190	150	120	90	70
100	150	130	100	80	60	50
50	80	70	50	40	30	30
20	30	25	20	15	12	10
5	10	7	5	4	3	2

METHYL CELLOSOLVE ACETATE



1. PERFORMANCE

1) Measuring range 3-120 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit : -4) Shelf life : 2 years 5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl cellosolve at 3 pump strokes and

Methyl cellosolve acetate concentration is determined by using a conversion

chart at 3 pump strokes

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 10 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3CO_2CH_2CH_2OCH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

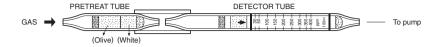
Substance	Interference	Coexistence	
Alcohols	Similar stain is produced.	Higher readings are given.	
Ethers	"	"	
Aliphatic hydrocarbons (more than C ₃)	"	"	
Ketones	"		
Aromatic hydrocarbons	Whole reagent is changed to Brown.	"	
Esters	"	"	
Halogenated hydrocarbons	"	"	

Methyl cellosolve acetate concentration (ppm)



No.190U Tube reading (ppm)

METHYL CHLOROFORM (1,1,1-TRICHLOROETHANE)



1. PERFORMANCE

1) Measuring range : 30-400 ppm 15-30 ppm Number of pump strokes $1(100m\ell)$ $2(200m\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit $10 \text{ ppm} (200 \text{m} \ell)$

4) Shelf life 3 years (Necessary to store in refrigerated conditions; $0 \sim 10 \, \text{°C}$)

5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading Direct reading from the scale calibrated by 1 pump stroke

7) Colour change White → Yellow orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chlorine is produced by an Oxidizer. By reacting between this Chlorine and o-Toluidine, Orthoguinone is produced.

 $CH_3CCI_3 + Cr_3O_3 + H_2SO_4 \cdot nSO_3 \rightarrow CI_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

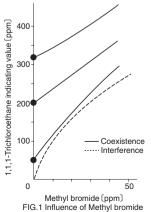
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Halogens	Similar stain is produced.	Higher readings are given.	
Halogenated hydrocarbons FIG.1,2	"	"	

(NOTE)

When the concentration is below 30 ppm, 2 pump strokes can be used to detremine the lower concentration. Following formula is available for the actual concentration.

Actual concentration = $1/2 \times$ Reading value



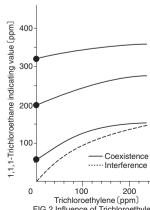


FIG.2 Influence of Trichloroethylene

METHYL CYCLOHEXANE



1. PERFORMANCE

1) Measuring range : 100-1,600 ppm Number of pump strokes : $1(100m\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by n-Hexane at 1 pump stroke

and Methyl cyclohexane concentration is determined by using a conversion

chart at 1 pump stroke.

7) Colour change : Orange → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

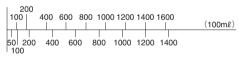
Chromium oxide is reduced

CH₃ (CH₂)₅CH₃ + Cr⁶⁺ + H₂SO₄ \rightarrow Cr³⁺

4. CALIBRATION OF THE TUBE

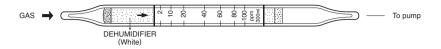
GAS CHROMATOGRAPHY

Methyl cyclohexane concentration (ppm)



No.113SB Tube reading (ppm)

METHYL CYCLOHEXANONE



1. PERFORMANCE

1) Measuring range 2-100 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

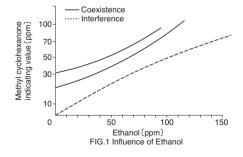
Chromium oxide is reduced. $CH_3C_6H_9O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

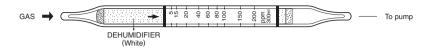
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Pale brown.	The accuracy of readings is no affected.	
Aromatic hydrocarbons	"	"	
Halogenated hydrocarbons	"	"	
Alcohols FIG.1	Similar stain is produced.	Higher readings are given.	
Esters	Pale ringed stain is produced near the bottom of the reagent.	The accuracy of readings is not affected.	



Tube	Corrected Concentration (ppm)						
Readings (ppm)	0°C (0°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)		
100	-	130	100	80	66		
80	130	105	80	65	52		
60	100	80	60	48	36		
40	68	54	40	30	22		
20	35	26	20	14	10		
10	18	14	10	7	5		
2	3	3	2	1	1		

METHYL CYCLOHEXANOL



1. PERFORMANCE

1) Measuring range 5-200 ppmNumber of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

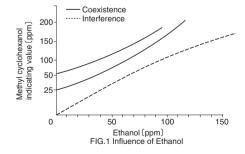
Chromium oxide is reduced. $CH_3C_6H_{10}OH + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

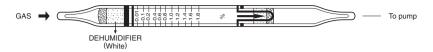
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Aliphatic hydrocarbons	Whole reagent is changed to Pale	The accuracy of readings is not	
(more than C ₃)	brown.	affected.	
Aromatic hydrocarbons	"	"	
Halogenated hydrocarbons	"	"	
Alcohols FIG.1	Similar stain is produced.	Higher readings are given.	
Esters	Pale ringed stain is produced near the bottom of the reagent.	The accuracy of readings is not affected.	



Tube			ed Concentration (ppm)			
Readings (ppm)	10 ℃ (50 °F)	20 °C (68 °F)	30℃ (86°F)	40 °C (104 °F)		
200	_	200	110	65		
150	_	150	85	52		
100	_	100	60	40		
80	170	80	50	30		
60	110	60	35	25		
40	70	40	25	15		
20	40	20	15	10		
10	20	10	7	5		
5	10	5	4	3		

METHYL ETHYL KETONE



1. PERFORMANCE

1) Measuring range 1.0-5.0% 0.05-2.2%Number of pump strokes $1/2(50m\ell)$ $1(100m\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit : 50 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \, ^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Ethylene oxide at 1 pump stroke

and Methyl ethyl ketone is determined by using a conversion chart.

7) Colour change : Orange → Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

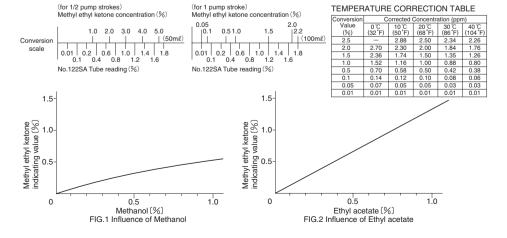
Potassium dichromate is reduced. $CH_3COC_2H_5 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substa	nce	Interference	ppm	Coexistence
Alcohols FIG.1		Similar stain is produced.		Higher readings are given.
Esters	FIG.2	"		"
Ketones		"		"
Aromatic hydrocarbons		"		"
Halogenated hydrocarbons		Whole stain is discoloured to Pale brown.	0.5%	"



METHYL ETHYL KETONE



1. PERFORMANCE

1) Measuring range 0.01-1.4%Number of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit 0.001%(10 ppm)4) Shelf life 3 years5) Operating temperature $0 \sim 40\%$

6) Reading : Direct reading from the scale calibrated by 2 pump strokes

7) Colour change : Orange → Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Dichromate is reduced.

 $CH_3COC_2H_5 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

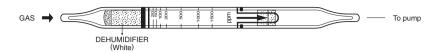
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is discoloured to Brown and higher readings are given.
Propane		0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons	Similar stain is produced.	50	Higher readings are given.

METHYL ETHYL KETONE



1. PERFORMANCE

1) Measuring range \therefore 20-1,500 ppm Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \vdots 10 ppm 4) Shelf life \vdots 2 years 5) Operating temperature \vdots 10 \sim 40 $^{\circ}$ C

6) Temperature compensation
7) Reading
1. Necessary (See "TEMPERATURE CORRECTION TABLE")
2. Direct reading from the scale calibrated by 1 pump stroke
3. Yellow→Pale blue (The top of discoloured layer is Brown.)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3COC_2H_5 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

N Substance	Interference	Coexistence
Alcohols	Similar or Brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Halogenated hydrocarbons FIG.2		Whole reagent is changed to Brown, but if the top of Pale blue stain is clear, the reading can be obtained.
Aliphatic hydrocarbons (more than C ₃) FIG.1		"

TEMPERATURE CORRECTION TABLE

(59°F) (68°F)

1750 1500 1260 1100 940

1200

120 100

1000

500 430

Crrected Concentration (ppm)

25 °C (77 °F)

850

(86°F)

720

70 60

30 30

820

Tube

1500

200

100

50

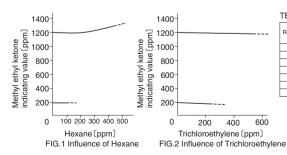
20

10℃ (50°F)

300 250 200 170 140 120

80 60 50 30 30 20

Reading (ppm)



METHYL IODIDE



1. PERFORMANCE

1) Measuring range : 0.4-8 ppm 1-20 ppm 2.5-50 ppm Number of pump strokes $2(200m\ell)$ $1(100m\ell)$ $1/2(50m\ell)$

2) Sampling time ∴ 1.5 minutes/1 pump stroke 3) Detectable limit ∴ 0.2 ppm (200mℓ)

4) Shelf life 0.2 ppin5) Operating temperature 0.2 ppin0.2 ppin0.2 ppin

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Grey

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. $CH_3I + V_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Carbon dioxide	The accuracy of readings is not affected.	50%	The accuracy of readings is not affected.
Methyl bromide	"	1	"
Acetone	"	200	"
Hexane	"	200	"
Hydrogen sulphide		0.5	Higher readings are given.
1,3-Dichloropropene		0.1	"
Toluene			Lower readings are given.

(NOTE)

1) In case of 1/2 pump strokes, following formula is available for the actual concentration. Actual concentration = $2.5 \times$ Temperature corrected value

2) In case of 2 pump strokes, following formula is available for the actual concentration. Actual concentration = $0.4 \times$ Temperature corrected value

TEMPERATURE CORRECTION TABLE

Temperature; To correct for temperature, multiply the tube reading by the following factors.

	Pump stroke	Temperature (℃)	0	5	10	15	20	25	30	35	40
	1		1.55	1.32	1.15			1.	00		
Г	1/2	Coefficient	2.20	1.80	1.50	1.20			1.00		
	2		1.30	1.22	1.15		1.00		1.15	1.22	1.30

METHYL IODIDE



1. PERFORMANCE

1) Measuring range ∴ 500-15,000 ppm Number of pump strokes 1/2 (50mℓ)

2) Sampling time : 45 seconds/1/2 pump strokes

3) Detectable limit \therefore 75 ppm 4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1/2 pump stroke

8) Colour change

∴ Yellowish orange → Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Bichromate is reduced. $CH_3I + Cr^{6+} + H_3PO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

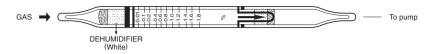
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	%	Interference
Carbon dioxide	60	The accuracy of readings is not affected.

Tube	Corrected Concentration (ppm)				
Readings	0°C	10 °C	20℃	30 °C	40°C
(ppm)	(32°F)	(50°F)	(68°F)	(80°F)	(104°F)
15000	19500	17250	15000	12325	9650
14000	18200	16100	14000	11525	9050
13000	16900	14950	13000	10750	8500
12000	15600	13800	12000	9950	7900
11000	14300	12650	11000	9200	7400
10000	13000	11500	10000	8400	6800
9000	11700	10350	9000	7600	6200
8000	10400	9200	8000	6800	5600
7000	9100	8050	7000	6000	5000
6000	7800	6900	6000	5150	4300
5000	6500	5750	5000	4350	3700
4000	5200	4600	4000	3540	3080
3000	3900	3450	3000	2750	2500
2000	2500	2250	2000	1960	1920
1000	1060	1030	1000	1000	1000
500	500	500	500	500	500

METHYL ISOBUTYL KETONE



1. PERFORMANCE

1) Measuring range 0.01-0.6 % Number of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit : 10 ppm 4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \, ^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Graduations printed on the tube are calibrated by Ethylene oxide at 1 pump stroke and Methyl isobuty ketone is determined by using a conversion chart at pump strokes.

8) Colour change : Orange → Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

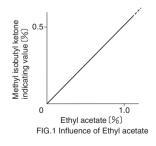
 $H_3COCH_2CH(CH_3)_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

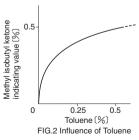
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

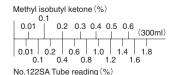
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Alcohols		Similar stain is produced.		Higher readings are given.
Esters	FIG.1	"		"
Ketones		"		"
Aromatic hydrocarbons	FIG.2	"		"
Halogenated hydrocarbons		Whole reagent is changed to Pale brown.	0.5%	"

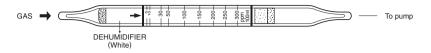




Conversion	Corrected Concentration (%)				
Value	0℃	10℃	20℃	30℃	40°C
(%)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)
0.6	_		0.60	0.55	0.52
0.5	_		0.50	0.46	0.43
0.4	_		0.40	0.37	0.35
0.3	_		0.30	0.28	0.27
0.2	_	0.23	0.20	0.19	0.18
0.1	0.16	0.11	0.10	0.10	0.10
0.05	0.05	0.05	0.05	0.05	0.05
•					



METHYL ISOBUTYL KETONE



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \text{5-300 ppm} \\ \text{Number of pump strokes} \end{array}$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $(CH_3)_2CHCH_2COCH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Aliphatic hydrocarbons (more than C ₃)	Whole reagent is changed to Pale brown.	"
Aromatic hydrocarbons	"	"
Esters	"	"
Ketones	"	"
Halogenated hydrocarbons	"	"

Tube	Corrected Concentration (ppm)					
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)	
300	_	_	300	230	190	
250	_	380	250	195	165	
200	_	290	200	155	130	
150	355	210	150	120	100	
100	220	130	100	80	70	
50	90	65	50	40	35	
30	55	40	30	25	25	
10	15	10	10	8	8	
5	7	5	5	5	5	

MITC (METHYL ISOTHIOCYANATE)



1. PERFORMANCE

1) Measuring range 200-10,000 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 50 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellowish orange→Pale green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Bichromate is reduced.

 $CH_3NCS + Cr^{6+} + H_3PO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	%	Interference		
Carbon dioxide	60	The accuracy of readings is not affected.		
Sulphuryl fluoride 10		"		

Tube	Corrected Concentration (ppm)					
Readings	0°C	10 °C	20°C	30°C	40 °C	
(ppm)	(32°F)	(50°F)	(68°F)	(80°F)	(104°F)	
10000	_	_	10000	7500	6100	
9000	_	_	9000	7000	5500	
8000	_	_	8000	6200	4900	
7000	_	_	7000	5400	4300	
6000	_	14400	6000	4600	3700	
5000	_	8600	5000	3900	3000	
4000	11400	6100	4000	3100	2500	
3000	7200	4200	3000	2400	1900	
2000	3800	2600	2000	1700	1400	
1000	1600	1200	1000	900	750	
500	650	575	500	450	400	
200	200	200	200	200	200	

MITC (METHYL ISOTHIOCYANATE)



1. PERFORMANCE

1) Measuring range 0.66-22 ppm Number of pump strokes $1 (100 \text{m} \ell)$ $1/2 (50 \text{m} \ell)$ 1/2 mpp stroke

3) Detectable limit : 0.1 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pink → Yellowish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By decomposing with an oxidizer, Sulphur dioxide is produced and PH indicator is discoloured.

 $CH_3NCS + V_2O_5 + H_2SO_4 \rightarrow SO_2$

SO₂ + pH indicator (Pink) → Yellowish orange

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence
Carbon dioxide		35	Higher readings are given.

(NOTE)

The scale is calibrated based on the temperature of $20\,\mathrm{C}$. Reading obtained in other temperature circumstances should be corrected with the following temperature correction coefficient table.

TEMPERATURE CORRECTION COEFFICIENT TABLE(AT 20°C)

Pump stroke	Temperature (℃)	0	5	10	15	20	25	30	35	40
1	Coefficient	1.30	1.23	1.15	1.08	1.00	0.95	0.90	0.85	0.80
1/2	Coemicient	1.70	1.48	1.30	1.11	1.00	0.97	0.95	0.92	0.90

Actual concentration = Reading value × Coefficient for temperature correction

MITC (METHYL ISOTHIOCYANATE)



1. PERFORMANCE

2) Sampling time : 1.5 minutes/1 pump stroke, 1 minute/1/2 pump strokes

3) Detectable limit $\begin{array}{c} \text{3 ppm} (100\text{m}\ell) \\ \text{4) Shelf life} \\ \text{5) Operating temperature} \\ \end{array} \begin{array}{c} \text{: 3 ppm} (100\text{m}\ell) \\ \text{: 1 year} \\ \text{: } 0 \sim 40\,\text{°C} \\ \end{array}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

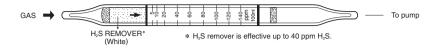
Bichromate is reduced. $CH_3NCS + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

Substance	%	Interference
Carbon dioxide	50	The accuracy of readings is not affected.

METHYL MERCAPTAN



1. PERFORMANCE

1) Measuring range 5-140 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (0 \sim 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Reddish yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Methyl mercaptan reacts with Palladium sulphate.

 $CH_3SH + PdSO_4 \rightarrow (CH_3S)_2Pd + H_2SO_4$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Carbon monoxide	150	Dark grey stain is produced.		
Ethylene	200	"		
Hydrogen sulphide	40	Dark brown stain is produced.		
Acetylene	20	Pale brown stain is produced.		
Ethyl mercaptan	1	Reddish yellow stain is produced.		
Methyl sulphide			1	Lower readings are given.
Chlorine			0.2	"
Nitrogen dioxide			1	"

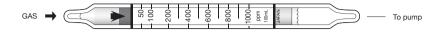
(NOTE)

Max. 40 ppm Hydrogen sulphide is eliminated by pretreat reagent.

TEMPERATURE CORRECTION TABLE

Tube	Correct	n (ppm)	
Readings (ppm)	0 ℃ (32 °F)	10 ℃ (50 °F)	20-40 °C (68-104 °F)
140	164	149	140
120	140	128	120
100	118	107	100
80	94	85	80
60	70	63	60
40	45	41	40
20	21	21	20
10	10	10	10

METHYL MERCAPTAN



1. PERFORMANCE

1) Measuring range 50-1,000 ppm Number of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : 10 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \text{ °C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→Orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

It reacts with Palladium sulphate. CH₃SH + PdSO₄→(CH₃S)₂Pd + H₂SO₄

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substance	ppm	Interference	ppm	Coexistence
Chlorine		The accuracy of readings is not affected.	Methyl mercaptan conc. X 1/3	Lower readings are given.
Carbon monoxide	500	Deep grey stain is produced.		The accuracy of readings is not affected If the top of the discolouration can be obtained.
Ethylene	500	"		"
Hydrogen sulphide	less than 500	The accuracy of readings is not affected.	650	Higher readings are given.
Nitrogen dioxide	5000	Yellow stain is produced.	1000	"

METHYL MERCAPTAN



1. PERFORMANCE

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by Methyl mercaptan at 1 pump stroke

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride, Hydrogen chloride is produced and PH indicator is discoloured. CH₃SH + HgCI₂→CH₃S (HgCI) + HCI

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

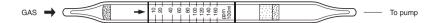
Substance	Interference	Coexistence
Arsine	Similar stain is produced.	Higher readings are given.
Hydrogen selenide	"	"
Phosphine	"	"
Hydrogen sulphide	"	"
Hydrogen cyanide	Whole reagent is changed to Red.	"
Sulphur dioxide		Whole reagent is changed to Pale red, but Pink stain indicates Mercaptans conc.

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times$ Reading value

METHYL METHACRYLATE



1. PERFORMANCE

1) Measuring range 10-160 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 15 % RSD-mid.: 10 % RSD-high: 5 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $H_2C = C(CH_3)CO_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

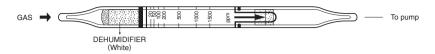
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Esters	Whole layer is discoloured to Pale brown.	"
Ketones	"	"
Aliphatic hydrocarbons (more than C ₃)	"	"
Aromatic hydrocarbons	"	"
Halogenated hydrocarbons	"	"

TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
160	190	175	160	150	140
140	165	155	140	130	125
120	140	130	120	110	105
100	120	110	100	95	90
80	95	90	80	75	70
60	70	65	60	55	50
40	47	44	40	37	35
20	24	22	20	19	17
10	12	- 11	10	9	9

METHYL PROPYL KETONE



1. PERFORMANCE

1) Measuring range 20-1,500 ppm Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years 5) Operating temperature : $10 \sim 40 \, \text{°C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke
8) Colour change : Yellow→Pale blue (The top of discoloured layer is Brown.)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

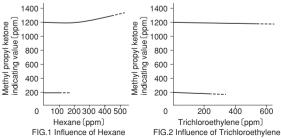
Chromium oxide is reduced.

 $CH_3CO(CH_2)_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	Coexistence
Alcohol	Similar or Brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Halogenated hydrocarbons FIG. 2		Whole reagent is changed to Brown, but if the maximum end point of Pale blue stain is discernable, the accuracy of readings is not affected.
Aliphatic hydrocarbons FIG. 1		"



Tube	l	Corrected Concentration (ppin)						
Readings (ppm)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	25 °C (77 °F)	30 ℃ (86 °F)	35 °C (95 °F)	40 ℃ (104 °F)	
1500	2050	1750	1500	1260	1100	940	820	
1000	1430	1200	1000	850	720	620	530	
500	750	620	500	430	360	320	280	
200	300	250	200	170	140	120	110	
100	160	120	100	90	70	60	50	
50	80	60	50	40	30	30	20	
20	30	30	20	20	10	10	10	

METHYL STYRENE



1. PERFORMANCE

1) Measuring range 10-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

A polymer of Styrene is produced by Sulphuric acid.

$$CH = CH_{2}$$

$$n \longleftrightarrow +H_{2}SO_{4} \longrightarrow \left(-CH - CH_{2} - \right)$$

$$CH_{3}$$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence	
Alcohols	The accracy of readings is not affected.	The accracy of readings is naffected.	
Esters	"	"	
Aromatic hydrocarbons (except Styrene)	"	"	
Halogenated hydrocarbons	"	"	
Aliphatic hydrocarbons (more than C ₃)	"	"	
Styrene	Similar stain is produced.	Higher readings are given.	

TEMPERATURE CORRECTION TABLE

Tube	С	orrected	Concentra	ation (ppr	n)
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 ℃ (68 °F)	30 ℃ (86 °F)	40 ℃ (104 °F)
500	300	380	500	680	850
400	270	320	400	560	730
300	210	250	300	390	500
200	140	160	200	250	300
100	70	80	100	130	150
50	30	40	50	70	80
30	17	24	30	40	54
10	4	6	10	12	18

MONOETHANOL AMINE



1. PERFORMANCE

4) Shelf life . 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pink→Pale purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Sulphuric acid, PH indicator is discoloured. H2NCH2CH2OH + H2SO4→(HOC2N5NH3)2SO4

4. CALIBRATION OF THE TUBE

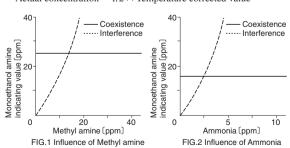
ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

,							
Substanc	е	Interference	Coexistence				
Other amines	FIG.1	Brownish yellow stain is produced.	Double-stain layer(Brownish yellow and Pale purple) is produced, but the maximum end point of the Pale purple stain is discernable.				
Ammonia	FIG.2	Similar stain is produced.	Double-stain layer(Yellow and Pale purple) is produced, but the maximum end point of the Pale purple stain is discernable.				
Hydrazine		"	"				
Atmospheric air (CO ₂ +H ₂ O)	r	"					

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration. Actual concentration = $1/2 \times$ Temperature corrected value



TEMPERATURE CORRECTION TABLE Tube | Corrected Concentration (ppm)

Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)	
50	_	_	50	35	29	
40	_	65	40	30	25	
30	_	49	30	23	20	
25	_	39	25	20	17	
20	65	30	20	16	14	
15	45	22	15	12	10	
10	29	14	10	8	7	
5	12	7	5	4	3	
3	6	4	3	3	2	
1	1	1	1	1	1	

MORPHOLINE



1. PERFORMANCE

1) Measuring range 2-22 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke

and Morpholine concentration is determined by using a conversion chart at 1

pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

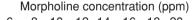
3. CHEMICAL REACTION

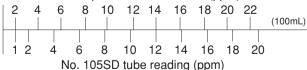
By reacting with Phosphoric acid, PH indicator is discoloured. C₄H₉NO + H₃PO₄→ (R₃NH)₃PO₄

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence	
Amines	Similar stain is produced.	Higher readings are given.	





NAPHTHALENE



1. PERFORMANCE

1) Measuring range : 10-100 ppm Number of pump strokes $1(100m\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 2ppm 4) Shelf life 1 year 0 ~ 40 °C 5) Operating temperature

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation:

7) Reading Graduations printed on the tube are calibrated by Isobutyl acetate at 1 pump stroke and Naphthalene is determined by using a conversion chart at pump strokes.

8) Colour change : Pale yellow → Blackish brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

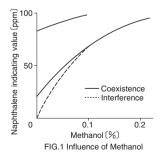
3. CHEMICAL REACTION

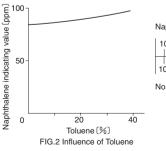
Chromium oxide is reduced. $C_{10}H_8 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

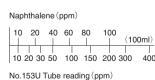
4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

Substance		Interference	Coexistence
Methanol	FIG.1	Pale blue stain is produced.	Higher readings are given.
Butyl acetate		"	"
Aromatic hydrocarbons	FIG.2	Brown stain is produced.	"
Aliphatic hydrocarbons (more than C ₃)		"	"
Esters		"	"







NICKEL CARBONYL



1. PERFORMANCE

1) Measuring range 20-700 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 3 minutes/1 pump stroke

3) Detectable limit : 10 ppm 4) Shelf life : 6 months 5) Operating temperature : $0 \sim 40^{\circ}$ C

6) Reading : Concentration chart method 7) Colour change : Pale yellow→ Dark purple

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 5 %

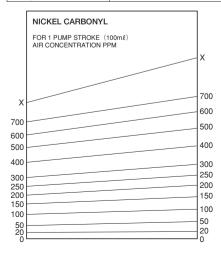
3. CHEMICAL REACTION

By reacting with Gold chloride (III), colloidal gold is liberated. Ni (CO) $_4+AuCI_3\!\!\to\!\!Au$

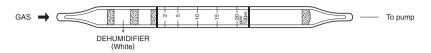
4. CALIBRATION OF THE TUBE

STANDARD GAS GYLINDER METHOD

Substance	Interference	ppm	Coexistence
Arsine	Similar stain is produced.	10	Higher readings are given.
Iron carbonyl	"	10	"
Mercury vapour	"	10	"
Hydrogen sulphide	Brown stain is produced.	10	"
Sulphur dioxide	Pale blue stain is produced.	10	"
Carbon monoxide	The accuracy of readings is not affected.	1,000	"
Acetylene	"	3%	"



NITRIC ACID VAPOUR



1. PERFORMANCE

1) Measuring range 2-20 ppm 1-10 ppm Number of pump strokes $1(100\text{m}\,\ell)$ $2(200\text{m}\,\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit : 0.5ppm (200m $\ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10 \,^{\circ}\text{C}$)

5) Operating temperature : $5 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale yellow→Purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 10%

3. CHEMICAL REACTION

PH indicator is discoloured.

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

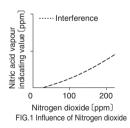
5. INTERFERENCE AND CROSS SENSITIVITY

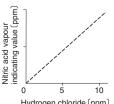
Substance		Interference	ppm	Coexistence
Hydrogen fluoride		Similar stain is produced.	8	The top of discoloured layer becomes unclear and higher readings are given.
Nitrogen dioxide	FIG.1	"	50	"
Hydrogen chloride	FIG.2	"		Higher readings are given.

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $1/2 \times$ Temperature corrected value





Hydrogen chloride (ppm)
FIG.2 Influence of Hydrogen chloride

CORRECTION FOR AMBIENT CONDITIONS:

Temperature; Correct the tube reading by following temperature correction table.

Temperature(℃)		_	_			5	6	7	8	9
Coefficient]			_		2.35	2.17	2.00	1.85	1.72
Temperature(℃)	10	11	12	13	14	15	16	17	18	19
Coefficient	1.60	1.52	1.44	1.37	1.31	1.25	1.19	1.13	1.08	1.04
Temperature(℃)	20	21	22	23	24	25	26	27	28	29
Coefficient	1.00	0.96	0.92	0.89	0.86	0.83	0.80	0.77	0.74	0.71
Temperature(℃)	30	31	32	33	34	35	36	37	38	39
Coefficient	0.70	0.68	0.66	0.64	0.62	0.60	0.58	0.56	0.55	0.54
Temperature(℃)	40									
Coefficient	0.53]								

NITROGEN DIOXIDE



1. PERFORMANCE

1) Measuring range \therefore 20-1,000 ppm Number of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 5 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White → Yellowish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with o-Toluidine, Nitroso-o-Toluidine (dyestuff) is produced.

$$NO_2+H_2N$$
 \longrightarrow NH_2 \longrightarrow $ON=$ \longrightarrow NC $ON=$ $ON=$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

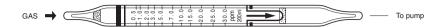
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Chlorine	Similar stain is produced.	5	Higher readings are given.
Bromine	"	5	"
Iodine	"	5	"
Ozone	"	5	"
Nitrogen monoxide	The accuracy of readings is not affected.	10	The accuracy of readings is not affected.

TEMPERATURE CORRECTION TABLE

Tube		Corrected Concentration (ppm)						
Readings (ppm)	0 °C (32 °F)	5 °C (41 °F)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)	
1,000	_	-	-	_	1,000	640	540	
800	_	-	-	_	800	560	480	
600	_	_	_	860	600	470	420	
500	_	_	1,000	700	500	410	380	
400	_	_	840	540	400	350	330	
300	_	1,000	570	390	300	280	270	
200	1,000	500	320	240	200	200	190	
100	200	160	120	110	100	100	100	
50	60	60	60	55	50	50	50	
20	20	10	20	20	20	20	20	

NITROGEN DIOXIDE



1. PERFORMANCE

1) Measuring range 0.5-30 ppmNumber of pump strokes $2(200\text{m}\ell)$

2) Sampling time : 1.5 minutes/2 pump strokes

3) Detectable limit : 0.1 ppm4) Shelf life : 1 year5) Operating temperature $: 0 \sim 40 \text{ °C}$

6) Reading : Direct reading from the scale calibrated by 2 pump strokes

7) Colour change : White→ Yellowish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

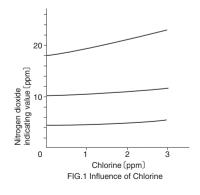
By reacting with o-Toluidine, Nitroso-o-Toluidine (dyestuff) is produced.

$$NO_2+H_2N$$
 \longrightarrow NH_2 \longrightarrow $ON=$ \longrightarrow NO_3 NO_4 NO_5 NO_5

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance		Interference	ppm	Coexistence
Chlorine	FIG.1	Similar stain is produced.	2	Higher readings are given.
Bromine		"	2	"
Iodine		"	2	"
Nitrogen monoxide	FIG.2	The accuracy of readings is not affected.	15	"



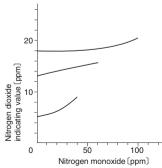
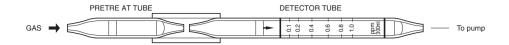


FIG.2 Influence of Nitrogen monoxide

NITROGEN DIOXIDE



1. PERFORMANCE

1) Measuring range 0.1-1.0 ppmNumber of pump strokes $3(300\text{m}\ell)$

2) Sampling time : 135 seconds (45 seconds/1 pump stroke)

3) Detectable limit $\begin{array}{c} \hbox{: } 0.01 \text{ ppm } (300 \text{m} \, \ell) \\ \hbox{4) Shelf life} \\ \hbox{5) Operating temperature} \\ \end{array}$ $\begin{array}{c} \hbox{: } 0.01 \text{ ppm } (300 \text{m} \, \ell) \\ \hbox{: } 1.5 \text{ years} \\ \hbox{: } 0 \sim 40 \, ^{\circ} \text{C} \\ \end{array}$

6) Reading : Direct reading from the scale calibrated by 3 pump strokes

7) Colour change : White→Pale purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

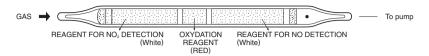
NO₂ + 3, 3'-Dimethylnaphthidine → Nitroso-compounds

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	ppm	Interference ppm 0		Coexistence
Chlorine	3	Similar stain is produced.	3	Higher readings are given.
Ozone		The accuracy of readings is not affected.	2	"
Sulphur dioxide		"	7	Lower readings are given.
Formaldehyde		"		The accuracy of readings is not affected.
Formic acid		"		"
Carbon monoxide		"		"
Carbon dioxide		"		"
Ethyl alcohol		"		"

NITROGEN OXIDE / NITROGEN DIOXIDE



1. PERFORMANCE

1) Measuring range : NO ; 10-300 ppm NO2 ; 1-40 ppm Number of pump strokes $1 (100 \text{m} \, \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : NO; 1 ppm NO2; 0.5 ppm

4) Shelf life \therefore 2 years \therefore 5) Operating temperature \therefore 10 \sim 30 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Concentration chart method 8) Colour change : NO; White → Yellowish orange NO₂; White → Pale yellowish orange

2. RELATIVE STANDARD DEVIATION

NO; RSD-low: 10% RSD-mid.: 10% RSD-high: 10% NO2; RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

NO; By reacting with an Oxidizer, NO2 is produced.

 $NO + CrO_3 + H_2SO_4 \rightarrow NO_2$

NO2; By reacting with o-Toluidine, Nitroso-o-Toluidine is produced.

$$NO_2+H_2N$$
 \longrightarrow NH_2 \longrightarrow $ON=$ \longrightarrow NC OH_3 OH_3

4. CALIBRATION OF THE TUBE

NO ; STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

NO2; PERMEATION TUBE METHOD

O2, FERMEATION TUBE METHOD

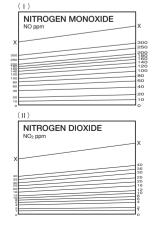
Temperature Corrected Concentration	(I) NO Temperature Correcte Concentration (ppm)						
(ppm)	20	60	100	150	200	250	300
40			34	33	32	31	30
38			32	31	30	29	28
36			30	29	28	27	26
34			28	27	26	25	24
32			26	25	24	23	22
30		25	24	23	22	21	20
28		23	22	21	20	20	19
26		20	20	20	19	19	18
24		18	18	18	17	17	16
22		16	16	16	15	15	15
20		15	15	15	14	14	14
18		13	13	13	12	12	12
16		12	12	12	11	11	11
14		10	10	10	9	9	9
12		9	9	9	8	8	8
10	8	7	7	7	6	6	6
8	6	5	5	5	5	5	5
6	5	4	4	4	4	4	4
4	3	3	3	3	3	3	3
2	2	2	2	2	2	2	2

(III) CORRECTION TABLE OF NO₂
CONCENTRATION COEXISTING WITH NO
(II) NO₂ NO₂ Actual Concentration (ppm)

Substance	Interference	ppm	Coexistence
Chlorine	Similar stain is produced.	1	Higher readings are given.

NITROGEN OXIDES CONCENTRATION CHART

Sampling Gas Volum 100ml



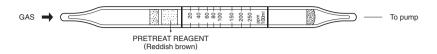
(I) NO TEMPERATURE CORRECTION TABLE

Chart			- 1	IO Tru	ie Co	ncent	ration	(ppm)			
Readings	10℃	12°C	14 °C	16°C	18 °C	20°C	22°C	24 °C	26 °C	28 °C	30°C
(ppm)	50°F	53.6°F	57.2°F	60.8°F	64.4 F	68°F	71.6°F	75.2°F	78.8°F	82.4°F	86°F
300	-	-	-	-	-	300	270	245	220	200	190
280	-	-	-	-	-	280	250	230	210	190	180
260	-	-	-	-	300	260	235	215	195	180	170
240	-	-	-	-	280	240	215	195	180	170	160
220	-	-	-	300	250	220	195	180	170	160	150
200	-	-	-	290	230	200	180	165	155	145	140
180	-	-	300	250	200	180	165	150	140	135	130
160	-	310	260	215	180	160	150	140	130	125	120
140	310	260	220	185	155	140	130	122	117	113	110
120	260	215	180	155	135	120	112	106	101	98	95
100	210	185	150	130	110	100	95	90	86	83	80
80	160	140	120	100	87	80	77	75	72	71	70
60	100	90	80	70	64	60	57	54	52	51	50
40	60	56	52	48	43	40	38	37	36	35	35
20	30	28	26	24	22	20	20	20	20	20	20
10	10	10	10	10	10	10	10	10	10	10	10

(II) NO2 TEMPERATURE CORRECTION TABLE

Chart			IN.	IO ₂ IT	ue Co	ncent	ration	(ppm)		
Readings	10°C	12 °C	14 °C	16°C	18°C	20°C	22 °C	24°C	26°C	28°C	30 °C
(ppm)	50°F	53.6°F	57.2°F	60.8°F	64.4°F	68°F	71.6°F	75.2°F	78.8°F	82.4°F	86°F
40	-	-	-	-	-	40	39	38	37	37	36
38	-	-	-	-	39	38	37	36	35	35	34
36	_	-	<u> </u>	38	37	36	35	34	34	33	33
34	-	39	37	36	35	34	33	33	32	32	31
32	39	37	35	34	33	32	31	31	30	30	30
30	36	34	33	32	31	30	30	29	29	28	28
28	33	32	31	30	29	28	28	27	27	26	26
26	31	30	29	28	27	26	26	25	25	25	24
24	28	27	26	25	25	24	24	24	23	23	22
22	26	25	24	23	23	22	22	22	21	21	21
20	24	23	22	21	21	20	20	20	19	19	19
18	21	21	20	19	19	18	18	18	18	17	17
16	19	18	18	17	17	16	16	16	16	16	15
14	17	16	15	15	14	14	14	14	14	14	14
12	14	14	13	13	12	12	12	12	12	12	12
10	12	12	11	11	10	10	10	10	10	10	10
8	10	9	9	8	8	8	8	8	8	8	8
6	7	7	6	6	6	6	6	6	6	6	6
4	4	4	4	4	4	4	4	4	4	4	4
2	2	2	2	2	2	2	2	2	2	2	2

NITROGEN OXIDES



1. PERFORMANCE

1) Measuring range 20-250 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 1 year 5) Operating temperature \therefore 5 \sim 45 °C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

NO; By reacting with an Oxidizer, NO2 is produced.

 $NO + CrO_3 + H_2SO_4 \rightarrow NO_2$

 $NO_2 + (C_6H_5)_2NH \rightarrow (C_6H_5)_2NNO$

NO2; By reacting with Diphenylamine, N-Nitroso-diphylamine is produced.

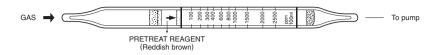
 $NO_2 + (C_6H_5)_2NH \rightarrow (C_6H_5)_2NNO$

4. CALIBRATION OF THE TUBE

NO ; STANDARD GAS CYLINDER METHOD

Substance	ppm	Interference	ppm	Coexistence
	Over 600	Dark blue stain is produced.		
Hydrogen chloride		Brownish red/Yellow/Dark blue stain is produced.		Lower readings are given.
Sulphur dioxide			100	"

NITROGEN OXIDES



1. PERFORMANCE

1) Measuring range : 100-2,500 ppm Number of pump strokes : $1(100m\ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit \therefore 10 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 5 \sim 45 °C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

NO; By reacting with an Oxidizer, NO2 is produced.

 $NO + CrO_3 + H_2SO_4 \rightarrow NO_2$

 $NO_2 + (C_6H_5)_2NH \rightarrow (C_6H_5)_2NNO$

NO2; By reacting with Diphenylamine, N-Nitroso-diphylamine is produced.

 $NO_2 + (C_6H_5)_2NH \rightarrow (C_6H_5)_2NNO$

4. CALIBRATION OF THE TUBE

NO ; STANDARD GAS CYLINDER METHOD

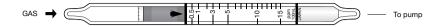
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen chloride	The bottom of discoloured layer is changed to Dark blue.	500	Higher readings are given.
Sulphur dioxide			The accurecy of readings is not affected.

(NOTE)

When the concentration of Nitrogen oxides is high (over 2,000 ppm) in the measuring range, a green ring may occur in the discoloured layer or double-stained layer may occur. The total stain length should be read, even if the stained layer gets multi-colour discolouration.

NITROGEN OXIDES



1. PERFORMANCE

1) Measuring range 1.0-30 ppm 0.5-15 ppm Number of pump strokes $1/2(50\text{m}\,\ell)$ $1(100\text{m}\,\ell)$ 2) Sampling time $1/2(50\text{m}\,\ell)$ 1 minute/1 pump stroke $1/2(50\text{m}\,\ell)$ 2. ppm $1/2(50\text{m}\,\ell)$ 2. ppm $1/2(50\text{m}\,\ell)$ 2. ppm $1/2(50\text{m}\,\ell)$ 3) Detectable limit $1/2(50\text{m}\,\ell)$ 2. ppm $1/2(50\text{m}\,\ell)$ 3. ppm $1/2(50\text{m}\,\ell)$ 4. ppm $1/2(50\text{m}\,\ell)$ 3. ppm $1/2(50\text{m}\,\ell)$ 4. ppm $1/2(50\text{m}\,\ell)$ 4. ppm $1/2(50\text{m}\,\ell)$ 5. ppm $1/2(50\text{m}\,\ell)$ 5. ppm $1/2(50\text{m}\,\ell)$ 6. ppm $1/2(50\text{m}\,$

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Pale purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

NO; By reacting with an Oxidizer, NO2 is produced.

 $NO + CrO_3 + H_2SO_4 {\rightarrow} NO_2$

NO₂ + 3, 3'-Dimethylnaphthidine→Nitroso-compound

NO2; By reacting with 3, 3'-Dimethylnaphthidine, Nitroso-compound is produced.

NO₂ + 3, 3'-Dimethylnaphthidine → Nitroso-compound

4. CALIBRATION OF THE TUBE

NO; STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Chlorine	Similar stain is produced.	1	Higher readings are given.
Hydrogen chloride	"	300	The accurecy of readings is not affected.
Sulphur dioxide	The accuracy of reading is not affected.	500	Lower readings are given.
Hydrogen sulphide	"	5	"
Ozone	"	Nox conc. X 1/10	Higher readings are given.
Hexane	"	Nox conc. X10	"
Laughing gas	"		

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times \text{Reading value}$

n-NONANE



1. PERFORMANCE

1) Measuring range Sumber of pump strokes Sampling time S-80 ppm 10-160 ppm $1(100m\ell)$ $1/2(50m\ell)$ Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and n-Nonane concentration is determined by using a conversion chart

at 1/2 and 1 pump strokes

7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

CH₃ (CH₂)₇CH₃ + Cr^{6+} + $H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

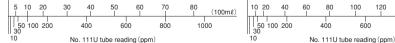
Substance	Interference	Coexistence
Alcohols	Similar or brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons	"	//

n-Nonane concentration (ppm)

(50mℓ)

1000





ORGANIC GAS CHECKER



1. PERFORMANCE

1) Measuring range : Refer to under-mentioned table

Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 20 sec./1 pump stroke

3) Shelf life : 3 years : 0 Operating temperature : 0 \sim 40 $^{\circ}$ C : 0 Reading : 0 Qualitatively

6) Colour change : Orange → Black or Dark green

2. RELATIVE STANDARD DEVIATION

RSD-low: % RSD-mid.: % RSD-high: %

3. CHEMICAL REACTION

Chromium oxide is reduced. $C_nH_m + H_2SO_4 + Cr^{6+} \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

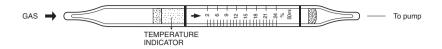
Hydrocarbons except Methane and Ethane, Alcohols, Esters, Ketones, Aromatic hydrocarbons and Hydrogen sulphide can be detected qualitatively.

6. DETECTABLE GAS CONCENTRATION TABLE

A: U.S.A B: U.K. J: JAPAN

NO.	Classification	Detective Gas	Detectable Limit of Gas Concentration (ppm)	TLV (ppm)
1	Alcohol	Methanol I.P.A. Butanol	20 50 50	200 (A, B, J) 400 (A, B, J) 50 (A, B, J)
2	Ester	Ethyl Acetate Butyl Acetate	400 200	400 (A, B, J) 150 (A, B),200 (J)
3	Ketone	Acetone M.E.K. M.I.B.K.	600 200 100	750 (A), 1,000 (B), 200 (J) 200 (A, B, J) 50 (A, J), 100 (B)
4	Aromatic	Benzene Toluene Xylene	40 5 5	10 (A, B, J) 100 (A, B, J) 100 (A, B, J)
5	Aliphatic	Propane Butane Pentane Hexane Heptane Octane Acetylene Ethylene Propylene	1,500 100 10 10 10 5 5 5 2,500 100	800 (A), 600 (B) 600 (A,B) 50 (A), 100 (B), 40 (J) 400 (A, B) 300 (A, B) —
6	Chlorination Hydrocarbon	Trichloroethylene Tetrachloroethylene	25 200	50(A, J), 100(B) 50(A, J), 100(B)
7	Mixed solvent	Gasoline (motor fuel) Kerosene Mineral Turpentine	10 8 (mg/m ₃) 8 (mg/m ₃)	300(A), 100(J)
8	Others	Ethyl acrylate Ether(ethyl) Ethylene oxide	10 10 50	5(A), 25(B) 400(A, B) 1(A), 5(B)

OXYGEN



1. PERFORMANCE

1) Measuring range : 2-24 % Number of pump strokes 1/2 (50m ℓ)

2) Sampling time : 2 minutes/1/2 pump strokes

3) Shelf life \therefore 2 years \therefore 0 \sim 40 $^{\circ}$ C

5) Reading : Direct reading from the scale calibrated by 1/2 pump strokes

6) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Oxygen reacts with alkaline pyrogallol.

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence
Hydrogen sulphide	The accuracy of reading is not affected.	2	Higher readings are given.
Nitrogen dioxide	"	2	"
Sulphur dioxide	Similar stain is produced.	2	"
Carbon dioxide	"	5	"

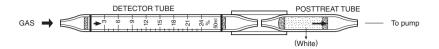
6.NOTE

- Coexistence of more than 5% of Carbon dioxide (CO₂) gives higher reading and the reading from the scale should be corrected with the following correction table.
- 2) The No.159SA and SB tubes are necessary to heat the part of the temperature indicator until the indicator discolours from Red to Purple first. In case of No.159SA, it is necessary to use direct flame such as a match or lighter to heat it up a non-hazardous area. No.159SB consists of a thin aluminum belt rolled on the indicator and chemical liquid in a small bottle. When the liquid is dropped on the aluminum fully, a chemical reaction occurs with generating smoke and temperature comes high. Therefore, the No.159SB is possible to use in a hazardous area.

CO2 CORRECTION TABLE

Tube	Corrected Concentration (%)					
Readings	CO2	CO ₂	CO ₂	CO ₂		
(%)	5%	10%	15%	20 %		
2	2.0	_	_	_		
6	6.0	2.5	_	_		
9	9.0	6.4	_	_		
12	12.0	10.0	7.0	6.0		
15	15.0	12.8	10.0	7.8		
18	18.0	16.0	14.0	12.1		
21	21.0	19.0	16.6	14.4		
24	24.0	21.2	19.6	17.9		

OXYGEN



1. PERFORMANCE

1) Measuring range 3-24% 1.5-3 % Number of pump strokes 2) Sampling time $1/2 (50m\ell)$ 1 $(100m\ell)$ 1.5 minute/1/2 pump strokes 1.5 minutes/1 pump stroke

3) Shelf life \therefore 2 years \therefore 0 \sim 45 $^{\circ}$ C

5) Reading : Direct reading from the scale calibrated by 1/2 pump stroke

6) Colour change : Black→White

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Titanium trichloride is oxidized and Titanium oxide is produced. $O_2 + TiCI_3 \rightarrow TiO_2$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Sulphur dioxide	The accuracy of readings is not affected.	The accuracy of readings is not affected.		
Carbon dioxide	"	"		
Nitrogen dioxide	"	"		
Hydrogen sulphide	"	"		

(NOTE)

When the concentration is below 3%, 1 pump stroke can be used to determine the lower concentration with following formula.

Actual concentration = $1/2 \times$ Reading value

OXYGEN · CARBON DIOXIDE



1. PERFORMANCE

1) Measuring range:

Oxygen 2-10%Carbon dioxide 1-20%Number of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 8 minutes/1 pump stroke

3) Detectable limit : Oxygen 0.1 % Carbon dioxide 0.2 %

4) Shelf life : 1.5 years5) Operating temperature $: 10 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation:

Oxygen Necessary (See "TEMPERATURE CORRECTION TABLE")

Carbon dioxide No temperature correction is necessary.

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change:

Oxygen White→Brown Carbon dioxide Pink→Yellow

2. RELATIVE STANDARD DEVIATION

 Oxygen
 RSD-low : 10 %
 RSD-mid : 10 %
 RSD-high : 10 %

 Carbon dioxide
 RSD-low : 10 %
 RSD-mid : 10 %
 RSD-high : 10 %

3. CHEMICAL REACTION

Oxygen: Oxygen reacts with alkaline pyrogallol. Carbon dioxide: CO₂ + 2KOH → K₂CO₃ + H₂O

4. CALIBRATION OF THE TUBE

Oxygen · Carbon dioxide STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Ordinary combustion gases do not affect the reading value and discolouration.

6. CONCENTRATION CORRECTION

On account of interaction of indication between Carbon dioxide and Oxygen, the true concentration is calculated by the following equations.

[EXAMPLE]

At the measuring temperature of $30\,^\circ\text{C}$, when Oxygen indication is $5\,\%$ (temperature corrected value $5.3\,\%$) and Carbon dioxide indication is $10\,\%$, each true concentration is calculated as bellow.

CO2 %=	10 —	5.3 × 10	- =	10	_	0.53	=	9.47
O2 % =	5.3 —	5.3 × 10	- =	5.3	_	0.53	=	4.77

Accordingly true concentration: Carbon dioxide 9.47 %

Carbon dioxide 9.47 % Oxygen 4.77 %

Tube	Corrected Oxygen Concentration (%)			
Readings (%)	10 ℃ (50 °F)	20 ℃ (68*F)	30 °C (86 °F)	40 °C (104 °F)
10.0	9.7	10.0	10.3	10.5
9.0	8.7	9.0	9.3	9.5
8.0	7.7	8.0	8.3	8.5
7.0	6.7	7.0	7.3	7.5
6.0	5.7	6.0	6.3	6.5
5.0	4.7	5.0	5.3	5.5
4.0	3.9	4.0	4.2	4.5
3.0	2.9	3.0	3.1	3.3
2.0	2.0	2.0	2.0	2.1

OZONE



1. PERFORMANCE

1) Measuring range 100-1,000 ppm S0-500 ppm Number of pump strokes $1/2(50\text{m}\,\ell)$ $1(100\text{m}\,\ell)$ 2) Sampling time $1/2(50\text{m}\,\ell)$ $1/2(50\text{m}\,\ell)$ 1/2

2) Sampling time $1 \min(1 \text{ pur})$ 1 minute/1 pur 1 som 2 per $100 \text{m} \ell$ 2 years 100 som 2 years 100 som 3 Operating temperature 100 som 3 per $100 \text{ m} \ell$ 4 years 100 som 3 per 100 som 4 years 100 som 4 years 100 som 5 years 100 som 4 years $100 \text{ s$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Dark blue → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Indigo is oxidized and Isatin is produced.

$$\begin{array}{cccc}
O & H & O & O \\
C & C & N & +2O_3 \longrightarrow 2 & C & C & +2O_2
\end{array}$$

$$\begin{array}{ccccc}
O & C & C & C & C & C & C & C & C
\end{array}$$

4. CALIBRATION OF THE TUBE

COLOURIMETRY METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Nitrogen dioxide		The top of discoloured layer becomes unclear and higher readings are given.
Chlorine	"	"

(NOTE)

In case of 1/2 pump strokes, following formula is available for actual concentration.

Actual concentration = $2 \times$ Reading value

OZONE



1. PERFORMANCE

1) Measuring range : 10-100 ppm 5-50 ppm 2.5-25 ppm Number of pump strokes $1/2(50 \text{m} \ell)$ $1(100 \text{m} \ell)$ $2(200 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit $1 \text{ ppm}(100\text{m}\ell)$ 4) Shelf life 2 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Blue→Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Indigo is oxidized and Isatin is produced.

$$\bigcup_{\substack{C \\ N \\ H}} \bigcup_{C=C} \bigcup_{\substack{C \\ C \\ O}} \bigcup_{M} +2O_3 \longrightarrow 2 \bigcup_{\substack{N \\ M}} \bigcup_{C=O+2O_2}$$

4. CALIBRATION OF THE TUBE

COLOURIMETRY METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	ppm Interference Coexistence	
Nitrogen dioxide	10	Similar stain is produced.	The top of discoloured layer becomes unclear and higher readings are given.

(NOTE)

- 1) In case of 1/2 pump strokes, following formula is available for actual concentration. Actual concentration = 2 × Reading value
- 2) In case of 2 pump strokes, following formula is available for actual concentration. Actual concentration = 1/2 × Reading value

OZONE



1. PERFORMANCE

1) Measuring range : 0.15-3.0 ppm 0.05-1.0 ppm 0.025-0.5 ppm Number of pump strokes $1(100m\ell)$ $3(300m\ell)$ $6(600 \text{m} \ell)$

2) Sampling time : 3 minutes/3 pump strokes 3) Detectable limit : 0.01 ppm (600mℓ)

4) Shelf life 2 years 5) Operating temperature : 0 ~ 40 ℃

6) Reading : Direct reading from the scale calibrated by 3 pump strokes

7) Colour change : Blue → White

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Indigo is oxidized and Isatin is produced.

4. CALIBRATION OF THE TUBE

COLOURIMETRY METHOD

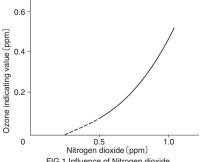
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Nitrogen dioxide FIG.1	Similar stain is produced.	0.5	The top of discoloured layer becomes unclear and higher readings are given.
Chlorine	"	10	"
Oxidant	Similar stain is produced and this has same sensitivity with Ozone.		

(NOTE)

In case of 1 or 6 pump strokes, following formula is available for actual concentration.

Actual concentration = Reading value $\times \frac{S}{\text{Number of strokes}}$





1. PERFORMANCE

1) Measuring range 50-1,000 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 10 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \, ^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by n-Hexane at 1 pump stroke

and Pentane concentration is determined by using a conversion chart.

7) Colour change : Orange → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low:10 % RSD-mid.: 10 % RSD-high: 5 %

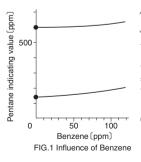
3. CHEMICAL REACTION

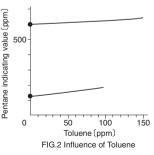
 $CH_3 (CH_2)_3 CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

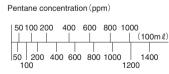
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	ppm	Coexistence
Alcohols	Similar stain is produced.	6%	Higher readings are given.
Ketones	"	11	"
Esters	"	11	"
Aromatic hydrocarbons FIG.1,2	"		The bottom of the discoloured layer is stained to Black and higher readings are given.
Aliphatic Hydrocarbons (more than C ₃)	"		Higher readings are given.

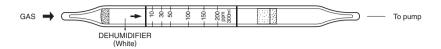






No.113SB Tube reading (ppm)

PENTYL ACETATE



1. PERFORMANCE

1) Measuring range 10-200 ppmNumber of pump strokes $3(300\text{m}\ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit \therefore 4 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 3 pump strokes

8) Colour change : Pale yellow→Pale blue (more than 20 ppm)

→ Dark brown (20 ppm or less)

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3CO_2(CH_2)_4CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

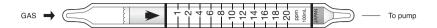
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar or Brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	The accuracy of readings is not affected.	Whole reagent is changed to Brown, but if the maximum end point of the stain is discernable, the accuracy of readings is not affected.
Aliphatic hydrocarbons (more than C ₃)	"	"
Halogenated hydrocarbons		"

TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)						
Readings (ppm)	10 °C (50 °F)	20 °C (68 °F)	30°C (86°F)	40 °C (104 °F)			
200	240	200	160	120			
150	180	150	130	100			
100	120	100	80	70			
50	60	50	40	30			
30	36	30	24	18			
10	10	10	8	6			

PENTYL AMINE



1. PERFORMANCE

1) Measuring range 2-22 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke and Pentyl amine concentration is determined by using a conversion chart at 1

pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

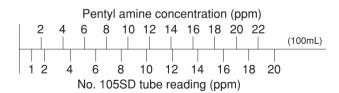
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. CH₃ (CH₂)₃CH₂NH₂ + H₃PO₄ → (R₁NH₃)₃PO₄

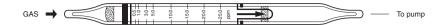
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



PHENOL



1. PERFORMANCE

1) Measuring range 0.5-25.0 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit 0.3 ppm 4) Shelf life 2 years 5) Operating temperature $10 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Pale yellow→Pale light brown (Pale brown)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

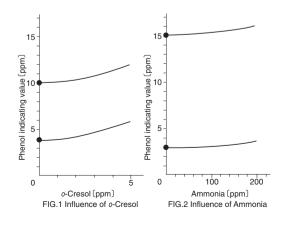
Phenol is oxidized and the Polymer is produced. $C_6H_5OH + C_6H_5O \cdot \rightarrow (C_6H_5O)_n$

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

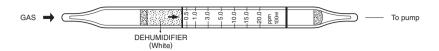
Substance		Interference	ppm	Coexistence
Other phenols	FIG.1	Similar stain is produced.	2.5	Higher readings are given.
Ammonia	FIG.2	White stain is produced.	200	Discolouration of gas inlet side is faded and higher readings are given.
Aliphatic amines		"	50	"
Aromatic amines		Blue stain is produced.	50	Two layers discolouration of Pale brown and Blue are produced and higher readings are given.



TEMPERATURE CORRECTION TABLE

Today	0						
_ Tube	Corrected Concentration (ppm)						
Readings	0℃	0℃ 10℃ 20℃		30℃	40℃		
(ppm)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)		
25.0	31.2	27.8	25.0	21.8	18.8		
20.0	24.5	22.3	20.0	17.5	15.0		
15.0	18.4	16.7	15.0	13.1	11.3		
10.0	12.3	11.1	10.0	8.8	7.5		
5.0	6.1	5.6	5.0	4.4	3.8		
3.0	3.7	3.3	3.0	2.6	2.3		
1.0	1.2	1.1	1.0	0.9	0.8		
0.5	0.5	0.5	0.5	0.5	0.5		

PHOSGENE



1. PERFORMANCE

1) Measuring range : 0.5-20 ppm 0.1 - 4.0 ppmNumber of pump strokes 1(100ml) 5 (500ml) 2) Sampling time : 1 minute/1 pump stroke 3) Detectable limit $0.05 \text{ ppm} (500 \text{m} \ell)$

4) Shelf life 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10^{\circ}$ C)

0 ~ 40 °C 5) Operating temperature

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation Direct reading from the scale calibrated by 1 pump stroke 7) Reading

8) Colour change

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Nitro-benzyl pyridine, urea derivative is produced. This urea derivative reacts with Benzyl aniline and dyestuff is produced.

: White→Red

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Nitrogen dioxide FIG.	Yellow stain is produced.	100	Yellow stain is produced and higher readings are given.
Chlorine		5	A stained layer at the side of gas inlet is bleached out and higher readings are given.
Hydrogen chloride		10	"
Sulphur dioxide		0.2%	"

(NOTE)

When the concentration is below 0.5 ppm, 5 pump strokes can be used to determine the lower concentration with following formula.

Actual concentration = $1/5 \times$ Temperature corrected value

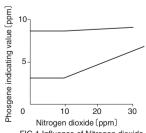
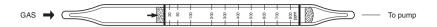


FIG.1 Influence of Nitrogen dioxide

TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)					
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)	
20	13.5	17.0	20.0	21.8	23.0	
15	10.5	12.8	15.0	16.5	17.5	
10	7.0	8.6	10.0	11.0	11.8	
5	3.5	4.3	5.0	5.5	5.8	
3	3.0	3.0	3.0	3.0	3.0	

PHOSPHINE IN ACETYLENE



1. PERFORMANCE

1) Measuring range 20-800 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 3 minutes/1 pump stroke with orifice

3) Detectable limit : 5 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale blue → Reddish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride (II) and Cupric sulphate (II), Mercuric phosphorus chloride and Cupric phosphate are produced respectively.

Moreover, Cupric phosphide reacts with Acetylene and Copper acetylene is produced.

 $PH_3 + HgCI_2 + H_2O \rightarrow Hg_3P_2 \cdot 3HgCI_2 \cdot 3H_2O$

 $PH_3 + CuSO_4 \rightarrow Cu_3P_2 + H_2O$ $Cu_3P_2 + HC \equiv CH \rightarrow CuC \equiv CCu$

4. CALIBRATION OF THE TUBE

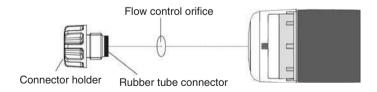
ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

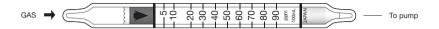
Substance	Interference	ppm	Coexistence
Hydrogen sulphide	Black stain is produced.	10	Higher readings are given.
Arsine	Dim yellow stain is produced.	10	"

6.NOTE

- 1) In case that Acetylene does not exist, lower readings are given.
- 2) A flow control orifice (an extra option) is required to attach as shown in the following drawing.



PHOSPHINE IN ACETYLENE



1. PERFORMANCE

1) Measuring range 5-90 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 3 minutes/1 pump stroke with orifice

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale blue → Yellowish brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride (II) and Cupric sulphate (II), Mercuric phosphorus chloride and Cupric phosphate are produced respectively.

Moreover, Cupric phosphide reacts with Acetylene and Copper acetylene is produced.

 $PH_3 + HgCI_2 + H_2O \rightarrow Hg_3P_2 \cdot 3HgCI_2 \cdot 3H_2O$

 $PH_3 + CuSO_4 \rightarrow Cu_3P_2 + H_2O$ $Cu_3P_2 + HC \equiv CH \rightarrow CuC \equiv CCu$

4. CALIBRATION OF THE TUBE

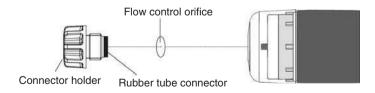
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

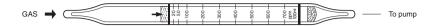
Substance	Interference	ppm	Coexistence
Hydrogen sulphide	Black stain is produced.	10	Higher readings are given.
Arsine	Dim yellow stain is produced.	10	"

6.NOTE

- 1) In case that Acetylene does not exist, lower readings are given.
- 2) A flow control orifice (an extra option) is required to attach as shown in the following drawing.



PHOSPHINE



1. PERFORMANCE

1) Measuring range 40-1400 ppm 20-700 ppmNumber of pump strokes $1/2 (50 \text{m} \ell) (100 \text{m} \ell)$ 2) Sampling time $1/2 (50 \text{m} \ell) (100 \text{m} \ell)$ 3) Detectable limit $10 \text{ ppm} (100 \text{m} \ell)$

4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride (II), Mercuric phosphorus chloride is produced.

 $PH_3 + HgCI_2 + H_2O \rightarrow Hg_3P_2 \cdot 3HgCI_2 \cdot H_2O$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Substance Interference		opm Coexistence	
Arsine	Brown stain is produced.	30	Higher readings are given.	
Hydrogen selenide	"	50	"	
Hydrogen sulphide	"	40	"	

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times \text{Reading value}$

PHOSPHINE



1. PERFORMANCE

1) Measuring range : 0.5-10.0 ppm 0.25-5.0 ppm 1-20 ppm Number of pump strokes 1(100ml) $2(200m\ell)$ $1/2(50 \text{m}\ell)$

: 0 ~ 40 °C

2) Sampling time : 1 minute/1 pump stroke 3) Detectable limit : $0.1 \text{ ppm} (200 \text{m} \ell)$ 4) Shelf life : 1 year

5) Operating temperature 6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Yellow → Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride (II), Hydrogen chloride is produced and PH indicator is discoloured. $PH_3 + 3HgCI_2 \rightarrow P(HgCI)_3 + 3HCI$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Ammonia		20	A stained layer at the side of gas inlet is bleached out and lower readings are given.
Hydrogen sulphide	Similar stain is produced.	50	Higher readings are given.
Mercaptans	"		"

(NOTE)

When the concentration is below 0.5ppm, 2 pump strokes can be used to determine the lower concentration with the following formula;

Actual concentration = $1/2 \times$ Reading value

When the concentration is over 10.0ppm, 1/2 pump strokes can be used to determine the higher concentration with the following formula;

Actual concentration = $2 \times$ Reading value

PHOSPHINE



1. PERFORMANCE

1) Measuring range 5-150 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minute/1 pump stroke

3) Detectable limit : 3 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride, Mercuric phosphorus chloride is produced.

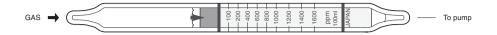
 $PH_3 + HgCI_2 + H_2O \rightarrow Hg_3P_2 \cdot 3HgCI_2 \cdot 3H_2O$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substance	%	Interference	Coexistence
Hydrogen selenide	5	Brown stain is produced.	Higher readings are given.
Hydrogen sulphide	5	"	"

PHOSPHINE-high range



1. PERFORMANCE

2) Sampling time : 1 minute/1 pump stroke
3) Detectable limit : 5 ppm (100mℓ)
4) Shelf life : 3 years
5) Operating temperature : 0 ~ 40 °C

6) Temperature compensation : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Potassium iodate is reduced. $PH_3 + KIO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Nitrogen monoxide	The accuracy of reading is not affected.	Lower readings are given.
Nitrogen dioxide	"	Higher readings are given.
Hydrogen sulphide	Oranage stain is produced.	"
Sulphur dioxide	"	"

(NOTE)

In case of 1/2 pump strokes, following formula is available for actual concentration.

Actual concentration = $2 \times$ Reading value

PHOSPHINE



1. PERFORMANCE

1) Measuring range 200-3,000 ppm 400-6,000 ppm Number of pump strokes $1(100 \mathrm{m}\ell)$ $1/2 (50 \mathrm{m}\ell)$

2) Sampling time $\begin{array}{c} : \ 1 \ minute/1 \ pump \ stroke \\ : \ 5 \ ppm \ (100m \, \ell) \\ : \ 3 \ years \\ : \ 0 \ \sim 40 \, ^{\circ} C \\ \end{array}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Potassium iodate is reduced. $PH_3 + KIO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

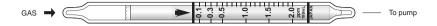
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence
Carbon dioxide	The accuracy of readings is not affected.	100	The accuracy of readings is not affected.
Methyl bromide	"	less than	"
Hydrogen cyanide	"	3	The stained layer at the side of the gas inlet is bleached out and higher readings are given.
Ammonia	"	0.6	"

(NOTE)

In case of 1/2 pump strokes, the following formula is available for the actual concentration. Actual concentration $= 2 \times \text{Reading value}$

PHOSPHINE



1. PERFORMANCE

1) Measuring range 0.05-1.0 ppm Number of pump strokes 0.1-2.0 ppm 0.05-1.0 ppm 0.05-1.0

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride (II), Hydrogen chloride is produced and PH indicator is discoloured. $PH_3 + 3HgCI_2 \rightarrow P(HgCI)_3 + HCI$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Hydrogen sulphide	Similar stain is produced.	Higher readings are given.
Hydrogen selenide	"	"
Mercaptans	"	"
Arsine	"	"
Hydrogen cyanide	Whole reagent is changed to Red.	"
Sulphur dioxide	4	Whole reagent is changed to Pale red, but Purplish red stain indicates Phosphine concentration.

(NOTE)

When the concentration is below 0.5 ppm, 2 pump strokes can be used to determine the lower concentration with the following formula.

Actual concentration = $1/2 \times$ Reading value

α-PINENE



1. PERFORMANCE

1) Measuring range 20-300 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : The tube scale is calibrated based on Styrene at 1 pump stroke and

 α -Pinene concentration is determined by using a conversion chart

at 1 pump stroke
7) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 10%

3. CHEMICAL REACTION

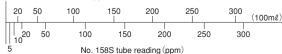
A polymer of α -Pinene is produced by Sulphuric acid.

4. CALIBRATION OF THE TUBE

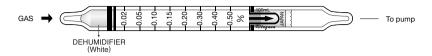
GAS CHROMATOGRAPHY

Substance	Interference	ppm	Coexistence
Acrylonitrile	The accuracy of readings is not affected.	400	Lower readings are given.
Butadiene	Similar stain is produced and higher readings are given.	5	Uneven discolouration is produced and higher readings are given.
Formaldehyde	"	15	Yellowish orange stain is produced and higher readings are given.
Acetaldehyde	"	350	Similar stain is produced and higher readings are given.
Methyl alcohol	The accuracy of readings is not affected.	0.35%	Pale discolouration is produced and higher readings are given.
Ethyl alcohol	"	0.18%	"
Ethyl acetate	"	700	"
Butyl acetate	"	700	"





PROPANE



1. PERFORMANCE

1) Measuring range 0.02-0.5%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 4 minutes/1 pump stroke 3) Detectable limit : 0.002 % (20 ppm)

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Orange→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3CH_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

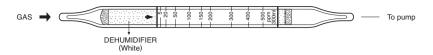
STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Toluene	Similar stain is produced.	Higher readings are given.
Hexane	"	"
Trichloroethylene	"	"
Ethyl alcohol	The accuracy of readings is not affected.	

Tube		Corrected Concentration (ppm)				
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)	
0.50	0.40	0.45	0.50	0.60	_	
0.40	0.33	0.37	0.40	0.50	0.65	
0.30	0.26	0.28	0.30	0.35	0.42	
0.25	0.21	0.23	0.25	0.28	0.32	
0.20	0.17	0.19	0.20	0.22	0.26	
0.15	0.13	0.14	0.15	0.16	0.19	
0.10	0.09	0.10	0.10	0.11	0.13	
0.05	0.05	0.05	0.05	0.05	0.05	

1-PROPANOL



1. PERFORMANCE

1) Measuring range 20-300 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl cellosolve at 3 pump strokes and

1-Propanol concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

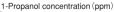
 $CH_3CH_2CH_2OH + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Aliphatic hydrocarbons (more than C ₃)	,	"
Ketones	"	"
Aromatic hydrocarbons	Whole reagent is changed to Brown.	"
Esters	"	"
Halogenated hydrocarbons	"	"





No. 190U tube reading (ppm)

PROPIONIC ACID



1. PERFORMANCE

7) Colour change

1) Measuring range 3-50 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years 5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Propionic acid concentration is determined by using a conversion chart at

1 pump stroke.: Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

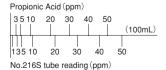
3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. CH₃CH₂COOH + NaOH → CH₃CH₂CO₂Na + H₂O

4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

Substance	ppm	Interference	ppm	Coexistence
Sulphur dioxide		Similar stain is produced.	HCO₂H conc. × 1/20	Higher readings are given.
Nitrogen dioxide	300	"	10	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	HCO ₂ H conc. ×2	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	5	"
Acetic acid		Similar stain is produced.		"



PROPYL ACETATE



1. PERFORMANCE

1) Measuring range 0.01-1.4%Number of pump strokes $2(200m\ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit : 10 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \text{ °C}$

6) Reading Graduations printed on the tube are calibrated by Methyl ethyl ketone at 2 pump strokes and Propyl acetate concentration is determined by using a conversion chart.

7) Colour change : Orange→Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Dichromate is reduced.

 $CH_3CO_2 (CH_2)_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

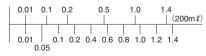
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		3%	Whole reagent is changed to Brown.
Propane		0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons	Similar stain is produced.		Higher readings are given.

Propyl acetate (%)



No.139SB Tube reading (%)

PROPYL ACETATE



1. PERFORMANCE

1) Measuring range \therefore 20-1,000 ppm Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 10 ppm 4) Shelf life : 2 years 5) Operating temperature : $10 \sim 40^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_3CO_2 (CH_2)_2CH_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

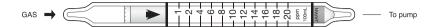
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stains or Blue stains are produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons (over C ₃)		Whole reagent is changed to Pale brown, but if the maximum end point of stained layer is discernable, the accuracy of readings is not affected.
Halogenated hydrocarbons		"

Tube	Corrected Concentration (ppm)				
Readings (ppm)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)	
1000	_	1000	480	320	
800	_	800	420	280	
600	1450	600	340	240	
400	850	400	240	160	
200	380	200	120	80	
100	200	100	60	40	
50	100	50	30	20	
20	45	20	10	6	

PROPYL AMINE



1. PERFORMANCE

1) Measuring range 1-20 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke

and Propyl amine concentration is determined by using a conversion chart at 1

pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

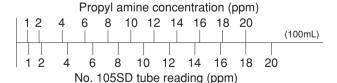
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. CH₃CH₂CH₂NH₂ + H₃PO₄ → (R₁NH₃)₃PO₄

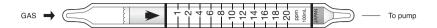
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



Di-iso-PROPYL AMINE



1. PERFORMANCE

1) Measuring range 1-16 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke

and Di-iso-Propyl amine concentration is determined by using a conversion

chart at 1 pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 5 % RSD-high : 5 %

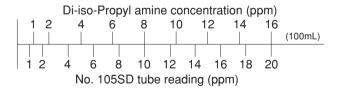
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. [(CH₃)₂CH]₂NH + H₃PO₄→ (R₂NH₂)₃PO₄

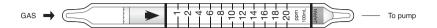
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



Di-n-PROPYL AMINE



1. PERFORMANCE

1) Measuring range 1-14 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke

and Di-n-Propyl amine concentration is determined by using a conversion

chart at 1 pump stroke.
7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

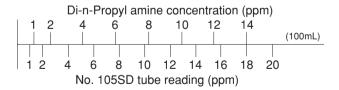
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $[CH_3 (CH_2)_2]_2NH + H_3PO_4 \rightarrow (R_2NH_2)_3PO_4$

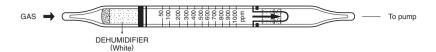
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance Interference		Coexistence	
Amines	Similar stain is produced.	Higher readings are given.	



PROPYLENE



1. PERFORMANCE

1) Measuring range 50-1,000 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 10 ppm 4) Shelf life : 2 years 5) Operating temperature : $10 \sim 40 \, \text{°C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Dark blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Molybdate is reduced and molybdeum blue is produced. CH₃CH = CH₂ + PdSO₄ + (NH₄)₂MoO₄ → Mo₃O₈

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Carbon monoxide	45	Similar stain is produced.	200	Higher readings are given.
Acetylene		Brown stain is produced.	50	"
Ethyene		Similar stain is produced.		"
Hydrogen sulphide	10	Black stain is produced.	50	Double-layer stain is produced and higher readings are given.

Tube	Corrected Concentration (ppm)				
Readings (ppm)	10 °C (50 °F)	20 ℃ (68 °F)	30 ℃ (86 °F)	40 °C (104 °F)	
1000	870	1000	_	_	
800	760	800	900	1000	
600	540	600	660	720	
400	380	400	420	440	
200	200	200	200	200	

PROPYLENE GLYCOL

1. PERFORMANCE

1) Measuring range 5-50 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethylene oxide at 3 pump strokes and

Propylene glycol concentration is determined by using a conversion chart

at 1 pump stroke

7) Colour change : Pale pink→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Formic acid is produced and PH indicator is discoloured.

 $CH_3CH(OH)CH_2OH + 2HIO_4 \rightarrow 3HCHO + 2HIO_3 + H_2O$

 $HCHO + HIO_4 + H_2SO_4 \rightarrow HCOOH + HIO_3$ $HCOOH + NaOH \rightarrow Na(HCOO) + H_2O$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

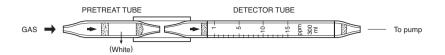
Substance	Interference	Coexistence
Aldehydes	Similar stain is produced.	Higher readings are given.
Hydrogen sulphide	Pale yellow stain is produced.	"
Sulphur dioxide	"	"





No. 122SC tube reading (ppm)

PROPYLENE OXIDE



1. PERFORMANCE

7) Colour change

1) Measuring range 3-70 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading . The tube scale is calibrated based on Ethylene oxide at 3 pump strokes and

Propylene oxide concentration is determined by using a conversion chart

at 1 pump stroke ∴ Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Formic acid is produced and PH indicator is discoloured.

 $CH_3CH_OCH_3 + 2HIO_4$ → 3HCHO + 2HIO₃ HCHO + HIO₄ + H₂SO₄ → HCOOH + HIO₃ HCOOH + N₈OH → N₈(HCOO) + H₂O

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Aldehydes	Similar stain is produced.	Higher readings are given.
Hydrogen sulphide	Pale yellow stain is produced.	"
Sulphur dioxide	"	"

Propylene oxide concentration (ppm)



No. 122SC tube reading (ppm)

PROPYLENE OXIDE



1. PERFORMANCE

1) Measuring range 1.0-5.0% 0.05-3.0% Number of pump strokes $1/2(50m\ell)$ $1(100m\ell)$ 2) Sampling time $1.5 \pm 1.5 \pm 1$

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Orange → Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3CHCH_2O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

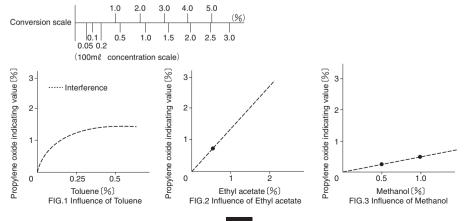
5. INTERFERENCE AND CROSS SENSITIVITY

(50mℓ actual value)

Substance		Interference	%	Coexistence
Aromatic hydrocarbons	FIG.1	Similar stain is produced.		Higher readings are given.
Esters	FIG.2	"		"
Ketones		"		"
Alcohols	FIG.3	"		"
Halogenated hydrocarbons		Whole reagent is changed to Brown.	0.5	"

(NOTE)

In case of 1/2 pump strokes, following conversion scale is available for actual concentration.



PROPYLENE OXIDE



1. PERFORMANCE

1) Measuring range 0.2-5.0 ppmNumber of pump strokes $2(200\text{m}\ell)$

2) Sampling time : 4 minute s/2 pump strokes

3) Detectable limit : 0.1 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $5 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 2 pump strokes

7) Colour change : Yellow→Pale pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Aldehydes generated through the pretreat tube are detected.

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Ethylene oxide	Similar stain is produced.		Higher readings are given.
Formaldehyde		0.5	"

Tube	Corrected Concentration (ppm)				
Readings (ppm)	5℃ (41°F)	10 °C (50 °F)	20 ℃ (68 °F)	30 ℃ (86*F)	40 ℃ (104 °F)
5.0	7.9	6.3	5.0	4.5	4.0
4.5	7.1	5.6	4.5	4.1	3.6
4.0	6.3	5.0	4.0	3.6	3.2
3.5	5.5	4.4	3.5	3.1	2.8
3.0	4.7	3.8	3.0	2.7	2.4
2.5	3.8	3.1	2.5	2.2	1.9
2.0	3.0	2.5	2.0	1.8	1.5
1.5	2.2	1.9	1.5	1.3	1.1
1.0	1.4	1.3	1.0	0.86	0.71
0.8	1.1	1.0	0.80	0.68	0.56
0.6	0.83	0.75	0.60	0.50	0.40
0.4	0.55	0.50	0.40	0.32	0.25
0.2	0.27	0.25	0.20	0.16	0.12

n-PROPYL MERCAPTAN



1. PERFORMANCE

4) Shelf life \therefore 2 years \therefore 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Reading : The tube scale is calibrate based on Methyl mercaptan at 1 pump stroke and

the tube has the same sensitivity for n-Propyl mercaptan.

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercuric chloride, Hydrogen chloride is produced and PH indicator is discoloured. RSH + HgCI₂→RS(HgCI) + HCI

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

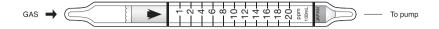
Substance	Interference	Coexistence
Arsine	Similar stain is produced.	Higher readings are given.
Hydrogen selenide	"	"
Phosphine	"	"
Hydrogen sulphide	"	"
Hydrogen cyanide	Whole reagent is changed to Red.	"
Sulphur dioxide		Whole reagent is changed to Pale red,but Pink stain indicates Mercaptans conc.

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times$ Reading value

PYRIDINE



1. PERFORMANCE

1) Measuring range 0.5-10 ppmNumber of pump strokes $1(100\text{m}\,\ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : The tube scale is calibrated based on Ammonia at 1 pump stroke and
Pyridine concentration is determined by deviding the tube reading by 2

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

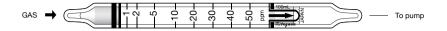
By reacting with Phosphoric acid, PH indicator is discoloured. $C_5H_5N + H_3PO_4 \rightarrow (R_2NH_2)_3PO_4$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.

SILANE



1. PERFORMANCE

 $\begin{array}{cccc} \text{1) Measuring range} & \text{1.1-50 ppm} & 0.5\text{-}25 \text{ ppm} \\ \text{Number of pump strokes} & 1 \left(100\text{m}\ell\right) & 2 \left(200\text{m}\ell\right) \\ \text{2) Sampling time} & \text{5 minutes/1 pump stroke} \\ \end{array}$

3) Detectable limit : $0.3 \text{ ppm} (200 \text{m} \ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10 \, ^{\circ}\mathrm{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (0 \sim 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Red

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Mercuric chloride, Hydrogen chloride is liberated and PH indicator is discoloured. SiH₄ + HgCI₂→ HCI

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence		
Phosphine	2 layers of greyish white and red stain are produced.	20	Higher readings are given.		
Arsine	2 layers of dark brown and red stain are produced.	50	"		
Disilane	A Similar stain is produced.	2	"		
Diborane	"	20	"		
Ammonia		100	Lower readings are given.		
Sulphur dioxide		Whole reagent is changed to Orange. But the accuracy of readings is not affected by Sulphur dioxide if the maximum end point of stained layer is discernable.			
IPA			Not affected.		
Hydrogen			"		
Hydrogen chloride		less than 250	Not affected.		
Dichlorosilane		less than 200	"		

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration.

Actual concentration = Temperature corrected value \times 1/2

Tube	Correcte	ed Concentration	
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20-40 °C (68-104 °F)
50	38	45	50
40	30	35	40
30	22	26	30
20	14	17	20
10	7	8	10
5	4	4	5
2	2	2	2
1	1	1	1

STYRENE



1. PERFORMANCE

1) Measuring range Number of pump strokes Sampling time S

4) Shelf life 3 years $0 \sim 40^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 10%

3. CHEMICAL REACTION

A polymer of Styrene is produced by sulphuric acid.

$$\begin{array}{ccc} CH: CH_2 & \longleftarrow CH - CH_2 - \\ H_2SO_4 + & \longrightarrow & \bigcirc \end{array}$$

4. CALIBRATION OF THE TUBE

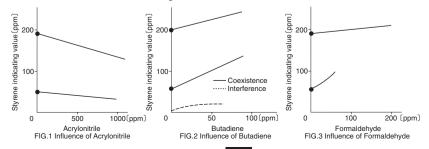
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substanc	е	ppm	Interference	ppm	Coexistence			
Acrylonitrile	FIG.1		The accuracy of readings is not affected.	400	Lower readings are given.			
Butadiene	FIG.2	3	Similar stain is produced.	5	Uneven discolouration is produced and higher readings are given.			
Formaldehyde	FIG.3		"	15	Yellowish orange stain is produced and higher readings are given.			
Acetaldehyde			"	350	Similar stain is produced and higher readings are given.			
Methyl alcohol			The accuracy of readings is not affected.	0.35%	Pale discolouration is produced and higher readings are given.			
Ethyl alcohol			"	0.18%	"			
Ethyl acetate			"	700	"			
Butyl acetate			"	700	"			

(NOTE)

In case of 2 pump strokes, following formula is available for actual concentration. Actual concentration = $1/2 \times \text{Reading value}$.



STYRENE



1. PERFORMANCE

1) Measuring range 2-100 ppm 1-50 ppm Number of pump strokes $2(200 \text{m} \, \ell)$ 4 $(400 \text{m} \, \ell)$ 2) Sampling time 2 minutes/2 pump strokes 3) Detectable limit 2 constant/2 pump strokes 2 constant/2 pump strokes

4) Shelf life 3 years $0 \sim 40^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

A polymer of Styrene is produced by fume sulphuric acid. C₆H₅C₂H₃ + H₂SO₄ ⋅ nSO₃ → Yellow compound

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acrylonitrile	The accuracy of readings is not affected.	400	Lower readings are given.
Butadiene	Pale orange stain is produced.	5	Higher readings are given.
Formaldehyde	The accuracy of readings is not affected.	15	"
Acetaldehyde	Orange stain is produced.	350	"
Methyl alcohol	The accuracy of readings is not affected.	3500	"
Ethyl alcohol	"	1800	"
Ethyl acetate	"	700	"
Butyl acetate	"	700	"

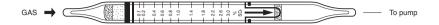
(NOTE)

In case of 4 pump strokes, following formula is available for actual concentration. Actual concentration = $1/2 \times \text{Reading value}$.

TEMPERATURE CORRECTION TABLE

Temperature: To correct for temperature, multiply the tube reading by the following factors.

1					_					
Temperature (℃)	0	1	2	3	4	5	6	7	8	9
Correction Factor	0.80	0.81	0.82	0.83	0.84	0.85	0.86	0.87	0.88	0.89
Temperature (℃)	10	11	12	13	14	15	16	17	18	19
Correction Factor	0.90	0.91	0.92	0.93	0.94	0.95	0.96	0.97	0.98	0.99
Temperature (℃)	20	21	22	23	24	25	26	27	28	29
Correction Factor	1.00	1.01	1.02	1.03	1.04	1.05	1.06	1.07	1.08	1.09
Temperature (°C)	30	31	32	33	34	35	36	37	38	39
Correction Factor	1.10	1.11	1.12	1.13	1.14	1.15	1.16	1.17	1.18	1.19
Temperature (°C)	40									
Correction Factor	1.20]								



1. PERFORMANCE

1) Measuring range 0.1-3.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke
3) Detectable limit : 0.01 % (100 ppm)
4) Shelf life : 3 years

5) Operating temperature $0 \sim 40^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Dichromate is reduced. $SO_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen sulphide		400	Brown stain is produced and higher readings are given.

Tube	Corrected Concentration (%)					
Readings	0°C	10 °C	20 °C	30 °C	40 °C	
(%)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)	
3.0	4.4	3.7	3.0	2.6	2.2	
2.6	3.8	3.2	2.6	2.3	2.0	
2.2	3.2	2.7	2.2	2.0	1.7	
1.8	2.6	2.2	1.8	1.6	1.4	
1.4	2.0	1.7	1.4	1.3	1.1	
1.0	1.4	1.2	1.0	0.9	0.9	
0.8	1.2	1.0	0.8	0.7	0.6	
0.6	0.8	0.7	0.6	0.6	0.5	
0.4	0.5	0.5	0.4	0.4	0.3	
0.2	0.3	0.2	0.2	0.2	0.2	
0.1	0.1	0.1	0.1	0.1	0.1	



1. PERFORMANCE

1) Measuring range 0.02-0.3 % Number of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke
3) Detectable limit : 0.001 % (10 ppm)
4) Shelf life : 3 years

4) Shelf life 3 years $0 \sim 40^{\circ}$ C $\sim 40^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Potassium iodate is reduced. $SO_2 + KIO_3 + H_2O \rightarrow I_2$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substance	Interference	ppm	Coexistence
Hydrogen sulphide	The accuracy of readings is not affected.	100	Higher readings are given.



1. PERFORMANCE

1) Measuring range 20-300 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit ∴ 3 ppm 4) Shelf life ∴ 2 years 5) Operating temperature ∴ 0 ~ 40 °C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Purple → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

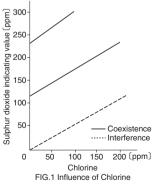
3. CHEMICAL REACTION

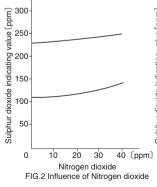
By reacting with alkali, PH indicator is discoloured. $SO_2 + 2NaOH \rightarrow Na_2SO_3 + H_2O$

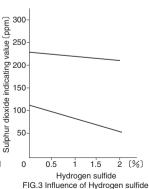
4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substa	ance	ppm	Interference	ppm	Coexistence
Chlorine	FIG.1		Similar stain is produced.	SO ₂ conc. × 1/5	Higher readings are given.
Nitrogen dio	xide FIG.2	100	Pink stain is produced.	"	"
Hydrogen si	ulphide FIG.3		The accuracy of readings is not affected.	SO ₂ conc. × 100	Lower readings are given.









1. PERFORMANCE

1) Measuring range 1-60 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 0.5 ppm4) Shelf life 3 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

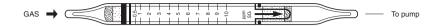
3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. SO₂ + 2NaOH → Na₂SO₃ + H₂O

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	ppm	Interference	ppm	Coexistence
Chlorine		Pale pink stain is produced.	SO ₂ conc.	Higher readings are given.
Nitrogen dioxide	20	"	SO ₂ conc. X 1	Higher readings with indiscernible maximum end point of stained layer are given.



1. PERFORMANCE

1) Measuring range Number of pump strokes 1 $(100 \text{m} \ell)$ 2 $(200 \text{m} \ell)$ 2 Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit : $0.1 \text{ ppm} (200 \text{m} \ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; 0 \sim 10 $^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Humidity compensation
7) Reading
: Necessary (See "R.H. CORRECTION COEFFICIENT TABLE")
: Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. SO₂ + 2NaOH → Na₂SO₃ + H₂O

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Nitrogen dioxide	Pale pink stain is produced.	3	Higher readings with indiscernible maximum end point of stained layer are given.
Hydrogen chloride	"		Higher readings are given.

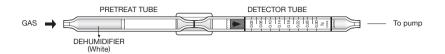
(NOTE)

- This detector tube is affected by abmbient relative humidity, therefore, it is necessary to compensate the reading
 of gas detector tube with the following formula and correction confficient table.
 Actual concentration = Reading Value (ppm) × Correction Coefficient
- 2) In case of 2 pump strokes, following formula is available for actual concentration. Actual concentration = 1/2 × Reading value corrected with above formula

R.H. CORRECTION COEFFICIENT TABLE

	10	30	50	60	70	80	90
coefficient of correction	0.9	0.95	1.0	1.1	1.2	1.3	1.4

SULPHUR DIOXIDE IN FLUE GAS



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \text{1.002-0.3 \%} \\ \text{Number of pump strokes} \end{array}$

2) Sampling time : 1 minute/1 pump stroke
3) Detectable limit : 0.001 % (10 ppm)
4) Shelf life : 3 years

5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$ 6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Potassium iodate is reduced. $SO_2 + KIO_3 + H_2O \rightarrow I_2$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substance	Interference	ppm	Coexistence
Hydrogen sulphide	The accuracy of readings is not affected.	100	Higher readings are given.

SULPHUR DIOXIDE IN CARBON DIOXIDE



1. PERFORMANCE

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Blue purple → White

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

lodine is reduced. $SO_2 + I_2 + 2H_2O \rightarrow 2HI + H_2SO_4$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen Sulphide	Whole reagent is changed to Pink.	0.5	Higher readings are given.
Nitrogen Dioxide	Whole reagent is changed to Black.	0.5	Lower readings are given.
Ammonia	The accuracy of readings is not affected.	1.0	"

Tube	Corrected Concentration (ppm)				
Readings (ppm)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 ℃ (104 °F)
25	40.0	32.5	25.0	22.5	19.0
20	28.0	24.0	20.0	17.7	15.5
15	20.0	17.5	15.0	13.5	12.0
10	12.5	11.3	10.0	9.3	8.5
8	10.0	9.0	8.0	7.5	7.0
6	7.5	6.8	6.0	5.8	5.5
4	4.4	4.2	4.0	3.8	3.6
2	2.0	2.0	2.0	2.0	2.0
1	1.0	1.0	1.0	1.0	1.0
0.5	0.5	0.5	0.5	0.5	0.5

SULPHURIC ACID



1. PERFORMANCE

1) Measuring range $0.5-5 \text{ mg/m}^3$ Number of pump strokes $5(500\text{m}\ell)$

2) Sampling time : 100 seconds/5 pump strokes

3) Detectable limit 0.2 mg/m^3 4) Shelf life 0.2 years5) Operating temperature 0.2 mg/m^3

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 5 pump strokes

8) Colour change : Yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

The pH indicator is discoloured.

4. CALIBRATION OF THE TUBE

ION CHROMATOGRAPHY.

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Hydrogen chloride	Similar stain is produced.	Higher readings are given.
Hydrogen flouride	"	"
Nitrogen dioxide	"	"
Chlorine	"	"
Nitric acid	"	"
Hydrogen sulphide	The accuracy of readings is not affected.	The accuracy of readings is not affected.
Sulphur dioxide	"	"

Temperature	5℃ (41*F)	10 °C (50 °F)	20 °C (68 °F)	30°C (86°F)	40 °C (104 °F)
Correction Factor	2.0	1.5	1.0	0.8	0.6



1. PERFORMANCE

1) Measuring range : 10-300 ppm 5-150 ppm Number of pump strokes $1/2(50 \text{m} \ell)$ $1(100m\ell)$ 2) Sampling time 2 minutes/1 pump stroke

3) Detectable limit $1 \text{ ppm} (100 \text{m} \ell)$

4) Shelf life 2 years (Necessary to store in refrigerated conditions; $0 \sim 10 \, ^{\circ}\mathrm{C}$)

5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation Direct reading from the scale calibrated by 1 pump stroke 7) Reading : Yellow → Red

8) Colour change

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. $CCI_2 = CCI_2 + PbO_2 + H_2SO_4 \rightarrow HCI$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence
Vinyl chloride		Similar stain is produced.	Higher readings are given.
Hydrogen chloride	FIG.1	"	"
1,2-Dichloroethylene	FIG.2	"	"
Trichloroethylene		"	"
Chlorine		Pale red stain is produced.	"

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times$ Temperature corrected value

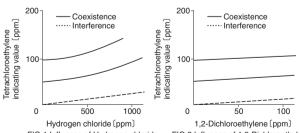
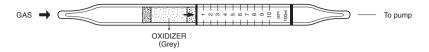


FIG.1 Influence of Hydrogen chloride FIG.2 Influence of 1,2-Dichloroethylene

TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
150	_	172	150	138	134
100	144	116	100	92	88
50	70	56	50	46	44
30	40	36	30	28	26
20	22	21	20	19	18
10	10	10	10	10	10



1. PERFORMANCE

1) Measuring range Number of pump strokes 1-10 ppm $1 (100 \text{m} \ell)$ 4 $(400 \text{m} \ell)$ 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit : $0.1 \text{ ppm} (400 \text{m} \ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10 \, ^{\circ}\text{C}$)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pale orange → Blueish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. $CI_2C = CCI_2 + PbO_2 + H_2SO_4 \rightarrow HCI$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

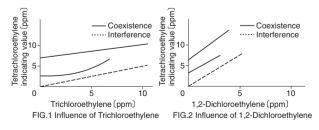
5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Trichloroethylene	FIG.1	Similar stain is produced.	2	Higher readings are given.
1,2-Dichloroethylene	FIG.2	"	2	"
Hydrogen chloride		"	2	"
Vinyl chloride		"	40	"

(NOTE)

In case of 4 pump strokes, following formula is available for the actual concentration.

Actual concentration = $1/5 \times$ Temperature corrected value



TEMPERATURE CORRECTION TABLE Tube | Corrected Concentration (ppm)

Readings (ppm)	0 °C (32 °F)	10 ℃ (50 °F)	20 ℃ (68 °F)	30 °C (86 °F)	40 °C (104 °F)
10	12.5	11.3	10.0	9.4	8.7
9	11.9	10.5	9.0	8.5	7.9
8	9.4	8.7	8.0	7.5	7.0
7	8.2	7.6	7.0	6.6	6.2
6	7.0	6.5	6.0	5.6	5.2
5	5.8	5.4	5.0	4.8	4.3
4	4.6	4.3	4.0	3.8	3.5
3	3.5	3.3	3.0	2.8	2.6
2	2.3	2.2	2.0	1.9	1.7
1	1.1	1.1	1.0	1.0	0.9



1. PERFORMANCE

1) Measuring range Number of pump strokes 1 $(100m\ell)$ 2 $(200m\ell)$ 2) Sampling time 2.3.5 minutes/1 pump stroke

3) Detectable limit $0.08\%(200\text{m}\ell)$ 4) Shelf life 2 years5) Operating temperature $0 \sim 40\%$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced.

 $CI_2C = CCI_2 + I_2O_5 + H_2SO_4 \cdot nSO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	%	Coexistence
Trichloroethylene	Yellow stain is produced.	0.2	Higher readings are given.
1,1,1-Trichloroetane	Orange stain is produced.	0.3	"
1,2-Dichloroethylene	"	0.1	"
Vinyl chloride	"	0.02	"
Aromatic hydrocarbons	Blackish brown stain is produced.	_	The accuracy of readings is not affected.
Carbon monoxide	Brownish-red stain is produced.	0.05	Higher readings are given.
Carbon tetrachloride	The accuracy of readings is not affected.		The accuracy of readings is not affected.



1. PERFORMANCE

1) Measuring range 125-1,250 ppm Number of pump strokes $1(100\text{m}\ell)$ $1/2(50\text{m}\ell)$ $1/2(50\text{m}\ell)$ $1/2(50\text{m}\ell)$ $1/2(50\text{m}\ell)$

3) Detectable limit : 5 ppm $(100 \text{m} \ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10 \, ^{\circ}\mathrm{C}$)

5) Operating temperature : $5 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change

. Mirect reading from the scale canonated by 1 pump shok.

Yellow→Red

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. $CI_2C = CCI_2 + PbO_2 + H_2SO_4 \rightarrow HCI$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Trichloroethylene	Similar stain is produced.	10	Higher readings are given.
1,2-Dichloroethylene	"	10	"
1,1,1-Trichloroetane		less than 300	The accuracy of readings is not affected.

(NOTE)

In case of 1/2 pump strokes, the following formula is available for the actual concentration.

Actual concentration = $2.5 \times$ Reading value

TEMPERATURE CORRECTION TABLE

Temperature; To correct for temperature, multiply the tube reading by the following factors.

Temperature (℃)	1	2	3	4	5	6	7	8	9	10
Correction Factor	_	_	_	_	1.40	1.36	1.32	1.28	1.24	1.20
Temperature (℃)	11	12	13	14	15	16	17	18	19	20
Correction Factor	1.18	1.16	1.14	1.12	1.10	1.08	1.06	1.04	1.02	1.00
Temperature (℃)	21	22	23	24	25	26	27	28	29	30
Correction Factor	0.99	0.98	0.97	0.96	0.95	0.94	0.93	0.92	0.91	0.90
Temperature (℃)	31	32	33	34	35	36	37	38	39	40
Correction Factor	0.89	0.88	0.87	0.86	0.85	0.84	0.83	0.82	0.81	0.80

TETRAETHOXYSILANE



1. PERFORMANCE

1) Measuring range Summer of pump strokes 1 (100m ℓ) 2 (200m ℓ) 2 Sampling time 2.5-200 ppm 5-80 ppm 1 (100m ℓ) 2 (200m ℓ) 2 in inutes/2 pump strokes

3) Detectable limit $1 \text{ ppm } (200\text{m}\ell)$ 4) Shelf life 3 years5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $Si(OC_2H_5)_4 + CrO_3 + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Phosphine	Similar stain is produced.	5	Higher readings are given.
Isopropyl alcohol	"	7	"
Ethyl alcohol	"	10	"
Silane			
Trichloroethylene	The accuracy of readings is not affected.		
Tetrachloroethylene			

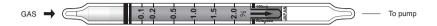
(NOTE)

In case of 1 pump stroke, following formula is available for the actual concentration.

Actual concentration = $2.5 \times$ Temperature corrected value

Tube		orrected	Concentr	ation (ppr	n)	
Readings (ppm)	0°C (32°F)	5℃ (41°F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
80	110	100	90	80	72	64
60	80	75	70	60	54	45
40	56	52	47	40	33	28
20	27	25	23	20	16	14
10	13	12	11	10	8	7
5	6	5	5	5	5	4

TETRAHYDROFURAN



1. PERFORMANCE

: 2.0-5.0 % 0.2-3.0% 1) Measuring range Number of pump strokes $1/2(50 \text{m} \ell)$ 1(100ml) 2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 20 ppm 4) Shelf life 3 years 5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation: Necessary (See "TEMPERATURE CORRECTION TABLE")

Graduations printed on the tube are calibrated by Acetone at 1 pump stroke and 7) Reading

Tetrahydrofuran is determined by using a conversion chart.

8) Colour change : Orange → Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 15 % RSD-mid.: 10 % RSD-high: 5 %

3. CHEMICAL REACTION

Chromium oxide is reduced.

CH₂-CH₂
$$O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$$

CH₂-CH₂

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm Coexistence	
Alcohols	Similar stain is produced.		Higher readings are given.
Aromatic hydrocarbons FIG	1 //		"
Ketones	"		"
Esters FIG	2 //		"
Halogenated hydrocarbons	Whole reagent is changed to Pale brown.	0.5%	"

Tetrahydrofuran concentration (%)

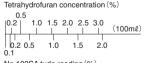
(50mℓ)

2.03.04.05.0

1.0 1.5 2.0

No.102SA tude reading (%)

0.2 0.5



No.102SA tude reading (%) ····· Interference

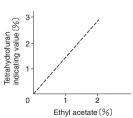
3

2

0

indicating value [%]

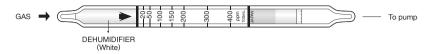
Tetrahydrofuran



Toluene (%) FIG.1 Influence of Toluene FIG.2 Influence of Ethyl acetate

Converstion	Corrected Concentration (%)				
Value (%)	0°C (32°F)	10 °C (50 °F)	20 °C (68 °F)	30 ℃ (86 °F)	40 °C (104 °F)
5.0	6.0	5.3	5.0	4.8	4.5
4.0	4.8	4.3	4.0	3.8	3.6
3.0	3.6	3.2	3.0	2.8	2.7
2.5	3.0	2.7	2.5	2.4	2.3
2.0	2.4	2.1	2.0	1.9	1.8
1.5	1.8	1.6	1.5	1.4	1.4
1.0	1.1	1.1	1.0	1.0	0.9
0.5	0.6	0.5	0.5	0.5	0.5

TETRAHYDROFURAN



1. PERFORMANCE

1) Measuring range 20-400 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 10 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

CH₂-CH₂
$$O + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$$

CH₂-CH₂

4. CALIBRATION OF THE TUBE

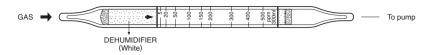
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Esters	Whole reagent is changed to Pale brown.	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons (more than C ₃)	"	"
Halogenated hydrocarbons	"	"

Tube	Corrected Concentration (ppm)					
Readings (ppm)	10 ℃ (50°F)	20 ℃ (68*F)	30 °C (86 °F)	40 °C (104 °F)		
500	680	500	390	320		
400	540	400	320	250		
300	400	300	240	190		
200	260	200	160	130		
150	190	150	120	90		
100	130	100	80	60		
50	60	50	40	30		
20	30	20	20	10		

TETRAHYDROTHIOPHENE



1. PERFORMANCE

1) Measuring range \therefore 4-100 ppm Number of pump strokes $3(300 \text{m} \ell)$

2) Sampling time : 4.5 minutes/3 pump strokes

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl cellosolve at 3 pump strokes and

Tetrahydrothiophene concentration is determined by using a conversion chart

at 3 pump strokes

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced. $C_4H_8S + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

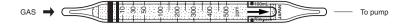
Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Ethers	"	"
Aliphatic hydrocarbons (more than C ₃)	,	"
Ketones	"	"
Aromatic hydrocarbons	Whole reagent is changed to Brown.	"
Esters	"	"
Halogenated hydrocarbons	"	"

Tetrahydrothiophene concentration (ppm)



No. 190U tube reading (ppm)

TOLUENE



1. PERFORMANCE

1) Measuring range 10-500 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit 3 ppm 4) Shelf life 3 years 5) Operating temperature 0 ~ 40 °C

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Indine pentoxide is reduced. $C_6H_5CH_3 + I_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

Substance	Interference	ppm	Coexistence
Benzene	Similar stain is produced.		Higher readings are given.
Xylene	"		"
Ethyl benzene	"		"
Hexane	Whole layer is discoloured to Pale brown.	0.1%	Higher readings with indiscernible maximum end point of stained layer are given.
Methyl alcohol	The accuracy of readings is not affected.	1%	Higher readings are given.

TOLUENE



1. PERFORMANCE

1) Measuring range : 2-100 ppm Number of pump strokes $1(100m\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit 1 ppm 4) Shelf life 3 years 5) Operating temperature 0 ~ 40 ℃

Necessary $(0 \sim 10^{\circ}\text{C})$ (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation:

7) Reading Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

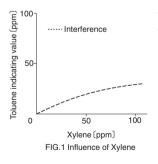
3. CHEMICAL REACTION

Iodine pentoxide is reduced. $C_6H_5CH_3 + I_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	ppm	Coexistence
Aromatic hydrocarbons FIG.1,FIG.2	Similar stain is produced.		Higher readings are given.
Hexane		1000	Double-layer stain (pale brown and brown) is produced, but if the maximum end point of the stained layer is discernable, the accuracy of readings is not affected.



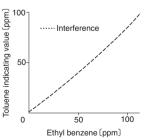
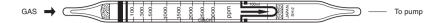


FIG.2 Influence of Ethyl benzene

TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)					
Readings (ppm)	0℃ (32°F)	5℃ (41°F)	10-40 °C (50-104 °F)			
100	150	125	100			
100			100			
80	120	100	80			
60	85	75	60			
40	55	45	40			
30	40	35	30			
20	28	23	20			
10	13	10	10			
5	6	5	5			
2	2	2	2			

TOLUENE



1. PERFORMANCE

1) Measuring range : 100-3000 ppm Number of pump strokes : $1(100 \text{me} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 10 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. $C_6H_5CH_3 + I_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

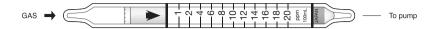
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference		Coexistence
Benzene	Dark green stain is produced.		Higher readings are given.
Ethyl benzene	Similar stain is produced.		"
Xylene	"		"
Hexane	Whole reagents is changed to Pale brown.		"
Methanol	The accuracy of readings is not affected.	1%	"

Tube		Corrected Concentration (ppm)						
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)			
3000	4000	3500	3000	2800	2600			
2500	3300	2900	2500	2300	2200			
2000	2700	2300	2000	1900	1800			
1500	2000	1700	1500	1400	1300			
1000	1300	1100	1000	970	940			
500	500	500	500	500	500			
300	300	300	300	300	300			
100	100	100	100	100	100			

105SDG o-TOLUIDINE



1. PERFORMANCE

1) Measuring range : 2-22 ppm Number of pump strokes 1 (100m l)

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 4) Shelf life 3 years 5) Operating temperature : 15 ~ 25 °C

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke

and o-Toluidine concentration is determined by using a conversion chart at 1

pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

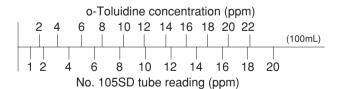
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. C_6H_4 (CH₃) (NH₂) + H₃PO₄ \rightarrow (R₁NH₃)₃PO₄

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



p-TOLUIDINE



1. PERFORMANCE

1) Measuring range 2-20 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}$ C

6) Reading : Graduations printed on the tube are calibrated by Ammonia at 1 pump stroke

and p-Toluidine concentration is determined by using a conversion chart at 1

pump stroke.

7) Colour change : Pale purple → Pale Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

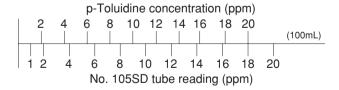
3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. C_6H_4 (CH₃) (NH₂) + H₃PO₄ \rightarrow (R₁NH₃)₃PO₄

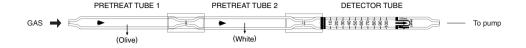
4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.



1,1,2-TRICHLOROETHANE



1. PERFORMANCE

1) Measuring range 10-100 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 5 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

Chlorine is produced by decomposing with an Oxidizer.

By reading between this Chlorine and 3, 3'-Dimethylnaphthidine, Nitroso-compound is produced.

 $CI_2CHCH_2CI + CrO_3 + H_2SO_4 \cdot nSO_3 {\rightarrow} CI_2$

CI₂ + 3, 3'-Dimethylnaphthidine → Nitroso-compound

4. CALIBRATION OF THE TUBE

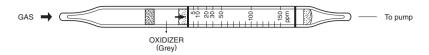
DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Nitrogen oxides	Similar stain is produced.		Higher readings are given.
Halogens	"		"
Halogenated hydrocarbons	"		"
Hexane	The accuracy of readings is not affected.	100	Lower readings are given.

Tube	Corrected Concentration (ppm)				
Readings (ppm)	0 °C (32 °F)	10 °C (50 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)
100	81	89	100	114	129
90	71	80	90	105	119
80	60	71	80	96	109
70	51	61	70	87	97
60	43	52	60	76	85
50	34	43	50	65	72
40	26	34	40	54	59
30	19	25	30	42	45
20	13	15	20	29	31
10	6	8	10	13	15

TRICHLOROETHYLENE



1. PERFORMANCE

1) Measuring range : 10-300 ppm 5-150 ppm Number of pump strokes $1/2(50 \text{m}\ell)$ $1(100m\ell)$ 2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit $1 \text{ ppm} (100 \text{m} \ell)$

4) Shelf life 2 years (Necessary to store in refrigerated conditions; $0 \sim 10 \, ^{\circ}\mathrm{C}$)

5) Operating temperature $0 \sim 40 \,^{\circ}\text{C}$

Necessary (See "TEMPERATURE CORRECTION TABLE") 6) Temperature compensation Direct reading from the scale calibrated by 1 pump stroke 7) Reading : Yellow → Red

8) Colour change

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. $CI_2C = CHCI + PbO_2 + H_2SO_4 \rightarrow HCI$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

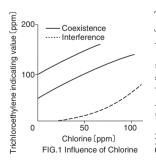
5. INTERFERENCE AND CROSS SENSITIVITY

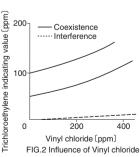
Substance		Interference	Coexistence
Vinyl chloride	FIG.2	Similar stain is produced.	Higher readings are given.
Hydrogen chloride		"	"
1,2-Dichloroethylene		"	"
Tetrachloroethylene		"	"
Chlorine	FIG.1	Pale red stain is produced.	

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

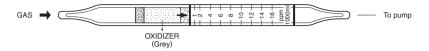
Actual concentration = $2 \times$ Temperature corrected value





Tube	C	Corrected Concentration (ppm)			
Readings	0°C	10 °C	20 °C	30 °C	40 °C
(ppm)	(32°F)	(50°F)	(68°F)	(86°F)	(104°F)
150	_	162	150	144	142
100	120	108	100	96	94
50	58	53	50	48	46
30	34	32	30	29	28
20	20	20	20	20	20

TRICHLOROETHYLENE



1. PERFORMANCE

1) Measuring range : 2.3-36.8 ppm 1-16 ppm 0.2-3.2 ppm Number of pump strokes $1/2(50m\ell)$ $1(100m\ell)$ $4(400m\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : $0.1 \text{ ppm} (400 \text{m} \ell)$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale orange → Blueish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. $Cl_2C = CHCl + PbO_2 + H_2SO_4 \rightarrow HCl$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	ppm	Coexistence
Tetrachloroethylene	FIG.1	Similar stain is produced.	2	Higher readings are given.
1,2-Dichloroethylene	FIG.2	"	2	"
Hydrogen chloride		"	2	"
Vinyl chloride		"	20	"

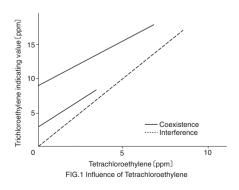
(NOTE)

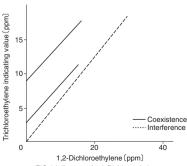
In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2.3 \times \text{Reading value}$

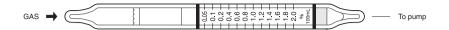
In case of 4 pump strokes, following formula is available for the actual concentration.

Actual concentration = $1/5 \times$ Reading value





TRICHLOROETHYLENE



1. PERFORMANCE

1) Measuring range 0.05-2.0 % Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 3 minutes/1 pump stroke

3) Detectable limit 0.01%4) Shelf life 2 years 5) Operating temperature $0 \sim 40\%$

6) Temperature compensation : Necessary (0 ~ 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $CI_2C = CHCI + I_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

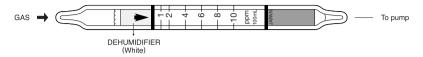
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence
Tetrachloroethylene	Brown stain is produced.	0.2	Higher readings are given.
1,1,1-Trichloroetane	Orange stain is produced.	0.2	"
1,2-Dichloroethylene	"	0.1	"
Vinyl chloride	"	0.02	"
Aromatic hydrocarbons	Dark brown stain is produced.	_	The accuracy of readings is not affected.
Carbon monoxide	Brownish-red stain is produced.	0.05	Higher readings are given.
Carbon tetrachloride	The accuracy of readings is not affected.		The accuracy of readings is not affected.

Tube	Corrected Concentration (%)			
Readings (%)	0°C	,5°C	10℃	20 ℃-40 ℃
(%)	(32°F)	(41°F)	(50°F)	(68-104°F)
2.0	_	2.7	2.3	2.0
1.8	_	2.4	2.1	1.8
1.6	_	2.1	1.9	1.6
1.4	2.4	1.8	1.6	1.4
1.2	1.9	1.55	1.4	1.2
1.0	1.5	1.3	1.15	1.0
0.8	1.2	1.0	0.9	0.8
0.6	0.9	0.75	0.7	0.6
0.4	0.6	0.5	0.45	0.4
0.2	0.3	0.25	0.22	0.2
0.1	0.15	0.13	0.11	0.1
0.05	0.06	0.05	0.05	0.05

TRIETHYL AMINE



1. PERFORMANCE

1) Measuring range 2-20 ppm 1-10 ppm Number of pump strokes 2) Sampling time $1/2(50m\ell)$ 1 $(100m\ell)$ 2 3) Detectable limit $1/2(50m\ell)$ 2 2-20 ppm $1/2(50m\ell)$ 1 $1/2(50m\ell)$ 2 1 minute/1 pump stroke $1/2(50m\ell)$ 2 2-20 ppm $1/2(50m\ell)$ 1 $1/2(50m\ell)$ 2 $1/2(50m\ell)$ 3 $1/2(50m\ell)$

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $2(C_2H_5)_3N + H_3PO_4 \rightarrow [(C_2H_5)_3NH_2]_2HPO_4$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

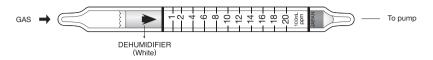
Substance	Interference	Coexistence
Ammonia	Similar stains are produced and higher readings are given.	Higher readings are given.
Other amines	"	"

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $2 \times \text{Reading value}$

TRIMETHYL AMINE



1. PERFORMANCE

1) Measuring range 1-20 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit 0.3 ppm4) Shelf life 3 years5) Operating temperature $0 \sim 40 \text{ °C}$

6) Reading : The tube scale is calibrated based on Diethyl amine at 1 pump stroke and the

tube has the same sensitivity for Trimethyl amine.

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Phosphoric acid, PH indicator is discoloured. 2(CH₃)₃N + H₃PO₄→[(CH₃)₃NH]₂HPO₄

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

Substance	Interference	Coexistence
Ammonia	Similar stains are produced	Higher readings are given.
Other amines	"	"

1,2,4-TRIMETHYL BENZENE



1. PERFORMANCE

1) Measuring range 20-250 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Reading : Graduations printed on the tube are calibrated by Ethyl acetate at 1 pump stroke

and 1,2,4-trimethyl benzene is determined by using a conversion chart.

7) Colour change : Yellow→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

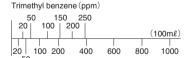
Chromium oxide is reduced. $C_6H_3 (CH_3)_3 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

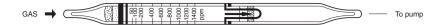
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Brown stain is produced.	Higher readings are given.
Esters	"	"
Ketones	"	"
Aromatic hydrocarbons	"	"
Aliphatic hydrocarbons (more than C ₃)		Double-layer stain is produced. If the dark brown stain is clear, the readings can be obtained by it.
Halogenated hydrocarbons		"



No. 111U Tube reading (ppm)

2,2,4-TRIMETHYL PENTANE



1. PERFORMANCE

1) Measuring range Number of pump strokes 2) Sampling time 200-4,000 ppm 100-1,400 ppm $1/2(50m\ell)$ $1(100m\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years 5) Operating temperature : $15 \sim 25$ °C

6) Reading : Graduations printed on the tube are calibrated by n-Hexane at 1 pump stroke and 2, 2, 4-Trimethyl Pentane concentration is determined by using a conversion

chart at 1/2 and 1 pump strokes.

7) Colour change : Orange → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

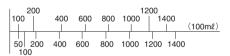
3. CHEMICAL REACTION

Chromium oxide is reduced $CH_3 (CH_2)_5 CH_3 + Cr^{6+} + H_2 SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

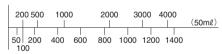
GAS CHROMATOGRAPHY

(for 1 pump stroke) 2, 2, 4-Trimethyl Pentane concentration (ppm)



No.113SB Tube reading (ppm)

(for 1/2 pump strokes) 2, 2, 4-Trimethyl Pentane concentration (ppm)



No.113SB Tube reading (ppm)

n-UNDECANE



1. PERFORMANCE

1) Measuring range 10-140 ppmNumber of pump strokes $1(100\text{m}\ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 2 years
5) Operating temperature : $15 \sim 25$ °C

6) Reading : The tube scale is calibrated based on Ethyl acetate at 1 pump stroke and

n-Undecane concentration is determined by using a conversion chart

at 1 pump stroke
7) Colour change : Yellow→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

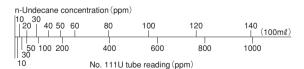
Chromium oxide is reduced.

 $CH_3 (CH_2)_9 CH_3 + Cr^{6+} + H_2 SO_4 \rightarrow Cr^{3+}$

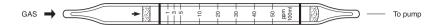
4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

Substance	Interference	Coexistence			
Alcohols	Similar or brown stain is produced.	Higher readings are given.			
Esters	"	"			
Ketones	"	"			
Aromatic hydrocarbons	"	"			
Aliphatic hydrocarbons	Whole reagent is discoloured to Pale brown.	If the maximum end point of stained layer is discernable, the accuracy of readings is not affected.			
Halogenated hydrocarbons	"	"			



n-VALERIC ACID



1. PERFORMANCE

7) Colour change

1) Measuring range 3-70 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years 5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and n-Valeric acid concentration is determined by using a conversion chart at

1 pump stroke.∴ Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

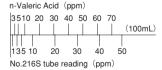
3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. CH₃ (CH₂) ₃CO₂H + N₃OH → CH₃ (CH₂) ₃COONa + H₂O

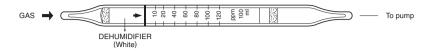
4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

Substance	ppm	Interference	Coexistence			
Sulphur dioxide		Similar stain is produced.	Higher readings are given.			
Nitrogen dioxide	300	//	The maximum end point of stained layer is indiscernible.			
Hydrogen chloride		Pink stain is produced.	Higher readings are given.			
Chlorine		Blueish yellow stain is produced.	"			
Acetic acid		Similar stain is produced.	"			



VINYL ACETATE



1. PERFORMANCE

1) Measuring range 10-120 ppm S-60 ppm Number of pump strokes $1(100\text{m}\ell)$ $2(200\text{m}\ell)$ 2) Sampling time 1.5 minutes/1 pump stroke

3) Detectable limit $1 \text{ ppm}(200\text{m}\ell)$ 4) Shelf life 2 years5) Operating temperature $0 \sim 40 \text{ }^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced. $CH_3CO_2CH = CH_2 + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetic acid			The accuracy of readings is not affected.
Ethylene	Pale brown or pale blue stain is produced.	150	Lower readings are given.
Alcohols	Similar stain is produced.		Higher readings are given.
Ethers	"		"
Aliphatic hydrocarbons (more than C ₃)	Whole layer is discoloured to Pale brown.		"
Aromatic hydrocarbons	"		"
Halogenated hydrocarbons	"		"
Esters	"		"
Ketones	"		"

(NOTE)

In case of 2 pump strokes, following formula is available for the actual concentration.

Actual concentration = $1/2 \times \text{Reading value}$

VINYL CHLORIDE



1. PERFORMANCE

1) Measuring range 0.05-1.0%Number of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 2 minutes/1 pump stroke

3) Detectable limit : 10 ppm4) Shelf life : 3 years5) Operating temperature $: 0 \sim 40 \text{ °C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Brownish orange → Brownish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Chromium oxide is reduced.

 $CH_2 = CHCI + Cr^{6+} + H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

Substance	Interference	ppm	Coexistence
Acetylene	Whole layer is discoloured to Brown.	3%	Whole reagent is discoloured to Brown and higher readings are given.
Propane	"	0.2%	"
Other organic gases or vapours except Halogenated hydrocarbons		50	Higher readings are given.

VINYL CHLORIDE



1. PERFORMANCE

1) Measuring range 5-500 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : 2 ppm

4) Shelf life : 1.5 years (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : White→Reddish orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Chlorine is produced. This Chlorine reacts with o-Toluidine and red Orthoquinone is produced.

$$CH_2 = CHCI + PdO_2 \rightarrow CI_2$$

$$H_2N \longrightarrow CH_3 \qquad CH_3 \qquad$$

4. CALIBRATION OF THE TUBE

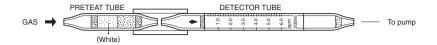
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Hydrogen chloride	Similar stain is produced.		Higher readings are given.
Chlorine	"		"
Other halogens or Halogenated hydrocarbons	"		"

Tube		Corrected Concentration (ppm)											
Readings (ppm)	0 °C (32 °F)	5 °C (41 °F)	10 °C (50 °F)	15 ℃ (59 °F)	20 °C (68 °F)	30 °C (86 °F)	40 °C (104 °F)						
500	_	_	820	600	500	430	400						
400	_	850	610	470	400	340	310						
300	_	640	440	350	300	250	230						
200	750	400	300	230	200	160	140						
100	340	200	150	120	100	80	70						
50	150	100	70	60	50	40	30						
20	50	40	30	25	20	15	10						
10	25	20	15	10	10	8	6						
5	15	10	10	5	5	5	5						

VINYL CHLORIDE



1. PERFORMANCE

1) Measuring range : 0.4-12.0 ppm 0.2-6.0 ppm 0.1-3.0 ppm Number of pump strokes $1\,(100\mathrm{m}\,\ell)$ $2\,(200\mathrm{m}\,\ell)$ $4\,(400\mathrm{m}\,\ell)$

2) Sampling time : 3 minutes/2 pump strokes 3) Detectable limit : 0.05 ppm (400mℓ)

4) Shelf life : 3 years 5) Operating temperature : $0 \sim 40^{\circ}$ C

6) Temperature compensation : Necessary (0 \sim 20 °C) (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Greenish yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By decomposing with an Oxidizer, Hydrogen chlorine is produced and PH indicator is discoloured.

 $CH_2 = CHCI + CrO_3 + H_2SO_4 \rightarrow HCI$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetylene		4%	Lower readings are given.
Ethylene		400	"
Hydrogen chloride		Less than 500	The accuracy of readings is not affected.
Chlorine		Less than 50	"

(NOTE)

In case of 1 or 4 pump strokes, following formula is available for the actual concentration.

Actual concentration = Temperature corrected value × 2/Number of strokes

Tube	Co	Corrected Concentration (ppm)									
Readings (ppm)	0 °C (32 °F)	5℃ (41°F)	10 ℃ (50 °F)	15 °C (59 °F)	20-40°C (68-104°F)						
6.0	9.4	8.3	7.4	6.8	6.0						
5.0	7.6	6.8	6.1	5.6	5.0						
4.0	6.1	5.5	5.0	4.5	4.0						
3.0	4.6	4.0	3.6	3.3	3.0						
2.0	3.0	2.7	2.4	2.2	2.0						
1.0	1.5	1.4	1.3	1.2	1.0						



1. PERFORMANCE

1) Measuring range : 1.7-33.8 mg/ ℓ Number of pump strokes : 1(100m ℓ)

From 10 % RH (20 $^{\circ}$ C = 68 $^{\circ}$ F) to 100 % RH (32 $^{\circ}$ C = 90 $^{\circ}$ F)

(40 % RH (40 % = 104 °F))

2) Sampling time : 20 seconds/1 pump stroke

3) Detectable limit $0.2 \text{ mg}/\ell$ 4) Shelf life 3 years5) Operating temperature $0 \sim 40 \text{ C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Greenish yellow→Purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Magnesium perchlomate, PH indicator is discoloured. H2O + Mg (CIO₄)2→Mg (CIO₄)2 · H2O

4. CALIBRATION OF THE TUBE

VAPOR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	%	Coexistence
Methanol		0.3	Unclear stain is produced.
Ethanol		0.3	"
Ethyl acetate		0.3	"
Acetone		0.5	"
Nitrogen dioxide		0.2	"
Ammonia		0.02	Purple and purplish red stains are produced and higher readings are given.

Tuba		Corrected Concentration																
Tube Readings	0℃ 32°F	5℃ 41°F	10℃ 50°F	12℃ 54°F	14 ℃ 57 °F	16℃ 61 F	18℃ 64°F	20℃ 68*F	22℃ 72˚F	24℃ 75°F	26℃ 79°F	28 ℃ 82 F	30 ℃ 86 F	32℃ 90°F	34℃ 93°F	36℃ 97°F	38℃ 100°F	40 ℃ 104 °F
0	100.0 4.8	50.0 3.4	30.0 2.8	20.0 2.1	18.0 2.2	15.0 2.0	12.0 1.8	10.0 1.7	9.0 1.7	8.0 1.7	7.5 1.8	7.0 1.9	6.5 2.0	6.0 2.0	5.5 2.1	5.0 2.1	5.0 2.3	4.0 2.0
5		100.0 6.8	50.0 4.7	30.0 3.2	26.0 3.1	20.0 2.7	18.0 2.8	15.0 2.5	12.0 2.3	9.0 1.9	8.0 2.0	8.0 2.2	7.5 2.3	7.0 2.4	6.5 2.4	6.0 2.5	5.5 2.5	5.0 2.6
10			100.0 9.4	70.0 7.5	48.0 5.8	35.0 4.8	28.0 4.3	24.0 4.2	18.0 3.5	15.0 3.3	13.0 3.2	10.0 2.7	9.0 2.7	8.5 2.9	8.0 3.0	7.5 3.1	7.0 3.2	6.5 3.3
12				100.0 10.7	80.0 9.6	50.0 6.8	40.0 4.8	30.0 5.1	25.0 4.9	20.0 4.4	17.0 4.1	14.0 3.8	12.0 3.6	10.0 3.4	9.5 3.6	9.0 3.8	8.0 3.5	7.0 3.5
14					100.0 12.0	80.0 10.9	55.0 8.5	40.0 6.9	35.0 6.8	26.0 5.7	20.0 4.9	18.0 4.9	15.0 4.5	13.0 4.4	11.0 4.1	10.0 4.2	9.0 4.2	9.0 4.6
16						100.0 13.6	80.0 12.3	55.0 9.5	45.0 8.7	35.0 7.6	27.0 6.6	21.0 5.7	18.0 5.5	15.0 5.4	14.0 5.3	12.0 5.0	11.0 5.1	10.0 5.1
18							100.0 15.4	80.0 13.8	55.0 10.7	45.0 9.8	35.0 8.5	27.0 7.3	22.0 6.7	19.0 6.4	17.0 6.4	15.0 6.3	13.0 6.0	12.0 6.1
20								100.0 17.2	80.0 15.5	60.0 13.1	48.0 11.7	38.0 10.3	29.0 8.8	24.0 8.1	20.0 7.5	18.0 7.5	16.0 7.4	14.0 7.2
22									100.0 19.4	80.0 17.4	62.0 15.1	50.0 13.6	40.0 12.1	30.0 10.1	25.0 9.4	22.0 9.2	19.0 8.8	17.0 8.7
24										100.0 21.8	80.0 19.5	62.0 16.9	50.0 15.2	40.0 13.5	30.0 11.3	27.0 11.3	23.0 10.6	20.0 10.2
26											100.0 24.4	80.0 21.8	64.0 19.4	50.0 17.0	40.0 15.0	32.0 13.3	28.0 12.9	25.0 12.8
28												100.0 27.2	80.0 24.2	64.0 21.6	50.0 18.8	40.0 16.7	32.0 14.8	28.0 14.3
30													100.0 30.3	80.0 27.0	64.0 24.1	50.0 20.1	40.0 18.5	34.0 17.4
32														100.0 33.8	80.0 30.1	64.0 26.7	48.0 22.2	40.0 20.4



1. PERFORMANCE

1) Measuring range 0.05-2.0 mg/ ℓ Number of pump strokes $1(100 \text{m}\ell)$

2) Sampling time : 1 minute/1 pump stroke

3) Detectable limit $0.03 \text{ mg/}\ell$ 4) Shelf life $0.03 \text{ mg/}\ell$ 5) Operating temperature $0.03 \text{ mg/}\ell$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke
8) Colour change : Yellow→Blue (over 0.6 mg/ℓ) or Yellowish green (below 0.6 mg/ℓ)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid: 10% RSD-high: 10%

3. CHEMICAL REACTION

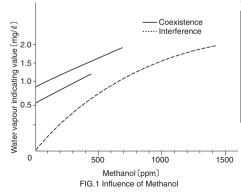
By reacting with Magnesium perchlomate, PH indicator is discoloured. $H_2O + Mg(CIO_4)_2 \rightarrow Mg(CIO_4)_2 \cdot H_2O$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence			
Alcohols	FIG.1	Similar stain is produced.	Higher readings are given.			
Aliphatic hydrocarbons		The accuracy of readings is not affected. affected.	The accuracy of readings is not			
Aromatic hydrocarbons		"	"			
Sulphur dioxide		"	"			
Hydrogen sulphide		"	"			



Tube	Corrected Concentration (mg/ℓ)								
Readings (mg/ℓ)	32°F	41 °F	50°F	59°F	68°F	77°F	86°F	95°F	104°F
2.0		3.70	3.06	2.46	2.00	1.60	1.28	0.92	0.66
1.8	4.40	3.26	2.66	2.20	1.80	1.44	1.10	0.82	0.58
1.6	4.00	2.94	2.40	1.95	1.60	1.30	0.98	0.74	0.52
1.4	3.30	2.52	2.08	1.72	1.40	1.15	0.85	0.64	0.44
1.2	2.70	2.06	1.80	1.48	1.20	0.98	0.72	0.54	0.38
1.0	1.92	1.66	1.50	1.24	1.00	0.82	0.60	0.46	0.30
0.8	1.58	1.38	1.24	1.00	0.80	0.66	0.50	0.36	0.25
0.6	1.10	1.02	0.90	0.74	0.60	0.50	0.37	0.26	0.18
0.4	0.72	0.68	0.60	0.50	0.40	0.34	0.25	0.18	0.12
0.2	0.38	0.36	0.32	0.26	0.20	0.18	0.14	0.08	0.06
0.1	0.18	0.17	0.15	0.13	0.10	0.08	0.06	0.05	0.04
0.05	0.09	0.08	0.07	0.06	0.05	0.05	0.04	0.03	0.02



1. PERFORMANCE

1) Measuring range : 3-80 LB/MMCF Number of pump strokes : 1(100mℓ)

2) Sampling time : 1 minute / 1 pump stroke

3) Detectable limit \therefore 2 LB/MMCF 4) Shelf life \therefore 3 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")
7) Reading : Direct reading from the scale calibrated by 1 pump stroke
8) Colour change : Yellow→Blue, Yellowish green (below 40 LB/MMCF)

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $H_2O + Mg(ClO_4)_2 \rightarrow Mg(ClO_4) \cdot H_2O$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Benzene	The accuracy of readings is not affected.	The accuracy of readings is not affected.
Toluene	"	"
Hexane	"	"
Sulphur dioxide	"	"
Hydrogen sulphide	"	"

CORRECTION FOR AMBIENT CONDITIONS:

Temperature; To correct for temperature, multiply the tube reading by the following factors.

Temperature(°F)	32	33	34	35	36	37	38	39	40	41
Correction Factor	1.85	1.83	1.81	1.78	1.76	1.73	1.73	1.69	1.66	1.63
Temperature(°F)	42	43	44	45	46	47	48	49	50	51
Correction Factor	1.61	1.59	1.56	1.53	1.50	1.48	1.45	1.43	1.40	1.38
Temperature(°F)	52	53	54	55	56	57	58	59	60	61
Correction Factor	1.36	1.33	1.30	1.28	1.26	1.24	1.21	1.19	1.16	1.14
Temperature(°F)	62	63	64	65	66	67	68	69	70	71
Correction Factor	1.12	1.10	1.08	1.06	1.04	1.02	1.00	0.98	0.96	0.94
Temperature(°F)	72	73	74	75	76	77	78	79	80	81
Correction Factor	0.92	0.91	0.89	0.87	0.86	0.84	0.82	0.81	0.79	0.77
Temperature(°F)	82	83	84	85	86	87	88	89	90	91
Correction Factor	0.76	0.75	0.73	0.72	0.71	0.69	0.68	0.67	0.66	0.64
Temperature(°F)	92	93	94	95	96	97	98	99	100	101
Correction Factor	0.63	0.62	0.61	0.60	0.59	0.58	0.57	0.56	0.55	0.54
Temperature(°F)	102	103	104							
Correction Factor	0.53	0.52	0.51							



1. PERFORMANCE

1) Measuring range : 2-12 LB/MMCF Number of pump strokes : 2(200m ℓ)

2) Sampling time : 2 minutes/2 pump strokes

3) Detectable limit ∴ 1 LB/MMCF 4) Shelf life ∴ 3 years 5) Operating temperature ∴ 0 ~ 35 °C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : Yellow → Yellowish green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $H_2O + Mg(ClO_4)_2 \rightarrow Mg(ClO_4) \cdot H_2O$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Alcohols	Similar stain is produced.	Higher readings are given.
Benzene	The accuracy of readings is not affected.	The accuracy of readings is not affected.
Toluene	"	"
Hexane	"	"
Sulphur dioxide	"	"
Hydrogen sulphide	"	"

TEMPERATURE CORRECTION TABLE

Temperature; To correct for temperature, multiply the tube reading by the following factors.

Temperature (°C)	0	1	2	3	4	5	6	7	8	9
Correction Factor	1.69	1.65	1.61	1.57	1.53	1.50	1.46	1.42	1.39	1.35
Temperature (°C)	10	11	12	13	14	15	16	17	18	19
Correction Factor	1.32	1.28	1.25	1.22	1.19	1.15	1.12	1.09	1.06	1.03
Temperature (°C)	20	21	22	23	24	25	26	27	28	29
Correction Factor	1.00	0.98	0.95	0.92	0.90	0.87	0.85	0.82	0.80	0.78
Temperature (°C)	30	31	32	33	34	35				
Correction Footox	0.75	0.70	0.71	0.00	0.67	0.05				

177UR2 WATER VAPOUR



1. PERFORMANCE

5-30 LB/MMCF 1) Measuring range : 2-12 LB/MMCF Number of pump strokes $2(200 \text{m} \ell)$ $1(100m\ell)$ 2) Sampling time : 4 minutes/2 pump strokes

3) Detectable limit : 2 LB/MMCF (200mℓ)

4) Shelf life : 3 years 5) Operating temperature : 0 ~ 35 °C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") : Direct reading from the scale calibrated by 2 pump strokes 7) Reading

8) Colour change : Yellow → Yellowish green or green

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

 $H_2O + Mg(ClO_4)_2 \rightarrow Mg(ClO_4) \cdot H_2O$

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Methanol	50	Similar stain is produced.	Higher readings are given.
Alcohols(except Methanol)		"	"
Benzene		The accuracy of readings is not affected.	
Toluene		"	
Hexane		"	
Sulphur dioxide		"	
Hydrogen sulphide		"	

(NOTE)

In case of 1 pump stroke, following formula is available for the actual concentration.

Actual concentration = $2.5 \times$ Temperature corrected value

TEMPERATURE CORRECTION TABLE

Temperature; To correct for temperature, multiply the tube reading by the following factors.

Tube	Corrected Concentration (LB/MMCF)				
Readings (LB/MMCF)	0℃ (32下)	10 ℃ (50 F)	20℃ (68干)	30 ℃ (86 F)	35℃ (95干)
12	17.8	14.9	12.0	9.1	7.8
10	15.3	12.7	10.0	7.4	6.3
8	12.9	10.4	8.0	5.9	5.1
6	10.3	8.0	6.0	4.5	3.9
4	6.8	5.1	4.0	3.1	2.8
2	2.7	2.3	2.0	-	-

XYLENE



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \text{S-1,000 ppm} \\ \text{Number of pump strokes} \end{array}$

2) Sampling time : 4 minutes/2 pump strokes

3) Detectable limit \therefore 2 ppm 4) Shelf life \therefore 1.5 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Reading : Direct reading from the scale calibrated by 2 pump strokes

7) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

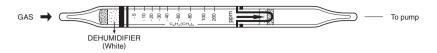
Iodine pentoxide is reduced. C_6H_4 (CH₃)₂ + I₂O₅ + H₂SO₄ \rightarrow I₂

4. CALIBRATION OF THE TUBE

DIFFUSION TUBE METHOD

Substance	Interference	ppm	Coexistence
Benzene	Similar stain is produced.		Higher readings are given.
Toluene	"		"
Ethyl Benzene	"		"
Methyl alcohol	The accuracy of readings is not affected.	1%	"
Hexane	Whole layer is discoloured to Pale brown.	0.1%	Higher readings with indiscernible maximum end point of stained layer are given.

XYLENE



1. PERFORMANCE

1) Measuring range 5-200 ppmNumber of pump strokes $2(200 \text{m} \ell)$

2) Sampling time : 3 minutes/2 pump strokes

3) Detectable limit \therefore 1 ppm 4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 $^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 2 pump strokes

8) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. C_6H_4 (CH₃)₂ + I₂O₅ + H₂SO₄ \rightarrow I₂

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Toluene	Xylene conc.1/5	Same stain is produced.	Higher reading are given.
Ethyl acetate		The accuracy of readings is not affected.	The accuracy of readings is not affected.
Methyl isobutyl ketone		"	"
Isobutyl alcohol		"	"

Tube	Corrected Concentration (PPm)			
Readings (ppm)	20°C (68°F)	30 °C (86 °F)	40 °C (104 °F)	
200	200	300	-	
150	150	190	400	
100	100	125	150	
80	80	100	120	
60	60	70	80	
40	40	45	50	
30	30	30	30	

3.2 A SUBSTANCES TO BE MEASURED BY USING CONVERSION CHARTS

Chemical substances listed in this section can be measured by using respective conversion charts and existing gas detector tubes on demands.

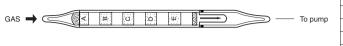
However, note that this conversion chart method is available for 20 °C (68 °F) in temperature and detailed conditions such as other temperatures, humidity and coexisting gases are not confirmed.

SUBSTANCE C	CHEMICAL FORMULA	MEASURING RANGE	USING TU	BE (PAGE)
Allyl chloride	CH2CHCH2CI	1-40 ppm	132SC	(344)
Benzyl chloride	C ₆ H ₅ CH ₂ CI	1-16 ppm	132SC	(344)
Bromochloromethane	CH₂BrCI	2-400 ppm	157SB	(233)
Bromoform	CHBr₃	0.5-20 ppm	157SB	(233)
1-Bromopropane	CH ₃ CH ₂ CH ₂ Br	10-500 ppm	157SA	(232)
1-Bromopropane	CH ₃ CH ₂ CH ₂ Br	5-80 ppm	157SB	(233)
2-Bromopropane	(CH ₃) ₂ CHBr	10-500 ppm	157SA	(232)
2-Bromopropane	(CH ₃) ₂ CHBr	5-80 ppm	157SB	(233)
m-Chlorotoluene	C ₆ H ₄ CI (CH ₃)	0.5-10 ppm	132SC	(344)
o-Chlorotoluene	CIC ₆ H ₄ CH ₃	1-50 ppm	132SC	(344)
p-Chlorotoluene	CIC ₆ H ₄ CH ₃	1-50 ppm	132SC	(344)
p-Cymene	CH ₃ C ₆ H ₄ CH (CH ₃) ₂	20-200 ppm	102SD	(28)
Dibromomethane	CH ₂ Br ₂	2.5-40 ppm	157SB	(233)
1,1-Dichloroethylene	$CH_2 = CCI_2$	1-22 ppm	132SC	(344)
1,2-Dichloropropane	CH3CHCICH2CI	20-250 ppm	157SB	(233)
Disilane	Si ₂ H ₆	1-50 ppm	240S	(307)
Ethyl bromide	C ₂ H ₅ Br	2-400 ppm	157SB	(233)
Ethylene chlorohydrine	CH2CICH2OH	5-300 ppm	119U	(228)
Iodine	l 2	0.7-42 ppm	117SB	(264)
Mineral turpentine	_	4-200 ppm	111U	(137)
Trichlorotoluene	C ₆ H ₅ CCl ₃	0.2-4 ppm	132SC	(344)
 Benzaldehyde 	C ₆ H ₅ CHO	5-70 ppm	190U	(143)
1,1,2,2-TetrachloroethauN.B.)	ne CHCl ₂ CHCl ₂	20-80 ppm	236SA	(331)

Upon these two kinds of measurement, the conversion chart and the measuring range may vary at each manufacturing lot.

3.3 KITAGAWA INORGANIC/ORGANIC GAS QUALITATIVE DETECTOR TUBES

INORGANIC GAS QUALITATIVE DETECTOR TUBE



Section	Original Colour
Α	Pale purple
В	Reddish purple
С	White
D	White
Е	Yellow

1. SPECIFICATIONS

1) Substances to be detected : Acetic acid, Amines, Ammonia, Carbon monoxide, Chlorine, Hydrogen chloride, Hydrogen sulphide, Nitrogen dioxide, Phosphine, Sulphur dioxide,

* Acetylene and * Metyl mercaptan (*: Organic gas)

2) Tube per box : 10tubes (10-time use)

3) Pump stroke : $1(100 \text{m} \ell)$ 4) Sampling time : 20 seconds5) Shelf life : 1 year6) Operating temperature : $0 \sim 40 \text{ }^{\circ}\text{C}$

7) Colour change : Refer to following" 3. DISCOLOURATION / QUALITATIVE CHART"

8) Non-discolouration : Carbon dioxide, Hydrogen cyanide, Nitric oxide and

confirmed substances *Ethylene (*: Organic gas)

2. CHEMICAL REACTION

SECTION	CHEMICAL REACTION PRINCIPLES
A	By reacting with Phosphoric acid, PH indicator is discoloured.
	$2NH_3 + H_3PO_4 \rightarrow (NH_4)_2HPO_4$
В	By reacting with an Alkaline, PH indicator is discoloured.
	$SO_2 + 2NaOH \rightarrow Na_2SO_3 + H_2O$
C	By reacting with o-Toluidine, Nitro- o -Toluidine (Dyestuff) is liberated.
D	By reacting with Lead Acetate (II), Lead sulphide is produced.
	$H_2S + Pd(CH_3COO)_2 \rightarrow PdS + 2CH_3COOH$
E	Potasium disulphide palladate (II) is reduced and Palladium is liberated.
	$CO + K_2Pd(SO_3)_2 \rightarrow K_2 (SO_3)_2PdCO$
	$K_2 (SO_3)_2PdCO \rightarrow CO_2 + SO_2 + K_2SO_3$

3. DISCOLOURATION / QUALITATIVE CHART

CHART 1. INORGANIC GAS QUALITATIVE DETECTION CHART

	Select	ion (Original C	olour)		
A (Pale purple)	B (Reddish purple)	C (White)	D (White)	E (Yellow)	* 1) Substances (* 2)
Yellow					1) Ammonia (5) 2) Amines (5)
	Yellow				3) SO ₂ (10) 4) Acetic Acid (15)
	Pink				5)Hydrogen chloride (20)
	White	Yellowish orange			6) Chlorine (5)
		Yellow		-	7) Nitrogen dioxide (5)
			Brown	-	8) H ₂ S (10)
				Pale blackish brown	9) CO (10)
				Dark black	10) Phosphine (2)
				Pale Yellowish green	11) Acetylene (10)
				Dark yellow	12)Methyl mercaptan (10)

NOTES: -

(1) — : Undiscoloured

(2) (*1) : Item No. for quick reference to details in CHART

(3) (*2) : Detectable gas concentration limit of the substance(Unit: ppm)

The discolouration length is approx.0.5 to 1.0 mm.

(4) Sybstance No.4),11) and 12) are organic substances.

CHART 2. CHART FOR GAS-CONCENTRATION LEVEL AND DISCOLOURATION

1) Ammonia	C C C C C C C C C C C C C C C C C C C	GAS CONCEN			SECTION		
50 Vellow (II)	INCHARAINC SOBSI AINCES	TRATION (PPM)	A (Pale purple)	B (Reddish purple)	C (White)	D (White)	E (Yellow)
5 Yellow (III) —— —— —— —— —— —— —— —— —— —— —— —— ——	1) Ammonio	20	Yellow (I)				
50 Yellow (II)		2	Yellow (III)				
5 Yellow (III) Yellow (IIII) Yellow (IIII) Yellow (IIII) Yellow (IIIII) Yellow (IIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	, (C	20	Yellow (I)				
20	z) Allilles	2	Yellow (III)				
10 Yellow (III) —— Yellow (II) —— —— Yellow (II) —— —— Yellow (II) —— —— —— Yellow (II) —— —— —— —— —— —— —— —— —— —— —— —— ——	(CO) objecto midali o (c	20		Yellow (I)			
15	a) Suipriul dioxide (SOS)	10		Yellow (III)			
15 Yellow (III)	7:00	30		Yellow (II)			
50	t) Aceilo Acid	15		Yellow (III)			
20		20		Pink (II)			
20 White (I) vellowish orange (I) vellowish orange (I) vellow shorte & vellow (I) shorte (II) shorted White & vellow (I) shorted — vellow (I) shorted — vellow (I) shorted (HeS) 100 — vellow (I) shorted — shown (II) shown (III) shown		20		Pink (Ⅲ)			
(CC) 100 — — — — — — — — — — — — — — — — — —		20		White (I)	Yellowish orange (I)		
(H ₂ S) 100	6) Chlorine	5		White & Pale purple (II)			
(H ₂ S) 100	Ocivoid accounting (2	20			Yellow (I)		
(HeS) 100 — Brown (II) Brown (II) (CO) 100 — Brown (II)		2			Yellow (I)		
(CO) 10 — Brown (II) (II) — Brown (II) (III) — Brown (II)		100				Brown (II)	Brown (II)
(CO) 50		10				Brown (II)	
(50) 10 — — — — — — — — — — — — — — — — — —		90					Blackish brown (I)
30 — — — 2 — — — 50 — — — 10 — — — captan 10 — —		10					Pale blackish brown (I)
2	0,00000	30					Black (II)
50 — — — — — — — — — — — — — — — — — — —		2					Pale black (III)
10 — — — — — — — — — — — — — — — — — — —		20					Yellowish
10 — — — — — — — — — — — — — — — — — — —	11) Acetylene						green (I)
100		10					Pale yellowish green ${ m (I)}$
10 — — — — — — —	19) Methyl mercantan	100				1	Pale yellow (I)
	(z) Metry mercaptan	10					Dark yellow (II)

NOTES: -

- 1) ----: Undiscoloured
- II; A half layer is discoloured. III; Approx. 0.5-1.0mm of the layer is discoloured. 2) Discolouration level: I; The whole layer is discoloured.
- 3) Sybstance No.4) ,11) and 12) are organic substances.

NON-DISCOLOURATION CONFIRMED SUBSTANCES

- 1) Hydrogen cyanide (HCN) 2) Carbon dioxide (CO2) 3) Ethylene
- 3) Ethylene (Organic substance) 4) Nitric oxide (NO)

ORGANIC GAS QUALITATIVE DETECTOR TUBE



Section	Original Colour
Α	Orange
В	White
С	Yellow
D	Yellow

1. SPECIFICATIONS

1) Substances to be detected

: Acetaldehyde, Actone, Acetylene, Aniline, Benzene, 1,3-Butadiene, Butane,1-Butanol, Butyl acetate, Carbon disulphide, Cresol, Ethyl acetate, Ethyl amine, Ethyl benzene, Ethyl cellosolve, Ethylene, Ethylene oxide, Formaldehyde, Gasoline, Heptane, Hexane, Isopropyl alcohol, Kerosine, Methyl alcohol, Methyl ethyl ketone, Methyl isobutyl ketone, Methyl mercaptan, Pentane, Phenol, Propane, Styrene, Tetrachloroethylene, Tetrahydrofuran, Toluene, 1,1,1-Trichloroethane, Trichloroethylene, Vinyl chloride, Xylene, *Arsine, *Carbon monoxide and *Hydrogen sulphide (*: Inorganic gas)

2) Tube per box : 10 tubes (5 -time use)3) Pump stroke $: 1(100 \text{m} \ell) + 1(100 \text{m} \ell)$ 4) Sampling time : 30 + 30 seconds5) Shelf life : 2 years6) Operating temperature $: 0 \sim 40 \text{ °C}$

Colour change
 Non-discolouration
 confirmed substances

Refer to following "3. DISCOLOURATION / QUALITATIVE CHART" Acetic acid, Carbon tetrachloride, Methane, Methyl bromide and Pyridine

2. CHEMICAL REACTION

SECTION	CHEMICAL REACTION PRINCIPLES
A	Chromium oxide is reduced.
	$CH_3 (CH_2)_4 CH_3 + Cr^6 + H_2 SO_4 \rightarrow Cr^{3+}$
В	Molybdate is reduced and Molybdeum blue is produced.
	$H_2C = CH_2 + PdSO_4 + (NH_4)_2MoM_4 \rightarrow Mo_3O_8$
C	Iodine pent-oxide is reduced.
	$C_2H_5CH_3 + I_2O_5 + H_2SO_4 \rightarrow I_2$
D	Phenol is oxidized and the polymer is produced.
	$C_6H_5OH + Ce^{4+} \rightarrow C_6H_5O \cdot \rightarrow (C_6H_5O)n$

3. DISCOLOURATION / QUALITATIVE CHART

CHART 1. ORGANIC GAS QUALITATIVE DETECTION CHART

"A" side sampling		"D" side	sampling		* 1) Substances
Selection		Sele	ction		-*2 (X) -
Α .	Α	В	C	D	-*3 (X/Y) -
(Orange)	(Orange)	(White)	(Yellow)	(Yellow)	
Dark brown	Dark brown				1) Propane (100)
					2) Butane (10)
					3) Pentane (10)
					4) Hexane (10)
					5) Trichloroethylene (10)
					6) Tetrachloroethylene (100)
					7) Vinyl chloride (10)
			White		8) Butadiene (100)
	Greenish brown		Pale blue		9) Gasoline (0.1mg/ ℓ)
		Pale brown			10) Benzene (10/100)
					11) Toluene (30/200)
					12) Xylene (60/1,000)
					13) Ethyl benzene
					(60/400)
			Pale blue		14) Ethylene (10)
					15) Acetylene
					(10,000/100)
			Yellowish	_	16) Styrene (100)
			orange		47) 4 - 4 - 7 (000)
				_	17) Acetone (600)
					18) Methyl ethyl ketone (100)
					19) Ethyl acetate (600)
					20) Butyl acetate (100)
					21) Ethylene oxide (100)
					22) Formaldehyde (600)
					23) Kerosene (0.1mg/ℓ)

"A" side sampling		"D" side	* 1) Substances		
Selection		Sele	ction		-*2 (X) -
A (Orange)	A (Orange)	B (White)	C (Yellow)	D (Yellow)	-*3 (X/Y) -
Greenish brown	Greenish brown				24) Heptane (10) 25) Carbon disulphide (100)
			Yellowish orange		26) Methyl mercaptan (100/20)
					27) Methyl alcohol (100)
					28) 1-Butanol (100)
					29) Acetaldehyde (100)
					30) Methyl isobutyl Ketone (100)
					31) Ethyl cellosolve (100)
					32) Tetrahydrofuran (100)
					33) 1,1,1-Trichloroethane (1,000)
Green			Black		34) Hydrogen sulphide (100,10)
					35) Arsine (100,20)
					36) Isopropyl alcohol (600)
			Pale blue		37) Carbon monoxide (100)
				Pale brown	38) Phenol (10)
					39) Cresol (20)
				Bluish green	40) Aniline (40)
				Pale blue	41) Ethyl amine (100)

NOTES: -

- (1) —: Undiscoloured
- (2) *1: Item No. for quick reference to details in CHART 2.
- (3) *2 (X) : Detectable gas concentration limit of the substance (Unit : ppm)
 - *3 (X/Y): "X" means detectable gas concentration limit (Unit: ppm) of "A" side sampling and "Y" means "D" side sampling's one.
- (4) The discolouration length is approx.0.5 to 1.0 mm.
- (5) Substance No.34) ,35) and 37) are inorganic gases."

CHART 2. CHART FOR GAS-CONCENTRATION LEVEL AND DISCOLOURATION

					:	
		"A" side sampling		"D" side sampling	sampling	
ORGANIC		SECTION		SECTION	NOI.	
000000000000000000000000000000000000000		A (Orange)	A (Orange)	B (White)	C (Yellow)	D (Yellow)
	I	Dark brown (I)	Dark brown (I)			
1) Propane	Σ	Dark brown (I)	Dark brown (I)			1
	I	Dark brown (I)	Dark brown (I)			
2) Butane	Σ	Dark brown (I)				
`	_	Dark brown (III)	Dark brown (III)			
	I	Greenish brown (I)	Greenish brown (I)			
3) Pentane	Σ	Dark brown (I)	Dark brown (I)			
	_	Dark brown (III)	Dark brown (III)			
	Н	Greenish brown (I)	Greenish brown (I)			
4) Hexane	Σ	Dark brown (II)	Dark brown (II)			
	L	Dark brown (III)	Dark brown (III)			
	I	Dark brown (I)				
5) Trichloroethylene	Σ	Dark brown (II)	Dark brown (II)			
	_	Dark brown (III)	Dark brown (III)			
	Н	Dark brown (I)				
6) Tetrachloroethylene	Σ	Dark brown (III)	Dark brown (III)			
	_					
	I	Dark brown (I)				
7) Vinyl chloride	Σ					
	_	Dark brown (III)	Dark brown (III)			
	T		Dark brown (II)	Yellow orange (I)	Brown (I)	
8) Butadiene	∑.	Dark brown (III)	Dark brown (III)		White (II)	
	_					
	I	Dark brown (I)	Greenish brown (II)	Brown (I)		
9) Gasoline	≥-	Dark brown (III)	Greenish brown (III)	Brown (I)	Pale blue (I)	
	_	Dalk Dlowii (III)	GIEGIIISII DIOWII (III)			
	Ι:	Dark brown (I)	Dark brown (III)	Pale brown (I)		
10) Benzene	Σ	Dark brown (II)		Pale brown (I)		
	_	Dark brown (III)				
	I	Dark brown (I)	Dark brown (III)	Brown (I)		
11) Toluene	≥.	Dark brown (III)		Pale brown (I)		
	_	Dark brown (III)				

		"A" side sampling		"D" side	"D" side sampling	
ORGANIC	CONC	SECTION		SEC	SECTION	
SUBSTANCES	LEVEL	A (Orange)	A (Orange)	B (White)	C (Yellow)	D (Yellow)
12) Xylene	ΙΣ	Dark brown (II)		Pale brown (II)		
	_					
	I	Dark brown (II)		Brown (I)		
13) Ethyl benzene	Σ	Dark brown (III)		Pale brown (II)		
	7					
	I	Dark brown (I)	Dark brown (I)		Deep blue (II)	
14) Ethylene	∑.	Dark brown (I)	Dark brown (II)		Deep blue (II)	
	_	Dark brown (III)			Pale blue (II)	
	I	Dark brown (I)	Dark brown (I)		Deep blue (II)	
15) Acetylene		loured to Brown(III).				
	Σ				Pale blue (II)	
	_					
	I	Dark brown (III)		Dark brown (III)	Yellowish orange (I)	
16) Styrene	Σ	Dark brown (III)				
	_					
17) Acotono	I	Dark brown (III)				
	Σ					
	I	Dark brown (III)				
18) Methyl ethyl ketone	Σ	Dark brown (III)				
	٦					
10) Ethyl acetate	I	Dark brown (III)		1		
19) Lillyl acetate	Σ					
	I	Dark brown (III)				
20) Butyl acetate	Σ	Dark brown (III)				
	_					
	I	Dark brown (III)				
21) Ethylene oxide	Σ	Dark brown (III)				
	7					

		"A" side sampling		"D" side	"D" side sampling	
ORGANIC	CONC.	SECTION		SECTION	NOIL	
SOBSTANCES	LEVEL	A (Orange)	A (Orange)	B (White)	C (Yellow)	D (Yellow)
	I	Greenish brown (I)				
22) Formaldehyde	Σ	Dark brown (III)				
	_	Dark brown (III)				
	Ŧ	Dark brown (II)				
23) Kerosene	Σ	Dark brown (III)				
	_	Dark brown (III)				
	I	Greenish brown (I)	Greenish brown (I)			
24) Heptane	Σ	Greenish brown (II)	Greenish brown (II)			
	_	Greenish brown (III)	Greenish brown (III)			
	I	Greenish brown (I)	Greenish brown (I)			
25) Carbon disulphide	Σ	Greenish brown (II)	Greenish brown (II)			
	_					
OG \ Mothyl morcopton	Σ	Greenish brown (III)			Deep blue (II)	
20) Methyl melcapian	_				Yellowish orange (III)	
	I	Greenish brown (III)				
27) Methyl alcohol	Σ	Greenish brown (III)				
	٦					
	エ	Greenish brown (III)				
28) 1-Butanol	Σ	Greenish brown (III)				
	Γ					
	I	Green (II)	1	1		1
29) Acetaldehyde	Σ	Greenish brown (III)				
	_					
	I	Greenish brown (III)	1			1
30) Methyl isobutyl ketone	Σ	Greenish brown (III)				
	_					
	I	Green (III)				
31) Ethyl cellosolve	Σ	Greenish brown (III)				
	Γ					
	I	Green (III)				
32) Tetrahydrofuran	Σ	Greenish brown (III)				
	7					

		"A" side sampling		"D" side	"D" side sampling	
ORGANIC	CONC.	SECTION		SECTION	NOIL	
	, ,	A (Orange)	A (Orange)	B (White)	C (Yellow)	D (Yellow)
33) 1 1 1-Trichloroethane	I	Greenish brown (III)		1		
00) 1, 1, 1	Σ					
-	I	Green (II)	Green (II)		Black (II)	
34) Hydrogen sulphide	Σ	Green (III)			Black (II)	
(H2S)	_				Black (III)	
OE) Aroino	Σ	Green (II)			Black (I)	
SO) Alsille	_				Black (II)	
Se) Isopone Ivagana (36)	I	Green (III)				
od) isopiopyi aicollol	Σ					
	I				Pale blue (II)	
37) Carbon monoxide	Σ				Pale blue (III)	
(00)	_					
38) Dhonol	Σ					Pale brown (I)
000	_					Pale brown (I)
30) (06	Σ					Pale brown (I)
oa) ciesol	٦					Pale brown (I)
40) Apilipo	Σ]	1	1	Bluish green (III)
	_					Bluish green (III)
	I			1		White (I)
41) Ethyl amine	Σ					Pale blue (III)
	_					

1) ---: Undiscoloured

2) CONC. LEVEL (Gas concentration level): H; approx. 1,000-5,000 ppm M; approx. 100-500 ppm L; approx. 10-50 ppm

3) Discolouration level: 1; The whole layer is discoloured.
2; A half layer is discoloured.
3; Approx.0.5-2.0 mm of the layer is discoloured.

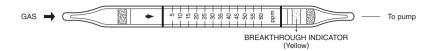
4) Substance No.34), 35) and 37) are inorganic substances.

NON-DISCOLOURATION CONFIRMED SUBSTANCES

- 3) Methyl bromide6) Ethane Pyridine
 Methane
- 1) Carbon tetrachloride4) Acetic acid

3.4 KITAGAWA TIME-WEIGHTED AVERAGE DETECTOR TUBES

TWA-CARBON MONOXIDE



1. PERFORMANCE

1) Measuring range : 5-400 ppm

(0.5 hr.) (4 hrs.) (8 hrs.) 50-400 ppm 5-100 ppm 5-60 ppm

2) Sampling time : 8 hrs. (6 m l/min.)

3) Shelf life : 3 years 4) Operating temperature : $10 \sim 30 \, ^{\circ}\text{C}$

5) Reading : Direct reading from the scale calibrated by 8 hrs. Sampling

6) Colour change : White→Brown ringed

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

Iodine pentoxide is reduced. $CO + I_2O_5 + H_2SO_4 \cdot nSO_3 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Butane		50	Higher readings are given.
Hexane		50	"

(NOTE)

- 1) Air sampler is required for this tube.
- 2) Flow Rate and Sampling Time
- In case of 8 hours, sampling with 6 mℓ/min., the TWA concentration can read directly by the scale printed on the tube at the top of Brown ring.
- (2) If the sampling duration is less than 8 hours, the actual TWA concentration can be obtained graphically from the chart provided below.
- (3) If the flow rate is not 6 m ℓ /min, divide the scale reading by the ratio of sampled air volume to 2880m ℓ .

Actual TWA concentration (ppm) = $I \times \frac{2880}{V}$

Sampled Air Volume 0 0.5 1.0 2.0 2.5 2.88 (mdd) 60 55 50 50 TWA Tube Scale Indication Actual TWA Concentration 45 45 40 40 35 35 30 30 25 25 20 20 15 15 10 5 Sampling Time (Hours)

SCALE CONVERSION CHART

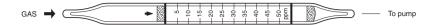
I = Scale reading

 $V = Sampled air volume in m \ell$

[Flow rate $(m\ell/min.) \times Sampling duration (min.)$]

- (a) If sampling time is 6 hours and scale reading is 30, the actual TWA concentration is 40 ppm.
- (b) If sampled air volume is 1.5ℓ and scale reading is 10, the actual TWA concentrationis 19.2 ppm.

TWA-AMMONIA



1. PERFORMANCE

1) Measuring range : 5-200 ppm

(1 hr.) (8 hrs.) 10-200 ppm 5-50 ppm

2) Sampling time : 8 hrs. (8 mℓ/min.)
3) Shelf life : 3 years
4) Operating temperature : 10 ~ 30 °C

4) Operating temperature $10 \sim 30 \, \text{C}$ 5) Reading Direct reading from the scale calibrated by 8 hrs. Sampling

6) Colour change : Purple → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

By reacting with Phosphoric acid, PH indicator is discoloured. $2NH_3 + H_3PO_4 \rightarrow (NH_4)_2HPO_4$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Sulphur dioxide		20	Lower readings are given.

(NOTE)

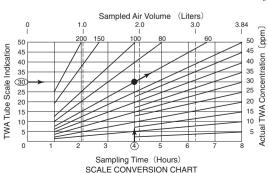
- 1) Air sampler is required for this tube.
- 2) Flow Rate and Sampling Time
- (1) In case of 8 hours, sampling with 8mℓ/min., the TWA concentration can be read directly by the scale printed on the tube at the top of Yellow stain.
- (2) If the sampling duration is less than 8 hours, the actual TWA concentration can be obtained graphically from the chart provided below.
- (3) If the flow rate is not $8m\ell/min$, divide the scale reading by the ratio of sampled air volume to $3840m\ell$.

Actual TWA concentration (ppm) = $I \times \frac{3840}{V}$

I = Scale reading

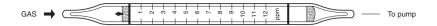
V = Sampled air volume in ml

[Flow rate($m\ell/min$.) \times Sampling duration(min.)]



- (a) If sampling time is 4 hours at 8mℓ/min and scale reading is 30, the actual TWA concentration is 60 ppm.
- (b) If sampled air volume is 2.0 ℓ and scale reading is 5, the actual TWA concentrationis 9.6 ppm.

TWA-HYDROGEN SULPHIDE



1. PERFORMANCE

1) Measuring range : 1-20 ppm

(1 hr.) (8 hrs.) 2-20 ppm 1-12 ppm

2) Sampling time : 8 hrs. (6 m ℓ /min.)

3) Shelf life : 1 year 4) Operating temperature : $10 \sim 30 \,^{\circ}\text{C}$

5) Reading : Direct reading from the scale calibrated by 8 hrs. Sampling

6) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

By reacting with Lead acetate (II), Lead sulphide is produced. H₂S + Pb (CH₃CO₂)₂→PbS + 2CH₃CO₂H

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Sulphur dioxide		10	Higher readings are given.

(NOTE)

- 1) Air sampler is required for this tube.
- 2) Flow Rate and Sampling Time
- (1) In case of 8 hours, sampling with 6mℓ/min., the TWA concentration can be read directly by the scale printed on the tube at the top of Brown stain.
- (2) If the sampling duration is less than 8 hours, the actual TWA concentration can be obtained graphically from the chart provided below.
- (3) If the flow rate is not $6m\ell/min$, divide the scale reading by the ratio of sampled air volume to $2880m\ell$.

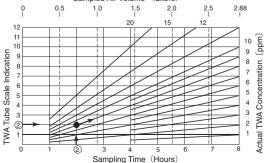
Actual TWA concentration (ppm) = $I \times \frac{2880}{V}$

I = Scale reading

V = Sampled air volume in ml

[Flow rate $(m\ell/\min) \times Sampling duration (\min)$]

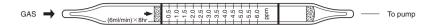




SCALE CONVERSION CHART

- (a) If sampling time is 2 hours at 6mℓ/min and scale reading is 2, the actual TWA concentration is 8 ppm.
- (b) If sampled air volume is 2.5 ℓ and scale reading is 6, the actual TWA concentrationis 7 ppm.

TWA-SULPHUR DIOXIDE



1. PERFORMANCE

1) Measuring range : 0.5-20 ppm

> (1 hr.) (8 hrs.) 1-20 ppm 0.5-6 ppm

2) Sampling time 8 hrs. (6 m l/min.)

3) Shelf life 3 years 10 ~ 30 ℃ 4) Operating temperature

5) Reading Direct reading from the scale calibrated by 8 hrs. Sampling

6) Colour change : Purple → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured.

 $SO_2 + 2NaOH \rightarrow Na_2SO_3 + H_2O$

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Carbon dioxide		1,000	Higher readings are given.
Nitrogen dioxide			The reagent colour fades, but the accuracy of the readings is not affected.

(NOTE)

- 1) Air sampler is required for this tube.
- 2) Flow Rate and Sampling Time
- (1) Read the scale printed on the tube at the top of Yellow stain.
- (2) Correct the reading value by average relative humidity of sampling atmosphere with humidity correction table. (Table 1)
- (3) In case of 8 hours, sampling with 6mℓ/min., the corrected value by Table 1 indicates actual TWA concentration.
- (4) If the sampling duration is less than 8 hours, the actual TWA concentration can be obtained graphically from the chart provided below.
- (5) If the flow rate is not 6mℓ/min, divide the corrected value by the ratio of sampled air volume to 2880mℓ. I =Corrected value by Table 1.

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Actual TWA concentration (ppm) = $I \times -$

2.0 6.0 (Wdd) 10.0 Indication 6.0 5.5 5.0 5.0 Concentration 45 4.5 4.0 Tube Scale 4.0 3.5 3.5 3.0 3.0 25 2.0 2.0 1.5 (1.5) 1.0 1.0 Actual 0.5 0.5 ٥

> Sampling Time (Hours) SCALE CONVERSION CHART

Example:

(a) If sampling time is 3 hours and corrected value by Table 1 is 1.5, the actual TWA concentration is 4.0 ppm.

V = Sampled air volume in ml[Flow rate($m\ell/min$.) \times Sampling duration(min.)]

> (b) If sampled air volume is 2.0ℓ and corre cted value by Table 1 is 3.5, the actual TWA concentration is 5.0 ppm

Table 1 Humidity Correction Table

Scale (ppm)	True Concentration (ppm)						
Readings	20%	30%	40%	50%	60%	70%	80%
6.0	4.5	5.0	5.5	6.0	_	_	_
5.5	4.1	4.6	5.1	5.5	5.9	_	_
5.0	3.8	4.2	4.6	5.0	5.4	5.8	_
4.5	3.4	3.8	4.1	4.5	4.8	5.2	5.5
4.0	3.0	3.4	3.7	4.0	4.3	4.6	4.9
3.5	2.6	2.9	3.2	3.5	3.8	4.0	4.3
3.0	2.3	2.5	2.8	3.0	3.2	3.5	3.7
2.5	1.9	2.1	2.3	2.5	2.7	2.9	3.1
2.0	1.5	1.7	1.9	2.0	2.2	2.3	2.4
1.5	1.2	1.3	1.4	1.5	1.6	1.7	1.8
1.0	0.8	0.8	0.9	1.0	1.1	1.2	1.2
0.5	0.4	0.4	0.5	0.5	0.5	0.6	0.6

TWA-TOLUENE



1. PERFORMANCE

: 20-200 ppm 1) Measuring range

> (1 hr.) (8 hrs.) 40-200 ppm 20-120 ppm

2) Sampling time 8 hrs. (10 m \(\ell \) min.)

3) Shelf life 3 years 4) Operation temperature 10 ~ 40 ℃

5) Reading Direct reading from the scale calibrated by 8 hrs. Sampling

6) Colour change White → Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 15% RSD-mid.: 15% RSD-high: 15%

3. CHEMICAL REACTION

Iodine pent-oxide is reduced. $C_6H_5CH_3 + I_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

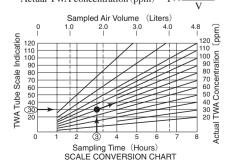
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Acetone	Similar stain is produced	Toluene conc. X 0.2	Higher readings are given.
Xylene	"	Toluene conc. X 0.7	"
Benzene	"	Toluene conc. X 1.8	"
Methyl ethyl ketone	"	Toluene conc. X 0.2	"
Hexane	Whole regent is discoloured to Brown.	50	Whole reagent is discoloured and readings cannot be obtained.

(NOTE)

- 1) Air sampler is required for this tube.
- 2) Flow Rate and Sampling Time
- (1) In case of 8 hours, sampling with 10m ℓ/min., the TWA concentration can be read directly by the scale printed on the tube at the top of Brack stain.
- (2) If the sampling duration is less than 8 hours, the actual TWA concentration can be obtained graphically from the chart provided below.
- (3) If the flow rate is not $10m\ell$ /min, divide the scale reading by the ratio of sampled air volume to $4800m\ell$. Actual TWA concentration (ppm) = $I \times_{-}$



 $I = Scale reading in m \ell$

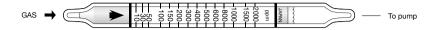
V = Sampled air volume

[Flow rate $(m\ell/min.) \times Sampling duration (min.)$]

- (a) If sampling time is 5 hours and scale reading is 50, the actual TWAconcentration is 80 ppm.
- (b) If sampled air volume is 4.0ℓ , and scale reading is 50, the actual TWA concentration is 60 ppm.

3.5 KITAGAWA DISSOLVED SUBSTANCE DETECTOR TUBES

CHLORIDE ION



1. PERFORMANCE

 1) Sampling method
 : Immersion method (Refer to Page 17)

 2) Measuring range
 : 10-2,000 ppm

 3) Sampling time
 : 1.5 minutes

 4) Sample volume
 : over $5 \text{ m} \ell$

 5) Detectable limit
 : 3 ppm

 6) Shelf life
 : 3 years

7) Operating temperature $5 \sim 80 \,^{\circ}\text{C}$ 8) Operating PH 3.5-13

9) Reading : Direct reading from the scale

10) Colour change : Brown → White

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Silver chromate, Silver chloride is produced. CI − + Ag₂CrO₄ → AgCI

4. CALIBRATION OF THE TUBE

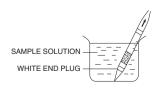
SODIUM CHLORIDE STANDARD SOLUTION METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

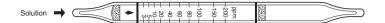
Substance	Interference	Coexistence
Bromide ion		Higher readings are given.
lodide ion		"
Cyanide ion		"
Sulphide ion	Brown stain is produced.	The bottom of discoloured layer is changed to Brown and higher readings are given.

6. SAMPLING METHOD

- 1) Cut both ends of a fresh detector tube with a file.
- 2) Immerse the filled end of the tube with white end plug into the prepared sample solution. Capillary action will occur immediately and the sample solution rise through the reagent. If Chloride ion is existed in the sample solution, a discolouration will be occurred in the detecting reagent layer from its inlet and the discoloured layer shall be given according to the concentration of Chloride ion.



CHLORIDE ION



1. PERFORMANCE

1) Sampling method : Immersion method (Refer to Page 17)

2) Measuring range 3 3-200 ppm 3) Sampling time 4.1.5 minutes 4) Sample volume 5 Detectable limit 6 Shelf life 7 Operating temperature 8 Operating PH 3-200 ppm 1.5 minutes 1.5 minutes 2 years 2 years 3.5-13

9) Reading : Direct reading from the scale

10) Colour change : Brown→White

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Silver chromate, Silver chloride is produced. CI − + Ag₂CrO₄ → AgCI

4. CALIBRATION OF THE TUBE

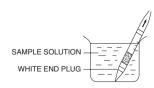
SODIUM CHLORIDE STANDARD SOLUTION METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

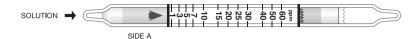
Substance	Interference	Coexistence
Bromide ion		Higher readings are given.
lodide ion		"
Cyanide ion		"
Sulphide ion	Brown stain is produced.	The bottom of discoloured layer is changed to Brown and higher reading are given.

6. SAMPLING METHOD

- 1) Cut both ends of a fresh detector tube with a file.
- 2) Immerse the filled end of the tube with white end plug into the prepared sample solution. Capillary action will occur immediately and the sample solution rise through the reagent. If Chloride ion is existed in the sample solution, a discolouration will be occurred in the detecting reagent layer from its inlet and the discoloured layer shall be given according to the concentration of Chloride ion.



CHLORIDE ION



1. PERFORMANCE

1) Sampling method : Immersion method

2) Measuring range 1-60 ppm

3) Sampling time Approx. 3 minutes 4) Sample volume Over 5 ml

5) Detectable limit 0.5 ppm 6) Shelf life 2 years : 5 ~ 40 ℃

2-13 8) Operating PH

9) Reading : Direct reading from the scale 10) Colour change : Brown → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

7) Operating temperature

By reacting with Silver chromate, Silver chloride is produced. CI - + Ag₂CrO₄ → AgCI

4. CALIBRATION OF THE TUBE

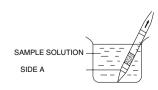
SODIUM CHLORIDE STANDARD SOLUTION METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

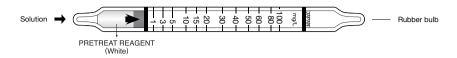
Substance	Interference	Coexistence
Bromide ion		Higher readings are given.
lodide ion		"
Cyanide ion		"
Sulphide ion	The bottom of reagents is changed to Brown and higher readings are given.	

6. SAMPLING METHOD

- 1) Cut both ends of a fresh detector tube with an ampule cutter.
- 2) Immerse the end of the tube with side A into the sample solution by capillary action so that the sample solution is rose through the reagent. If Chloride ion exists in the solution, a discolouration will be occurred in the detecting reagent layer from its inlet and the discoloured layer will be given according to the concentration of Chloride ion.



COPPER ION



1. PERFORMANCE

1) Sampling method : Direct sampling method

(Refer to Page 17)

2) Measuring range $1-100 \text{mg}/\ell$ 3) Sampling time 1 to 2 minutes4) Sample volume $0 \text{ cover } 5 \text{ m} \ell$ 5) Detectable limit $0 \text{ com}/\ell$ 6) Shelf life 0 cover 1 year7) Operating temperature $0 \text{ cover} 1 \text{ cov$

8) Operating PH : 2-11

9) Reading : Direct reading from the scale

10) Colour change : White→Orange

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Hydroxylamine sulphate, divalent Copper ion is reduced to monovalent Copper ion. This monovalent Copper ion is reacted with 2, 9-Diphenyl 1-4, 7-Dipheny 1-1, 10-phenanthroline and Chelate is produced.

 $Cu^{2+} + NH_2OH + H_2SO_4 \rightarrow Cu^+$

Cu⁺+C₁₂H₄N₂ (CH₃)₂ (C₆H₅)₂→Orange Chelate compound

4. CALIBRATION OF THE TUBE

CUPRIC SULPHATE STANDARD SOLUTION METHOD

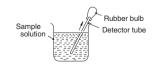
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	mg/ℓ	Interference	mg/ℓ	Coexistence
Ferric ion	20	Similar stain is produced.	Copper ion conc. X 2	Higher readings are given.
Zinc ion		The accuracy of readings in not affected.	100	"
Chlorine ion		"		"
Manganous ion		"		"

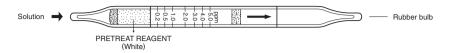
(NOTE)

6. SAMPLING METHOD

- 1) Cut both ends of a fresh detector tube with a file.
- 2) Squeeze the rubber bulb (an extra option), insert the tube end (B) into it as it is and immerse filled end (A) of the tube.
- 3) Put the thumb off the rubber bulb, and the sample solution shall rise up.
- 4) When the sample solution rises up to (C) of the tube, remove the tube from the rubber bulb and from the sample solution.
- The concentration can be obtained directly from the reading value of scale printed on the tube.
- 6) When the concentration is over 100mg/ℓ, dilute the sample solution and multiply the readings obtained by the dilution ratio.



CYANIDE ION



1. PERFORMANCE

1) Sampling method : Direct sampling method (Refer to Page 17)

2) Measuring range 0.2-5 ppm 3) Sampling time 0.2-5 2 to 4 minutes 4) Sample volume 0.05 0.05 ppm 6) Shelf life 0.05 2 years 7) Operating temperature 0.2-5 ppm 0.

8) Operating PH : 6-13

9) Reading : Direct reading from the scale

10) Colour change : White→Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with o-Toluidine and Cupric sulphate (II), complex salt is produced.

4. CALIBRATION OF THE TUBE

POTASSIUM CYANIDE STANDARD SOLUTION METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

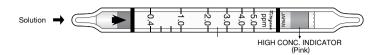
Substance	Interference	ppm	Coexistence
Carbonate ion		1,700	Higher readings are given.
Chloride ion		100	"
Sulphate ion		2,700	Lower readings are given.
Thiocyanate ion	Similar stain is produced.	200	Higher readings are given.
Sulphide ion			"
Dichromate ion			Pretreat reagent is discoloured and the accuracy of readings is not affected.
Permanganate ion			"
Ferricyanate ion			"
Residual chloride ion			"

6. SAMPLING METHOD

- 1) Make the sample solution at PH 6-13 before test.
- 2) Cut both ends of a fresh detector tube with a file.
- 3) Squeeze the rubber bulb (an extra option), insert the tube end (B) into it as it is and immerse filled end (A) of the tube.
- 4) Put the thumb off the rubber bulb, and the sample solution shall rise up.
- 5) When the sample solution rises up to (C) of the tube, remove the tube from the rubber bulb and from the sample solution.
- 6) The concentration can be obtained directly from the reading value of scale printed on the tube.
- 7) At concentration of over 5 ppm, dilute the sample solution and multiply the readings obtained by the dilution ratio.



FREE RESIDUAL CHLORINE



1. PERFORMANCE

1) Sampling method : Immersion method (Refer to Page 17)

2) Measuring range 3 (0.4-5 ppm 3) Sampling time 3 minutes 4) Sample volume 5 (0.4 ppm 6) Detectable limit 6) Shelf life 2 years 7) Operating temperature 8) Operating PH 12-10

9) Reading : Direct reading from the scale

10) Colour change : White→Purple

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 10 %

3. CHEMICAL REACTION

By reacting with 3, 3'-Dimethylnaphthidine, Nitroso-compound is produced.

4. CALIBRATION OF THE TUBE

IODOMETRY METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Chloride ion		200	Lower readings are given.
Calcium ion			The accuracy of readings is not affected.
Copper ion			"
Iron ion	Similar stain is produced.	20	Higher readings are given.

(NOTE)

- 1. This tube is for measuring FREE-RESIDUAL CHLORINE, not for COMBINED-RESIDUAL CHLORINE.
- This tube is not suitable for measuring sea water or sample solution which includes sea water. Because it is affected by Chloride ion.
- 3. When the concentration is over full scale (5 ppm), the HIGH CONC. INDICATOR (Pink) is discoloured to White. In this case, dilute the sample solution and multiply the reading value obtained, by the dilution ratio.

6. SAMPLING METHOD

- 1) Cut both ends of a fresh detector tube with a file.
- 2) Immerse the filled end of the tube with white end plug into the prepared sample solution. Capillary action will occur immediately and the sample solution rise through the reagent. If Chloride ion is existed in the sample solution, a discolouration will be occurred in the detecting reagent layer from its inlet and the discoloured layer shall be given according to the concentration of Chloride ion.

SALINITY



1. PERFORMANCE

1) Sampling method : Direct sampling method

(Refer to Page 17)

2) Measuring range : 0.01-0.8% 3) Sampling time : 30 seconds 4) Sample volume $: 0 \text{ ver } 5 \text{ m}\ell$ 5) Detectable limit : 0.002% 6) Shelf life : 2 years 7) Operating temperature $: 5 \sim 80\%$

8) Reading : Direct reading from the scale

9) Colour change : Brown→White

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Silver chromate, Silver chloride is produced.

 $CI^- + Ag_2CrO_4 \rightarrow AgCI$

4. CALIBRATION OF THE TUBE

SODIUM CHLORIDE STANDARD SOLUTION METHOD

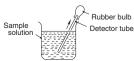
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Bromide ion		Higher readings are given.
lodide ion		"
Cyanide ion		"
Sulphide ion	Brown stain is produced.	The bottom of the discoloured layer is changed to Brown and higher readings are given.

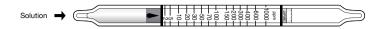
6. SAMPLING METHOD

(Direct sampling method)

- 1) Cut both ends of a fresh detector tube with a file.
- 2) Squeeze the rubber bulb (an extra option), insert the tube end (B) into it as it is and immerse filled end (A) of the tube.
- 3) Put the thumb off the rubber bulb, and the sample solution shall rise up.
- 4) When the sample solution rises up to (C) of the tube, remove the tube from the rubber bulb and from the sample solution.
- 5) The concentration can be obtained directly from the reading value of scale printed on the tube.
- At concentration over 0.8 %, dilute the sample solution and multiply the readings obtained by the dilution ratio.



SULPHIDE ION



1. PERFORMANCE

1) Sampling method : Immersion method (Refer to Page 17)
2) Measuring range : 2-1,000 ppm
3) Sampling time : 3 minutes

3) Sampling time 3. Similates 4) Sample volume 5. Over 5 m ℓ 5) Detectable limit 1 ppm 6) Shelf life 1 year 7) Operating temperature 5 \sim 60 °C 8) Operating PH 1-12

9) Reading : Direct reading from the scale

10) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Lead nitrate, Lead sulphide is produced. S2 + Pb(NO₃)2→PbS

4. CALIBRATION OF THE TUBE

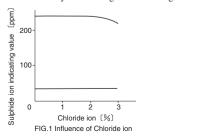
SODIUM SULPHIDE STANDARD SOLUTION METHOD

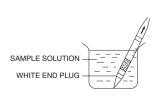
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Dichromate ion		500	Lower readings are given.
Mercaptans ion			Similar stain is produced.
Sulphate ion		less than 1.5 %	The accuracy of readings is not affected.
Iron ion		less than 0.2%	"
Chloride ion FIG.1		less than 2.5 %	"
Carbonic ion		less than 4.0 %	"

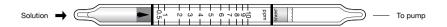
6. SAMPLING METHOD

- 1) Cut both ends of a fresh detector tube with a file.
- 2) Immerse the filled end of the tube with white end plug into the prepared sample solution. Capillary action will occur immediately and the sample solution shall rise through the reagent. If Sulphide ion is existed in the sample solution, a discolouration will be occurred in the detecting reagent layer from its inlet and the discoloured layer shall be given according to the concentration of Sulphide ion.





SULPHIDE ION



1. PERFORMANCE

1) Sampling method : Immersion method (Refer to Page 17)
2) Measuring range : 0.5-10 ppm

2) Measting range 2.2.5 minutes 3.9 Sampling time 2.2.5 minutes 4.9 Sample volume 5.0 Detectable limit 5.0.1 ppm 6.9 Shelf life 2.2 years 7.0 Operating temperature 5.5 \sim 40 $^{\circ}$ C 8.0 Operating PH 5.7-14

9) Reading : Direct reading from the scale 10) Colour change : White → Pale brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Lead acetate, Lead sulphide is produced. $S^2 - + (CH_3CO_2)_2Pb \rightarrow PbS$

4. CALIBRATION OF THE TUBE

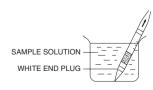
SODIUM SULPHIDE STANDARD SOLUTION METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Carbonate ion		1,000	Lower readings are given.
Chloride ion		2,000	Higher readings are given.
Copper ion			The accuracy of readings is not affected.

6. SAMPLING METHOD

- 1) Cut both ends of a fresh detector tube with a file.
- 2) Immerse the filled end of the tube with white end plug into the shall prepared sample solution. Capillary action will occur immediately and the sample solution shall rise through the reagent. If Chloride ion is existed in the sample solution, a discolouration will be occurred in the detecting reagent layer from its inlet and the discoloured layer shall be given according to the concentration of Chloride ion.



3.6 KITAGAWA COMPRESSED BREATHING AIR TEST TUBES

COMPRESSED BREATHING AIR TEST SYSTEM

This system is designed to measure impurities in compressed breathing air such as in scuba and rescue cylinders, as well as from an outlet of an air-change compressor.

● Model P-41R Compressed Breathing Air Sampling Kit



Composition

① Control assembly
(Including an adaptor with W22-14RH Femail thread for rescue and on-land cylinders)
2 International fitting yoke (For a scuba cylinder) 1pc.
3 Gas detector tube (an extra option) 1pc.
4 Tube protector
5 Tip cutter for Gas detector tube1pc.
⑥ Wrench 1pc.
⑦ Digital stopwatch1pc.
8 Carrying case (Aluminum)1pc.
9 Instruction manual1set.

Optional Accessiry for 604SP only

- 1 50ml plastic syringe
- 2 1m vinyl tube

Tube No. **600SP**

CARBON MONOXIDE



1. PERFORMANCE

1) Measuring range 2.5-5 ppm 5-100 ppm 2) Sampling time 4 minutes 2 minutes 3) Detectable limit 1 ppm (4 minutes)

4) Shelf life \therefore 2 years 5) Operating temperature \therefore 0 \sim 40 °C

6) Reading : Direct reading from the scale printed on the tube

7) Colour change : Yellow→Dark brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Pottasium disulphate palladate (II) is reduced and Palladium is liberated.

 $CO + K_2Pd(SO_3)_2 \rightarrow Pd$

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. PRESET

PRESET CONDITION

1) 2nd pressure : 0.6 kgf/cm² (60 Kpa)

2) Flow rate \therefore 2.0 ℓ /min.

CARBON DIOXIDE



1. PERFORMANCE

1) Measuring range : 100-3,000 ppm2) Sampling time : 2 minutes3) Detectable limit : 20 ppm4) Shelf life : 2 years5) Operating temperature $: 0 \sim 40 \text{ }^{\circ}\text{C}$

6) Reading : Direct reading from the scale printed on the tube

7) Colour change : Purplish blue → Pale pink

2. RELATIVE STANDARD DEVIATION

RSD-low : 10 % RSD-mid. : 10 % RSD-high : 10 %

3. CHEMICAL REACTION

By reacting with an Alkali, PH indicator is discoloured. CO₂ + 2NaOH → Na₂CO₃ + H₂O

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

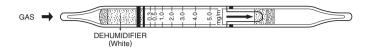
5. PRESET

PRESET CONDITION

1) 2nd pressure : 1.0 kgf/cm² (100 Kpa)

2) Flow rate : $1.0 \ell/\text{min}$.

COMPRESSED BREATHING AIR TEST TUBE FOR OIL MIST



1. PERFORMANCE

1) Measuring range $0.3-5 \text{ mg/m}^3$ 2) Sampling time $0.3-5 \text{ mg/m}^3$ 3) Detectable limit 0.1 mg/m^3 4) Shelf life 0.2 years5) Operating temperature $0.3-6 \text{ mg/m}^3$

6) Reading : Direct reading from the scale printed on the tube

7) Colour change : Yellow→Pale blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Chromium oxide is reduced. Oil + Cr^{6+} + $H_2SO_4 \rightarrow Cr^{3+}$

4. CALIBRATION OF THE TUBE

INFRARED SPECTOPHOTOMETRY

5. PRESET

PRESET CONDITION

1) 2nd pressure : 1.0 kgf/cm² (100Kpa)

2) Flow rate $3.0 \ell/\text{min}$.

COMPRESSED BREATHING AIR TEST TUBE FOR WATER VAPOUR



1. PERFORMANCE

1) Measuring range $2.0-160 \text{ mg/m}^3$ 2) Sampling time 2.0 1 minute3) Detectable limit 2.0 1 minute4) Shelf life 2.0 years5) Operating temperature 0.0 C

6) Reading : Direct reading from the scale printed on the tube

7) Colour change : Yellow → Yellowish green or Blue

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Magnesium perchlorate, complex salt is produced and PH indicator is discoloured. $H_2O + Mg(CIO_4)_2 \rightarrow Mg(CIO_4)_2 + H_2O$

4. CALIBRATION OF THE TUBE

DEW-POINT METER

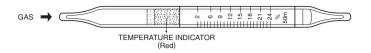
5. PRESET

PRESET CONDITION

1) 2nd pressure : 1.0 kgf/cm² (100 Kpa)

2) Flow rate \therefore 2.0 ℓ /min.

COMPRESSED BREATHING AIR TEST TUBE FOR **OXYGEN**



1. PERFORMANCE

1) Measuring range 2.24%2) Sampling time 1 minute3) Shelf life 2 years4) Operating temperature $0 \sim 40\%$

5) Temperature compensation : (Necessary (0 ~ 10/30 ~ 40 ℃) (Refer to "SPECIAL NOTE")

6) Reading : Direct reading from the scale printed on the tube

7) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

Oxygen reacts with alkaline pyrogallol.

4. CALIBRATION OF THE TUBE

STANDARD GAS CYLINDER METHOD

5. PRESET/MEASUREMENT CONDITIONS

PRESET CONDITION

1) 2nd pressure : 0.6 kgf/cm² (60 Kpa)

2) Flow rate \therefore 2.0 ℓ /min.

MEASUREMENT CONDITION

1) Sampling amount : $50 \text{ m}\ell$

SPECIAL NOTE

1) TEMPERATURE COMPENSATION:

At 0° (32°) to 10° (86°), multiply the reading value by 1.05.

At 30 $^{\circ}$ C (86 $^{\circ}$ C) to 40 $^{\circ}$ C (104 $^{\circ}$ C), multiply the reading value by 0.95.

- 2) The sampling is made with an optional $50m\ell$ plastic syringe which collects accurate sampling volume in $50m\ell$ and $1m\ell$ vinyl tube which connects the syringe and the detector rubber, is recommendable in order to prevent influence by ambient oxygen.
- 3) Pressure regulator of which diaphragm is made of stainless steel or Teflon coated rubber, is recommendable in order to prevent influence by ambient oxygen.

3.7 KITAGAWA INDOOR AIR POLLUTANTS MEASUREMENT DETECTOR TUBES

FORMALDEHYDE

DETECTOR TUBE



1. PERFORMANCE

1) Measuring range : 0.01-0.12 ppm 0.04-0.48 ppm 2) Sampling time : 300mℓ/min × 30min 300mℓ/min × 10min

3) Detectable limit : $0.005 \text{ ppm} (300 \text{m} \ell \times 30 \text{min})$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10 \, ^{\circ}\mathrm{C}$)

5) Operating temperature : $10 \sim 35 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated at the sampling of $300 \text{m} \ell \times 30 \text{min}$

8) Colour change : Yellowish orange → Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Hydroxylamine phosphate, Phosphoric acid is liberated. HCHO + (NH₂OH)₃⋅ H₃PO₄→ H₃PO₄ + HCN = NOH + H₂O

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Ammonia	The accuracy of readings is not affected.	0.5	Discolouration layer fades from the bottom of stained layer.
Amines	"	0.5	"
Nitrogen dioxide	Similar stain is produced.	0.5	Higher readings with indiscernible maximum end point of stained layer are given.
Acetaldehyde	"		Higher readings are given.
Acetone	"		"

(NOTE)

Air sampler is required for this tube.

TABLE OF THE COEFFICIENT FOR TEMPERATURE CORRECTION (20 °C standard)

THE COLL FIGURE OF THE COLL FOR TELL FOR THE COLL FOR THE										
Temp(°C)	0	1	2	3	4	5	6	7	8	9
10	1.40	1.36	1.32	1.28	1.24	1.20	1.16	1.12	1.08	1.04
20	1.00	0.97	0.94	0.91	0.88	0.85	0.82	0.79	0.76	0.73
30	0.70	0.67	0.64	0.61	0.58	0.55		_		_

FORMALDEHYDE

DETECTOR TUBE



1. PERFORMANCE

1) Measuring range : 0.05-1.0 ppm 0.10-2.0 ppm 2) Sampling time : 300mℓ/min × 30min 300mℓ/min × 15min

3) Detectable limit : $0.005 \text{ ppm} (300 \text{m} \ell \times 30 \text{min})$

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10 \, ^{\circ}\mathrm{C}$)

5) Operating temperature : $10 \sim 35 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated at the sampling of $300 \text{m} \ell \times 30 \text{min}$

8) Colour change : Yellowish orange→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Hydroxylamine phosphate, Phosphoric acid is liberated. $HCHO + (NH_2OH)_3 \cdot H_3PO_4 \rightarrow H_3PO_4 + HCN = NOH + H_2O$

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Ammonia		0.5	Discolouration layer fades from the bottom of stained layer.
Amines		0.5	"
Nitrogen dioxide	Similar stain is produced.	0.5	Higher readings with indiscernible maximum end point of stained layer are given.
Acetaldehyde	"		Higher readings are given.
Acetone	"		"

(NOTE)

Air sampler is required for this tube.

TABLE OF THE COEFFICIENT FOR TEMPERATURE CORRECTION(20°C standard)

THE COLL TICLETT FOR TELL ENVIOUE CONTINUON CONTINUON										
Temp(℃)	0	1	2	3	4	5	6	7	8	9
10	1.16	1.14	1.13	1.11	1.10	1.08	1.06	1.05	1.03	1.02
20	1.00	0.98	0.97	0.95	0.94	0.92	0.90	0.89	0.87	0.86
30	0.84	0.82	0.81	0.79	0.78	0.76	_			

FORMALDEHYDE



1. PERFORMANCE

1) Measuring range : $0.01 \sim 0.50 \text{ ppm} (12.5 \sim 625 \mu \text{g/m}^3)$

2) Sampling volume 3) Sampling time 3.350mℓ/min × 10min 4) Detectable limit 5.0.005 ppm 3.350mℓ/min × 10min 5.0005 ppm 3.350mℓ/min × 10min × 10min 5.0005 ppm 3.350mℓ/min × 10min × 10min

5) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10^{\circ}$ C)

6) Operating temperature : $10 \sim 35 \,^{\circ}\text{C}$

7) Temperature compensation: Necessary (See "TEMPERATURE CORRECTION TABLE")

8) Reading ∴ Direct reading from the scale calibrated at sampling volume of 350m × 10min

9) Colour change : Yellowish orange → Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Hydroxylamine phosphate, Phosphoric acid is liberated. HCHO + (NH₂OH)₃ · H₃PO₄ → H₃PO₄ + HCN = NOH + H₂O

4. CALIBRATION OF THE TUBE

ABSORPTIOMETRIC METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	ppm	Coexistence
Ammonia		The accuracy of readings is not affected.	1.0	Discolouration layer fades from the bottom of stained layer.
Amines		"	1.0	//
Ethanol		"	200	Higher readings are given.
Nitrogen dioxide	1	Similar stain is produced.	1.0	Higher readings with indiscernible maximum end point of stained layer are given.
Acetaldehyde		"		Higher readings are given.
Acetone		"		"

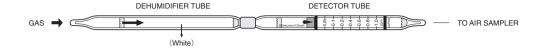
(NOTE)

Air sampler is required for this tube.

TEMPERATURE CORRECTION TABLE

Tube	Corrected Concentration (ppm)								
Readings (ppm)	10 ℃ (50 F)	15℃ (59F)	20 ℃ (68 F)	25℃ (77F)	30 ℃ (86 F)	35℃ (95°F)			
	(301)								
0.50	_	0.780	0.500	0.390	0.340	0.290			
0.40	0.900	0.520	0.400	0.310	0.270	0.230			
0.30	0.550	0.370	0.300	0.230	0.200	0.170			
0.20	0.330	0.250	0.200	0.155	0.135	0.115			
0.10	0.150	0.120	0.100	0.080	0.070	0.060			
0.05	0.070	0.060	0.050	0.040	0.035	0.030			
0.01	0.020	0.015	0.010	0.008	0.007	0.006			

TOLUENE



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \hbox{:} \ 0.05\text{-}1.0 \ ppm \\ \hbox{:} \ 200 \text{m}\ell/\text{min} \times 20 \text{min} \\ \end{array}$

3) Detectable limit : 0.01 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions ; 0 \sim 10 $^{\circ}\mathrm{C})$

5) Operating temperature : $5 \sim 35 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated at the sampling of $300 \text{m} \ell \times 30 \text{min}$

7) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced and Iodine pentoxide is liberated.

 $C_6H_5CH_3 + I_2O_5 + H_2SO_4 \rightarrow I_2$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

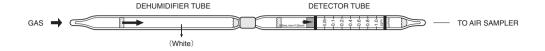
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence		
Xylene	Similar stain is produced.	Higher readings are given.		
Ethyl Benzene	"	"		

(NOTE)

Air sampler is required for this tube.

ETHYL BENZENE



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \hbox{:} \ 0.05\text{-}1.2 \ ppm \\ \hbox{:} \ 200 \text{m$\ell/min} \times 20 \text{min} \end{array}$

3) Detectable limit : 0.01 ppm : 1 year (Necessary to store in refrigerated conditions ; $0 \sim 10^{\circ}$ C)

5) Operating temperature : $5 \sim 35 \,^{\circ}\text{C}$

6) Reading : Scales printed on the tube are calibrated by Toluene at the sampling of

 $200 \text{m}\,\ell \times 20 \text{min}$ and Ethyl benzene concentration is determined by using

a conversion chart.

7) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced and Iodine pentoxide is liberated. C6H5CH3 + I2O5 + H2SO4→I2

4. CALIBRATION OF THE TUBE

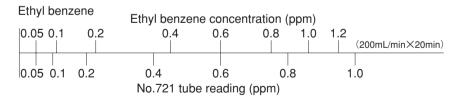
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

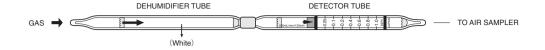
Substance	Interference	Coexistence
Toluene	Similar stain is produced.	Higher readings are given.
Xylene	"	"

(NOTE)

Air sampler is required for this tube.



XYLENE



1. PERFORMANCE

1) Measuring range : 0.1-1.4 ppm

2) Sampling time : $200 \text{m}\ell/\text{min} \times 20 \text{min}$

3) Detectable limit : 0.01 ppm

4) Shelf life : 1 year (Necessary to store in refrigerated conditions; $0 \sim 10 \, ^{\circ}\text{C}$)

5) Operating temperature : $5 \sim 35 \,^{\circ}\text{C}$

6) Reading : Scales printed on the tube are calibrated by Toluene at the sampling of

 $200 \text{m}\,\ell \times 20 \text{min}$ and Xylene concentration is determined by using

a conversion chart.

7) Colour change : White→Brown

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

Iodine pentoxide is reduced and Iodine pentoxide is liberated. C6H5CH3 + I2O5 + H2SO4→I2

4. CALIBRATION OF THE TUBE

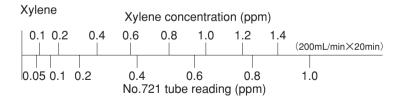
GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Toluene	Similar stain is produced.	Higher readings are given.
Ethyl Benzene	"	"

(NOTE)

Air sampler is required for this tube.



p-DICHLOROBENZENE



1. PERFORMANCE

1) Measuring range $\begin{array}{c} \hbox{:} \ 0.01\text{-}0.40 \ ppm \\ \hbox{:} \ 200 \text{mℓ/min} \times 15 \text{min} \\ \end{array}$

3) Detectable limit : 0.002 ppm4) Shelf life : 1 year5) Operating temperature $: 10 \sim 35 \text{ }^{\circ}\text{C}$

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated at the sampling of $200 \text{m} \ell \times 15 \text{min}$

8) Colour change : Orange → Reddish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. C₆H₄Cl₂ + PbO₂ + H₂SO₄ → HCI

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Trichloroethylene	Similar stain is produced.	Higher readings are given.
Tetrachloroethylene	"	"
1,2-Dichloroethylene	"	"
Vinyl chloride	"	"

(NOTE)

Air sampler is required for this tube.

TABLE OF THE COEFFICIENT FOR TEMPERATURE CORRECTION (20 °C standard)

Temp(°C)	0	1	2	3	4	5	6	7	8	9
10	2.13	1.95	1.78	1.63	1.50	1.38	1.28	1.19	1.11	1.05
20	1.00	0.95	0.92	0.88	0.84	0.81	0.78	0.75	0.73	0.72
30	0.70	0.69	0.68	0.68	0.67	0.66	_	_	_	_

3.8 ATMOSPHERIC ENVIRONMENT MEASUREMENT DETECTOR TUBES

NITROGEN DIOXIDE



1. PERFORMANCE

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Temperature compensation : Necessary (30 \sim 40°C) (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated at the sampling of 200m ℓ /min \times 20min

8) Colour change : White → purple-red

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with 3,3'-Dimethylnaphthidine, Nitroso compound is produced. NO₂ + C₂₂H₂₀N₂ → Purple red reaction products

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	ppm	Coexistence
Sulphur dioxide		0.3	Lower readings are given.
Ozone		0.2	"
Carbon monoxide			The accuracy of readings is not affected.
Carbon dioxide			"
Nitrogen monoxide			"
Formaldehyde			"
Toluene			"
Xylene			"
Ethyl benzene			"
Styrene			"
p-Dichlorobenzene			"
Acetone			"

(NOTE)

Air sampler is required for this tube.

TABLE OF THE COEFFICIENT FOR TEMPERATURE CORRECTION

Temperature(℃)	31	32	33	34	35	36	37	38	39	40
Correction factor	1.02	1.04	1.06	1.08	1.10	1.12	1.14	1.16	1.18	1.20

TRICHLOROETHYLENE



1. PERFORMANCE

1) Measuring range : 30-400 μg/m³ 69-920 μg/m³ 2) Sampling time : 100mℓ/min × 30min 1000mℓ/min × 15min

3) Detectable limit : $5 \mu g/m^3 (100 \text{m} \ell/\text{min} \times 30 \text{min})$

4) Shelf life : 1 year 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

 $\hbox{\bf 6) Temperature compensation } \hbox{\bf :} \ \ Necessary (See "TEMPERATURE CORRECTION TABLE")}$

7) Reading : Direct reading from the scale calibrated at the sampling of $100 \text{m} \ell \times 30 \text{min}$

8) Colour change : Yellow orange → Purple red

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. CCl₂ = CHCl + PbO₂ + H₂SO₄→ HCl

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Tetrachloroethylene	Similar stain is produced.	Higher readings are given.
1,2-Dichloroethylene	"	"
Vinyl chloride	"	"

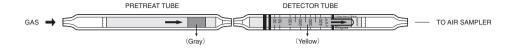
(NOTE)

Air sampler is required for this tube.

TABLE OF THE COEFFICIENT FOR TEMPERATURE CORRECTION(20°C standard)

Temp(℃)	0	1	2	3	4	5	6	7	8	9
0	1.39	1.36	1.34	1.32	1.30	1.28	1.26	1.24	1.22	1.20
10	1.18	1.16	1.14	1.13	1.11	1.09	1.07	1.05	1.04	1.02
20	1.00	0.98	0.97	0.95	0.93	0.92	0.90	0.88	0.87	0.85
30	0.84	0.52	0.81	0.79	0.78	0.76	0.75	0.73	0.72	0.71
40	0.69	_	_	_	_	_	_	_	_	_

TETRACHLOROETHYLENE



1. PERFORMANCE

1) Measuring range : $30\text{-}400~\mu\text{g/m}^3$ 69-920 $\mu\text{g/m}^3$ 2) Sampling time : $100\text{m}\ell/\text{min} \times 30\text{min}$ $100\text{m}\ell/\text{min} \times 15\text{min}$

3) Detectable limit : $5 \mu g/m^3 (100 \text{m} \ell/\text{min} \times 30 \text{min})$

4) Shelf life : 1 year 5) Operating temperature : $0 \sim 40^{\circ}$ C

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE")

7) Reading : Direct reading from the scale calibrated at the sampling of $100 \text{m} \ell \times 30 \text{min}$

8) Colour change : Yellow→Purple red

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with an Oxidizer, Hydrogen chloride is produced and PH indicator is discoloured. $Cl_2C = CCl_2 + PbO_2 + H_2SO_4 \rightarrow HCl$

4. CALIBRATION OF THE TUBE

GAS CHROMATOGRAPHY

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Tetrachloroethylene	Similar stain is produced.	Higher readings are given.
1,2-Dichloroethylene	"	"
Vinyl chloride	"	"

(NOTE)

Air sampler is required for this tube.

TABLE OF THE COEFFICIENT FOR TEMPERATURE CORRECTION(20 °C standard)

Temp(°C)	0	1	2	3	4	5	6	7	8	9
0	1.71	1.66	1.62	1.58	1.54	1.50	1.46	1.42	1.39	1.35
10	1.31	1.28	1.24	1.21	1.18	1.15	1.12	1.09	1.06	1.03
20	1.00	0.97	0.95	0.92	0.90	0.87	0.85	0.83	0.81	0.79
30	0.77	0.75	0.73	0.71	0.70	0.68	0.67	0.65	0.64	0.63
40	0.61	_	_	_	_	_	_	_	_	_

3.9 SUPER-HIGH SENSITIVITY DETECTOR TUBES FOR AMMONIA IN ART GALLERIES/ MUSEUM AND CLEAN ROOMS

AMMONIA



1. PERFORMANCE

1) Measuring range : $10-80 \mu g/m^3$

2) Sampling time : $24000 \text{m} \ell (400 \text{m} \ell / \text{min} \times 60 \text{min})$

3) Detectable limit :—
4) Shelf life : 2

4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 40 °C

6) Reading : Direct reading from the scale calibrated at the sampling of $400 \text{m} \ell \times 60 \text{min}$

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 10%

3. CHEMICAL REACTION

Indicator is discoloured by neutralization of Ammonia with acid.

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

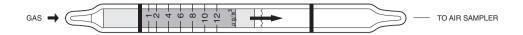
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.

(NOTE)

Air sampler is required for this tube.

AMMONIA



1. PERFORMANCE

1) Measuring range : $1-12 \mu g/m^3$

2) Sampling time : $24000 \text{m} \ell (400 \text{m} \ell/\text{min} \times 60 \text{min})$

3) Detectable limit :—

4) Shelf life \therefore 2 years 5) Operating temperature \therefore 10 \sim 40 °C

6) Reading : Direct reading from the scale calibrated at the sampling of $400 \text{m} \ell \times 60 \text{min}$

7) Colour change : Pale purple → Pale yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 5% RSD-mid.: 5% RSD-high: 10%

3. CHEMICAL REACTION

Indicator is discoloured by neutralization of Ammonia with acid.

4. CALIBRATION OF THE TUBE

PERMEATION TUBE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Amines	Similar stain is produced.	Higher readings are given.

(NOTE)

Air sampler is required for this tube.

4. REFERENCE FOR MEASUREMENT

4.1 GENERAL KNOWLEDGE FOR GAS MEASUREMENT

4.1.1 UNIT OF CONCENTRATION

%: percentage of volumetric ratio of gas, corresponding to millilitres of gas per 100 millilitres of air.

ppm: volumetric ratio of gas expressed by parts per million, corresponding to microlitres ($\mu \ell$) per 1 litre of air, e.g. 0.1 % = 1,000 ppm

mg/m³: milligrams of substance per 1 cubic metre $(1,000\ell)$ of air, mainly used for concentration of particulate matter

(1) Conversion of gas volume

Volume of gas, V at t C under a pressure of phPa is derived from the volume, Vo at 0° C under 1 atmospheric pressure (1013 hPa) according to the following formula:

$$V = V_0 \times \frac{273 + t}{273} \times \frac{1013}{p}$$

 $V=V_0\times\frac{273+t}{273}\times\frac{1013}{p}$ For example, volume of gas, V at 20 °C under 1 atm. is 1.073Vo.

(2) Relationship between volume and weight of gas

Volume of gas vapourized from M grams of a M molecular weight substance, i.e. 1 mol., is 22.4 litre at 0°C under 1 atmospheric pressure. Preparing the standard gas vapourized from a liquid sample, the concentration of gas in a fixed volume vessel is obtained from the following formula:

$$C = \frac{x}{M} \times \frac{22.4}{V} \times \frac{273 + t}{273} \times \frac{1013}{p} \times 100$$

 $C = \frac{x}{M} \times \frac{22.4}{V} \times \frac{273 + t}{273} \times \frac{1013}{p} \times 100$ where C = concentration of gas(%); x = quantity of sample(g); M = molecularweight of sample [g] and $V = internal volume of vessel(<math>\ell$). For example, the quantity of sample required for preparing the standard gas of 1,000 ppm (0.1%) of toluene in a vessel of 1 litre at 20 °C under 1 atm. is as follows:

$$0.1 = \frac{x}{92.14} \times \frac{22.4}{1} \times \frac{273 + 20}{273} \times \frac{1013}{1013} \times 100$$
 Then, $x = 0.00383$ (g) $= 3.83$ (mg)

The relationship in gaseous state of a substance (molecular weight = M) between ppm and mg/m³ at t°C under phPa is as follows:

$$ppm = mg/m^3 \times \frac{22.4}{M} \times \frac{273 + t}{273} \times \frac{1013}{p} \qquad mg/m^3 = ppm \times \frac{M}{22.4} \times \frac{273}{273 + t} \times \frac{P}{1013}$$
 (4) Concentration derived from saturated vapour pressure

Vapour concentration on the surface of liquid, the upper limit at the temperature, is obtained according to the following formula:

$$C = \frac{Pt}{1013} \times 100$$
 where Pt = standard vapour pressure of substance at t°C (hpa) and C = vapour concentration(%)

4.1.2 NORMAL COMPOSITION OF AIR

Many studies on the chemical composition of atmosphere are made so far. The standard chemical composition of air is shown in Table 4.1. Various gaseous and particulate matters besides the nomal Components are contained in the atmosphere on the surface of the earth. Air in urban districts contains the substances shown in Table 4.2.

4.1.3 PROPERTIES OF SUBSTANCES

There are two groups of substances in air, uniformaly dispersed gas or vapour and suspended particles. The names and properties of the particles are shown in Table 4.3. The substances measurable by gas detector tubes are gaseous and vapourized matter.

TABLE 4.1 NORMAL COMPOSITION OF AIR

GASES	CONCENTRATION			
GASES	VOLUME [ppm]	WEIGHT [ppm]		
Nitrogen	780,900	775,100		
Oxygen	209,500	231,500		
Argon	9,300	12,800		
Carbon dioxide	300	400		
Neon	18	12.5		
Helium	5.2	0.72		
Methane	2.2	1.2		
Krypton	1	2.9		
Nitrous oxide	1	1.5		
Hydrogen	0.5	0.03		
Xenon	0.08	0.36		

TABLE 4.2 POLLUTANTS IN URBAN DISTRICT AIR

POLLUTANTS	CONCENTRATION [ppm]
Sulfur dioxide	0.001 ~ 0.7
Carbon monoxide	2.0~10.0
Aldehyde, as formaldehyde	0.02~0.2
Nitrogen oxides	0.02~0.9
Ozone	0.009~0.3
Ammonia	0.02~0.2
Hydrogen sulfide	0.002~0.1
Hydrogen fluoride	0.001 ~ 0.02

4.1.4 HARMFULNESS

Some chemicals are harmful to the touch, depending on exposure time, quantity and human characteristics such as health condition, sex, age and others.

When a high concentration chemical substance is inhaled in a short time, symptoms of acute poisoning are developed. When a relatively low concentration chemical substance is inhaled for many hours, symptoms of chronic poisoning are developed. Acute poisoning is relatively well known, but chronic poisoning is not so well known, because it has various types of symptoms in subjective and objective symptoms. Since the symptoms by long exposure to low-concentration chemicals are not characteristic, the subjective symptoms must be carefully observed.

Classification of chemical substances by symptom is as follows:

- (1) Simple suffocative substances ... carbon dioxide, methane, propane
- (2) Chemical suffocative substances ··· carbon monoxide, hydrogen cyanide
- (3) Upper air-passage irritating substances ··· ammonia, sulphur dioxide, styrene, formaldehyde
- (4) Pulmonary issue irritating substances ··· chlorine, phosgene, ozone
- (5) Toxic substances to central nervous system ··· toluene, xylene, chloroform
- (6) Hematogenous organs and hematic system poisoning substances ⋅ ⋅ ⋅ benzene, arsine, nitrobenzene

TABLE 4.3 POLLUTANTS IN URBAN DISTRICT AIR

SUBSTANCE	PROPERTIES	STATE	PARTICLE DIAMETER (μ)	EXAMPLES
Gas	Gas at 25 ℃ under I atm. (critical temperature or higher)	Gas	Molecules	Ammonia, Chlorine, Ozone, Carbon dioxide, Sulphur dioxide
Vapour	Solid (critical temperature or lower) or liquid at 25 $^{\circ}$ C under 1 atm.	Gas	Wolecules	Acetone, Ethyl alcohol, Carbon disulphide, Ethyl acetate
Mist	Suspended fine particles of solid	Liquid	5-100	Sulphuric acid (0.8-5.5 μ), Chromic acid , Sodium hydroxide , Cyanide , Hydrochloric acid, Fog (50 μ)
Fog	Finer particles than mist (5-50 μ particles condensed in air)			
Dust	Suspended fine particles of solid Suspended fine particulate com-		1-150	Pigment (1-4 μ). Talc (10 μ), Coal dust (10 μ), Grain flour (15-20 μ), Cement mill dust (exhaust from kiln, 10 μ), Dust in grain flour elevator
ruille	pound derived from gas in air	Solid		$(15 \ \mu)$, P.C.P., Mica, Asbestos
			0.1-1	Zinc oxide (0.05 μ), Ammonium chloride (0.1 μ)
Smoke	Originally meaning of organic particles with a certain black colour	Mixed	0.01-1	Tobacco smoke (0. 25 μ)

- (7) Carcinogens ... vinyl chloride, benzidine
- (8) General poisoning substances ··· lead, mercury, arsenic, aniline
- (9) Hepatic system poisoning substances ··· carbon tetrachloride, trichloroethylene, tetrachloroethylene

Exposure to high concentration of these substances gives shortly symptoms, but generally the substances (1) through (4) develop acute poisoning, while those (5) through (9) develop chronic poisoning.

4.1.5 THRESHOLD LIMIT VALUE

Threshold limit values of chemicals have been published from various organizations in the world as indicators of exposure for prevention of troubles so far.

The threshold limit values used most widely in the world are Threshold Limit value, TLV, published from American Conference of Governmental Industrial Hygienists (ACGIH). TLV is stated in the same manner as in the preceding documents as follows:

"Threshold limit values refer to airborne concentrations of substances and represent conditions under which it is believed that nearly all workers may be repeatedly exposed day after day without adverse effect. Because of wide variation in individual susceptibility, however, a small percentage of workers may experience discomfort from some substances at concentrations at or below the threshold limit; a smaller percentage may be affected more seriously by aggravation of a pre-existing condition or by development of an occupational illness."

The threshold limit value is expressed in the unit of ppm or mg/m³ which is generally calculated from the concentrations by the actual working hours in a day, time weighted average concentration.

The threshold limit values are classified into the following three items by the ACGIH.

- (1) The Threshold Limit Value-Time Weighted Average (TLV-TWA) the time-weighted average concentration for a normal 8-hour workday and a 40-hour workweek, to which nearly all workers may be repeatedly exposed, day after day, without adverse effect.
- (2) Threshold Limit Value-Short Term Exposure Limit (TLV-STEL) the concentration to which workers can be exposed continuously for a short period of time without suffering from irritation, chronic or irreversible tissue damage, or narcosis of sufficient degree to increase the likelihood of accidental injury, impair self-rescue or materially reduce work efficiency, and provided that the daily TLV-TWA is not exceeded. It is not a separate independent exposure limit, rather it supplements the time-weighted average (TWA) limit where there are recognized acute effects from a substance whose toxic effects are primarily of a chronic nature. STELs are recommended only where toxic effects have been reported from high short-term exposures in either humans or animals.

A STEL is defined as a 15-minute time-weighted average exposure which should not be exceeded at any time during a work day even if the eight-hour time-weighted average is within the TLV. Exposures at the STEL should not be longer than 15 minutes and should not be repeated more than four times per day. There should be at least 60 minutes between successive exposures at the STEL. An averaging period other than 15 minutes may be recommended when this is warranted by observed biological effects.

(3) Threshold Limit Value-Ceiling (TLV-C) — the concentration that should not be exceeded even instantaneously. It is necessary to investigate the synergistic ill effect rather than individual ill effects of two or more coexisting injurious materials. Unless any information concerning the synergistic effect is available, the weight of the individual injurious effect of each component may be considered as the total injurious effect for convenience as follows: if the sum total of Formula (1) exceeds 1, the overall concentration of injurious materials is considered to exceed the threshold limit value in the environment.

$$C_1/T_1 + C_2/T_2 + C_3/T_3 \cdots + C_n/T_n$$

where C₁, C₂, C₃... and C_n = concentration of each component and T₁, T₂, T₃... and T_n = threshold limit value of each component. When only the concentration of one injurious material can be determined among various injurious ones escaping from the process, the harmfulness in the whole environment may be conservatively estimated from the threshold limit value of that material determined.

4.1.6 DANGEROUSNESS

The material causing a fire or explosion is called dangerous material including combustible gas and liquid. If combustible liquid flows out, the danger range widely expands because of liquified properties. Besides, combustible vapour is always volatilised from the surface of the combustible liquid with high vapour pressure, so an ignition source causes ignition or explosion. Combustible liquid is particularly dangerous compared with other dangerous materials due to two characteristics mentioned above.

There is a range of mixture of certain characteristics of combustible gas or vapour and air in which an ignition source causes explosion. The range of the concentrations is called range of explosion. Combustion bolstering gas such as oxygen, chlorine and nitrogen dioxide causes explosion together with combustible gas. The higher the concentration of combustion bolstering gas is, the larger the range of explosion is. The range of explosion varies according to temperature, pressure and coexisting gas, but it is generally expressed by a triangular diagram based on the experimental values.

FIG. 4.1 illustrates the triangular diagram of three component system of combustible gas (benzene), bolstering gas (oxygen) and inert gas (nitrogen). The area enclosed with solid lines is the range of explosion and the intercept of the broken line between the points of intersection

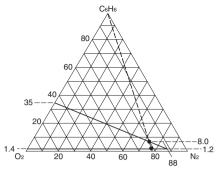


FIG. 4.1 EXPLOSION RANGE OF MIXED GASES (BENZENE, O2,N2)

shows the range of explosion of the mixture of combustible gas and air. The range of explosion of benzene in the air ranges 1.2% to 8.0% and that in oxygen ranges 1.4% to 35%. Existence of nitrogen over 88% does not cause explosion. Since explosion is generally caused by an ignition source, the concentration measuring instrument must be explosion-proof. The measurement by a gas detector tube is based on chemical reaction, so it is used even in an explosive place.

4.2 SAMPLE COLLECTION

Careful sampling is especially important obtaining correct results even in highly accurate gasanalysis. The optimum sampling method varies depending upon sample gases and analytical methods. There are three sampling methods, liquid, solid and direct sampling methods. Sampling is eliminated in the gas detector tube method in which measurement is made at the same time as sampling. Errors caused by sampling and changing of the sample with time are, therefore, available. Since the direct sampling method accompanied by the gas detector tube method is widely applied, the sampling method and the precautions are stated hereinafter.

4.2.1 SAMPLING METHOD

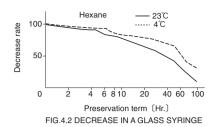
The following attention must be paid to the sampling by a bag or a glass syringe:

- (1) Since the concentration of sample decreases with time depending upon the material and size of bag and the substance and concentration of sample, the sample shall be fed to the gas detector tube confirming the stabilization of the concentration based on the decreasing rate of it (see FIG. 4.2 and 4.3).
- (2) The sample is sometimes adsorbed to the diaphragm of the diaphragm pump for sampling into a bag. Instead of the diaphragm pump, therefore, the bag shall be placed in a vessel which will be evacuated as shown in FIG. 4.4.

4.2.2 SAMPLING SITE

Gas and vapour accumulate in different places depending upon the difference in specific gravity from that of air. Lighter gas such as hydrogen, methane and ammonia accumulates in the vicinity of ceiling, while heavier gas such as toluene, gasoline and hydrogen sulfide accumulates on the floor, especially in depressions such as ditch and pit. The specific gravities of typical gases are listed in Table 4.4.

- (1) Sampling shall be carried out at the possible gas leak position such as flange and slide section and the possible gas accumulating area to detect the gas leak.
- (2) Sampling of injurious gas shall be carried out in a close range of workers' breathing and of the source for the environmental management of working place. A rubber extension hose shall be used for sampling in a place where acute poisoning is possible to confirm whether or not the area is safe for entry.



Teflon (tetrafluoroethylene fluorocation resins)

Mylar (polyester film)

Tedlar (polyvinylfluoride film)

FIG.4.3 DECREASE IN A SAMPLING BAG

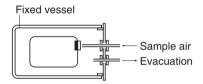


FIG 4.4 COLLECTION BY BAG

TABLE 4.4 SPECIFIC GRAVITY

SUBSTANCES	SPECIFIC GRAVITY (AIR=1)
Hydrogen	0.07
Methane	0.55
Ammonia	0.59
Carbon monoxide	0.97
Hydrogen sulfide	1.19
Propane	1.52
Butane	2.0
Chlorine	2.45

4.3 DETERMINATION RESULTS AND MEASUREMENTS

Measurements shall be made to determine contaminant leads in the environment. The measurements to prevent gas poisoning and explosion will be mentioned hereinafter.

4.3.1 PREVENTIVE MEASURES FOR POISONING

People shall immediately leave the area where lack of oxygen and acute poisoning are imminent redistribution. A respirator or an air mask shall be put on to enter the dangerous area. Safety shall be confirmed by repeated measurements before entering the area even after ventilation. The general procedure for examining the acute poisoning sources and for reviewing the environmental management is described as follows:

- (1) Exposure derived from changes in operation and the secondary pollution shall be prevented.
- (2) Redistribution of pollutants by local ventilation shall be prevented.
- (3) Redistribution of pollutants by push-pull ventilation shall be prevented.
- (4) Dilution of exhaust gas by total ventilation shall be promoted.
- (5) Facilities handling injurious materials shall be enclosed and automated.
- (6) Hazardous production plants shall be isolated and remote control shall be applied.
- (7) Process and operation shall be improved.
- (8) Production or use of hazardous chemical materials shall be discontinued or converted to less hazardous materials.

4.3.2 PREVENTIVE MEASURES FOR EXPLOSION

When combustible gas is detected, measurements shall be taken to determine the concentration. People shall immediately leave the area where the combustible gas concentration is within or close to the range of explosion. Since explosion is possible even if the concentration is low, the leak point shall be immediately examined with measuring equipment or gas detector tubes to connect it.

4.4 APPLICATIONS OF GAS DETECTOR TUBE METHOD

4.4.1 INSTANTANEOUS VALUE, HIGH TEMPERATURE GAS OR HIGH CONCENTRATION GAS MEASUREMENT

(1) Instantaneous value or high temperature gas: Gas sampling shall be made with a glass syringe the inner wall of which will have been cleaned and dried shown in FIG. 4.5. Then the syringe shall be connected with a gas detector tube and an aspirating pump as shown in FIG.4.6 to determine the gas concentration. If the gas condenses on the inner surface of the syringe, care shall be taken that the piston of syringe will not stick to the cylinder.

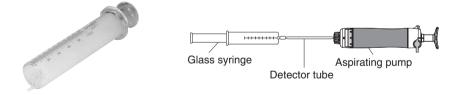


FIG.4.5 GLASS SYRINGE

FIG.4.6 MEASUREMENT

- (2) Gas of high concentration exceeding the measuring range of gas detector tube: After sampling a constant volume of gas into a glass syringe, it shall be arbitrarily diluted with fresh air. Measurement shall be made in such a manner as described in FIG. 4.6. The value obtained shall be multiplied by the magnification of dilution to determine the real concentration.
- (3) Gas with a concentration as high as 5 % or more: Since the amount of gas absorbed in a gas detector tube is not negligible, the sample pump shall be purged with fresh air. The inside of the glass syringe shall not be greased, because analytical error will be caused by

dissolving gas into the grease or the gas adsorption onto the grease.

4.4.2 REMOTE MEASUREMENT

A rubber extension hose shown in FIG. 4.7 shall be used for detection of potentially harmful gases prior to entering a confined space such as a manhole or a tank. A gas detector tube is connected with the rubber extension hose and the aspirating pump to determine the gas concentration. The rubber extension hose is available in lengths of 5 or 10 metres.



FIG. 4.7 RUBBER EXTENSION HOSE

4.4.3 FLUE GAS

The flue gas sampling instrument shown in FIG.4.8 and 4.9 is used in conjunction with the AP-1S pump and specific detector tubes for analyzing flue gas. The equipment consists of the stainless steel sampling probe equipped with a thermometer, heated by a ribbon heater for preventing condensation.



FIG. 4.8 FLUE GAS SAMPLING INSTRUMENT

4.4.4 VAPOUR OF A MIXED SOLVENT

(1) Mixed solvent

Mixed solvent thinners, are used for dissolving paint and/or cleaning painting equipment.

Thinners are prepared by mixing organic solvents to meet the requirements of the paint. Thinner for paint, mainly comprised of petroleum hydrocarbons such as mineral turpentine and solvent naphtha is a typical one used for all the oil based paints and oil modified resin coatings. (Mineral spirit is a synonym of mineral turpentine.) Lacquer thinner comprised of aromatic hydrocarbons such as toluene and xylene, ester, ketones and alcohol is used for some synthetic resin paints as well as lacquer.

Table 4.5 lists the paints and main solvents to assist in choosing gas detector tubes for measurement.

- (2) Examples of analysis for mixed solvents by gas chromatography
 - (a) Paint thinner

Paint thinner is mainly composed of petroleum hydrocarbons (C₂ to C₁₁) and may be

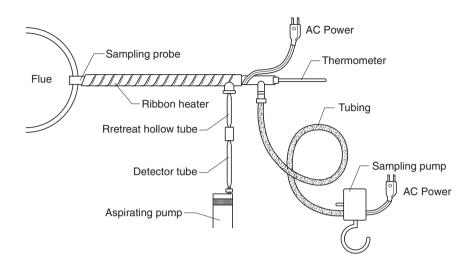


FIG.4.9 FLUE GAS MONITOR

TABLE 4.5 PAINTS AND MAIN SOLVENTS

PAINTS AND MATERIALS	MAIN SOLVENTS
①OIL PAINTS	
Compounding paint	Mineral turpentine
Corrosion-resistant paint	Xylene, Mineral Turpentine
Ship-hull paint	Mineral turpentine, Solvent naphtha
Others	Xylene, Mineral turpentine
② LACQUER	Toluene, Xylene, Methanol, Isopropanol, Butanol, Ethyl acetate, Butyl acetate, Methyl isobutyl ketone, Cellosolve
3 SYNTHETIC RESIN PAINTS	
Phenolic resins	Xylene, Mineral turpentine
Alkyd resins	Xylene, Mineral turpentine
Aminoalkyd resins	Toluene, Xylene, Methanol, Butanol, Butylacetate
Vinyl resins	Toluene, Xylene, Ethanol, Isopropanol, Butanol, Ethyl acetate, Butyl acetate, Methyl isobutyl ketone, Cellosolve
Epoxy resins	Toluene, Xylene, Isopropanol, Methyl isobutyl ketone, Cellosolve
Urethane resins	Toluene, Ethyl acetate, Butyl acetate, Metyl ethyl ketone, Cellosolve
 Corrosion-Resist paint 	Toluene, Xylene, Mineral turpentine
 Soluble resins 	Isopropanol, Butanol, Cellosolve

called mineral tupentine or mineral spirit. See FIG.4.10 for the example of analysis.

(b) Lacquer thinner

The FIG.4.11 is for a lacquer thinner which contains ethyl acetate and MIBK as well as the main component of toluene (for industrial use)

The FIG.4.12 is for a lacquer thinner which contains MIBK as well as the main component of xylene (for house use)

(3) Analysis of mixed solvent

The composition of solvent vapour (see FIG.4.14) is not consistent with that of liquified solvent (see FIG.4.13), because low boiling components are volatilised first and then higher

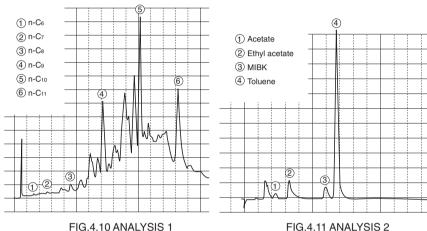


FIG.4.11 ANALYSIS 2



² p-Xylene

³ m-Xylene



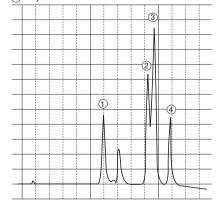


FIG.4.12 ANALYSIS 3

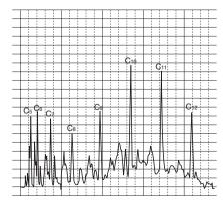


FIG.4.13 COMPOSITION OF LIQUID SOLVENT

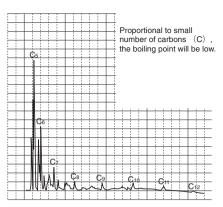


FIG.4.14 COMPOSITION OF SOLVENT VAPOUR

boiling ones are volatilized later with the passing of time.

- (a) Gas detector tube method
 - The procedure for analyzing mixed solvent vapour like thinner in a work place using gas detector tube is as follows:
- ① Estimate the components of thinner from the type of paint (see TABLE 4.5). Consult the material safety data sheet, ask the manufacturer about it or analyse it by gas chromatography unless the composition is known.
- ② Select the components of solvent, of which boiling point is low and/or TLV is low.
- ③ Investigate the effect of interfering gas on the gas detector to be used. Choose detector tubes according to the instruction where there is no interfering gas or the effect of interfering gas is very low.
- Make detector tube measurements of each component noting any that list negative
 interferences. Record the values. Record "zero" responses as being less than the listed detecting
 limit
- (5) When there are positive interfering gases overstating the reading and a certain value is indicated, convert or calibrate the measured value according to the analytical value by gas chromatography for future use.
- (b) Solid sampling

A charcoal tube or a silica gel tube is used for solid media sampling which is an effective means for determining the components of mixed solvent (refer to 4.5.1 and 4.5.2).

4.4.5 RESIDUAL SALINITY

Steel structures in coastal areas are damaged by defects such as swelling, peeling and rusting caused by salt deposits. The chloride ion detector tube is used for determining the salinity of the water washing the surface coating.

- (1) Measuring apparatus
- Chloride ion detector tube (No. 201SA)
- Deionized water (or distilled water) 150 mℓ
- Gauze 30 × 30 cm
- · Rubber glove
- Masking tape (approx. 20 mm width)
- · Measuring scale
- Polyethylene beaker 250 mℓ
 - (2) Sampling procedure

- ① Mark off a division of 0.25m2 (e.g. 50×50 cm) with masking tape after correctly measuring with a measuring scale.
- (2) Put on a rubber glove on a hand.
- (3) Put 150 mℓ of deionized water into a polyethylene beaker.
- (4) Wet gauze folded in a proper size with water in the beaker.
- ⑤ Wipe the measuring surface in a parallel direction with wet gauze (see FIG. 4.15). Be sure not to drop water while wiping.
- (6) Rinse the gauze in the water in the beaker. Repeat 5 and 6 three times.
- (3) Measuring procedure
- ① Cut off both ends of a chloride ion detector tube with a file. Put the detector tube into the sample water turning the arrow upward (see FIG. 4.16).
- ② Take the detector tube out from the beaker when the water permeates up to the upper end of the reagent. Read the concentration at the end of discolored layer which has been developed by the salinity contained in the sample water.

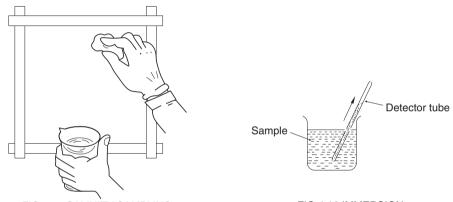


FIG.4.15 SALINITY SAMPLING

FIG.4.16 IMMERSION

(4) Relationship between concentration of chloride ion and amount of salinity The relationship between concentration [ppm] of chloride ion and amount of salinity [mg/m³] is expressed by the following formula:

$$N = LC \times \frac{NaCl}{Cl} \times \frac{1}{M}$$

where N = amount of salinity mg/m²; L = amount of deionized water [ℓ]; C = concentration of chloride ion [ppm] [NaCl] = molecular weight of sodium chloride (58.5); Cl = molecular weight of chlorine (35.5) and M = area of salinity sampling [m²]. Using 0.15 for L, 0.25 for M and 58.5 and 35.5 for NaCl and Cl, respectively gives

$$N = 0.15 \text{ C} \times \frac{58.5}{35.5} \times \frac{1}{0.25} = \text{ C}$$

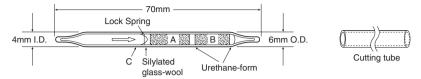
The concentration of chloride ion [ppm] obtained by the detector tube is equal to the amount of salinity $[mg/m^2]$.

4.5 OTHER WORKING ENVIRONMENT MEASURING DEVICES 4.5.1 No. 800B CHARCOAL TUBE (CONFORMED TO NIOSH REQUIREMENTS)

A charcoal tube is used for sampling and measurement of organic solvents. Sample air containing vapour of organic solvents is passed through a glass tube packed with charcoal which adsorbs the solvent. Adsorbed solvent is recovered with desorption solvent leeching followed by gaschromatographic analysis.

The charcoal tube consists of 10 to 30-mesh charcoal packed into a glass tube about 4 mm I.D., separated into front and back up sections of 100mg and 50mg, respectively. The ends of the tube are then fore-sealed by fusion as shown in FIG.4 17.

Every organic solvent has its own desorption efficiency as shown in Table 4.6.



A: 100mg 20/40 mesh coconut shell charcoal (first section)
B: 50mg 20/40 mesh coconut shell charcoal (back up section)

Fig.4.17 No.800B Charcoal Tube

TABLE 4.6 DESORPTION EFFICIENCY

Substance	Desorption Efficiency(%)	Desorption Solvent
Toluene	100	CS ₂
Trichloroethylene	100	"
Acetone	69	"
Iso-propanol	77	1% of 2-Butanol in CS2
Butyl cellosolve	98	5% of Methanol in Dichloromethane

4.5.2 No. 801 SILICA GEL TUBE

A silica gel tube is used for sampling organic solvents in environmental air, particularly for polar solvents such as methanol and acetone which cannot be sufficiently adsorbed onto charcoal. Organic solvent vapor is adsorbed onto the silica gel while passing through the tube. Adsorbed solvent is recovered with a desorption solvent followed by gas-chromatographic analysis.

The silica gel tube contains 20 to 40-mesh silica gel of 300 milligrams and a breakthrough indicator packed separately into the glass tube. The ends of the tube are fire sealed as shown in FIG. 4.18.

The breakthrough indicator was first introduced by Komyo. When sample air corresponding to 60 to 80 % of the amount causing breakthrough is passed, the colour begins to change from blue to pink by the moisture from the silica gel side predicting breakthrough.

Every organic solvent and every desorption solvent have their own recovery rates as shown in Table 4.7.

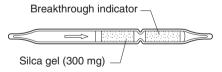


FIG.4.18 SILICA GEL TUBE

TABLE 4.7 RECOVERY RATE OF SILICA GEL

SUBSTANCE	DESORPTION SOLVENT		RECOVERY RATE(%)
Methanol	Distilled water	3 mℓ	90
IPA	Distilled water	3 mℓ	83
	DMF	2 m ℓ	92
Acetone	Methanol	2 m ℓ	98
MEK	DMF	2 m ℓ	96
Ethyl acetate	DMF	2 m ℓ	95
Toluene	Acetone	2 m ℓ	99

4.5.3 AIR FLOW INDICATOR

There are many areas in work places where air flow must be examined for testing the performance of ventilating and exhaust systems. Although tobacco smoke and incense have been used for detecting air flow both have the problem of insufficient smoke or exclusion from use in fire-restricted areas. The air flow indicator develops smoke without combustion. The smoke is generated by the reaction of the reagent with the moisture contained in the air passing through the tube, (see FIG. 4.20).

(1) Special features

- 1) Dry system with no need of a heat source
- (2) Smoke is produced simply by squeezing the rubber bulb (Model AS-1)
- (3) Compact, lightweight and hand-held
- (4) Indicates the direction and velocity of air flow
- (5) Continuous Smoke production (Model AS-2)



FIG.4.20 MODEL AS-1 AIR FLOW INDICATOR

5.NUMERICAL INDEX WITH REFERENCES

TUBE NAME OF GAS SYNONYM CHEMICAL CROPP C								
MCETYLENE			SYNONYM					PAGE
MCETYLENE	100	CARBON MONOXIDE		со	25(A), 30(B), 50(J)	300(B)	12.5	83
1925AD TETRAHYDROPURAN							_	_
1928AB TETRAHYDROPIRAN	102SA	ACETONE		CH3COCH3	500(B), 200(A, J)	750(A), 1,500(B)	2.1	26
19250								_
1925					, .,			_
1935A		"						28
1938B		SULPHUR DIOXIDE		SO ₂		0.25(A), 5(B)		_
1935C	103SB	"						311
1935D		,,						_
1935E		"						_
1038G		,,						_
1038G	103SF	SULPHUR DIOXIDE IN FLUE GAS						315
1958B								
1958B	104SA	ETHYL ALCOHOL	ETHANOL	C ₂ H ₅ OH	1.000(A. B)		3.5	_
1658B						35(A, B)	_	_
105SC					2011, 2, 27		10.0	_
10SSD								_
105SDD		"						_
105SD@ Di-n-BUTYL AMINE CaHaN(CHs)2 5(J) C 5(A) 65		CYCLOHEXYL AMINE		CeH++NH2	10(A B)			
1055D© P.BUTYL AMINE CoH-N(CHs) 5(J) C 5(A) 65				0011111112	10(A, D)			_
105SDO					5(1)	C 5(A)		_
105D© n-METHYL ANILINE C.dHsNHCHs 0.5(A)				CeHeN(CHe)e				_
105D© MORPHOLINE C.H⊌NO 20(A), 10(B)						10(A)		_
105SD© PENTYL AMINE CH5 (CH2) 3 CH3NH2 280 105SD© PROPYL AMINE CH5 (CH2) 3 CH3NH2 297 105SD© DH3P PROPYL AMINE (CH3) 2 CH3 2 MINE <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>								
105D© PROPYL AMINE CH3CH3CH2NHz 5(A, B) 297 105SD© Di+so-PROPYL AMINE [(CH4)2CH]3NH 5(A, B) 298 105SD© Di+-PROPYL AMINE C3HaN 1(A), 5(J) 1.8 306 105SD© PYRIDINE C3HaN 1(A), 5(J) 1.8 306 105SD© PTOLUIDINE C3Ha (CHa) (NHz) 2(A), 1(J) 329 105SD© AMMONIA C3Ha (CHa) (NHz) 2(A) 330 105SH AMMONIA C3Ha (CHa) (NHz) 2(A) 42 105SH AMMONIA C3Ha (CHa) (NHz) 2(A) 42 105SH A C3Ha (CHa) (NHz) 2(A) 42 105SH A C3 44 44 106B CARBON MONOXIDE CO 25(A), 30(B), 50(J) 300(B) 1.2 85 106C A C3 C3 C3 66 66 66 66 66 66 66 66 66 66 66 66 66 <td></td> <td></td> <td></td> <td></td> <td>20(A), 10(B)</td> <td></td> <td></td> <td>_</td>					20(A), 10(B)			_
10SSD© Di-iso-PROPYL AMINE [(Chs)s CH]sNH 5(A, B) 298 10SSDD Di-n-PROPYL AMINE CaHaN 1(A), 5(J) 1.8 306 10SSD© PYRIDINE CaHa (CHa) (NHa) 2(A), 1(J) ————————————————————————————————————								_
105SD© Di-n-PROPYL AMINE CaHaN 1 (A), 5 (J) 1.8 299 105SD© PYRIDINE CaHa (CHa) (NHa) 2 (A), 1 (J) 1.8 308 105SD© p-TOLUIDINE CaHa (CHa) (NHa) 2 (A) 339 105SE AMMONIA CaHa (CHa) (NHa) 2 (A) 42 105SM A 43 105SM A 42 105SM A					F(A D)		_	_
105SD PYRIDINE CcHaN 1(A), 5(J) 1.8 300 105SD© ∞TOLUIDINE CcHa (CHa) (NHz) 2(A), 1(J) 2329 105SD p-TOLUIDINE CcHa (CHa) (NHz) 2(A) 333 105SE AMMONIA 242 43 105SH ∞ 43 43 105SM ∞ 44 44 106B CARBON MONOXIDE CO 25(A), 30(B), 50(J) 300(B) 12.5 84 106C ∞ 6 6 85 85 85 106S ∞ 85 86 106S ∞ 6 8 8 86 106S ∞ 88 8 106S ∞ 88 8 106SH ∞ 1 1 1 89 1 1 88 106SH ∞ 1 1 1 89 1 1 89 106SH ∞ 1 4 2 2 9 <td></td> <td></td> <td></td> <td>[(GH3/2GH]2NH</td> <td>5(A, b)</td> <td></td> <td></td> <td>_</td>				[(GH3/2GH]2NH	5(A, b)			_
105SD© o-TOLUIDINE Cell (CHa) (NHz) 2(A), 1(J) 4 329 105SD© p-TOLUIDINE Cell (CHa) (NHz) 2(A) 4 330 105SE AMMONIA 4 4 42 105SH γ 4 4 42 105SM γ 6 4 4 106B CARBON MONOXIDE CO 25(A), 30(B), 50(J) 300(B) 12.5 84 106C γ 6 6 6 6 85 106S γ 6 6 6 86 106SA γ 6 6 6 88 106SA γ 6 6 6 88 106SA γ 6 6 9 9 106SS γ 6 6 9 9 106SB γ 6 6 9 9 106SB γ 6 6 9 1 1 15<				CHN	1/4) 5/1)		1.0	
105SD© ρ-TOLUIDINE C6H4 (CH3) (NH2) 2 (A) 330 105SE AMMONIA							1.8	_
105SE AMMONIA 42 105SH φ 105SH φ 43 105SM φ 105SH 244 106B CARBON MONOXIDE CO 25(A), 30(B), 50(J) 300(B) 12.5 84 106C φ 106C 25(A), 30(B), 50(J) 300(B) 12.5 84 106S φ 106SH 107 107 107 106 106SH 107 106 107 107 107 107 106 106							_	
105SH ⋄ 105SM ⋄ 44 105SM ⋄ 44 44 106B CARBON MONOXIDE CO 25(A), 30(B), 50(J) 300(B) 12.5 84 106C ⋄ E CO 25(A), 30(B), 50(J) 300(B) 12.5 84 106C ⋄ E CO 25(A), 30(B), 50(J) 300(B) 12.5 84 106C ⋄ E C CO 25(A), 30(B), 50(J) 300(B) 12.5 84 106S ⋄ E C C 86 86 106SA ⋄ E C C 88 86 86 106SC ⋄ E C C C 89 90				G6F14 (GF13) (NF12)	2(A)		-	_
105SM γ 44 106B CARBON MONOXIDE CO 25(A), 30(B), 50(J) 300(B) 12,5 84 106C γ CO 25(A), 30(B), 50(J) 300(B) 12,5 84 106S γ CO 25(A), 30(B), 50(J) 300(B) 12,5 84 106SA γ CO 25(A), 30(B), 50(J) 300(B) 12,5 84 106SA γ CO CO 25(A), 30(B), 50(J) 300(B) 12,5 85 106SA γ CO CO CO CO CO 200(A) CO 88 106SC γ CO CO 400(A, J), 100(B) 500(A, B) 1,7 145 107U γ CO CO 400(A, J), 100(B) 500(A, B) 1,7 145 108B ETHYLERE CH₂ CH₂ 200(A) 2,7 147 108SA γ CH₂ CH₂ 200(A) 1(A, B) 94 109SA <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td>_</td>								_
106B CARBON MONOXIDE CO 25(A), 30(B), 50(J) 300(B) 12.5 84 106C γ Image: Control of the part of t		"						_
106C β 106S β 106SA β 106SA β 106SC β 106SH β 106SH β 106SH β 106SH β 106SH β 106SS β 106UH β 107SA ETHYL ETHER DIETHYL ETHER (C₂H₂)₂O 400(A, J), 100(B) 500(A, B) 17 145 107U β 108B ETHYL ETHER DIETHYL ETHER (C₂H₂)₂O 400(A, J), 100(B) 500(A, B) 1,7 145 107U β 108B ETHYL ETHER CH₂ ECH₂ 200(A) 108B ETHYL ETHER CH₂ ECH₂ 200(A) 108B ETHYL ETHER CH₂ ECH₂ 200(A) 108B ETHYL ETHER CH₂ ECH₂ 0.5(A, B, J) 11(A, B) 94		/			(1)(2)(1)	202(5)		
106S β 106SA β 106SA β 106SC β 106SH β 106SH β 106SH β 106SS β 106SS β 106SS β 106SH β 106SH β 106SH β 106WH β 107W β 107W β 108B ETHYLETHER 108B ETHYLENE 108B ETHYLENE 108B CHYLENE 108SA β 108SA β 108SA β 108SA β 109SA CHLORINE Cl2 0.5(A, B, J) 114B 1995 109U β 109W β 110S GASOLINE PETROL Collan 300(A), 100(J) 500(A) 111SA				CO	25(A), 30(B), 50(J)	300(B)	12.5	_
106SA β 106SC β 106SC β 106SH β 106SS β 106SS β 106UH β 107SA ETHYL ETHER DIETHYL ETHER (CaH ₀) ₂ O 400(A, J), 100(B) 500(A, B) 1.7 145 107U β CH2 CH2 200(A) 2.7 147 108B ETHYLENE CH2=CH2 200(A) 2.7 147 108SA β CH2 CH2 200(A) 2.7 147 108SA β CH2 CH2 200(A) 2.7 147 108SA β CH2 CH2 0.5(A, B, J) 1(A, B) 94 109SB β CH2 CH2 0.5(A, B, J) 1(A, B) 94 109SB β CH2 0.5(A, B, J) 1(A, B) 94 109SB β CH3 CH3 95 95 109SB <								
106SC β 106SH β 106SH β 106SH β 106SH β 106SH β 106SH β 106UH β 107SA ETHYLETHER 107SA ETHYLETHER 107U β 107U β 108B ETHYLENE 108B ETHYLENE 108SA β 108SA β 108SA β 109SA CHLORINE 109SB β 109SB β <							_	_
106SH β 106SS β 106SS β 106WH β 107SA ETHYLETHER 107SA ETHYLETHER 107U β 107U β 108B ETHYLETHE 108B ETHYLETHE 108B ETHYLENE 108SC β 108SC β 109SA CHLORINE 109SA CHLORINE 109U β 110S GASOLINE 110S GASOLINE 110S GASOLINE 111SA ETHYLACETATE 111SA ETHYLACETATE 111U								_
106SS								_
106UH ∅ 91 107SA ETHYL ETHER DIETHYL ETHER (C₂H₅)₂O 400(A, J), 100(B) 500(A, B) 1.7 145 107U ⋄ CH₂ = CH₂ 200(A) 2.7 147 108B ETHYLENE CH₂ = CH₂ 200(A) 2.7 147 108SA ⋄ 148 168 148 168 168 168 168 168 168 168 168 168 168 168 168 168 178 94 169 94 169 95 95 96 169 96 169 96 169 96 169 169 169 169 169 169 169 169 169 169 169 169 169 169 169							_	_
107SA ETHYL ETHER DIETHYL ETHER (C₂H₅)₂O 400(A, J), 100(B) 500(A, B) 1.7 14s 107U γ CH₂=CH₂ 200(A) 2.7 14f 108B ETHYLENE CH₂=CH₂ 200(A) 2.7 147 108SA γ CH₂ 0.5(A, B, J) 1(A, B) 148 109SA CHLORINE Cl₂ 0.5(A, B, J) 1(A, B) 94 109SB γ CH₂ 0.5(A, B, J) 1(A, B) 94 109U γ CH₃ CH₃ 0.5(A, B, J) 1(A, B) 96 110S GASOLINE PETROL CnHm 300(A), 100(J) 500(A) 1.0 168 111SA ETHYL ACETATE CH₃CO₂CH₃ 400(A), 200(B, J) 250(A) 2.1 136 111U ETHYL ACETATE CH₃CO₂CCH 200(A, B, J) 250(A) 224 111U Isopropyl ACETATE CH₃CO₂CH(CH₃)₂ 100(A, J) 200(A, B) 1.8 211 111U® Iter-BUTANO								_
107U ∅ 146 108B ETHYLENE CH₂=CH₂ 200(A) 2.7 147 108SA ∅ 148 148 109SA CHLORINE Cl₂ 0.5(A, B, J) 1(A, B) 94 109SB ∅ Cl₂ 0.5(A, B, J) 1(A, B) 94 109SB ∅ SOLINE 96 96 110S GASOLINE PETROL CnHm 300(A), 100(J) 500(A) 1.0 168 111SA ETHYL ACETATE CH₃CO₂C₃H₃ 400(A), 200(B, J) 2.1 136 111SW METHYL ACETATE CH₃CO₂CH₃ 200(A, B, J) 250(A) 224 111U ISOPROPYL ACETATE CH₃CO₂CH(CH₃)₂ 100(A, J) 200(A, B) 1.8 211 111U® Iter-BUTANOL (CH₃)GOCH 100(A, B), 50(J) 2.4 60 111U® BUTYL ETHER (CH₃CH₂CH₂CH₂CH₂CO - - 67 111U® BUTYL ETHER CH₂CCCH⟨CH₂CCH₂COCCCH₀ - -								_
108B ETHYLENE CH₂=CH₂ 200(A) 2.7 147 108SA γ CH2=CH₂ 200(A) 2.7 148 108SC γ CHORINE 149 149 109SA CHLORINE Cl₂ 0.5(A, B, J) 1(A, B) 94 109SB γ Cl₂ 0.5(A, B, J) 1(A, B) 95 109U γ Cl₂ 0.5(A, B, J) 500(A) 1.0 168 110S GASOLINE PETROL Chłm 300(A), 100(J) 500(A) 1.0 168 111SA ETHYL ACETATE CH3CO2C2łs 400(A), 200(B, J) 2.1 136 111SA METHYL ACETATE CH3CO2CH3 200(A, B, J) 250(A) 224 111U ETHYL ACETATE CH3CO2CH(CH3)2 100(A, J) 200(A, B) 1.8 211 111U ISOPROPYL ACETATE CH3CO2CH(CH3)2 100(A, J) 200(A, B) 1.8 211 111U® Istributed CH3CO3CH4(CH3)2 100(A, B), 50(J) <t< td=""><td></td><td></td><td>DIETHYL ETHER</td><td>(C₂H₅)₂O</td><td>400(A, J), 100(B)</td><td>500(A, B)</td><td>1.7</td><td>_</td></t<>			DIETHYL ETHER	(C ₂ H ₅) ₂ O	400(A, J), 100(B)	500(A, B)	1.7	_
108SA β 148 108SC β 149 109SA CHLORINE Cl₂ 0.5(A, B, J) 1(A, B) 94 109SB β 95 95 95 110S GASOLINE PETROL CnHm 300(A), 100(J) 500(A) 1.0 168 111SA ETHYL ACETATE CH₃CO₂C2H₃ 400(A), 200(B, J) 2.1 136 111SA METHYL ACETATE CH₃CO₂CH₃ 200(A, B, J) 250(A) 224 111U ETHYL ACETATE CH₃CO₂CH(CH₃)₂ 100(A, J) 200(A, B) 1.8 211 111U ISOPROPYL ACETATE CH₃CO₂CH(CH₃)₂ 100(A, J) 200(A, B) 1.8 21 111U® Isorroward (CH₃)₃COH 100(A, B), 50(J) 2.4 60 111U® BUTYL ETHER (CH₃CH₃CH₃CH₂)₂O								_
108SC ° 149 109SA CHLORINE Cl₂ 0.5(A, B, J) 1(A, B) 94 109SB ° 5 95 95 96 96 109U ° 96				CH ₂ = CH ₂	200(A)		2.7	_
109SA CHLORINE Cl₂ 0.5(A, B, J) 1 (A, B) 94 109SB γ C 0.5(A, B, J) 1 (A, B) 95 109U γ 96 96 96 110S GASOLINE PETROL CnHm 300(A), 100(J) 500(A) 1.0 168 111SA ETHYL ACETATE CH₃CO₂CH₂ 400(A), 200(B, J) 250(A) 2.1 136 111SA METHYL ACETATE CH₃CO₂CH₃ 200(A, B, J) 250(A) 224 111U ISOPROPYL ACETATE CH₃CO₂CH(CH₃)₂ 100(A, J) 200(A, B) 1.8 211 111U® tert-BUTANOL (CH₃)₃COH 100(A, B), 50(J) 2.4 60 111U® BUTYL ETHER (CH₃CH₂CH₂CH₂CO 67 67 111U® BUTYL METHACRYLATE CH₂C (CH₃) CO₂CdH₃ 2 70							_	_
109SB γ 95 109U γ 96 110S GASOLINE PETROL CnHm 300(A), 100(J) 500(A) 1.0 168 111SA ETHYL ACETATE CHsCOsCeHs 400(A), 200(B, J) 2.1 136 111SA® METHYL ACETATE CHsCOsCHs 200(A, B, J) 250(A) 224 111U ETHYL ACETATE T111U ISOPROPYL ACETATE 137 137 111U® IsoPROPYL ACETATE CHsCOsCH(CHs)2 100(A, J) 200(A, B) 1.8 211 111U® Iter-BUTANOL (CHsCHsCHsCHs)CO 100(A, B), 50(J) 2.4 60 111U® BUTYL ETHER (CHsCHsCHsCHs)CO 67 67 111U® BUTYL METHACRYLATE CHsCCCCHsCHsCHs 2 70								_
109U ø 96 110S GASOLINE PETROL CnHm 300(A), 100(J) 500(A) 1.0 168 111SA ETHYL ACETATE CH₃CO₂C₂H₅ 400(A), 200(B, J) 2.1 136 111SA® METHYL ACETATE CH₃CO₂CH₅ 200(A, B, J) 250(A) 224 111U ETHYL ACETATE 137 131 137 111U ISOPROPYL ACETATE CH₃CO₂CH(CH₃)₂ 100(A, J) 200(A, B) 1.8 211 111U® tert-BUTANOL (CH₃)GCOH 100(A, B), 50(J) 2.4 60 111U® BUTYL ETHER (CH₃CH₂CH₂CH₂)O 67 67 111U® BUTYL METHACRYLATE CH₂C(CH₃)CO₂CdH₃ 2 70				Cl2	0.5(A, B, J)	1(A, B)		_
110S GASOLINE PETROL CoHm 300(A), 100(J) 500(A) 1.0 168 111SA ETHYL ACETATE CHsCOsCsHs 400(A), 200(B, J) 2.1 136 111SA® METHYL ACETATE CHsCOsCHs 200(A, B, J) 250(A) 224 111U ETHYL ACETATE 137 111U ISOPROPYL ACETATE CHsCOsCH(CHs)2 100(A, J) 200(A, B) 1.8 211 111U® tert-BUTANOL (CHs)SCOH 100(A, B), 50(J) 2.4 60 111U® BUTYL ETHER (CHsCHsCHsCHs)2O 67 67 111U® BUTYL METHACRYLATE CHsCCCCHsCHs)COsCdHs 2 70								_
111SA ETHYL ACETATE CH3CO2CCH5 400(A), 200(B, J) 2.1 136 111SA© METHYL ACETATE CH3CO2CH5 200(A, B, J) 250(A) 224 111U ETHYL ACETATE 137 111U ISOPROPYL ACETATE 100(A, J) 200(A, B) 1.8 211 111U© tert-BUTANOL (CH3)2COH 100(A, B), 50(J) 2.4 60 111U© BUTYL ETHER (CH3CH2CH3CH2)2O 67 111U© BUTYL METHACRYLATE CH2 C(CH3)CO2C4H5 2 70 70 70 70 70 70 70		"						_
111SA® METHYL ACETATE CH3CO2CH3 200(A, B, J) 250(A) 224 111U ETHYL ACETATE 137 111U ISOPROPYL ACETATE CH3CO2CH(CH3)2 100(A, J) 200(A, B) 1.8 211 111U® tert-BUTANOL (CH3)3COH 100(A, B), 50(J) 2.4 60 111U® BUTYL ETHER (CH3CH2CH3CH2)2O 67 111U® BUTYL METHACRYLATE CH2=C(CH3)CO2C4H9 2 70	110S	GASOLINE	PETROL	C _n H _m	300(A), 100(J)	500(A)	1.0	168
111U ETHYL ACETATE 137 111U ISOPROPYL ACETATE CH3CO2CH(CH3)2 100(A, J) 200(A, B) 1.8 211 111U© tert-BUTANOL (CH3)3COH 100(A, B), 50(J) 2.4 60 111U© BUTYL ETHER (CH3CH2CH2CH2)2O 67 111U© BUTYL METHACRYLATE CH2=C(CH3)CO2C4H9 2 70		ETHYL ACETATE					2.1	136
111U ISOPROPYL ACETATE CH3CO2CH(CH3)2 100(A, J) 200(A, B) 1.8 211 111U© tert-BUTANOL (CH3)3COH 100(A, B), 50(J) 2.4 60 111U© BUTYL ETHER (CH3CH2CH2CH2)2O 67 111U© BUTYL METHACRYLATE CH2=C(CH3)CO2C4H3 2 70	111SA©	METHYL ACETATE		CH3CO2CH3	200(A, B, J)	250(A)		224
111U© tert-BUTANOL (CHs)sCOH 100(A, B), 50(J) 2.4 60 111U© BUTYL ETHER (CHsCHsCHsCHs)cO 67 111U© BUTYL METHACRYLATE CHs=C(CHs)COsCdHs 2 70	111U	ETHYL ACETATE						137
111U© BUTYL ETHER (CH3CH2CH2CH2)2O 67 111U© BUTYL METHACRYLATE CH2 = C(CH3)CO2CdH3 2 70	111U	ISOPROPYL ACETATE		CH ₃ CO ₂ CH(CH ₃) ₂	100(A, J)	200(A, B)	1.8	211
111U© BUTYL METHACRYLATE CH2=C(CH3)CO2C4H9 2 70	111U©	tert-BUTANOL		(CH ₃) ₃ COH	100(A, B), 50(J)		2.4	60
		BUTYL ETHER		(CH ₃ CH ₂ CH ₂ CH ₂) ₂ O				67
111U© tert-BUTYL METHYL ETHER(MTBE) CH ₃ OC (CH ₃) ₃ 50(A), 25(B) 16.1 71	111U©	BUTYL METHACRYLATE		CH2 = C (CH3) CO2C4H9			2	70
	111U©	tert-BUTYL METHYL ETHER (MTBE)		CH ₃ OC (CH ₃) ₃	50(A), 25(B)		16.1	71

TUBE NO.	NAME OF GAS TO BE DETECTED	SYNONYM	CHEMICAL FORMULA	T.L.VTWA (ppm)	T.L.VSTEL (ppm)	LEL (%)	PAGE
111U©	CUMENE		C6H5CH(CH3)2	50(A), 25(B)		0.8	104
111U©	CYCLOHEXENE		C6H10	300(A)		1.2	108
111U©	DECAHYDRONAPHTHALENE		C10H18				110
111U©	n-DECANE		CH ₃ (CH ₂) ₈ CH ₃				111
111U©	DIETHYL BENZENE		C ₆ H ₄ (C ₂ H ₅) ₂			0.8	125
111U©	ETHYL METHACRYLATE		CH ₂ =C(CH ₃)CO ₂ C ₂ H ₅			1.8	160
111U©	ISOPROPYL ETHER		[(CH ₃) ₂ CH] ₂ O	250(A, B)	310(A)	1.4	215
111U©	n-NONANE		CH ₃ (CH ₂) ₇ CH ₃	200(A, J)	010(A)	1.4	270
111U©	1,2,4-TRIMETHYL BENZENE		OF IS (OF IZ) / OF IS	25(A, B, J)			337
111U©	n-UNDECANE		CH ₃ (CH ₂) ₉ CH ₃	25(A, B, J)		0.7	
112SA	HYDROGEN CYANIDE		HCN	5(1) 40(D)	O 4 7(A) 40(B)	5.6	339 179
	HYDROGEN CYANIDE		HCN	5(J), 10(B)	C 4.7(A), 10(B)	0.0	
112SB	,						180
112SC			/>	(-) (-)			181
113SA	n-HEXANE		CH ₃ (CH ₂) ₄ CH ₃	50(A), 20(B), 40(J)		1.1	172
113SB	"						173
113SB©	HEPTANE		CH ₃ (CH ₂) ₅ CH ₃	400(A), 500(B),200(J)	500(A)		171
113SB©	ISOBUTANE		(CH ₃) ₃ CH				199
113SB©	ISOBUTYLENE						204
113SB©	METHYL CYCLOHEXANE		C ₆ H ₁₁ (CH ₃)	400(A, J)			239
113SB©	2,2,4-TRIMETHYL PENTANE						338
113SB©	PENTANE		CH3 (CH2)3 CH3	600(A, B), 300(J)			278
113SC	n-HEXANE						174
114	BROMINE		Br ₂	0.1 (A, B, J)	0.2(A), 0.3(B)		52
115S	CYCLOHEXANE		C6H12	100(A, B), 150(J)	300(B)	1.2	105
116	CHLORINE DIOXIDE		CIO ₂	0.1(A, B)	0.3(A, B)		97
117SA	NITROGEN DIOXIDE		NO ₂	0.2(A)	5(A, B)		263
117SB	//		1402	U.E(A)	3 (A, B)		264
117SD	,						_
			0.11	0.5(4), 4(D), 40(1)	0.5(4)	- 10	265
118SB	BENZENE		C ₆ H ₆	0.5(A), 1(B), 10(J)	2.5(A)	1.2	48
118SC	"						49
118SD	"						50
118SE	"						51
119LPG	METHYL ALCOHOL	METHANOL	CH₃OH	200(A, B, J)	250(A, B)	5.5	226
119SA	"						227
119U	"						228
119U©	1,4-DIOXANE			20(A),25(B),10(J)			133
120GR	HYDROGEN SULPHIDE		H ₂ S	1(A), 5(B, J)	5(A), 15(B)	4.0	186
120GT	"						187
120SB	*						188
120SC	"						189
120SD	"						190
120SE	"						191
120SF	,,						192
120SH	,,						193
120SM	"						194
1200W	,						195
120UH	"						_
							196
120UT	PURCOPURING IN A DETY OF THE		P::	0.0(4.1) 0.1(5)	1(4) 22(2)		197
121SA	PHOSPHINE IN ACETYLENE		PH ₃	0.3(A, J), 0.1(B)	1(A), 0.3(B)		283
121SB	"						284
121SC	PHOSPHINE						285
121SD	"						286
121SG	"						287
121SH	PHOSPHINE-high range						288
121SS	PHOSPHINE						289
121U	"						290
121U	ARSINE		AsH ₃	0.005(A), 0.05(B), 0.1(J)			47
122SA	ETHYLENE OXIDE		CH ₂ CH ₂ O	1(A, J),5(B)		3.0	153
122SA©	FURAN	FURFURAN					165
122SA©	ISOPROPYL ALCOHOL	2-PROPANOL	CH ₃ CH(OH)CH ₃	200(A),400(J)	400(A)		212
.2200	LIGOT TE ALCOHIOL	L HOI ANOL	O1 13 O1 1 (O1 1 / O1 13	200 (A),400 (d)	100(A)		1 414

TUBE NO.	NAME OF GAS TO BE DETECTED	SYNONYM	CHEMICAL FORMULA	T.L.VTWA (ppm)	T.L.VSTEL (ppm)	LEL (%)	PAGE
122SA©	METHYL ETHYL KETONE	2-BUTANONE	CH3COC2H5	200(A, B, J)	300(A)		242
122SA©	METHYL ISOBUTYL KETONE	ISOPROPYL ACETONE	CH ₃ COH ₂ CH(CH ₃) ₂	30(A),50(B, J)	75(A)		247
122SC	ETHYLENE OXIDE	IOOT HOT TE MOETONE	GT ISO GT IE GTT (GT IS) E	00(//),00(2),0/	7000		154
122SC©	PROPYLENE GLYCOL		CH ₃ CH(OH)CH ₂ OH	150(B)		2.6	301
122SC©	PROPYLENE OXIDE			2(A), 5(B)		1.9	302
	PROFILENE OXIDE		CH ₃ CHCH ₂ O	2(A), 5(B)		1.9	
122SL	,						155
122SM		METING ETHER	011 0011	400(B)		0.4	156
123S	DIMETHYL ETHER	METHYL ETHER	CH ₃ OCH ₃	400(B)	1=2(5)	3.4	130
124SA	TOLUENE	METHYL BENZENE	C6H5CH3	20(A, J),50(B)	150(B)	1.2	326
124SB	"						327
124SH	"						328
125SA	PROPANE		C ₃ H ₈	1,000(A)		2.1	292
126B	CARBON DIOXIDE		CO ₂	5,000 (A, B, J)	30,000(A), 15,000(B)		73
126SA	"						74
126SB	"						75
126SF	"						76
126SG	"						77
126SH	"						78
126UH	"						79
128SA	ACRYLONITRILE	VINYL CYANIDE	CH2=CHCN	2(A, B, J)		2.8	33
128SB	"						34
128SC	"						35
128SD	,						36
129	NICKEL CARBONYL	NICKEL TETRACARBONYL	Ni(CO) ₄	0.05(A), 0.001(J)		2.0	261
130U	tert-BUTYL MERCAPTAN	NICKEL TETTACATIBOTITE	(CH ₃) ₃ CSH	0.5(A)		2.0	68
130U	ETHYL MERCAPTAN	ETHANETHIOL	C2H5SH		2(B)	2.0	159
		ETHANETHIOL		0.5(A, B)	Z(B)	2.8	
130U	ISOPROPYL MERCAPTAN		(CH ₃) ₂ CHSH	. = (1 =)			216
130U	METHYL MERCAPTAN	METHANETHIOL	CH₃SH	0.5(A, B)		3.9	254
130U	n-PROPYL MERCAPTAN		CH ₃ CH ₂ CH ₂ SH				305
131	INORGANIC GAS QUALITATIVE						355
132SA	VINYL CHLORIDE	CHLOROETHYLENE	CH ₂ = CHCI	1(A),3(B),2.5(J)		3.6	342
132SB	"						343
132SC	"						344
133A	ACETALDEHYDE		CH₃CHO	20(B), 50(J)	C 25(A)	40.0	22
133SB	"						23
134SA	TRICHLOROETHYLENE		CI ₂ C = CHCI	10(A), 100(B), 25(J)	25(A), 150(B)	8.0	332
134SB	"						333
134SG	*						334
135SA	TETRACHLOROETHYLENE	PERCHLOROETHYLENE	CI ₂ C = CCI ₂	25(A), 50(B)	100(A), 150(B)		318
135SB	"						319
135SG	"						320
135SM	"						321
136	ACROLEIN	ALLYL ALDEHYDE	CH ₂ CHCHO	0.1(B, J)	C 0.1(A)	2.8	31
137U	HYDROGEN		H ₂			4.0	176
138U	BUTYL ACETATE		CH ₃ CO ₂ C ₄ H ₉	150(A,B),100(J)	200(A, B)		62
139SB	METHYL ETHYL KETONE	2-BUTANONE	CH3COC2H5	200 (A, B, J)	300(A, B)	1.7	243
139SB©	BUTYL ACETATE		CH3CO2C4H9	150(A,B), 100(J)	200(A, B)	•••	61
139SB©	1,4-DIOXANE		O1 15002041 19	20(A), 25(B), 10(J)	200 (A, D)		132
	ISOBUTYL ACETATE		CH-CO-CH-CH(Ch /				_
139SB© 139SB©	ISOBUTYL ACETATE		CH ₃ CO ₂ CH ₂ CH (CH ₃) ₂ CH ₃ CO ₂ CH (CH ₃) ₂	150(A, B) 100(A, J)	200(A)		200
							
139SB©	PROPYL ACETATE		CH ₃ CO ₂ (CH ₂)CH ₃	200(A, B, J)	250(A)		295
139U	METHYL ETHYL KETONE		au aa/au \		1==(1)		244
139U	METHYL PROPYL KETONE		CH ₃ CO(CH ₂) ₂ CH ₃		150(A)		256
139U©	DIISOBUTYL KETONE		[(CH ₃) ₂ CHCH ₂] ₂ CO	25(A)			126
140SA	ARSINE		AsH ₃	0.005(A, B),0.1(J)			46
141SA	CARBON DISULPHIDE		CS ₂	1(A),10(B, J)		1.0	80
141SB	"						81
141SC	"						82
142S	MERCURY VAPOUR		Hg	0.025mg/m ³ (A, J)	0.15mg/m ³ (B)		220
143SA	XYLENE	DIMETHYL BENZENE	C ₆ H ₄ (CH ₃) ₂	100(A), 50(B, J)	150(A, B)	1.1	350

TUBE NO.	NAME OF GAS TO BE DETECTED	SYNONYM	CHEMICAL FORMULA	T.L.VTWA (ppm)	T.L.VSTEL (ppm)	LEL (%)	PAGE
143SB	XYLENE						351
145SA	1, 2-DICHLOROETHYLENE	ACETYLENE DICHLORIDE	CHCI = CHCI	200(A, B),150(J)	250(B)	9.7	119
146S	PHOSGENE	CARBONYL CHLORIDE	COCI ₂	0.1(A, J), 0.02(B)			282
147S	CARBON TETRACHLORIDE	TETRACHLOROMETHANE	CCI ₄	5(A, J), 2(B)	10(A)		92
150U	ISOPROPYL ALCOHOL	2-PROPANOL	CH ₃ CH(OH)CH ₃	200(A), 400(J)	400(A), 500(B)	2.0	213
151U	PROPYL ACETATE		CH ₃ CO ₂ C ₃ H ₇	200(A, B, J)	250(A, B)	1.7	296
152S	CHLOROFORM	TRICHLOROMETHANE	CHCl ₃	10(A), 2(B), 3(J)	200 (1, 2)		99
153U	ISOBUTYL ACETATE		CH ₃ CO ₂ CH ₂ CH(CH ₃) ₂	150(A, B)	187(B)	2.4	201
153U©	NAPHTHALENE		C10H8	5(A)	15(A)		260
155U	METHYL ISOBUTYL KETONE	ISOPROPYL ACETONE, HEXONE	(CH ₃) ₂ CHCH ₂ COCH ₃	50(A, B, J)	75(A)	1.2	248
156S	HYDROGEN FLUORIDE	IOOTTOT TEXOETONE, TEXONE	HF	0.5(A), 1.8(B), 3(J)	C 2(A), 3(B)	1.2	183
157JS	METHYL BROMIDE	BROMOMETHANE	CH ₃ Br	1(A, J), 5(B)	15(B)	10.0	231
1575S 157SA	//	BROWOWETHANE	СПЗБІ	I(A, J), 5(B)	15(b)	10.0	232
	"						
157SB	"						233
157SD				(- 1) ((-)	12(1) 222(2)		234
158S	STYRENE	VINYL BENZENE	C ₆ H ₅ CH = CH ₂	20(A, J), 100(B)	40(A), 250(B)	1.1	308
158S©	α-PINENE		C10H16				291
158S©	DIVINYL BENZENE		C ₆ H ₄ (CHCH ₂) ₂	10(A)		1.1	134
158SB	STYRENE						309
159SA	OXYGEN		O ₂				272
159SB	"						272
159SC	"						273
160S	METHYL CHLOROFORM	1,1,1-TRICHLOROETHANE	CH ₃ CCI ₃	350(A), 100(B), 200(J)	450(A, B)		238
162U	TETRAHYDROFURAN		C ₄ H ₈ O	50(A,B), 200(J)	100(A), 250(B)	1.5	324
163SA	PROPYLENE OXIDE	1,2-EPOXYPROPANE	C ₃ H ₆ O	2(A), 5(B)		1.9	303
163SD	"						304
164SA	METHYL MERCAPTAN	METHANETHIOL	CH₃SH	0.5(A,B)		3.9	252
164SH	"						253
165SA	ETHYL MERCAPTAN	ETHANETHIOL	C ₂ H ₅ SH	0.5(A,B)	2(B)	2.8	157
165SB	"						158
165SB	tert-BUTYL MERCAPTAN		(CH ₃) ₃ CSH				69
166S	ETHYLENE DIBROMIDE	1,2-DIBROMOETHANE	BrCH ₂ CH ₂ Br	0.5(B)			150
167S	HYDROGEN SELENIDE	1,E BIBITOMOETTIME	H ₂ Se	0.05(A,J), 0.02(B)			185
168SA	1,3-BUTADIENE		CH ₂ = CHCH= CH ₂	2(A), 10(B)		1.1	53
168SB	//		OTIZ OTIOTT OTIZ	2(A), 10(B)			54
168SC	,						55
168SE							
	"	A OUT ORODUTADIENE	011 001011 011	40(4)		4.0	56
169S	CHLOROPRENE	2-CHLOROBUTADIENE	CH ₂ =CCICH=CH ₂	10(A)	0.00(4).0(D)	4.0	101
171SA	FORMALDEHYDE		нсно	2(B), 0.5(J)	C 0.3(A), 2(B)	7.0	161
171SB	"						162
171SC	/						163
172S	CHLOROPICRIN	NITRO-TRICHLOROMETHANE	Cl ₃ CNO ₂	0.1(A, J)			100
173SA	HYDROGEN CHLORIDE		HCI	1(B), 5(J)	C 2(A), 5(B)		177
173SB	"						178
174A	NITROGEN OXIDE/DIOXIDE		NO & NO ₂	NO; 25(A)	NO; 35(B)		266
174B	-SEPARATELY			NO ₂ ; 3(A)	NO ₂ ; 5(A, B)		
175SA	NITROGEN OXIDES		NOx				267
175SH	"						268
175U	"						269
176SC	METHYL IODIDE	IODOMETHANE	CH₃I	2(A, B)			245
176UH	"						246
177SA	WATER VAPOUR		H₂O				345
177U	"						346
177UL	"						347
177UR	"						348
177UR2	"						349
178SB	CHLOROBENZENE		C ₆ H ₅ CI	10(A, J), 1(B)		1.3	98
				20(A), 100(B), 50(J)	125(A, B)		141
179S	ETHYL BENZENE		C6H5C2H5	20(A), 100(D), 30(3)			
179S 180S	ETHYL BENZENE DICHLOROMETHANE	METHYLENE CHLORIDE	C6H5C2H5 CH2Cl2	50(A, J),100(B)	120(7, 0)		120

TUBE NO.	NAME OF GAS TO BE DETECTED	SYNONYM	CHEMICAL FORMULA	T.L.VTWA (ppm)	T.L.VSTEL (ppm)	LEL (%)	PAGE
182SA	OZONE		O ₃	0.05(A), 0.1(J)	0.3(B)		275
182SB	//			0.00 (77, 0.170)	0.0(2)		276
182U	,						277
183U	CRESOL		C ₆ H ₄ (CH ₃)OH	20mg/m³ (A), 5(J)		1.1	102
183U	PHENOL		C6H5OH	5(A, J), 2(B)	10(B)	1.8	281
184S	METHYL METHACRYLATE		CH ₂ = C(CH ₃)CO ₂ CH ₃	50(A, B)	10(B) 100(A), 125(B)	2.1	255
							_
184S©	ALLYL ALCOHOL		CH2 = CHCH2OH	0.5(A), 2(B), 1(J)	4(B)	2.5	37
185S	PROPYLENE		CH ₂ = CHCH ₃	500(A)			300
186	ORGANIC GAS CHECKER						271
186B	ORGANIC GAS QUALITATAIVE						358
187S	GENERAL HYDROCARBONS	i-C4H10,n-C5H12,n-C6H14,n-C7H	., ,				169
188U	ISOPENTYL ACETATE	ISOAMYL ACETATE	CH ₃ CO ₂ CH ₂ CH ₂ CH(CH ₃) ₂	50(A),100(B, J)	100(A), 125(B)	1.1	206
189U	2-BUTANOL	sec-BUTYL ALCOHOL	CH ₃ CH ₂ CH(OH)CH ₃	100(A, B, J)		1.7	59
190U	ETHYL CELLOSOLVE	ETHYLENE GLYCOL MONOETHYL ETHER/ 2-ETHOXYETHANOL	C2H5OCH2CH2OH	5(A, J),10(B)		1.8	143
190U	METHYL CELLOSOLVE	ETHYLENE GLYCOL MONOMETHYL ETHER/ 2-METHOXYETHANOL	CH3OCH2CH2OH	0.1(A), 5(B, J)		2.3	236
190U©	BUTYL CELLOSOLVE	ETHYLENE GLYCOL MONOBUTYL ETHER/ 2-BUTOXYETHANOL	C4H9OCH2CH2OH	20(A), 25(B)		1.1	66
190U©	1-BUTANOL	n-BUTYL ALCOHOL	CH ₃ (CH ₂)	20(A), 50(J)	C50(A)		58
190U©	CROTONALDEHYDE		CH₃CH = CHCHO		C0.3(A)	2.1	103
190U©	DIACETONE ALCOHOL			50(A, B)			112
190U©	DICYCLOPENTADIENE		C10H12	5(A, B)			123
190U©	ETHYL CELLOSOLVE ACETATE	ETHYLENE GLYCOL ETHYL ETHER ACETATE	CH3CO2CH2OC2H5	5(A, J), 10(B)		1.7	144
190U©	FURFURAL	(2-FURALDEHYDE)		2(A, B), 2.5(J)			166
190U©	ISOPRENE		CH2 = C(CH3)CH = CH2				209
190U©	ISOPROPYL CELLOSOLVE		(CH ₃) ₂ HCO(CH ₂) ₂ COH				214
190U©	MESITYL OXIDE	4-METHYL-3-PENTEN-2-ONE	CH ₃ COCH = C(CH ₃) ₂	15(A), 50(B)	25(A)		221
190U©	METHYL CELLOSOLVE ACETATE		CH3CO2CH2CH2OCH3	0.1(A, J)	2000		237
190U©	1-PROPANOL		CH ₃ CH ₂ CH ₂ OH	100(A), 200(B)		2.1	293
190U©	TETRAHYDROTHIOPHENE		C ₄ H ₈ S	100 (10), 200 (2)		1.1	325
192S	EPICHLOROHYDRIN	1-CHLORO-2,3- EPOXYPROPANE	CH2CHCH2CIO	0.5(A, B)		2.3	135
193S	METHYL STYRENE		CH ₃ C ₆ H ₄ CH = CH ₂	5(A)	150(B)	1.9	257
194S	1,3-DICHLOROPROPANE		CICH2CH2CH2CI		100 (2)	93	121
196S	N,N-DIMETHYL FORMAMIDE		HCON(CH ₃) ₂	10(A, B, J)	20(B)		131
197U	CYCLOHEXANONE		C6H10O	20(A), 25(J), 10(B)	50(A),100(B)	1.1	107
197U©	ISOPHORONE		C9H14O	20(17), 20(0), 10(0)	C5(A)	0.8	208
197U©	1-METHOXY-2-PROPANOL		CH3CHOHCH2OCH3	50(A, B)	150(A)	1.9	223
198U	METHYL CYCLOHEXANONE		CH3C6H9O		75(A, B)	1.15	240
				50(A, B, J)			_
199U	METHYL CYCLOHEXANOL		CH ₃ C ₆ H ₁₀ OH	50(A, B, J)	75(B)	1.0	241
200SA	SULPHIDE ION		S ² -				380
200SB	//		-				381
201SA	CHLORIDE ION		CI-				373
201SB	"						374
201SC	"						375
203S	COPPER ION		Cu ²⁺				376
204S	CYANIDE ION		CN ⁻				377
205SL	SALINITY		NaCl				379
206U	CYCLOHEXANOL		C ₆ H ₁₁ OH	50(A, B), 25(J)		1.2	106
208U	ISOBUTYL ALCOHOL	ISOBUTANOL	(CH ₃) ₂ CHCH ₂ OH	50(A, B, J)		1.7	203
209U	ISOPENTYL ALCOHOL	ISOAMYL ALCOHOL	(CH ₃) ₂ CH(CH ₂) ₂ OH	100(A, B, J)	125(A)	1.2	207
210U	PENTYL ACETATE	AMYL ACETATE	CH ₃ CO ₂ (CH ₂) ₄ CH ₃	50(A, B),100(J)	100(A), 150(B)	1.0	279
211U	BUTYL ACRYLATE		CH2=CHCO2 (CH2)3CH3	2(A), 1(B)		1.5	63
211U	METHYL ACRYLATE		CH2=CHCO2CH3	2(A)		2.4	225
211U©	ETHYL ACRYLATE		CH2 = CHCO2C2H5	5(A, B)	15(A)		138
211U©	ISOBUTYL ACRYLATE		CH2 = CHCO2CH2CH(CH3)2				202

TUBE	NAME OF GAS	SYNONYM	CHEMICAL	T.L.VTWA	T.L.VSTEL	LEL	PAGE
NO.	TO BE DETECTED		FORMULA	(ppm)	(ppm)	(%)	
213S	TRIETHYL AMINE		(C ₂ H ₅) ₃ N	1(A), 2(B)	3(A), 15(B)	1.2	335
214S	o-DICHLOROBENZENE	1,2-DICHLOROBENZENE	C6H4Cl2	25(A, B, J)	50(A)	2.2	114
215S	p-DICHLOROBENZENE	1,4-DICHLOROBENZENE	C6H4Cl2	10(A, J), 25(B)	50(B)		115
216S	ACETIC ACID		CH₃COOH	10(A, B, J)	15(A, B)	4.0	24
216S	FORMIC ACID		НСООН	5(A, B, J)	10(A)		164
216S	MALEIC ANHYDRIDE		C ₄ H ₂ O ₃	0.01mg/m ³ (A)			219
216S©	ACETIC ANHYDRIDE		(CH ₃ CO) ₂ O	1(A), 5(J), 0.5(B)			25
216S©	ACRYLIC ACID		CH2 = CHCOOH	2(A)			32
216S©	BUTYRIC ACID		CH3CH2CH2COOH				72
216S©	ISOBUTYRIC ACID		CH3CH2CH2COOH				205
216S©	ISOVALERIC ACID		(CH ₃) ₂ CHCH ₂ COOH				218
216S©	METHACRYLIC ACID		CH ₂ =C(CH ₃)COOH	20(A, B)			222
216S©	PROPIONIC ACID		CH ₃ CH ₂ COOH	10(A, B)			294
216S©	n-VALERIC ACID		CH3 (CH2)3CO2H	10(A, B)			340
		AMIDDAZONE		0.04(A), 0.00(B), 0.4(1)		4.7	_
2198	HYDRAZINE	AMIDRAZONE	NH ₂ NH ₂	0.01(A), 0.02(B), 0.1(J)	750/D)	4.7	175
221SA	n-BUTANE		C4H10	1,000(A), 600(B), 500(J)	750(B)	1.5	57
222S	DIETHYL AMINE		(C ₂ H ₅) ₂ NH	5(A, B), 10(J)	15(A), 25(B)	1.8	124
222S	TRIMETHYL AMINE		(CH ₃) ₃ N	5(A), 2(B)	15(A, B)	2.0	336
222S©	ISOPROPYLAMINE		(CH ₃) ₂ CHNH ₂	5(A)	10(A)	2.0	217
223S	2,2'-DICHLOROETHYL ETHER		(CICH ₂ CH ₂) ₂ O	5(A), 15(J)	10(A)		118
224SA	MONOETHANOL AMINE	2-AMINOETHANOL	H2NCH2CH2OH	3(A, J), 1(B)	6(A, B)	2.1	258
227S	DIMETHYL AMINE		(CH ₃)₂NH	5(A), 2(B), 10(J)	15(A)	2.8	128
227S	ETHYL AMINE		CH3CH2NH2	5(A), 2(B), 10(J)	15(A)	3.5	140
227S	METHYL AMINE		CH3NH2	5(A), 10(J)	15(A)	5	229
229S	N,N-DIMETHYL ACETAMIDE		CH ₃ CON(CH ₃) ₂	10(A, B, J)	20(B)		127
230SA	1,2-DICHLOROETHANE	ETHYLIDENE DICHLORIDE	CICH2CH2CI	10(A, J), 5(B)		6.2	117
232SA	ETHYLENE GLYCOL						151
232SB	ETHYLENE GLYCOL	MONOETHYLENE GLYCOL	HOCH2CH2OH		C100mg/m³ (A)	3.2	152
233S	NITRIC ACID VAPOUR		HNO ₃	2(A, J)	4(A, B)		262
234SA	FREE RESIDUAL CHLLORINE		Cl2	201,07	101, 57		378
235SA	1,1-DICHLOROETHANE	ETHYLIDENE CHLORIDE	CH ₃ CHCl ₂	100(A, B, J)	400(B)	5.6	116
236SA	1,1,2-TRICHLOROETHANE	ETTTEIDENE OFFICITIBE	CI2CHCH2CI	10(A, J)	400(B)	0.0	331
				10(A, 3)	45(A) 00(B)	0.0	
237S 237S©	VINYL ACETATE METHYL BUTYL KETONE		CH ₃ CO ₂ CH = CH ₂ CH ₃ (CH ₂) ₃ COCH ₃		15(A), 20(B) 10(A)	2.6 1.2	341 235
				5(A, B, J)			
238S	FURFURYL ALCOHOL	2-FURYLCARBINOL	C ₄ H ₃ OCH ₂ OH	10(A), 5(B, J)	15(A,B)	1.8	167
239S	CARBONYL SULPHIDE	CARBON OXYSULPHIDE	cos	5(A)		12	93
240S	SILANE	MONOSILANE	SiH ₄	5(A),0.5(B),100(J)	1.5(B)		307
242S	DIBORANE	BOROETHANE	B ₂ H ₆	0.1(A), 0.01(J)		0.9	113
242S©	HYDROGEN SELENIDE		H₂Se	0.05(A, J), 0.02(B)			182
243U	TETRAETHOXYSILANE		Si(OC ₂ H ₅) ₄	10(J)			322
244U	SULPHURIC ACID		H ₂ SO ₄	0.2mg/m ³ (A), 1mg/m ³ (J)			317
245UH	METHYL ISOTHIOCYANATE		CH₃NCS	0.02(A)			249
245UL	"						250
245UM	"						251
247S	HYDROGEN PEROXIDE		H ₂ O ₂	1(A, B)			184
248U	ETHYL-tert-BUTYL ETHER		C ₂ H ₅ OC(CH ₃) ₃	25(A)		1.2	142
2498	1,3-DICHLOROPROPENE		CICH ₂ -CH = CHCI	1(A)		5.3	122
280S	ACETYLENE • ETHYLENE-SEF	PARATION MEASUREMENT		1 22			30
281S	OXYGEN • CARBON DIOXIDE-		IT	CO ₂ 5,000(A, B, J)			274
	HYDROGEN SULPHIDE • MER			002 3,000 (A, D, J)			
282S		UNI TANO-SEPARATION MEA	NOUNEWENT				198 417
301	AIR FLOW INDICATOR TUBE		00	05(A) 00(B) 55(1)	000(5)	10.5	+
500	TWA-CARBON MONOXIDE		CO	25(A), 30(B), 50(J)	300(B)	12.5	366
501	TWA-AMMONIA		NH ₃	25(A, B, J)	35(A, B)	15.0	367
502	TWA-HYDROGEN SULPHIDE		N ₂ S	1(A), 5(B, J)	5(A), 15(B)	4.3	368
503	TWA-SULPHUR DIOXIDE		SO ₂		0.25(A), 5(B)		369
504	TWA-TOLUENE	TWA-METHYL BENZENE	C ₆ H ₅ CH ₃	20(A), 50(B, J)	150(B)		371
600SP	COMPRESSED BRETHING AIR	TEST TUBE(CO)					384
	COMPRESSED BRETHING AIR TEST TUBE(CO ₂)						385
601SP	COMPRESSED BRETHING AIR	TEGT TOBE (OG2)					
601SP 602SP	COMPRESSED BRETHING AIR						386

TUBE NO.	NAME OF GAS TO BE DETECTED	SYNONYM	CHEMICAL FORMULA	T.L.VTWA (ppm)	T.L.VSTEL (ppm)	LEL (%)	PAGE
604SP	COMPRESSED BRETHING AIR TEST TUBE(O2)						388
710	FORMALDEHYDE						390
710A	"						391
713	"						392
721	TOLUENE						393
721©	ETHYL BENZENE						394
721©	XYLENE						395
730	p-DICHLOROBENZENE						396
740	NITROGEN DIOXIDE						398
750	TRICHLOROETHYLENE						399
760	TETRACHLOROETHYLENE						400
800B	CHARCOAL TUBE (CONFORMED TO NIOSH REQUIREMENTS)						415
801	SILICA-GEL TUBE						416
900NHH	AMMONIA						402
901NHL	"						403

^{*} Additionally, refer to page 315 for the conversion chart method.

(NOTE)

- T.L.V. (A) : Threshold Limit Value for Chemical Substances in the Work Environment Adopted by ACGIH (American Governmental Industrial Hygienists, Inc.) with Intended Change for 2012
- T.L.V. (B) : Occupational Exposure Limits Listed Guidance Note EH 40/2007 from the Health and Safety Executive in U. K.
- T.L.V. (J) : Occupational Exposure Limits (OELs) recommended in Journal of Occupational Health issued by the Japan Society for Occupational Health in 2011
- T.L.V.-TWA (The Threshold Limit Value-Time Weighted Average): the time-weighted average concentration for a 8-hour workday and a 40-hour workweek, to which nearly all workers may be repeatedly exposed, day after day, without adverse effect
- T.L.V.-STEL (The Threshold Limit Value-Short Term Exposure Limit): the concentration to which workers can be exposed continuously for a short period of time without suffering from 1) irritation, 2) chronic or irreversible tissue damage, or 3) narcosis of sufficient degree to increase the likelihood of accidental injury, impair self-rescue or materially reduce work efficience, and provided that the daily T. L. V.-TWA is not exceeded. It is not a separate in dependent exposure limit, rather it supplements the time-weighted average (TWA) limit where there are recognized acute effects from a substance whose toxic effects are primarily of a chronic nature. STELs are recommended only where toxic effects have been reported from high short-term exposures in either humans or animals. The period of the short term is defined as 15 minutes in U.S.A. and 10 minutes in U. K. The Japanese STELs are advised that a 15-minute average exposure concentration does not exceed 1.5 times of the 8-hour TWA.
- $\label{eq:T.L.V.-STEL} \begin{tabular}{ll} \textbf{T.L.V.-STEL} (C) & (Threshold Limit Value-Ceiling) : the concentration that should not be exceeded during any part of the working exposure. \end{tabular}$
- LEL (vol.%): Lower Explosive Limit-refer from "Recommended Practice for Explosion Protected Electrical Installations in General Industries" written by Ministry of Labour in Japan

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